

2-*n*-Butyl-6-chloro-1-(2,4-dimethylphenylsulfonyl)-1*H*-benzimidazole–2-*n*-butyl-5-chloro-1-(2,4-dimethylphenylsulfonyl)-1*H*-benzimidazole (0.759/0.241)

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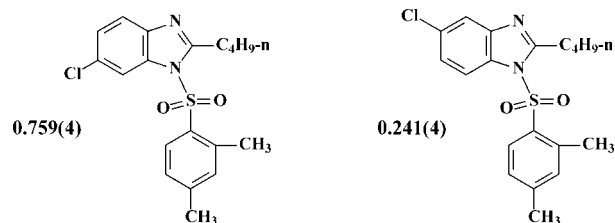
Received 18 October 2011; accepted 11 November 2011

Key indicators: single-crystal X-ray study; $T = 290$ K; mean $\sigma(\text{C}-\text{C}) = 0.007$ Å; disorder in main residue; R factor = 0.069; wR factor = 0.191; data-to-parameter ratio = 11.3.

The title compound, $0.759\text{C}_{19}\text{H}_{21}\text{ClN}_2\text{O}_2\text{S}\cdot 0.241\text{C}_{19}\text{H}_{21}\text{ClN}_2\text{O}_2\text{S}$, was synthesized by arylsulfonylation of 2-*n*-butyl-5-chloro-1*H*-benzimidazole in the presence of triethylamine. The crystal structure is composed of two molecules, 2-*n*-butyl-6-chloro-1-(2,4-dimethylphenylsulfonyl)-1*H*-benzimidazole and 1-(2,4-dimethylphenylsulfonyl)-2-*n*-butyl-5-chloro-1*H*-benzimidazole, in the refined ratio of 0.759 (4):0.241 (4) disordered at the same position in the unit cell. The molecule has three essentially planar fragments *viz.* benzimidazole, dimethylbenzene and *n*-butyl (r.m.s. deviations of 0.009, 0.024 and 0.003 Å, respectively). The angle between the benzimidazole and dimethylbenzene fragments is 86.0 (1)°. In the crystal, pairs of intermolecular C–H··· π interactions form centrosymmetrical dimers, which are linked by weak intermolecular C–H···O hydrogen bonds.

Related literature

For the biological and pharmaceutical properties of benzimidazole derivatives, see: Koči *et al.* (2002); Matsuno *et al.* (2000); Garuti *et al.* (1999). For the synthesis, biological activity and related structures of 2-*n*-butylbenzimidazole derivatives, see: Kubo *et al.* (1993*a,b*); For the arylsulfonylation of benzimidazole derivatives, see: Abdireimov *et al.* (2010).



Experimental

Crystal data

$0.759\text{C}_{19}\text{H}_{21}\text{ClN}_2\text{O}_2\text{S}\cdot 0.241\text{C}_{19}\text{H}_{21}\text{ClN}_2\text{O}_2\text{S}$

$M_r = 376.89$

Triclinic, $P\bar{1}$

$a = 8.7340$ (17) Å

$b = 10.251$ (2) Å

$c = 11.390$ (2) Å

$\alpha = 71.29$ (3)°

$\beta = 78.38$ (3)°

$\gamma = 76.75$ (3)°

$V = 931.1$ (3) Å³

$Z = 2$

Cu $K\alpha$ radiation

$\mu = 2.98$ mm⁻¹

$T = 290$ K

$0.68 \times 0.45 \times 0.20$ mm

Data collection

Stoe Stadi-4 four-circle diffractometer

Absorption correction: ψ scan

(*X-RED*; Stoe & Cie, 1997)

$T_{\min} = 0.250$, $T_{\max} = 0.551$

2722 measured reflections

2714 independent reflections

2460 reflections with $I > 2\sigma(I)$

$\theta_{\max} = 60.0^\circ$

3 standard reflections every 60 min

intensity decay: 10.4%

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.069$

$wR(F^2) = 0.191$

$S = 1.12$

2714 reflections

240 parameters

H-atom parameters constrained

$\Delta\rho_{\max} = 0.35$ e Å⁻³

$\Delta\rho_{\min} = -0.45$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

Cg3 is the centroid of the C12–C17 ring.

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
C19–H19B···O2 ⁱ	0.96	2.62	3.554 (7)	165
C4–H4A···Cg3 ⁱⁱ	0.93	2.76	3.665 (8)	163

Symmetry codes: (i) $x + 1, y, z$; (ii) $-x + 1, -y + 1, -z + 1$.

Data collection: *STADIA* (Stoe & Cie, 1997); cell refinement: *STADIA*; data reduction: *X-RED* (Stoe & Cie, 1997); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *XP* in *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *pubCIF* (Westrip, 2010).

We thank the Academy of Sciences of the Republic of Uzbekistan for supporting this study (grants FA-F3-T045 and FA-A6-T114).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: IM2330).

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supporting information

Acta Cryst. (2011). E67, o3345–o3346 [https://doi.org/10.1107/S1600536811047957]

2-*n*-Butyl-6-chloro-1-(2,4-dimethylphenylsulfonyl)-1*H*-benzimidazole–2-*n*-butyl-5-chloro-1-(2,4-dimethylphenylsulfonyl)-1*H*-benzimidazole (0.759/0.241)

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S1. Comment

Benzimidazole (Koči *et al.*, 2002; Matsuno *et al.*, 2000; Garuti *et al.*, 1999) and 2-*n*-butylbenzimidazole (Kubo *et al.*, 1993*a*; 1993*b*) derivatives are important heterocyclic compounds which have attracted great attention due to their biological and pharmaceutical activities.

Reaction of 2-*n*-butyl-5-chloro-1*H*-benzimidazole with 2,4-dimethylbenzenesulfonyl chloride in the presence of triethylamine results in a mixture of 1-(2,4-dimethylbenzenesulfonyl)-2-*n*-butyl-6-chloro-1*H*-benzimidazole and 1-(2,4-dimethylbenzenesulfonyl)-2-*n*-butyl-5-chloro-1*H*-benzimidazole, in the refined ratio of 0.759 (4):0.241 (4) (Abdireimov *et al.*, 2010). The structure of the received product is investigated by ¹H NMR and X-ray diffraction.

As a whole the molecule consists of three flat fragments: benzimidazole (N1/C2/N3/C3A–C7A), dimethylbenzene (C12–C19) and *n*-butyl (C8–C11) (r.m.s. deviations are 0.009, 0.024 and 0.003 Å, respectively). The angle between flat benzimidazole and dimethylbenzene fragment is 86.0 (1)°, and between benzimidazole and *n*-butyl is 4.4 (2)° (Fig. 1).

The crystal structure is stabilized by intermolecular C—H⋯π interactions observed between the atoms of two benzene rings of neighboring molecules with distance C4—H⋯Cg3ⁱ = 3.665 (2) Å [symmetry code: (i) 1 - x, 1 - y, 1 - z; Cg3 is centroid of the C12–C17 benzene ring]. Observable C—H⋯π interactions form centrosymmetric dimers, another weak intermolecular H-bond such as C19—H⋯O2 sew these dimers (Table 1).

S2. Experimental

In the three-necked round-bottomed flask, supplied with a mechanical mixer, dropping funnel and backflow condenser, were placed 2.04 g (10 mmol) 2,4-dimethylbenzenesulfonyl chloride in 15 ml of acetone and was added a solution of 2.08 g (10 mmol) 2-*n*-butyl-5-chloro-1*H*-benzimidazole and 1.01 g (10 mmol) triethylamine in 30 ml acetone by stirring and cooling. The reaction mixture was stirred at room temperature for 4 h. Afterwards acetone is evaporated. The residual product was washed with 100 ml of the water, obtained crystals were filtered and recrystallized from ethanol. 2.18 g (56%) mixed crystals of 1-(2,4-dimethylbenzenesulfonyl)-2-*n*-butyl-6-chloro-1*H*-benzimidazole (**A**) and 1-(2,4-dimethylbenzenesulfonyl)-2-*n*-butyl-5-chloro-1*H*-benzimidazole (**B**), melting in the temperature range of 108–117°C were obtained.

Colorless crystals suitable for XRD have been received from ethanol at room temperature.

S3. Refinement

The 10.4% decay correction was applied by using the programm *X-RED*. The H atoms bonded to C atoms were placed geometrically (with C—H distances of 0.97 Å for CH₂; 0.96 Å for CH₃; and 0.93 Å for C_{ar}) and included in the refinement in a riding motion approximation with $U_{iso}=1.2U_{eq}(C)$ [$U_{iso}=1.5U_{eq}(C)$ for methyl H atoms].

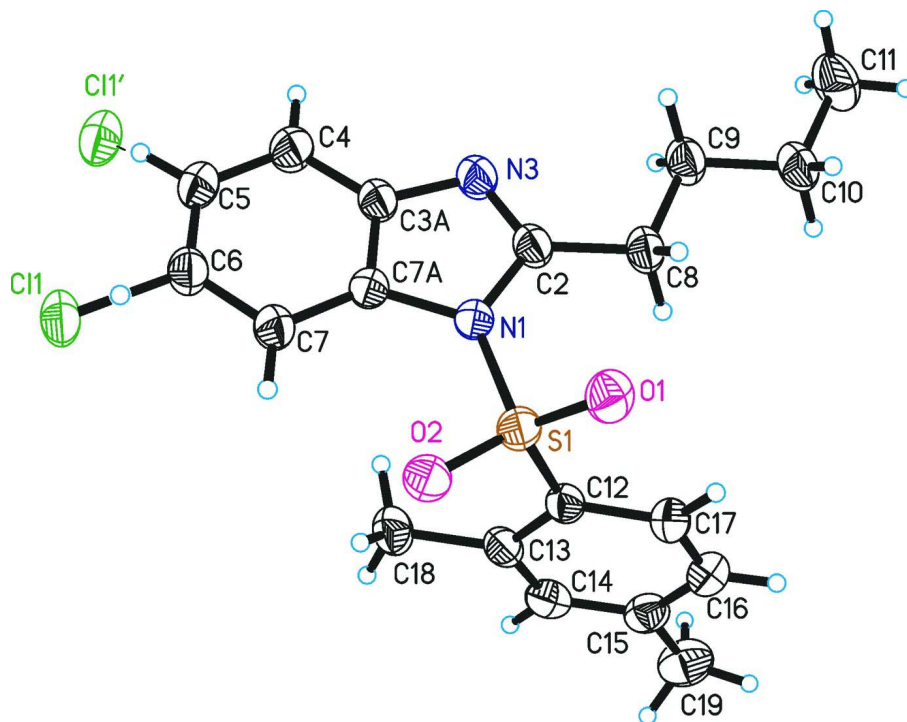


Figure 1

Molecular structure of the title compound, displacement ellipsoids are drawn at the 30% probability level.

2-*n*-Butyl-6-chloro-1-(2,4-dimethylphenylsulfonyl)-1*H*-benzimidazole–2-*n*-butyl-5-chloro-1-(2,4-dimethylphenylsulfonyl)-1*H*-benzimidazole (0.759/0.241)

Crystal data

0.7590.241C₁₉H₂₁ClN₂O₂S·0.2410.241C₁₉H₂₁ClN₂O₂S

$M_r = 376.89$

Triclinic, $P\bar{1}$

Hall symbol: -P 1

$a = 8.7340$ (17) Å

$b = 10.251$ (2) Å

$c = 11.390$ (2) Å

$\alpha = 71.29$ (3)°

$\beta = 78.38$ (3)°

$\gamma = 76.75$ (3)°

$V = 931.1$ (3) Å³

$Z = 2$

$F(000) = 396$

$D_x = 1.344$ Mg m⁻³

Melting point < 381(9) K

Cu $K\alpha$ radiation, $\lambda = 1.54184$ Å

Cell parameters from 13 reflections

$\theta = 10\text{--}20^\circ$

$\mu = 2.98$ mm⁻¹

$T = 290$ K

Prismatic, colorless

$0.68 \times 0.45 \times 0.20$ mm

Data collection

Stoe Stadi-4 four-circle
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Scan width (ω) = 1.56 – 1.80, scan ratio $2\theta:\omega =$

1.00 I(Net) and sigma(I) calculated according to
Blessing (1987)

Absorption correction: ψ scan

(*X-RED*; Stoe & Cie, 1997)

$T_{\min} = 0.250$, $T_{\max} = 0.551$

2722 measured reflections

2714 independent reflections

2460 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.000$

$\theta_{\max} = 60.0^\circ$, $\theta_{\min} = 4.1^\circ$

$h = -9 \rightarrow 9$

$k = -10 \rightarrow 11$

$l = 0 \rightarrow 12$

3 standard reflections every 60 min

intensity decay: 10.4%

*Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.069$ $wR(F^2) = 0.191$ $S = 1.12$

2714 reflections

240 parameters

0 restraints

Primary atom site location: structure-invariant
direct methodsSecondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0919P)^2 + 1.1929P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} < 0.001$ $\Delta\rho_{\max} = 0.35 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{\min} = -0.45 \text{ e } \text{\AA}^{-3}$ Extinction correction: *SHELXL*, $F_c^* = kF_c[1 + 0.001 \times F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.026 (3)

*Special details***Experimental.** Empirical absorption correction using ψ Scan. Reflections used $\text{Mu} * R = 0.00$ H K L, θ , χ , I_{\min}/I_{\max} : -1 -2 4 45.0 82.7 0.455**¹H NMR (400 MHz, CDCl₃- δ):** 1-(2,4-dimethylbenzenesulfonyl)-2-*n*-butyl-6-chloro-1*H*-benzimidazole (**A**). 7.97 (1*H*, d, $J=2.1$ Hz, H-7), 7.49 (1*H*, d, $J=8.3$ Hz, H-17), 7.49 (1*H*, d, $J=8.5$ Hz, H-4), 7.37 (2*H*, d, $J=7.9$ Hz, H-14, 16), 7.29 (1*H*, dd, $J=2.0$, $J=8.5$ Hz, H-5), 3.08 (2*H*, m, CH₂-8), 2.34 (6*H*, s, CH₃-18, 19), 1.73 (2*H*, m, CH₂-9), 1.37 (2*H*, m, CH₂-10), 0.89 (3*H*, t, $J=7.3$ Hz, CH₃-11).1-(2,4-dimethylbenzenesulfonyl)-2-*n*-butyl-5-chloro-1*H*-benzimidazole (**B**). 7.95 (1*H*, d, $J=8.7$ Hz, H-7), 7.78 (1*H*, d, $J=8.3$ Hz, H-17), 7.52 (1*H*, d, $J=2.0$ Hz, H-4), 7.35 (2*H*, d, $J=7.9$ Hz, H-14, 16), 7.31 (1*H*, dd, $J=2.0$, $J=8.5$ Hz, H-6), 3.08 (2*H*, m, CH₂-8), 2.34 (6*H*, s, CH₃-18, 19), 1.73 (2*H*, m, CH₂-9), 1.37 (2*H*, m, CH₂-10), 0.90 (3*H*, t, $J=7.5$ Hz, CH₃-11).**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.**Refinement.** Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
S1	0.25613 (12)	0.65554 (11)	0.19565 (9)	0.0537 (4)	
O1	0.2520 (4)	0.5625 (3)	0.1267 (3)	0.0660 (9)	
O2	0.1230 (4)	0.7622 (3)	0.2077 (3)	0.0647 (9)	
N1	0.2783 (4)	0.5558 (3)	0.3424 (3)	0.0511 (8)	
N3	0.3767 (4)	0.3796 (4)	0.4993 (3)	0.0606 (10)	
C2	0.3863 (5)	0.4310 (4)	0.3785 (4)	0.0552 (10)	
C3A	0.2567 (5)	0.4705 (4)	0.5479 (4)	0.0552 (10)	
C4	0.1997 (6)	0.4606 (5)	0.6731 (4)	0.0688 (13)	
H4A	0.2428	0.3874	0.7370	0.083*	
C5	0.0780 (6)	0.5624 (5)	0.6986 (4)	0.0658 (12)	
H5A	0.0377	0.5584	0.7815	0.079*	0.759 (4)
C6	0.0142 (5)	0.6701 (5)	0.6052 (4)	0.0604 (11)	
H6A	-0.0684	0.7372	0.6269	0.072*	0.241 (4)
C7	0.0675 (5)	0.6835 (5)	0.4791 (4)	0.0573 (11)	
H7A	0.0226	0.7564	0.4158	0.069*	

C7A	0.1909 (5)	0.5817 (4)	0.4545 (4)	0.0498 (10)	
C8	0.4990 (5)	0.3662 (5)	0.2876 (4)	0.0606 (11)	
H8A	0.5704	0.4300	0.2385	0.073*	
H8B	0.4397	0.3519	0.2306	0.073*	
C9	0.5963 (6)	0.2267 (5)	0.3510 (4)	0.0640 (12)	
H9A	0.6586	0.2409	0.4061	0.077*	
H9B	0.5255	0.1631	0.4016	0.077*	
C10	0.7063 (7)	0.1625 (5)	0.2552 (5)	0.0754 (14)	
H10A	0.7774	0.2261	0.2052	0.090*	
H10B	0.6439	0.1496	0.1997	0.090*	
C11	0.8036 (7)	0.0226 (6)	0.3163 (6)	0.0918 (18)	
H11A	0.8725	-0.0141	0.2528	0.138*	
H11B	0.7337	-0.0417	0.3638	0.138*	
H11C	0.8662	0.0349	0.3710	0.138*	
C12	0.4289 (5)	0.7259 (4)	0.1379 (4)	0.0498 (10)	
C13	0.4696 (5)	0.8270 (4)	0.1800 (4)	0.0553 (11)	
C14	0.6081 (6)	0.8746 (4)	0.1239 (4)	0.0603 (11)	
H14A	0.6356	0.9413	0.1512	0.072*	
C15	0.7100 (5)	0.8296 (5)	0.0291 (4)	0.0615 (11)	
C16	0.6664 (6)	0.7320 (5)	-0.0118 (4)	0.0657 (12)	
H16A	0.7317	0.6998	-0.0757	0.079*	
C17	0.5288 (6)	0.6823 (5)	0.0405 (4)	0.0609 (11)	
H17A	0.5009	0.6178	0.0105	0.073*	
C18	0.3722 (6)	0.8773 (5)	0.2876 (5)	0.0702 (13)	
H18A	0.4075	0.9581	0.2906	0.105*	
H18B	0.3850	0.8043	0.3646	0.105*	
H18C	0.2623	0.9014	0.2761	0.105*	
C19	0.8634 (6)	0.8809 (7)	-0.0233 (6)	0.0881 (16)	
H19A	0.9019	0.8631	-0.1029	0.132*	
H19B	0.9399	0.8329	0.0331	0.132*	
H19C	0.8471	0.9796	-0.0340	0.132*	
Cl1	-0.1412 (2)	0.79798 (18)	0.64266 (16)	0.0782 (7)	0.759 (4)
Cl1'	0.0475 (7)	0.5668 (7)	0.8540 (5)	0.081 (2)	0.241 (4)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0575 (7)	0.0540 (7)	0.0473 (6)	-0.0039 (5)	-0.0118 (5)	-0.0129 (5)
O1	0.077 (2)	0.073 (2)	0.0594 (18)	-0.0196 (16)	-0.0128 (15)	-0.0283 (16)
O2	0.0621 (18)	0.0621 (18)	0.0554 (18)	0.0096 (14)	-0.0143 (14)	-0.0076 (14)
N1	0.056 (2)	0.0460 (18)	0.0433 (18)	-0.0003 (15)	-0.0038 (15)	-0.0104 (15)
N3	0.068 (2)	0.053 (2)	0.054 (2)	0.0043 (17)	-0.0096 (17)	-0.0153 (17)
C2	0.056 (2)	0.049 (2)	0.059 (3)	0.0009 (19)	-0.0074 (19)	-0.020 (2)
C3A	0.058 (2)	0.050 (2)	0.054 (2)	0.0005 (19)	-0.0065 (19)	-0.0173 (19)
C4	0.082 (3)	0.066 (3)	0.049 (3)	0.000 (2)	-0.008 (2)	-0.013 (2)
C5	0.073 (3)	0.070 (3)	0.049 (2)	-0.009 (2)	0.002 (2)	-0.019 (2)
C6	0.054 (2)	0.062 (3)	0.063 (3)	-0.006 (2)	0.001 (2)	-0.025 (2)
C7	0.053 (2)	0.056 (2)	0.056 (3)	0.0005 (19)	-0.0031 (19)	-0.016 (2)

C7A	0.051 (2)	0.049 (2)	0.046 (2)	-0.0048 (18)	-0.0043 (17)	-0.0128 (18)
C8	0.063 (3)	0.058 (3)	0.058 (3)	0.003 (2)	-0.004 (2)	-0.024 (2)
C9	0.063 (3)	0.060 (3)	0.065 (3)	0.006 (2)	-0.008 (2)	-0.025 (2)
C10	0.084 (3)	0.058 (3)	0.078 (3)	0.011 (2)	-0.008 (3)	-0.028 (3)
C11	0.082 (4)	0.075 (4)	0.112 (5)	0.023 (3)	-0.021 (3)	-0.041 (3)
C12	0.057 (2)	0.045 (2)	0.045 (2)	-0.0020 (18)	-0.0097 (18)	-0.0132 (17)
C13	0.067 (3)	0.045 (2)	0.053 (2)	0.001 (2)	-0.015 (2)	-0.0169 (19)
C14	0.071 (3)	0.049 (2)	0.063 (3)	-0.011 (2)	-0.021 (2)	-0.011 (2)
C15	0.063 (3)	0.062 (3)	0.053 (2)	-0.008 (2)	-0.013 (2)	-0.006 (2)
C16	0.071 (3)	0.068 (3)	0.054 (3)	-0.013 (2)	0.002 (2)	-0.016 (2)
C17	0.076 (3)	0.057 (3)	0.051 (2)	-0.010 (2)	-0.004 (2)	-0.021 (2)
C18	0.085 (3)	0.065 (3)	0.065 (3)	-0.004 (2)	-0.009 (2)	-0.033 (2)
C19	0.076 (4)	0.096 (4)	0.086 (4)	-0.026 (3)	-0.013 (3)	-0.010 (3)
Cl1	0.0714 (11)	0.0803 (12)	0.0733 (11)	0.0080 (8)	0.0057 (8)	-0.0333 (9)
Cl1'	0.077 (4)	0.100 (4)	0.069 (3)	-0.018 (3)	0.010 (3)	-0.039 (3)

Geometric parameters (Å, °)

S1—O2	1.419 (3)	C9—H9A	0.9700
S1—O1	1.427 (3)	C9—H9B	0.9700
S1—N1	1.679 (3)	C10—C11	1.517 (7)
S1—C12	1.739 (4)	C10—H10A	0.9700
N1—C2	1.401 (5)	C10—H10B	0.9700
N1—C7A	1.420 (5)	C11—H11A	0.9600
N3—C2	1.298 (6)	C11—H11B	0.9600
N3—C3A	1.387 (5)	C11—H11C	0.9600
C2—C8	1.488 (6)	C12—C17	1.394 (6)
C3A—C7A	1.389 (6)	C12—C13	1.410 (6)
C3A—C4	1.392 (6)	C13—C14	1.366 (6)
C4—C5	1.366 (7)	C13—C18	1.508 (6)
C4—H4A	0.9300	C14—C15	1.383 (7)
C5—C6	1.365 (7)	C14—H14A	0.9300
C5—Cl1'	1.750 (7)	C15—C16	1.382 (7)
C5—H5A	0.9300	C15—C19	1.493 (7)
C6—C7	1.390 (6)	C16—C17	1.362 (7)
C6—Cl1	1.747 (5)	C16—H16A	0.9300
C6—H6A	0.9300	C17—H17A	0.9300
C7—C7A	1.371 (6)	C18—H18A	0.9600
C7—H7A	0.9300	C18—H18B	0.9600
C8—C9	1.522 (6)	C18—H18C	0.9600
C8—H8A	0.9700	C19—H19A	0.9600
C8—H8B	0.9700	C19—H19B	0.9600
C9—C10	1.515 (6)	C19—H19C	0.9600
O2—S1—O1	119.5 (2)	C8—C9—H9B	109.4
O2—S1—N1	105.51 (18)	H9A—C9—H9B	108.0
O1—S1—N1	106.36 (18)	C9—C10—C11	112.0 (5)
O2—S1—C12	110.7 (2)	C9—C10—H10A	109.2

O1—S1—C12	108.22 (19)	C11—C10—H10A	109.2
N1—S1—C12	105.50 (18)	C9—C10—H10B	109.2
C2—N1—C7A	106.4 (3)	C11—C10—H10B	109.2
C2—N1—S1	126.9 (3)	H10A—C10—H10B	107.9
C7A—N1—S1	126.7 (3)	C10—C11—H11A	109.5
C2—N3—C3A	105.6 (3)	C10—C11—H11B	109.5
N3—C2—N1	112.4 (4)	H11A—C11—H11B	109.5
N3—C2—C8	124.4 (4)	C10—C11—H11C	109.5
N1—C2—C8	123.2 (4)	H11A—C11—H11C	109.5
N3—C3A—C7A	112.1 (4)	H11B—C11—H11C	109.5
N3—C3A—C4	128.0 (4)	C17—C12—C13	118.9 (4)
C7A—C3A—C4	119.9 (4)	C17—C12—S1	116.9 (3)
C5—C4—C3A	117.5 (4)	C13—C12—S1	124.2 (3)
C5—C4—H4A	121.3	C14—C13—C12	117.5 (4)
C3A—C4—H4A	121.3	C14—C13—C18	119.6 (4)
C6—C5—C4	121.5 (4)	C12—C13—C18	122.8 (4)
C6—C5—C11'	122.7 (4)	C13—C14—C15	124.2 (4)
C4—C5—C11'	114.0 (4)	C13—C14—H14A	117.9
C6—C5—H5A	119.2	C15—C14—H14A	117.9
C4—C5—H5A	119.2	C16—C15—C14	117.3 (4)
C5—C6—C7	122.8 (4)	C16—C15—C19	121.5 (5)
C5—C6—C11	119.8 (4)	C14—C15—C19	121.2 (5)
C7—C6—C11	117.4 (4)	C17—C16—C15	120.8 (4)
C5—C6—H6A	118.6	C17—C16—H16A	119.6
C7—C6—H6A	118.6	C15—C16—H16A	119.6
C7A—C7—C6	115.2 (4)	C16—C17—C12	121.4 (4)
C7A—C7—H7A	122.4	C16—C17—H17A	119.3
C6—C7—H7A	122.4	C12—C17—H17A	119.3
C7—C7A—C3A	123.0 (4)	C13—C18—H18A	109.5
C7—C7A—N1	133.5 (4)	C13—C18—H18B	109.5
C3A—C7A—N1	103.5 (3)	H18A—C18—H18B	109.5
C2—C8—C9	112.6 (4)	C13—C18—H18C	109.5
C2—C8—H8A	109.1	H18A—C18—H18C	109.5
C9—C8—H8A	109.1	H18B—C18—H18C	109.5
C2—C8—H8B	109.1	C15—C19—H19A	109.5
C9—C8—H8B	109.1	C15—C19—H19B	109.5
H8A—C8—H8B	107.8	H19A—C19—H19B	109.5
C10—C9—C8	111.0 (4)	C15—C19—H19C	109.5
C10—C9—H9A	109.4	H19A—C19—H19C	109.5
C8—C9—H9A	109.4	H19B—C19—H19C	109.5
C10—C9—H9B	109.4		
O2—S1—N1—C2	174.6 (3)	C4—C3A—C7A—N1	180.0 (4)
O1—S1—N1—C2	46.7 (4)	C2—N1—C7A—C7	-177.5 (5)
C12—S1—N1—C2	-68.1 (4)	S1—N1—C7A—C7	2.5 (7)
O2—S1—N1—C7A	-5.4 (4)	C2—N1—C7A—C3A	0.9 (4)
O1—S1—N1—C7A	-133.3 (3)	S1—N1—C7A—C3A	-179.1 (3)
C12—S1—N1—C7A	111.9 (4)	N3—C2—C8—C9	3.6 (7)

C3A—N3—C2—N1	1.0 (5)	N1—C2—C8—C9	-176.6 (4)
C3A—N3—C2—C8	-179.2 (4)	C2—C8—C9—C10	178.5 (4)
C7A—N1—C2—N3	-1.2 (5)	C8—C9—C10—C11	-179.4 (5)
S1—N1—C2—N3	178.8 (3)	O2—S1—C12—C17	-133.0 (3)
C7A—N1—C2—C8	179.0 (4)	O1—S1—C12—C17	-0.2 (4)
S1—N1—C2—C8	-1.0 (6)	N1—S1—C12—C17	113.3 (3)
C2—N3—C3A—C7A	-0.4 (5)	O2—S1—C12—C13	44.3 (4)
C2—N3—C3A—C4	179.3 (5)	O1—S1—C12—C13	177.1 (3)
N3—C3A—C4—C5	-179.1 (5)	N1—S1—C12—C13	-69.4 (4)
C7A—C3A—C4—C5	0.6 (7)	C17—C12—C13—C14	-1.5 (6)
C3A—C4—C5—C6	0.0 (8)	S1—C12—C13—C14	-178.7 (3)
C3A—C4—C5—C11'	-164.9 (4)	C17—C12—C13—C18	-177.9 (4)
C4—C5—C6—C7	0.1 (8)	S1—C12—C13—C18	4.9 (6)
C11'—C5—C6—C7	163.8 (4)	C12—C13—C14—C15	0.0 (7)
C4—C5—C6—C11	-179.9 (4)	C18—C13—C14—C15	176.5 (4)
C11'—C5—C6—C11	-16.3 (7)	C13—C14—C15—C16	1.0 (7)
C5—C6—C7—C7A	-0.9 (7)	C13—C14—C15—C19	-176.8 (4)
C11—C6—C7—C7A	179.2 (3)	C14—C15—C16—C17	-0.4 (7)
C6—C7—C7A—C3A	1.5 (6)	C19—C15—C16—C17	177.4 (5)
C6—C7—C7A—N1	179.6 (4)	C15—C16—C17—C12	-1.1 (7)
N3—C3A—C7A—C7	178.3 (4)	C13—C12—C17—C16	2.1 (7)
C4—C3A—C7A—C7	-1.4 (7)	S1—C12—C17—C16	179.5 (4)
N3—C3A—C7A—N1	-0.3 (5)		

Hydrogen-bond geometry (\AA , $^\circ$)

$Cg3$ is the centroid of the C12–C17 ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C19—H19B \cdots O2 ⁱ	0.96	2.62	3.554 (7)	165
C4—H4A \cdots Cg3 ⁱⁱ	0.93	2.76	3.665 (8)	163

Symmetry codes: (i) $x+1, y, z$; (ii) $-x+1, -y+1, -z+1$.