

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

2-[(2-Chlorobenzylidene)amino]-4,5,6,7-tetrahydro-1-benzothiophene-3-carbonitrile

 Abdullah M. Asiri,^{a,b} Salman A. Khan^b and M. Nawaz Tahir^{c*}

^aThe Center of Excellence for Advanced Materials Research, King Abdulaziz University, Jeddah 21589, PO Box 80203, Saudi Arabia, ^bDepartment of Chemistry, Faculty of Science, King Abdulaziz University, Jeddah 21589, PO Box 80203, Saudi Arabia, and ^cUniversity of Sargodha, Department of Physics, Sargodha, Pakistan
Correspondence e-mail: dmntahir_uos@yahoo.com

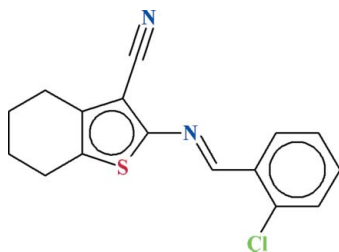
Received 24 July 2011; accepted 9 August 2011

Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.037; wR factor = 0.105; data-to-parameter ratio = 14.4.

In the title compound, $\text{C}_{16}\text{H}_{13}\text{ClN}_2\text{S}$, the mean planes fitted through all non-H atoms of the heterocyclic five-membered and the benzene rings are oriented at a dihedral angle of 5.19 (7)°. In the crystal, a weak $\text{C}-\text{H}\cdots\pi$ interaction occurs, along with weak $\pi-\pi$ interactions [centroid-centroid distance = 3.7698 (11) Å].

Related literature

For information on the use of Schiff bases in pharmaceutical chemistry, see: Lewinski *et al.* (2005). For related structures, see: Asiri *et al.* (2011a,b).



Experimental

Crystal data

$\text{C}_{16}\text{H}_{13}\text{ClN}_2\text{S}$
 $M_r = 300.79$
Triclinic, $P\bar{1}$

$a = 8.3383$ (4) Å
 $b = 8.6885$ (4) Å
 $c = 10.5746$ (5) Å

$\alpha = 85.975$ (2)°
 $\beta = 80.806$ (2)°
 $\gamma = 73.003$ (2)°
 $V = 723.00$ (6) Å³
 $Z = 2$

Mo $K\alpha$ radiation
 $\mu = 0.40$ mm⁻¹
 $T = 296$ K
 $0.40 \times 0.25 \times 0.25$ mm

Data collection

Bruker Kappa APEXII CCD diffractometer
Absorption correction: multi-scan (SADABS; Bruker, 2005)
 $T_{\min} = 0.931$, $T_{\max} = 0.951$

10003 measured reflections
2600 independent reflections
2308 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.019$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.037$
 $wR(F^2) = 0.105$
 $S = 1.03$
2600 reflections

181 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.30$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.34$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

C_g is the centroid of the C11–C16 ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C5}-\text{H5A}\cdots\text{C}g^i$	0.97	2.87	3.744 (3)	151

Symmetry code: (i) $-x, -y + 1, -z$.

Data collection: APEX2 (Bruker, 2009); cell refinement: SAINT (Bruker, 2009); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997) and PLATON (Spek, 2009); software used to prepare material for publication: WinGX (Farrugia, 1999) and PLATON.

The authors would like to thank the Chemistry Department, King Abdul Aziz University, Jeddah, Saudi Arabia, for providing research facilities and for the financial support of this work via grant No. 3-045/430.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NK2106).

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supporting information

Acta Cryst. (2011). E67, o2355 [doi:10.1107/S1600536811032302]

2-[(2-Chlorobenzylidene)amino]-4,5,6,7-tetrahydro-1-benzothiophene-3-carbonitrile

Abdullah M. Asiri, Salman A. Khan and M. Nawaz Tahir

S1. Comment

Schiff-base compounds have been used as intermediate for various reactions (Lewinski, *et al.*, 2005) and medical substrates. The title compound (I), (Fig. 1) has been prepared as a Pharmaceutical intermediate.

We have reported the crystal structure of 2-[(benzo[1,3]dioxol-5-ylmethylene)-amino]-4,5,6,7-tetrahydro-benzo[*b*]thiophene -3-carbonitrile (Asiri *et al.*, 2011*a*) and 2-[(4-Chloro-benzylidene)-amino]-4,5,6,7-tetrahydro-benzo[*b*]thiophene -3-carbonitrile (Asiri *et al.*, 2011*b*) which are related to the title compound.

In (I), the five membered ring A (C1/C2/C3/C8/S1) of 2-amino-4,5,6,7- tetrahydro-1-benzothiophene-3-carbonitrile and the group B (C10–C16/CL1) of 2-chlorobenzaldehyde are planar with r. m. s. deviation of 0.0097 and 0.0020 Å, respectively. The dihedral angle between A/B is 5.19 (7)°. A C—H··· π interaction between the six membered rings of 2-amino-4,5,6,7-tetrahydro-1- benzothiophene-3-carbonitrile and the 2-chlorobenzaldehyde group is present (Table 1). π – π interactions [separation: 3.7698 (11) Å] between the heterocyclic five membered and benzene rings are also present.

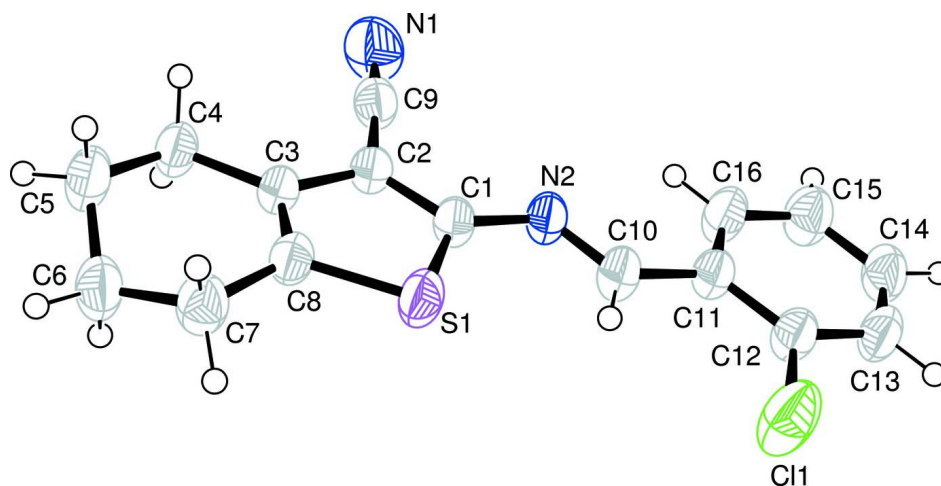
S2. Experimental

A mixture of 2-chloro benzaldehyde (0.46 g, 2.4 mmol) and 2-amino-4,5,6,7-tetrahydro-benzo[*b*]thiophene-carbonitrile (0.32 g, 3.3 mmol) in ethanol (15 ml) was heated for 3 h. The progress of the reaction was monitored by TLC. The solid that separated from the cooled mixture was collected and recrystallized from a methanol-chloroform mixture (8:2) to give yellow needles of the title compound (I).

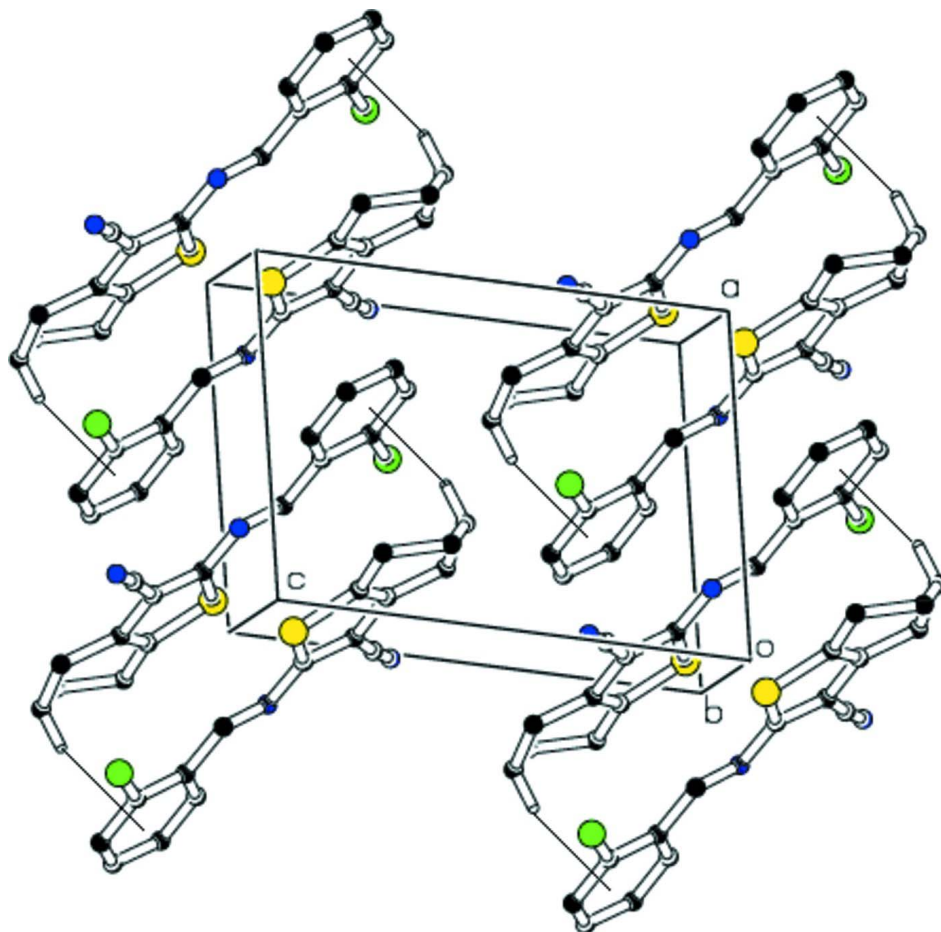
Yellow solid: Yield: 82%, m.p. 450 K.

S3. Refinement

The H-atoms were positioned geometrically (C–H = 0.93–0.97 Å) and refined as riding with $U_{\text{iso}}(\text{H}) = xU_{\text{eq}}(\text{C})$, where $x = 1.2$ for all H-atoms.

**Figure 1**

The title compounds with displacement ellipsoids at the 50% probability level. H-atoms are shown as small spheres of arbitrary radii.

**Figure 2**

The partial packing (*PLATON*; Spek, 2009) which shows that C—H... π interactions are present. Only the H-atom involved in π -interaction is present.

2-[(2-Chlorobenzylidene)amino]-4,5,6,7-tetrahydro-1-benzothiophene-3-carbonitrile

Crystal data

C₁₆H₁₃ClN₂S $M_r = 300.79$ Triclinic, $P\bar{1}$

Hall symbol: -P 1

 $a = 8.3383$ (4) Å $b = 8.6885$ (4) Å $c = 10.5746$ (5) Å $\alpha = 85.975$ (2)° $\beta = 80.806$ (2)° $\gamma = 73.003$ (2)° $V = 723.00$ (6) Å³ $Z = 2$ $F(000) = 312$ $D_x = 1.382$ Mg m⁻³Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 2308 reflections

 $\theta = 3.0$ – 25.3 ° $\mu = 0.40$ mm⁻¹ $T = 296$ K

Rod, yellow

 $0.40 \times 0.25 \times 0.25$ mm

Data collection

Bruker Kappa APEXII CCD

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Detector resolution: 8.20 pixels mm⁻¹ ω scans

Absorption correction: multi-scan

(SADABS; Bruker, 2005)

 $T_{\min} = 0.931$, $T_{\max} = 0.951$

10003 measured reflections

2600 independent reflections

2308 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.019$ $\theta_{\max} = 25.3$ °, $\theta_{\min} = 3.0$ ° $h = -9 \rightarrow 10$ $k = -10 \rightarrow 10$ $l = -12 \rightarrow 12$

Refinement

Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.037$ $wR(F^2) = 0.105$ $S = 1.03$

2600 reflections

181 parameters

0 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0542P)^2 + 0.253P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} < 0.001$ $\Delta\rho_{\max} = 0.30$ e Å⁻³ $\Delta\rho_{\min} = -0.34$ e Å⁻³

Special details

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Cl1	0.51410 (9)	0.59840 (7)	-0.31788 (6)	0.0866 (2)
S1	0.05883 (6)	0.76273 (6)	0.05536 (4)	0.0534 (2)
N1	0.0435 (3)	0.2524 (2)	0.3132 (2)	0.0764 (7)
N2	0.24264 (18)	0.44103 (18)	0.02432 (13)	0.0474 (5)

C1	0.1187 (2)	0.5590 (2)	0.09452 (16)	0.0444 (5)
C2	0.0275 (2)	0.53040 (19)	0.20896 (15)	0.0409 (5)
C3	-0.0859 (2)	0.67215 (19)	0.26797 (15)	0.0402 (5)
C4	-0.1898 (2)	0.6775 (2)	0.39739 (16)	0.0481 (6)
C5	-0.3072 (3)	0.8451 (2)	0.42743 (19)	0.0596 (6)
C6	-0.2276 (3)	0.9751 (2)	0.3795 (2)	0.0639 (7)
C7	-0.1776 (3)	0.9748 (2)	0.23394 (18)	0.0563 (6)
C8	-0.0812 (2)	0.8060 (2)	0.19563 (16)	0.0450 (5)
C9	0.0415 (2)	0.3740 (2)	0.26452 (17)	0.0490 (6)
C10	0.3285 (2)	0.4785 (2)	-0.07655 (17)	0.0515 (6)
C11	0.4641 (2)	0.3591 (2)	-0.15131 (16)	0.0467 (5)
C12	0.5574 (2)	0.4006 (2)	-0.26298 (18)	0.0505 (6)
C13	0.6858 (2)	0.2868 (3)	-0.33329 (18)	0.0553 (6)
C14	0.7236 (2)	0.1295 (3)	-0.29270 (19)	0.0581 (6)
C15	0.6354 (3)	0.0838 (3)	-0.1829 (2)	0.0631 (7)
C16	0.5064 (3)	0.1976 (2)	-0.11318 (19)	0.0580 (6)
H4A	-0.11493	0.64359	0.46182	0.0577*
H4B	-0.25684	0.60246	0.40122	0.0577*
H5A	-0.40913	0.86058	0.38900	0.0715*
H5B	-0.34027	0.85337	0.51947	0.0715*
H6A	-0.12747	0.96148	0.41971	0.0766*
H6B	-0.30654	1.07855	0.40410	0.0766*
H7A	-0.27819	1.01182	0.19234	0.0676*
H7B	-0.10754	1.04641	0.20812	0.0676*
H10	0.30434	0.58581	-0.10391	0.0618*
H13	0.74600	0.31718	-0.40782	0.0664*
H14	0.80998	0.05264	-0.33990	0.0697*
H15	0.66233	-0.02358	-0.15544	0.0757*
H16	0.44656	0.16560	-0.03915	0.0696*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C11	0.0933 (4)	0.0609 (3)	0.0809 (4)	-0.0090 (3)	0.0284 (3)	0.0135 (3)
S1	0.0620 (3)	0.0531 (3)	0.0391 (3)	-0.0172 (2)	0.0119 (2)	-0.0022 (2)
N1	0.0988 (15)	0.0503 (10)	0.0746 (12)	-0.0181 (10)	-0.0035 (11)	0.0008 (9)
N2	0.0421 (8)	0.0579 (9)	0.0393 (8)	-0.0130 (7)	0.0043 (6)	-0.0117 (6)
C1	0.0419 (9)	0.0528 (10)	0.0373 (8)	-0.0138 (7)	0.0020 (7)	-0.0100 (7)
C2	0.0393 (8)	0.0441 (9)	0.0375 (8)	-0.0108 (7)	-0.0007 (7)	-0.0052 (7)
C3	0.0376 (8)	0.0446 (9)	0.0365 (8)	-0.0113 (7)	0.0013 (6)	-0.0051 (7)
C4	0.0494 (10)	0.0483 (10)	0.0400 (9)	-0.0107 (8)	0.0082 (7)	-0.0042 (7)
C5	0.0584 (11)	0.0571 (11)	0.0507 (11)	-0.0076 (9)	0.0146 (9)	-0.0069 (9)
C6	0.0718 (13)	0.0486 (10)	0.0587 (12)	-0.0092 (9)	0.0165 (10)	-0.0123 (9)
C7	0.0635 (12)	0.0425 (10)	0.0542 (11)	-0.0095 (8)	0.0064 (9)	-0.0009 (8)
C8	0.0466 (9)	0.0467 (9)	0.0384 (8)	-0.0130 (7)	0.0044 (7)	-0.0044 (7)
C9	0.0521 (10)	0.0460 (10)	0.0447 (9)	-0.0099 (8)	0.0009 (8)	-0.0091 (8)
C10	0.0462 (10)	0.0592 (11)	0.0445 (10)	-0.0129 (8)	0.0065 (8)	-0.0092 (8)
C11	0.0413 (9)	0.0547 (10)	0.0413 (9)	-0.0130 (8)	0.0040 (7)	-0.0077 (7)

C12	0.0461 (10)	0.0547 (10)	0.0462 (10)	-0.0130 (8)	0.0041 (8)	-0.0027 (8)
C13	0.0453 (10)	0.0701 (12)	0.0447 (10)	-0.0151 (9)	0.0099 (8)	-0.0062 (9)
C14	0.0477 (10)	0.0625 (12)	0.0548 (11)	-0.0054 (9)	0.0064 (8)	-0.0156 (9)
C15	0.0634 (12)	0.0543 (11)	0.0604 (12)	-0.0074 (9)	0.0070 (10)	-0.0042 (9)
C16	0.0578 (11)	0.0596 (12)	0.0486 (10)	-0.0146 (9)	0.0120 (9)	-0.0018 (9)

Geometric parameters (Å, °)

C11—C12	1.7277 (18)	C12—C13	1.381 (3)
S1—C1	1.7322 (17)	C13—C14	1.364 (3)
S1—C8	1.7243 (18)	C14—C15	1.371 (3)
N1—C9	1.139 (2)	C15—C16	1.380 (3)
N2—C1	1.383 (2)	C4—H4A	0.9700
N2—C10	1.265 (2)	C4—H4B	0.9700
C1—C2	1.371 (2)	C5—H5A	0.9700
C2—C3	1.428 (2)	C5—H5B	0.9700
C2—C9	1.422 (2)	C6—H6A	0.9700
C3—C4	1.494 (2)	C6—H6B	0.9700
C3—C8	1.353 (2)	C7—H7A	0.9700
C4—C5	1.521 (2)	C7—H7B	0.9700
C5—C6	1.491 (3)	C10—H10	0.9300
C6—C7	1.529 (3)	C13—H13	0.9300
C7—C8	1.500 (2)	C14—H14	0.9300
C10—C11	1.457 (2)	C15—H15	0.9300
C11—C12	1.393 (3)	C16—H16	0.9300
C11—C16	1.390 (2)		
C1—S1—C8	92.06 (8)	C3—C4—H4A	109.00
C1—N2—C10	120.23 (15)	C3—C4—H4B	109.00
S1—C1—N2	126.07 (13)	C5—C4—H4A	109.00
S1—C1—C2	109.88 (12)	C5—C4—H4B	109.00
N2—C1—C2	124.04 (15)	H4A—C4—H4B	108.00
C1—C2—C3	114.07 (14)	C4—C5—H5A	109.00
C1—C2—C9	123.61 (15)	C4—C5—H5B	109.00
C3—C2—C9	122.32 (15)	C6—C5—H5A	109.00
C2—C3—C4	125.31 (14)	C6—C5—H5B	109.00
C2—C3—C8	111.71 (15)	H5A—C5—H5B	108.00
C4—C3—C8	122.86 (15)	C5—C6—H6A	109.00
C3—C4—C5	112.01 (14)	C5—C6—H6B	109.00
C4—C5—C6	112.83 (19)	C7—C6—H6A	109.00
C5—C6—C7	112.49 (17)	C7—C6—H6B	109.00
C6—C7—C8	108.21 (14)	H6A—C6—H6B	108.00
S1—C8—C3	112.24 (13)	C6—C7—H7A	110.00
S1—C8—C7	122.74 (13)	C6—C7—H7B	110.00
C3—C8—C7	124.94 (16)	C8—C7—H7A	110.00
N1—C9—C2	175.9 (2)	C8—C7—H7B	110.00
N2—C10—C11	122.19 (16)	H7A—C7—H7B	108.00
C10—C11—C12	122.04 (15)	N2—C10—H10	119.00

C10—C11—C16	120.95 (16)	C11—C10—H10	119.00
C12—C11—C16	117.01 (16)	C12—C13—H13	120.00
C11—C12—C11	120.35 (13)	C14—C13—H13	120.00
C11—C12—C13	117.94 (15)	C13—C14—H14	120.00
C11—C12—C13	121.71 (17)	C15—C14—H14	120.00
C12—C13—C14	119.53 (18)	C14—C15—H15	120.00
C13—C14—C15	120.6 (2)	C16—C15—H15	120.00
C14—C15—C16	119.8 (2)	C11—C16—H16	119.00
C11—C16—C15	121.36 (19)	C15—C16—H16	119.00
C8—S1—C1—N2	-176.35 (16)	C4—C3—C8—C7	-0.2 (3)
C8—S1—C1—C2	2.04 (14)	C3—C4—C5—C6	-38.3 (2)
C1—S1—C8—C7	175.39 (17)	C4—C5—C6—C7	60.9 (2)
C1—S1—C8—C3	-1.37 (15)	C5—C6—C7—C8	-48.8 (3)
C10—N2—C1—C2	-175.64 (17)	C6—C7—C8—S1	-156.62 (16)
C1—N2—C10—C11	177.94 (16)	C6—C7—C8—C3	19.7 (3)
C10—N2—C1—S1	2.5 (3)	N2—C10—C11—C12	179.42 (17)
S1—C1—C2—C3	-2.3 (2)	N2—C10—C11—C16	-0.9 (3)
S1—C1—C2—C9	176.99 (14)	C10—C11—C12—C11	-0.3 (2)
N2—C1—C2—C3	176.19 (16)	C10—C11—C12—C13	179.96 (17)
N2—C1—C2—C9	-4.6 (3)	C16—C11—C12—C11	-179.97 (15)
C9—C2—C3—C8	-177.98 (16)	C16—C11—C12—C13	0.3 (3)
C9—C2—C3—C4	5.9 (3)	C10—C11—C16—C15	-179.6 (2)
C1—C2—C3—C4	-174.81 (16)	C12—C11—C16—C15	0.1 (3)
C1—C2—C3—C8	1.3 (2)	C11—C12—C13—C14	179.92 (15)
C2—C3—C4—C5	-175.60 (17)	C11—C12—C13—C14	-0.3 (3)
C8—C3—C4—C5	8.7 (2)	C12—C13—C14—C15	-0.1 (3)
C2—C3—C8—S1	0.3 (2)	C13—C14—C15—C16	0.5 (3)
C2—C3—C8—C7	-176.34 (18)	C14—C15—C16—C11	-0.5 (4)
C4—C3—C8—S1	176.53 (13)		

Hydrogen-bond geometry (Å, °)

Cg is the centroid of the C11—C16 ring.

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C5—H5A \cdots Cg ⁱ	0.97	2.87	3.744 (3)	151

Symmetry code: (i) $-x, -y+1, -z$.