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6-Hydroxy-2*H*-1,3-benzodioxole-5-carbaldehyde

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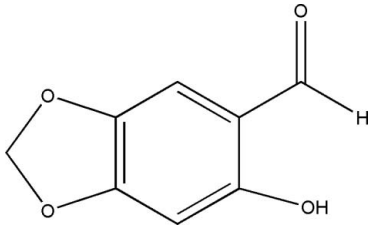
Key indicators: single-crystal X-ray study; T = 173 K; mean σ(C–C) = 0.004 Å; R factor = 0.045; wR factor = 0.125; data-to-parameter ratio = 12.4.

The title compound, $C_8H_6O_4$, crystallizes with two independent molecules in the asymmetric unit. The benzodioxole ring system is almost planar in each molecule, with maximum deviations of 0.008 (1) and 0.007 (1) Å. The molecular structure is characterized by strong electrostatic intramolecular O···O contacts [2.649 (3) Å] and intramolecular O—H···O hydrogen-bonding interactions. Intermolecular O···O interactions [3.001 (2) Å] are observed in the crystal structure.

Related literature

For the preparation, see: Juhász *et al.* (2007); Akselsen *et al.* (2009). For hydrogen-bond motifs, see: Bernstein *et al.* (1995). The title compound is a starting material and an intermediate in the synthesis of biologically active compounds. These compounds have shown HIV-1 integrase inhibitory activity (Bailly *et al.*, 2005), dopamine D1 receptor full agonist (Cueva, *et al.* 2006) and glycogen phosphorylase inhibitory activity (Juhász *et al.*, 2007).



Experimental

Crystal data

$C_8H_6O_4$
M_r = 166.13
Monoclinic, $P2_1/c$
a = 6.4916 (3) Å
b = 12.8242 (7) Å
c = 16.7122 (8) Å
β = 96.258 (3)°
V = 1382.99 (12) Å^3
Z = 8
Mo $K\alpha$ radiation
μ = 0.13 mm^{-1}
T = 173 K
0.38 × 0.11 × 0.10 mm

Data collection

Bruker APEXII CCD diffractometer
10746 measured reflections
2708 independent reflections
1348 reflections with $I > 2\sigma(I)$
R_{int} = 0.081

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.045
$wR(F^2)$ = 0.125
S = 0.94
2708 reflections
219 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.19 e Å^{-3}
$\Delta\rho_{min}$ = −0.31 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
O1—H1···O4	0.84	1.92	2.652 (3)	146
O5—H5···O8	0.84	1.91	2.645 (3)	145

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT-Plus* (Bruker, 2009); data reduction: *SAINT-Plus*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

We wish to thank Dr Manuel Fernandes (University of the Witwatersrand) for the data collection and the NRF and the University of KwaZulu-Natal for financial support. This work is based upon research supported by the South African Research Chairs Initiative of the Department of Science and Technology.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: QM2028).

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supporting information

Acta Cryst. (2011). E**67**, o2681 [https://doi.org/10.1107/S1600536811037019]

6-Hydroxy-2*H*-1,3-benzodioxole-5-carbaldehyde

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S1. Comment

The title compound 6- hydroxybenzo[*d*][1,3]dioxole-5-carbaldehyde was obtained as an intermediate product in our research effort aimed at the total synthesis of biologically active compounds. These compounds have been used for HIV-1 integrase inhibitory activities as reported by Bailly *et al.* (2005), Dopamine D1 receptor full agonist (Cueva, *et al.* 2006) and glycogen phosphorylase inhibitory activity reported by Juhász *et al.* (2007). The compound has been previously reported by Juhász *et al.* (2007) and Akselsen *et al.* (2009) with 45% yield when it was respectively utilized as a starting material and as an intermediate in the synthesis of the biologically active compounds. However, inspite of the varied biological applications of (I) the crystal structure of the title compound has not been reported to date. The compound has two independent molecules in the asymmetric unit that are related by a crystallographic centre of inversion and a glide plane perpendicular to the (0, 1, 0) axis. The benzodioxole ring systems in the title compound are almost planar and show strong pi-pi interactions in the unit cell. The molecule is stabilized by intra-molecular hydrogen bonding contacts which are however balanced by a network of O···O electrostatic contacts that are both intra- [O1···O4 & O5···O8 = 2.649 (3) Å] and inter-molecular [O4···O8 = 3.001 (2) Å] in nature.

S2. Experimental

The compound 2-yydroxy-4,5-methylenedioxybenzaldehyde was synthesized by following the literature method of Akselsen *et al.* (2009). Brown crystals suitable for X-ray diffraction were grown from hexane:ethyl acetate (95:5). m.p. 125–127 °C. ^{1}H NMR: δ (p.p.m.): 6.01 (2*H*,s, O–CH_2–O); 6.46 (1*H*, s, H-5); 6.86 (1*H*, s, H-8); 9.62 (1*H*, s, CHO); 11.79 (1*H*, s, OH). ^{13}C NMR: δ = 98.37, 102.15, 109.35, 113.65, 141.33, 155.17, 161.54, 193.69. HRMS m/*z* 166.0264 (calcd for $C_8H_6O_4$: 166.0266).

S3. Refinement

All H-atoms were refined using a riding model, with C—H = 0.95 Å and U_{iso}(H) = 1.2U_{eq}(C) for aromatic, C—H = 0.99 Å and U_{iso}(H) = 1.2U_{eq}(C) for CH_2.

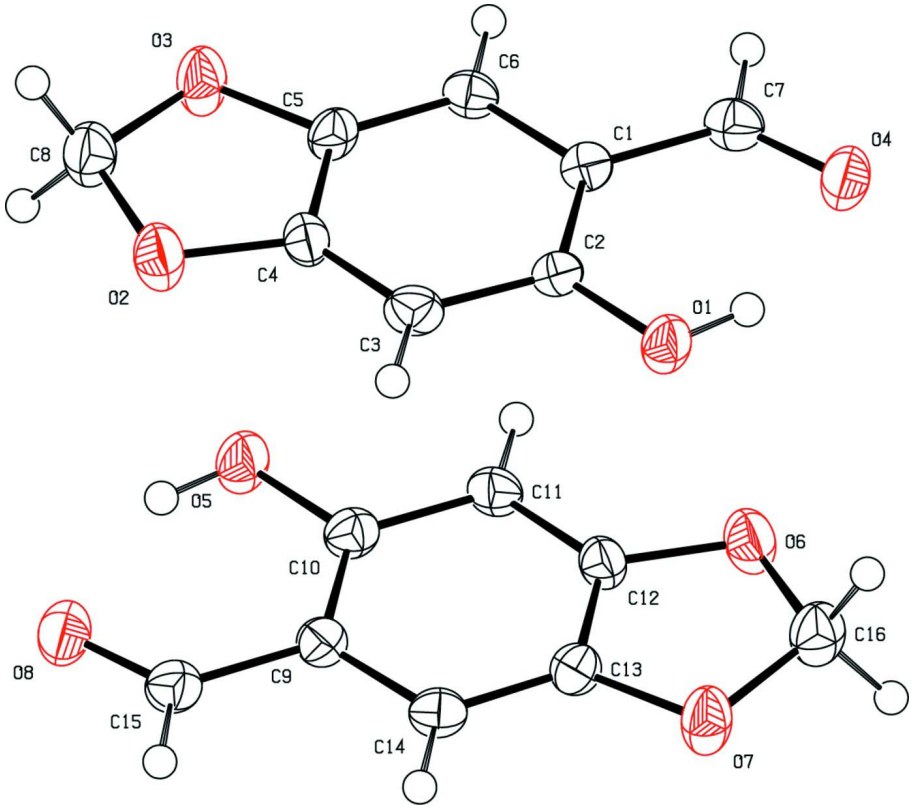

Figure 1

Molecular structure of the title compound with the atom numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are presented as small spheres of arbitrary radius.

6-Hydroxy-2*H*-1,3-benzodioxole-5-carbaldehyde

Crystal data

$C_8H_6O_4$
M_r = 166.13
Monoclinic, $P2_1/c$
Hall symbol: -P 2ybc
a = 6.4916 (3) Å
b = 12.8242 (7) Å
c = 16.7122 (8) Å
β = 96.258 (3)°
V = 1382.99 (12) Å^3
Z = 8

$F(000)$ = 688
D_x = 1.596 Mg m^{-3}
Mo $K\alpha$ radiation, λ = 0.71073 Å
Cell parameters from 1671 reflections
θ = 2.5–26.6°
μ = 0.13 mm^{-1}
T = 173 K
Needle, colourless
0.38 × 0.11 × 0.10 mm

Data collection

Bruker APEXII CCD diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
φ and ω scans
10746 measured reflections
2708 independent reflections

1348 reflections with $I > 2\sigma(I)$
R_{int} = 0.081
θ_{max} = 26.0°, θ_{min} = 2.0°
h = −7→8
k = −15→15
l = −18→20

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)] = 0.045$
$wR(F^2) = 0.125$
$S = 0.94$
2708 reflections
219 parameters
0 restraints
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.0556P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} < 0.001$
$\Delta\rho_{max} = 0.19$ e Å^{-3}
$\Delta\rho_{min} = -0.31$ e Å^{-3}

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	x	*y*	*z*	U_{iso}*/U_{eq}
C1	0.0179 (4)	0.62547 (19)	0.46651 (15)	0.0211 (6)
C2	0.2306 (4)	0.6144 (2)	0.49173 (15)	0.0221 (6)
C3	0.3046 (4)	0.6148 (2)	0.57346 (15)	0.0272 (7)
H3	0.4474	0.6059	0.5915	0.033*
C4	0.1604 (4)	0.6288 (2)	0.62576 (14)	0.0233 (6)
C5	−0.0500 (4)	0.6406 (2)	0.60194 (15)	0.0230 (6)
C6	−0.1269 (4)	0.63872 (19)	0.52350 (15)	0.0225 (6)
H6	−0.2710	0.6459	0.5073	0.027*
C7	−0.0596 (4)	0.6209 (2)	0.38261 (16)	0.0265 (7)
H7	−0.2052	0.6268	0.3691	0.032*
C8	−0.0031 (5)	0.6428 (2)	0.73701 (16)	0.0328 (7)
H8A	−0.0324	0.5811	0.7695	0.039*
H8B	−0.0048	0.7056	0.7714	0.039*
C9	0.4907 (4)	0.6181 (2)	0.12152 (15)	0.0240 (6)
C10	0.2786 (4)	0.6096 (2)	0.09607 (15)	0.0247 (6)
C11	0.2024 (4)	0.6141 (2)	0.01456 (14)	0.0248 (7)
H11	0.0591	0.6072	−0.0034	0.030*
C12	0.3487 (4)	0.6294 (2)	−0.03776 (15)	0.0238 (6)
C13	0.5577 (4)	0.6389 (2)	−0.01390 (15)	0.0236 (6)
C14	0.6353 (4)	0.6332 (2)	0.06454 (15)	0.0244 (6)
H14	0.7797	0.6389	0.0807	0.029*
C15	0.5696 (4)	0.6118 (2)	0.20578 (16)	0.0283 (7)
H15	0.7151	0.6174	0.2196	0.034*
C16	0.5093 (4)	0.6575 (2)	−0.14766 (16)	0.0316 (7)
H16A	0.5389	0.6056	−0.1887	0.038*

H16B	0.5077	0.7277	−0.1724	0.038*
O1	0.3710 (3)	0.60201 (16)	0.43841 (10)	0.0301 (5)
H1	0.3094	0.5985	0.3916	0.045*
O2	0.1955 (3)	0.63205 (16)	0.70789 (10)	0.0353 (5)
O3	−0.1564 (3)	0.65199 (16)	0.66900 (10)	0.0343 (5)
O4	0.0485 (3)	0.60984 (15)	0.32659 (10)	0.0314 (5)
O5	0.1369 (3)	0.59693 (17)	0.14948 (11)	0.0336 (5)
H5	0.1983	0.5920	0.1962	0.050*
O6	0.3117 (3)	0.63578 (15)	−0.11966 (11)	0.0331 (5)
O7	0.6644 (3)	0.65268 (16)	−0.08064 (10)	0.0330 (5)
O8	0.4600 (3)	0.59961 (16)	0.26055 (11)	0.0357 (5)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0199 (15)	0.0198 (14)	0.0233 (15)	0.0015 (12)	0.0015 (12)	0.0026 (11)
C2	0.0210 (15)	0.0221 (15)	0.0235 (15)	−0.0035 (12)	0.0041 (12)	0.0004 (11)
C3	0.0176 (15)	0.0323 (17)	0.0310 (17)	−0.0011 (13)	0.0000 (13)	0.0009 (12)
C4	0.0250 (16)	0.0266 (15)	0.0182 (15)	−0.0003 (13)	0.0010 (12)	−0.0011 (11)
C5	0.0215 (15)	0.0252 (15)	0.0238 (16)	0.0018 (13)	0.0088 (12)	−0.0026 (12)
C6	0.0158 (14)	0.0247 (15)	0.0262 (15)	0.0011 (13)	−0.0016 (12)	0.0006 (12)
C7	0.0236 (16)	0.0246 (16)	0.0302 (17)	0.0002 (13)	−0.0020 (13)	0.0032 (12)
C8	0.0281 (17)	0.0458 (19)	0.0248 (16)	0.0026 (15)	0.0042 (13)	−0.0019 (14)
C9	0.0256 (15)	0.0238 (15)	0.0225 (15)	0.0029 (14)	0.0014 (12)	−0.0019 (12)
C10	0.0223 (16)	0.0248 (15)	0.0276 (16)	0.0002 (13)	0.0055 (12)	0.0007 (12)
C11	0.0154 (14)	0.0319 (17)	0.0263 (16)	0.0005 (13)	−0.0009 (12)	0.0011 (12)
C12	0.0261 (16)	0.0254 (15)	0.0191 (14)	0.0010 (13)	−0.0006 (12)	0.0000 (12)
C13	0.0223 (15)	0.0252 (15)	0.0238 (16)	−0.0007 (13)	0.0049 (12)	0.0015 (12)
C14	0.0156 (14)	0.0278 (16)	0.0297 (16)	−0.0017 (13)	0.0020 (12)	−0.0001 (12)
C15	0.0228 (15)	0.0339 (17)	0.0282 (16)	0.0023 (14)	0.0024 (13)	−0.0019 (13)
C16	0.0290 (17)	0.0385 (17)	0.0281 (17)	−0.0045 (15)	0.0073 (14)	0.0025 (13)
O1	0.0191 (11)	0.0496 (13)	0.0217 (11)	0.0002 (10)	0.0035 (8)	0.0014 (10)
O2	0.0234 (12)	0.0598 (14)	0.0224 (11)	0.0034 (10)	0.0015 (9)	−0.0048 (10)
O3	0.0238 (11)	0.0573 (15)	0.0225 (11)	0.0067 (10)	0.0053 (9)	−0.0045 (10)
O4	0.0321 (12)	0.0414 (12)	0.0217 (11)	0.0012 (10)	0.0068 (9)	0.0020 (9)
O5	0.0208 (11)	0.0557 (14)	0.0252 (11)	−0.0006 (10)	0.0061 (9)	0.0015 (10)
O6	0.0271 (12)	0.0484 (13)	0.0232 (11)	−0.0025 (10)	−0.0003 (9)	0.0039 (9)
O7	0.0257 (12)	0.0508 (14)	0.0231 (11)	−0.0075 (10)	0.0051 (9)	0.0011 (9)
O8	0.0335 (13)	0.0493 (14)	0.0250 (11)	0.0020 (11)	0.0062 (9)	−0.0010 (10)

Geometric parameters (Å, º)

C1—C2	1.407 (4)	C9—C14	1.422 (3)
C1—C6	1.419 (3)	C9—C15	1.447 (4)
C1—C7	1.438 (4)	C10—O5	1.360 (3)
C2—O1	1.351 (3)	C10—C11	1.399 (3)
C2—C3	1.398 (3)	C11—C12	1.373 (4)
C3—C4	1.360 (3)	C11—H11	0.9500

C3—H3	0.9500	C12—O6	1.366 (3)
C4—O2	1.367 (3)	C12—C13	1.378 (4)
C4—C5	1.389 (4)	C13—C14	1.354 (4)
C5—C6	1.351 (3)	C13—O7	1.387 (3)
C5—O3	1.387 (3)	C14—H14	0.9500
C6—H6	0.9500	C15—O8	1.229 (3)
C7—O4	1.238 (3)	C15—H15	0.9500
C7—H7	0.9500	C16—O7	1.423 (3)
C8—O3	1.432 (3)	C16—O6	1.440 (3)
C8—O2	1.433 (3)	C16—H16A	0.9900
C8—H8A	0.9900	C16—H16B	0.9900
C8—H8B	0.9900	O1—H1	0.8400
C9—C10	1.400 (4)	O5—H5	0.8400

C2—C1—C6	120.8 (2)	O5—C10—C11	116.8 (2)
C2—C1—C7	121.0 (2)	O5—C10—C9	121.5 (2)
C6—C1—C7	118.2 (3)	C11—C10—C9	121.6 (2)
O1—C2—C3	117.4 (2)	C12—C11—C10	115.5 (3)
O1—C2—C1	121.7 (2)	C12—C11—H11	122.3
C3—C2—C1	120.9 (2)	C10—C11—H11	122.3
C4—C3—C2	116.2 (3)	O6—C12—C11	126.0 (3)
C4—C3—H3	121.9	O6—C12—C13	110.2 (2)
C2—C3—H3	121.9	C11—C12—C13	123.8 (2)
C3—C4—O2	126.7 (3)	C14—C13—C12	121.7 (2)
C3—C4—C5	123.7 (2)	C14—C13—O7	128.3 (3)
O2—C4—C5	109.6 (2)	C12—C13—O7	109.9 (2)
C6—C5—O3	128.5 (3)	C13—C14—C9	116.9 (3)
C6—C5—C4	121.6 (2)	C13—C14—H14	121.5
O3—C5—C4	109.9 (2)	C9—C14—H14	121.5
C5—C6—C1	116.8 (3)	O8—C15—C9	124.0 (3)
C5—C6—H6	121.6	O8—C15—H15	118.0
C1—C6—H6	121.6	C9—C15—H15	118.0
O4—C7—C1	125.1 (3)	O7—C16—O6	108.3 (2)
O4—C7—H7	117.5	O7—C16—H16A	110.0
C1—C7—H7	117.5	O6—C16—H16A	110.0
O3—C8—O2	108.1 (2)	O7—C16—H16B	110.0
O3—C8—H8A	110.1	O6—C16—H16B	110.0
O2—C8—H8A	110.1	H16A—C16—H16B	108.4
O3—C8—H8B	110.1	C2—O1—H1	109.5
O2—C8—H8B	110.1	C4—O2—C8	106.6 (2)
H8A—C8—H8B	108.4	C5—O3—C8	105.6 (2)
C10—C9—C14	120.4 (2)	C10—O5—H5	109.5
C10—C9—C15	121.6 (2)	C12—O6—C16	105.8 (2)
C14—C9—C15	118.0 (3)	C13—O7—C16	105.4 (2)

C6—C1—C2—O1	−179.8 (2)	C10—C11—C12—O6	179.6 (2)
C7—C1—C2—O1	1.9 (4)	C10—C11—C12—C13	0.6 (4)
C6—C1—C2—C3	0.8 (4)	O6—C12—C13—C14	−178.7 (2)

C7—C1—C2—C3	−177.5 (2)	C11—C12—C13—C14	0.4 (4)
O1—C2—C3—C4	179.1 (2)	O6—C12—C13—O7	0.4 (3)
C1—C2—C3—C4	−1.4 (4)	C11—C12—C13—O7	179.5 (2)
C2—C3—C4—O2	180.0 (2)	C12—C13—C14—C9	−0.8 (4)
C2—C3—C4—C5	1.1 (4)	O7—C13—C14—C9	−179.6 (2)
C3—C4—C5—C6	0.0 (4)	C10—C9—C14—C13	0.1 (4)
O2—C4—C5—C6	−179.0 (2)	C15—C9—C14—C13	−179.9 (2)
C3—C4—C5—O3	179.2 (3)	C10—C9—C15—O8	−0.5 (4)
O2—C4—C5—O3	0.2 (3)	C14—C9—C15—O8	179.5 (3)
O3—C5—C6—C1	−179.8 (2)	C3—C4—O2—C8	−177.1 (3)
C4—C5—C6—C1	−0.7 (4)	C5—C4—O2—C8	2.0 (3)
C2—C1—C6—C5	0.3 (4)	O3—C8—O2—C4	−3.3 (3)
C7—C1—C6—C5	178.7 (2)	C6—C5—O3—C8	176.9 (3)
C2—C1—C7—O4	−1.8 (4)	C4—C5—O3—C8	−2.2 (3)
C6—C1—C7—O4	179.8 (3)	O2—C8—O3—C5	3.4 (3)
C14—C9—C10—O5	−178.8 (3)	C11—C12—O6—C16	177.0 (3)
C15—C9—C10—O5	1.2 (4)	C13—C12—O6—C16	−3.9 (3)
C14—C9—C10—C11	0.9 (4)	O7—C16—O6—C12	6.0 (3)
C15—C9—C10—C11	−179.1 (3)	C14—C13—O7—C16	−177.7 (3)
O5—C10—C11—C12	178.5 (2)	C12—C13—O7—C16	3.4 (3)
C9—C10—C11—C12	−1.2 (4)	O6—C16—O7—C13	−5.7 (3)

Hydrogen-bond geometry (Å, º)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
O1—H1···O4	0.84	1.92	2.652 (3)	146
O5—H5···O8	0.84	1.91	2.645 (3)	145