

Dichlorido[(4*E*,11*E*)-5,7,12,14-tetra-benzyl-7,14-dimethyl-1,4,8,11-tetraaza-cyclotetradeca-4,11-diene]cobalt(III) perchlorate

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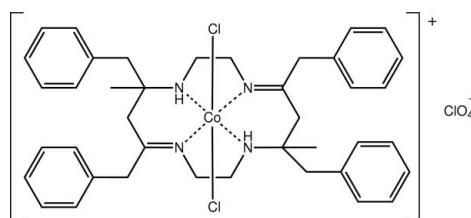
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Key indicators: single-crystal X-ray study; $T = 153\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.005\text{ \AA}$; disorder in main residue; R factor = 0.048; wR factor = 0.142; data-to-parameter ratio = 15.7.

The Co^{III} atom in the title complex, $[\text{CoCl}_2(\text{C}_{40}\text{H}_{48}\text{N}_4)]\text{ClO}_4$, is octahedrally coordinated within a *trans*-Cl₂N₄ donor set provided by the tetradeятate macrocyclic ligand and two chloride ions. The N–H atoms, which are orientated to one side of the N₄ plane, form hydrogen bonds with chloride ions and perchlorate-O atoms. These along with C–H···O interactions consolidate the three-dimensional crystal structure. One of the benzene rings was disordered. This was resolved over two positions with the major component of the disorder having a site-occupancy factor of 0.672 (4).

Related literature

For background to the synthesis, characterization, kinetic studies and biological activity of 14-membered methyl-substituted tetraazamacrocyclic ligands, their *N*-substituted derivatives and metal complexes, see: Bembi *et al.* (1990); Roy *et al.* (2007, 2011a); Hazari *et al.* (2008). For a related structure, see: Roy *et al.* (2011b).



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Experimental

Crystal data

| | |
|---|--|
| $[\text{CoCl}_2(\text{C}_{40}\text{H}_{48}\text{N}_4)]\text{ClO}_4$ | $\gamma = 68.65 (2)^\circ$ |
| $M_r = 814.10$ | $V = 1915.6 (5)\text{ \AA}^3$ |
| Triclinic, $P\bar{1}$ | $Z = 2$ |
| $a = 10.8111 (7)\text{ \AA}$ | Mo $K\alpha$ radiation |
| $b = 13.835 (2)\text{ \AA}$ | $\mu = 0.70\text{ mm}^{-1}$ |
| $c = 14.868 (3)\text{ \AA}$ | $T = 153\text{ K}$ |
| $\alpha = 73.66 (3)^\circ$ | $0.30 \times 0.20 \times 0.10\text{ mm}$ |
| $\beta = 70.06 (3)^\circ$ | |

Data collection

| | |
|--|--|
| Rigaku AFC12/SATURN724 diffractometer | 20336 measured reflections |
| Absorption correction: multi-scan (<i>ABSCOR</i> ; Higashi, 1995) | 7470 independent reflections |
| $T_{\min} = 0.627$, $T_{\max} = 1.000$ | 6940 reflections with $I > 2\sigma(I)$ |
| | $R_{\text{int}} = 0.027$ |

Refinement

| | |
|---------------------------------|--|
| $R[F^2 > 2\sigma(F^2)] = 0.048$ | H atoms treated by a mixture of independent and constrained refinement |
| $wR(F^2) = 0.142$ | $\Delta\rho_{\text{max}} = 0.72\text{ e \AA}^{-3}$ |
| $S = 1.05$ | $\Delta\rho_{\text{min}} = -0.50\text{ e \AA}^{-3}$ |
| 7470 reflections | |
| 476 parameters | |
| 2 restraints | |

Table 1
Hydrogen-bond geometry (\AA , $^\circ$).

| $D-\text{H}\cdots A$ | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|------------------------------|--------------|--------------------|-------------|----------------------|
| N2–H2n···O1 | 0.88 (3) | 2.24 (3) | 3.063 (4) | 155 (2) |
| N4–H4n···Cl2 ⁱ | 0.88 (3) | 2.64 (2) | 3.432 (2) | 150 (3) |
| C10–H10a···O3 ⁱⁱ | 0.99 | 2.50 | 3.437 (4) | 159 |
| C19–H19b···O1 ⁱⁱⁱ | 0.99 | 2.54 | 3.409 (4) | 147 |
| C38a–H38a···O4 ^{iv} | 0.95 | 2.57 | 3.480 (3) | 160 |

Symmetry codes: (i) $-x, -y + 1, -z + 1$; (ii) $x - 1, y, z$; (iii) $-x + 1, -y, -z + 1$; (iv) $x - 1, y, z + 1$.

Data collection: *CrystalClear* (Rigaku/MSC, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *PATTY* in *DIRDIF92* (Beurskens *et al.*, 1992); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997) & *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *publCIF* (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: PV2479).

References

- Bembi, R., Roy, T. G., Jhanji, A. K. & Maheswari, A. (1990). *J. Chem. Soc. Dalton Trans.* pp. 3531–3534.
- Beurskens, P. T., Admiraal, G., Beurskens, G., Bosman, W. P., Garcia-Granda, S., Gould, R. O., Smits, J. M. M. & Smykalla, C. (1992). The *DIRDIF* Program System. Technical Report. Crystallography Laboratory, University of Nijmegen, The Netherlands.
- Brandenburg, K. (2006). *DIAMOND*. Crystal Impact GbR, Bonn, Germany.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Hazari, S. K. S., Roy, T. G., Barua, K. K. & Tieckink, E. R. T. (2008). *J. Chem. Crystallogr.* **38**, 1–8.

- Higashi, T. (1995). *ABSCOR*. Rigaku Corporation, Tokyo, Japan.
- Rigaku/MSC. (2005). *CrystalClear*. MSC, The Woodlands, Texas, USA, and Rigaku Corporation, Tokyo, Japan.
- Roy, T. G., Hazari, S. K. S., Dey, B. K., Miah, H. A., Olbrich, F. & Rehder, D. (2007). *Inorg. Chem.* **46**, 5372–5380.
- Roy, T. G., Hazari, S. K. S., Dey, B. K., Nath, B. C., Dutta, A., Olbrich, F. & Rehder, D. (2011a). *Inorg. Chim. Acta*, **371**, 63–70.
- Roy, T. G., Hazari, S. K. S., Nath, B. C., Ng, S. W. & Tiekink, E. R. T. (2011b). *Acta Cryst. E* **67**, m1581–m1582.
- Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.
- Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.

supporting information

Acta Cryst. (2011). E67, m1722–m1723 [https://doi.org/10.1107/S1600536811046484]

Dichlorido[(4*E*,11*E*)-5,7,12,14-tetrabenzyl-7,14-dimethyl-1,4,8,11-tetraaza-cyclotetradeca-4,11-diene]cobalt(III) perchlorate

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S1. Comment

In continuation of on-going studies of the synthesis, characterization and biological activities of substituted tetraaza-macrocyclic ligands and their metal complexes (Bembi *et al.*, 1990; Roy *et al.*, 2007; Hazari *et al.*, 2008; Roy *et al.*, 2011a; Roy *et al.* 2011b), the synthesis and crystal structure of the title complex, (I), was investigated.

In (I), Fig. 1, the Co^{III} atom exists within a *trans*-Cl₂N₄ donor set defined by the four nitrogen atoms of the macrocyclic ligand and two chlorido atoms. The coordination geometry is based on an octahedron, with the greatest angular distortion manifested in the N2—Co—N3 angle of 83.81 (9)[°]. With respect to the central N₄ plane, the rings adopt three distinct orientations. Two rings adopt similar orientations lying approximately perpendicular and parallel to the N₄ plane: the dihedral angle between the N₄ and the C12—C17 and C20—C25 planes are 86.928 (8) and 78.645 (10)[°], respectively. The C27—C32 ring is also orientated in a perpendicular fashion (dihedral angle = 88.921 (10)[°]) but lies to one side of the N₄ plane, with the C6—C26—C27—C28 torsion angle = 117.15 (4)[°]. The final ring is disordered over two positions. The major component is approximately planar with the N₄ donor set, forming a dihedral angle of 20.644 (10)[°], whereas the minor component forms a dihedral angle of 13.400 (9)[°], *i.e.* even more co-planar. Within the N₄ donor set, the two amine-H atoms are orientated to one side of the plane. The N2—H atom forms a contact with the perchlorate-O1 atom, and the N4—H forms an intramolecular N—H···Cl hydrogen bond, Table 2. These interactions along with several C—H···O contacts lead to the formation of supramolecular arrays in the *ab* plane. The layers stack along the *c* axis with the closest connection being of the type C—H···O, involving the perchlorate-O4 atom (Fig. 2 and Table 1).

S2. Experimental

The macrocyclic ligand, (4*E*,11*E*)-5,7,12,14-tetrabenzyl-7,14-dimethyl-1,4,8,11-tetraazacyclotetradeca-4,11-diene (0.783 g, 1.0 mmol) was suspended in methanol (30 ml). Separately, cobaltous acetate (0.248 g, 1.0 mmol) was dissolved in methanol (30 ml). The combined solutions were heated on a water bath until the solution turned red. Concentrated HCl was added drop-wise so that the solution turned green. Then, about 1 ml HClO₄ was added whereupon a green product started to appear. The mixture was heated in order to reduce the volume to 15 ml. The resulting solution was kept at room temperature for about 1 h. The solid product, (I), was separated by filtration, washed with dry ethanol, followed by diethylether and dried in a vacuum desiccator over silica-gel. The yield was about 50%. The same complex was also prepared by using the acetonitrile as the solvent instead of methanol. However, the yield was about 42%. Green crystals of (I) were isolated from the slow evaporation of its methanol solution.

S3. Refinement

The H-atoms were placed in calculated positions (C—H = 0.95–0.99 Å) and were included in the refinement in the riding model approximation, with $U_{\text{iso}}(\text{H}) = 1.2\text{--}1.5U_{\text{eq}}(\text{C})$. The N—H atoms were located from a difference map and refined

with $N—H = 0.88 \pm 0.01 \text{ \AA}$, and with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{equiv}}(\text{N})$. The C35–C40 phenyl ring was found to be disordered over two positions with a dihedral angle of 24.9 (3) \AA between the orientations. After anisotropic refinement (pairs of atoms were constrained to have equivalent anisotropic displacement parameters), the major component had a site occupancy = 0.672 (4). A number of reflections, *i.e.* (3 2 11), (6 0 6), (10 5 0), (3 2 10), (2 1 12) and (2 2 11), were omitted from the final refinement owing to poor agreement.

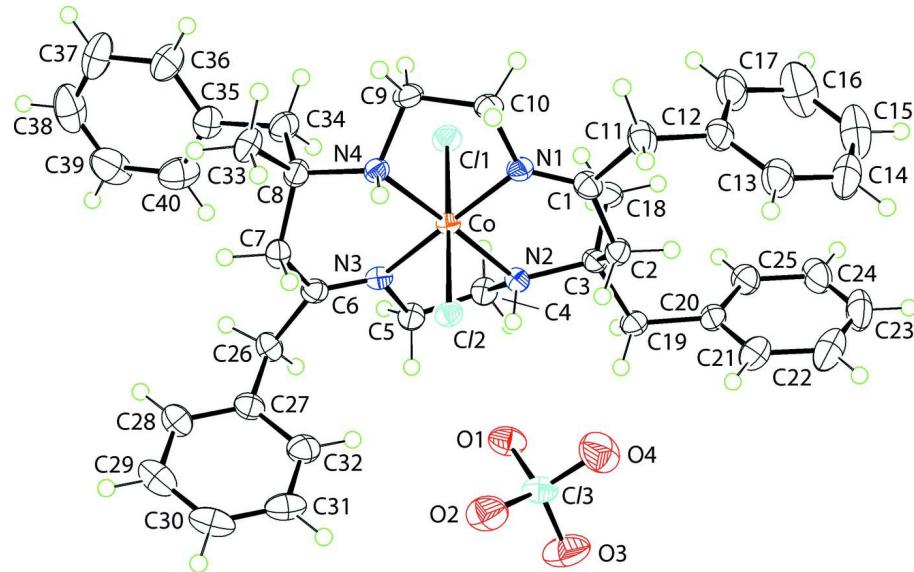
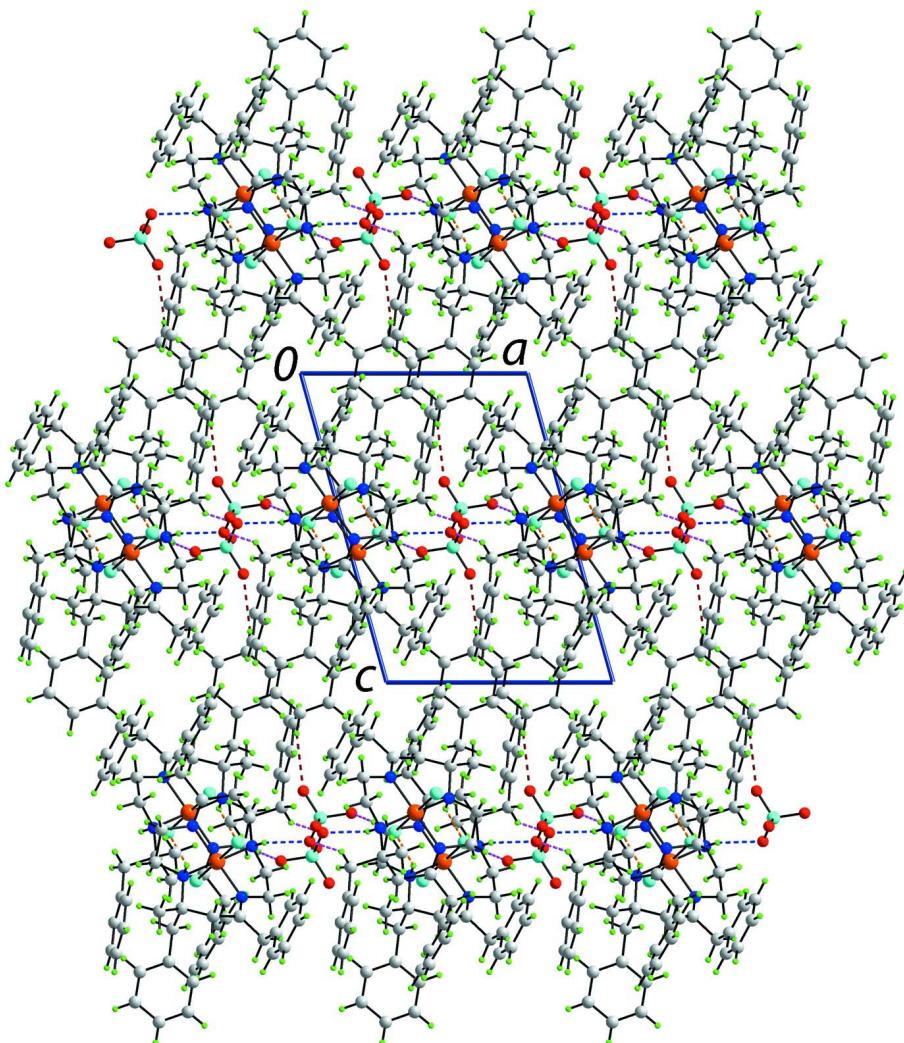


Figure 1

The molecular structure of (I) showing the atom-labelling scheme and displacement ellipsoids at the 50% probability level. Only the major orientation of the disordered C35–C40 ring is shown.

**Figure 2**

A view of the unit-cell contents in projection down the b axis in (I). The $\text{N}—\text{H}··\cdot\text{O}(\text{perchlorate})$, $\text{N}—\text{H}··\cdot\text{Cl}$, $\text{C}—\text{H}··\cdot\text{O}(\text{intra-layer})$ and $\text{C}—\text{H}··\cdot\text{O}(\text{inter-layer})$ interactions are shown as blue, orange, pink and brown dashed lines, respectively.

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Crystal data



$M_r = 814.10$

Triclinic, $P\bar{1}$

Hall symbol: -P 1

$a = 10.8111 (7)$ Å

$b = 13.835 (2)$ Å

$c = 14.868 (3)$ Å

$\alpha = 73.66 (3)^\circ$

$\beta = 70.06 (3)^\circ$

$\gamma = 68.65 (2)^\circ$

$V = 1915.6 (5)$ Å³

$Z = 2$

$F(000) = 852$

$D_x = 1.411 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 6113 reflections

$\theta = 2.2\text{--}30.4^\circ$

$\mu = 0.70 \text{ mm}^{-1}$

$T = 153$ K

Prism, green

$0.30 \times 0.20 \times 0.10$ mm

Data collection

Rigaku AFC12K/SATURN724
diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
 ω scans
Absorption correction: multi-scan
(*ABSCOR*; Higashi, 1995)
 $T_{\min} = 0.627$, $T_{\max} = 1.000$

20336 measured reflections
7470 independent reflections
6940 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.027$
 $\theta_{\max} = 26.0^\circ$, $\theta_{\min} = 2.2^\circ$
 $h = -13 \rightarrow 13$
 $k = -15 \rightarrow 17$
 $l = -18 \rightarrow 18$

Refinement

Refinement on F^2
Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.048$
 $wR(F^2) = 0.142$
 $S = 1.05$
7470 reflections
476 parameters
2 restraints
Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map
Hydrogen site location: inferred from
neighbouring sites
H atoms treated by a mixture of independent
and constrained refinement
 $w = 1/[\sigma^2(F_o^2) + (0.0779P)^2 + 2.2391P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.72 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.50 \text{ e } \text{\AA}^{-3}$

Special details

Experimental. Microanalysis: Calculated for $C_{40}H_{48}Cl_3CoN_4O_4$, C, 59.09; H, 5.78; N, 6.89; Co, 7.25%. Found, C, 59.25; H, 5.65; N, 6.89; Co, 7.05%. IR (cm^{-1}): 3161 $\nu(\text{N}-\text{H})$; 3024 $\nu(\text{Ar}-\text{H})$; 2949 and 2978 $\nu(\text{C}-\text{H})$; 1393 $\nu(\text{CH}_3)$; 1095 and 622 $\nu(\text{ClO}_4^-)$; 550 $\nu(\text{Co}-\text{N})$.

Geometry. All s.u.'s (except the s.u. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | <i>x</i> | <i>y</i> | <i>z</i> | $U_{\text{iso}}^*/U_{\text{eq}}$ | Occ. (<1) |
|-----|--------------|--------------|--------------|----------------------------------|-----------|
| Co | 0.03217 (3) | 0.25958 (2) | 0.57912 (2) | 0.01928 (12) | |
| Cl1 | -0.08339 (7) | 0.13989 (5) | 0.64106 (5) | 0.02986 (17) | |
| Cl2 | 0.14748 (6) | 0.38188 (5) | 0.51694 (4) | 0.02299 (15) | |
| N1 | -0.0314 (2) | 0.29843 (17) | 0.46384 (16) | 0.0230 (4) | |
| N2 | 0.2055 (2) | 0.15227 (16) | 0.52957 (15) | 0.0219 (4) | |
| H2N | 0.265 (2) | 0.186 (2) | 0.518 (2) | 0.026* | |
| N3 | 0.0985 (2) | 0.22239 (16) | 0.69411 (16) | 0.0227 (4) | |
| N4 | -0.1336 (2) | 0.37339 (17) | 0.62547 (15) | 0.0217 (4) | |
| H4N | -0.105 (3) | 0.4285 (16) | 0.5924 (19) | 0.026* | |
| C1 | 0.0414 (3) | 0.2780 (2) | 0.3799 (2) | 0.0272 (6) | |
| C2 | 0.1919 (3) | 0.2166 (2) | 0.36292 (19) | 0.0263 (5) | |
| H2A | 0.2430 | 0.2636 | 0.3621 | 0.032* | |
| H2B | 0.2250 | 0.1975 | 0.2976 | 0.032* | |

| | | | | |
|------|-------------|-------------|--------------|-------------|
| C3 | 0.2289 (3) | 0.1159 (2) | 0.43667 (19) | 0.0235 (5) |
| C4 | 0.2349 (3) | 0.0693 (2) | 0.61436 (19) | 0.0268 (6) |
| H4A | 0.3298 | 0.0215 | 0.5966 | 0.032* |
| H4B | 0.1696 | 0.0271 | 0.6367 | 0.032* |
| C5 | 0.2186 (3) | 0.1249 (2) | 0.69361 (19) | 0.0269 (6) |
| H5A | 0.2038 | 0.0775 | 0.7575 | 0.032* |
| H5B | 0.3034 | 0.1438 | 0.6821 | 0.032* |
| C6 | 0.0646 (3) | 0.2809 (2) | 0.75664 (19) | 0.0248 (5) |
| C7 | -0.0518 (3) | 0.3823 (2) | 0.75592 (19) | 0.0258 (5) |
| H7A | -0.0141 | 0.4381 | 0.7091 | 0.031* |
| H7B | -0.0823 | 0.4014 | 0.8211 | 0.031* |
| C8 | -0.1801 (3) | 0.3860 (2) | 0.73077 (19) | 0.0264 (6) |
| C9 | -0.2421 (3) | 0.3776 (2) | 0.5824 (2) | 0.0279 (6) |
| H9A | -0.3176 | 0.4446 | 0.5888 | 0.033* |
| H9B | -0.2811 | 0.3183 | 0.6163 | 0.033* |
| C10 | -0.1742 (3) | 0.3700 (2) | 0.4760 (2) | 0.0261 (5) |
| H10A | -0.2277 | 0.3430 | 0.4518 | 0.031* |
| H10B | -0.1729 | 0.4410 | 0.4375 | 0.031* |
| C11 | -0.0180 (3) | 0.3201 (3) | 0.2918 (2) | 0.0397 (7) |
| H11A | -0.1191 | 0.3332 | 0.3150 | 0.048* |
| H11B | -0.0006 | 0.3886 | 0.2589 | 0.048* |
| C12 | 0.0400 (3) | 0.2479 (2) | 0.2186 (2) | 0.0331 (6) |
| C13 | 0.1397 (4) | 0.2649 (3) | 0.1340 (2) | 0.0432 (8) |
| H13 | 0.1690 | 0.3257 | 0.1200 | 0.052* |
| C14 | 0.1982 (4) | 0.1960 (3) | 0.0691 (2) | 0.0521 (9) |
| H14 | 0.2669 | 0.2094 | 0.0112 | 0.063* |
| C15 | 0.1565 (5) | 0.1085 (3) | 0.0888 (3) | 0.0630 (12) |
| H15 | 0.1967 | 0.0604 | 0.0447 | 0.076* |
| C16 | 0.0565 (6) | 0.0897 (3) | 0.1725 (3) | 0.0714 (13) |
| H16 | 0.0278 | 0.0287 | 0.1860 | 0.086* |
| C17 | -0.0029 (4) | 0.1600 (3) | 0.2374 (2) | 0.0534 (9) |
| H17 | -0.0729 | 0.1473 | 0.2947 | 0.064* |
| C18 | 0.1422 (3) | 0.0454 (2) | 0.4501 (2) | 0.0305 (6) |
| H18A | 0.1701 | -0.0196 | 0.4958 | 0.046* |
| H18B | 0.1556 | 0.0280 | 0.3873 | 0.046* |
| H18C | 0.0447 | 0.0824 | 0.4758 | 0.046* |
| C19 | 0.3844 (3) | 0.0560 (2) | 0.4032 (2) | 0.0274 (6) |
| H19A | 0.4384 | 0.1015 | 0.4010 | 0.033* |
| H19B | 0.4063 | -0.0079 | 0.4528 | 0.033* |
| C20 | 0.4313 (3) | 0.0230 (2) | 0.3052 (2) | 0.0281 (6) |
| C21 | 0.4946 (3) | 0.0826 (3) | 0.2235 (2) | 0.0398 (7) |
| H21 | 0.5067 | 0.1452 | 0.2289 | 0.048* |
| C22 | 0.5411 (4) | 0.0521 (3) | 0.1329 (3) | 0.0528 (9) |
| H22 | 0.5844 | 0.0938 | 0.0771 | 0.063* |
| C23 | 0.5241 (4) | -0.0387 (3) | 0.1248 (3) | 0.0519 (10) |
| H23 | 0.5557 | -0.0598 | 0.0633 | 0.062* |
| C24 | 0.4615 (3) | -0.0988 (3) | 0.2055 (3) | 0.0451 (8) |
| H24 | 0.4493 | -0.1611 | 0.1995 | 0.054* |

| | | | | |
|------|--------------|--------------|--------------|-----------------------|
| C25 | 0.4161 (3) | -0.0692 (2) | 0.2956 (2) | 0.0349 (6) |
| H25 | 0.3743 | -0.1119 | 0.3512 | 0.042* |
| C26 | 0.1434 (3) | 0.2615 (2) | 0.8307 (2) | 0.0298 (6) |
| H26A | 0.0801 | 0.2578 | 0.8971 | 0.036* |
| H26B | 0.2173 | 0.1933 | 0.8275 | 0.036* |
| C27 | 0.2058 (3) | 0.3502 (2) | 0.8096 (2) | 0.0291 (6) |
| C28 | 0.1666 (3) | 0.4162 (2) | 0.8764 (2) | 0.0355 (7) |
| H28 | 0.0996 | 0.4058 | 0.9364 | 0.043* |
| C29 | 0.2248 (4) | 0.4971 (3) | 0.8560 (3) | 0.0450 (8) |
| H29 | 0.1979 | 0.5414 | 0.9022 | 0.054* |
| C30 | 0.3218 (3) | 0.5132 (3) | 0.7689 (3) | 0.0441 (8) |
| H30 | 0.3616 | 0.5683 | 0.7552 | 0.053* |
| C31 | 0.3605 (3) | 0.4491 (3) | 0.7019 (3) | 0.0388 (7) |
| H31 | 0.4265 | 0.4607 | 0.6417 | 0.047* |
| C32 | 0.3035 (3) | 0.3676 (2) | 0.7217 (2) | 0.0340 (6) |
| H32 | 0.3312 | 0.3235 | 0.6752 | 0.041* |
| C33 | -0.2828 (3) | 0.4958 (2) | 0.7397 (2) | 0.0336 (6) |
| H33A | -0.3678 | 0.4998 | 0.7268 | 0.050* |
| H33B | -0.2427 | 0.5491 | 0.6924 | 0.050* |
| H33C | -0.3038 | 0.5088 | 0.8055 | 0.050* |
| Cl3 | 0.54134 (7) | 0.26751 (5) | 0.43796 (5) | 0.03305 (18) |
| O1 | 0.4658 (2) | 0.20304 (16) | 0.51435 (16) | 0.0382 (5) |
| O2 | 0.5109 (3) | 0.36627 (18) | 0.4651 (2) | 0.0499 (6) |
| O3 | 0.6847 (2) | 0.2140 (2) | 0.4224 (2) | 0.0592 (7) |
| O4 | 0.4999 (3) | 0.2853 (2) | 0.35077 (18) | 0.0575 (7) |
| C34A | -0.2460 (18) | 0.2928 (8) | 0.8000 (11) | 0.0270 (17) 0.672 (3) |
| H34A | -0.3205 | 0.2923 | 0.7758 | 0.032* 0.672 (3) |
| H34B | -0.1740 | 0.2238 | 0.7954 | 0.032* 0.672 (3) |
| C35A | -0.3028 (3) | 0.3053 (3) | 0.90364 (18) | 0.0331 (9) 0.672 (3) |
| C36A | -0.4428 (3) | 0.3556 (3) | 0.9363 (2) | 0.0439 (11) 0.672 (3) |
| H36A | -0.4986 | 0.3835 | 0.8922 | 0.053* 0.672 (3) |
| C37A | -0.5011 (3) | 0.3649 (3) | 1.0337 (2) | 0.0550 (13) 0.672 (3) |
| H37A | -0.5968 | 0.3993 | 1.0560 | 0.066* 0.672 (3) |
| C38A | -0.4194 (4) | 0.3240 (3) | 1.09829 (16) | 0.0597 (16) 0.672 (3) |
| H38A | -0.4593 | 0.3304 | 1.1648 | 0.072* 0.672 (3) |
| C39A | -0.2795 (4) | 0.2738 (3) | 1.0656 (2) | 0.0600 (16) 0.672 (3) |
| H39A | -0.2236 | 0.2458 | 1.1098 | 0.072* 0.672 (3) |
| C40A | -0.2212 (3) | 0.2644 (3) | 0.9683 (2) | 0.0477 (13) 0.672 (3) |
| H40A | -0.1255 | 0.2300 | 0.9459 | 0.057* 0.672 (3) |
| C34B | -0.245 (4) | 0.314 (2) | 0.789 (3) | 0.0270 (17) 0.328 (3) |
| H34C | -0.3364 | 0.3354 | 0.7759 | 0.032* 0.328 (3) |
| H34D | -0.1923 | 0.2458 | 0.7662 | 0.032* 0.328 (3) |
| C35B | -0.2705 (8) | 0.2912 (6) | 0.9041 (4) | 0.0331 (9) 0.328 (3) |
| C36B | -0.3766 (7) | 0.3593 (5) | 0.9616 (5) | 0.0439 (11) 0.328 (3) |
| H36B | -0.4417 | 0.4163 | 0.9335 | 0.053* 0.328 (3) |
| C37B | -0.3876 (8) | 0.3441 (6) | 1.0601 (5) | 0.0550 (13) 0.328 (3) |
| H37B | -0.4601 | 0.3907 | 1.0994 | 0.066* 0.328 (3) |
| C38B | -0.2923 (10) | 0.2607 (7) | 1.1012 (4) | 0.0597 (16) 0.328 (3) |

| | | | | | |
|------|-------------|------------|------------|-------------|-----------|
| H38B | -0.2998 | 0.2503 | 1.1686 | 0.07 (4)* | 0.328 (3) |
| C39B | -0.1861 (8) | 0.1926 (6) | 1.0438 (5) | 0.0600 (16) | 0.328 (3) |
| H39B | -0.1210 | 0.1356 | 1.0719 | 0.072* | 0.328 (3) |
| C40B | -0.1752 (7) | 0.2078 (6) | 0.9452 (5) | 0.0477 (13) | 0.328 (3) |
| H40B | -0.1027 | 0.1613 | 0.9060 | 0.057* | 0.328 (3) |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|--------------|-------------|--------------|---------------|---------------|---------------|
| Co | 0.01748 (19) | 0.0189 (2) | 0.02131 (19) | -0.00528 (14) | -0.00556 (14) | -0.00311 (13) |
| Cl1 | 0.0289 (3) | 0.0277 (3) | 0.0344 (4) | -0.0131 (3) | -0.0044 (3) | -0.0063 (3) |
| Cl2 | 0.0211 (3) | 0.0206 (3) | 0.0262 (3) | -0.0075 (2) | -0.0044 (2) | -0.0033 (2) |
| N1 | 0.0188 (10) | 0.0235 (11) | 0.0259 (11) | -0.0040 (8) | -0.0070 (9) | -0.0053 (9) |
| N2 | 0.0227 (11) | 0.0197 (10) | 0.0241 (11) | -0.0051 (8) | -0.0084 (9) | -0.0041 (8) |
| N3 | 0.0197 (10) | 0.0214 (11) | 0.0266 (11) | -0.0068 (8) | -0.0066 (9) | -0.0022 (8) |
| N4 | 0.0195 (10) | 0.0238 (11) | 0.0209 (10) | -0.0072 (8) | -0.0039 (8) | -0.0034 (8) |
| C1 | 0.0277 (14) | 0.0248 (13) | 0.0302 (14) | -0.0046 (10) | -0.0121 (11) | -0.0055 (11) |
| C2 | 0.0238 (13) | 0.0283 (14) | 0.0251 (13) | -0.0045 (10) | -0.0056 (11) | -0.0075 (11) |
| C3 | 0.0206 (12) | 0.0228 (13) | 0.0286 (13) | -0.0025 (10) | -0.0090 (11) | -0.0093 (10) |
| C4 | 0.0270 (13) | 0.0225 (13) | 0.0273 (13) | -0.0023 (10) | -0.0099 (11) | -0.0023 (10) |
| C5 | 0.0255 (13) | 0.0254 (13) | 0.0267 (13) | -0.0023 (10) | -0.0106 (11) | -0.0026 (10) |
| C6 | 0.0247 (13) | 0.0274 (13) | 0.0236 (12) | -0.0118 (10) | -0.0064 (10) | -0.0012 (10) |
| C7 | 0.0289 (14) | 0.0254 (13) | 0.0229 (12) | -0.0085 (11) | -0.0045 (11) | -0.0063 (10) |
| C8 | 0.0247 (13) | 0.0288 (14) | 0.0233 (13) | -0.0063 (11) | -0.0034 (11) | -0.0071 (10) |
| C9 | 0.0175 (12) | 0.0323 (14) | 0.0323 (14) | -0.0053 (10) | -0.0062 (11) | -0.0064 (11) |
| C10 | 0.0202 (13) | 0.0262 (13) | 0.0317 (14) | -0.0016 (10) | -0.0110 (11) | -0.0071 (11) |
| C11 | 0.0395 (17) | 0.0425 (17) | 0.0331 (16) | 0.0050 (13) | -0.0184 (14) | -0.0131 (13) |
| C12 | 0.0319 (15) | 0.0384 (16) | 0.0264 (14) | -0.0027 (12) | -0.0128 (12) | -0.0058 (12) |
| C13 | 0.0477 (19) | 0.0477 (19) | 0.0307 (16) | -0.0109 (15) | -0.0127 (14) | -0.0036 (14) |
| C14 | 0.054 (2) | 0.058 (2) | 0.0284 (16) | -0.0016 (17) | -0.0061 (15) | -0.0110 (15) |
| C15 | 0.093 (3) | 0.047 (2) | 0.039 (2) | 0.000 (2) | -0.023 (2) | -0.0148 (17) |
| C16 | 0.125 (4) | 0.056 (3) | 0.048 (2) | -0.045 (3) | -0.026 (3) | -0.0022 (19) |
| C17 | 0.070 (3) | 0.070 (3) | 0.0291 (16) | -0.036 (2) | -0.0096 (17) | -0.0064 (16) |
| C18 | 0.0267 (14) | 0.0287 (14) | 0.0398 (15) | -0.0083 (11) | -0.0094 (12) | -0.0109 (12) |
| C19 | 0.0214 (13) | 0.0280 (14) | 0.0320 (14) | -0.0033 (10) | -0.0078 (11) | -0.0087 (11) |
| C20 | 0.0196 (12) | 0.0307 (14) | 0.0306 (14) | 0.0020 (10) | -0.0076 (11) | -0.0118 (11) |
| C21 | 0.0331 (16) | 0.0392 (17) | 0.0391 (17) | -0.0060 (13) | -0.0028 (13) | -0.0097 (13) |
| C22 | 0.045 (2) | 0.057 (2) | 0.0339 (17) | -0.0023 (16) | 0.0015 (15) | -0.0065 (16) |
| C23 | 0.046 (2) | 0.064 (2) | 0.0354 (17) | 0.0091 (17) | -0.0118 (15) | -0.0256 (17) |
| C24 | 0.0412 (18) | 0.0457 (19) | 0.050 (2) | 0.0034 (15) | -0.0164 (16) | -0.0280 (16) |
| C25 | 0.0323 (15) | 0.0325 (15) | 0.0393 (16) | -0.0025 (12) | -0.0100 (13) | -0.0144 (13) |
| C26 | 0.0348 (15) | 0.0318 (14) | 0.0255 (13) | -0.0103 (12) | -0.0115 (12) | -0.0044 (11) |
| C27 | 0.0285 (14) | 0.0307 (14) | 0.0303 (14) | -0.0081 (11) | -0.0144 (12) | -0.0018 (11) |
| C28 | 0.0411 (17) | 0.0379 (16) | 0.0324 (15) | -0.0121 (13) | -0.0155 (13) | -0.0061 (12) |
| C29 | 0.057 (2) | 0.0374 (17) | 0.054 (2) | -0.0145 (15) | -0.0316 (18) | -0.0073 (15) |
| C30 | 0.0443 (18) | 0.0365 (17) | 0.063 (2) | -0.0184 (14) | -0.0333 (17) | 0.0064 (15) |
| C31 | 0.0281 (15) | 0.0401 (17) | 0.0474 (18) | -0.0133 (13) | -0.0167 (14) | 0.0059 (14) |
| C32 | 0.0292 (15) | 0.0380 (16) | 0.0352 (15) | -0.0082 (12) | -0.0139 (12) | -0.0027 (12) |

| | | | | | | |
|------|-------------|-------------|-------------|--------------|--------------|--------------|
| C33 | 0.0285 (14) | 0.0323 (15) | 0.0311 (14) | -0.0018 (11) | -0.0012 (12) | -0.0103 (12) |
| Cl3 | 0.0299 (4) | 0.0291 (4) | 0.0425 (4) | -0.0119 (3) | -0.0124 (3) | -0.0022 (3) |
| O1 | 0.0405 (12) | 0.0344 (11) | 0.0437 (12) | -0.0187 (9) | -0.0179 (10) | 0.0050 (9) |
| O2 | 0.0455 (13) | 0.0350 (12) | 0.0713 (17) | -0.0174 (10) | -0.0075 (12) | -0.0154 (11) |
| O3 | 0.0303 (12) | 0.0483 (15) | 0.093 (2) | -0.0083 (11) | -0.0085 (13) | -0.0182 (14) |
| O4 | 0.0720 (18) | 0.0681 (17) | 0.0413 (14) | -0.0345 (14) | -0.0223 (13) | 0.0043 (12) |
| C34A | 0.0324 (16) | 0.021 (5) | 0.023 (4) | -0.018 (4) | 0.001 (2) | 0.005 (4) |
| C35A | 0.038 (3) | 0.0297 (19) | 0.0295 (15) | -0.0150 (18) | -0.0039 (16) | -0.0024 (13) |
| C36A | 0.041 (3) | 0.046 (2) | 0.033 (2) | -0.011 (2) | 0.0013 (19) | -0.0047 (19) |
| C37A | 0.061 (3) | 0.051 (3) | 0.035 (2) | -0.015 (3) | 0.011 (2) | -0.012 (2) |
| C38A | 0.097 (5) | 0.060 (4) | 0.029 (2) | -0.051 (3) | 0.002 (3) | -0.005 (2) |
| C39A | 0.079 (4) | 0.082 (4) | 0.037 (3) | -0.054 (3) | -0.022 (3) | 0.011 (3) |
| C40A | 0.042 (3) | 0.053 (3) | 0.043 (3) | -0.023 (3) | -0.010 (2) | 0.010 (2) |
| C34B | 0.0324 (16) | 0.021 (5) | 0.023 (4) | -0.018 (4) | 0.001 (2) | 0.005 (4) |
| C35B | 0.038 (3) | 0.0297 (19) | 0.0295 (15) | -0.0150 (18) | -0.0039 (16) | -0.0024 (13) |
| C36B | 0.041 (3) | 0.046 (2) | 0.033 (2) | -0.011 (2) | 0.0013 (19) | -0.0047 (19) |
| C37B | 0.061 (3) | 0.051 (3) | 0.035 (2) | -0.015 (3) | 0.011 (2) | -0.012 (2) |
| C38B | 0.097 (5) | 0.060 (4) | 0.029 (2) | -0.051 (3) | 0.002 (3) | -0.005 (2) |
| C39B | 0.079 (4) | 0.082 (4) | 0.037 (3) | -0.054 (3) | -0.022 (3) | 0.011 (3) |
| C40B | 0.042 (3) | 0.053 (3) | 0.043 (3) | -0.023 (3) | -0.010 (2) | 0.010 (2) |

Geometric parameters (Å, °)

| | | | |
|--------|------------|----------|-----------|
| Co—N1 | 1.927 (2) | C20—C21 | 1.381 (4) |
| Co—N3 | 1.943 (2) | C20—C25 | 1.394 (4) |
| Co—N4 | 1.969 (2) | C21—C22 | 1.396 (5) |
| Co—N2 | 1.977 (2) | C21—H21 | 0.9500 |
| Co—Cl1 | 2.2395 (9) | C22—C23 | 1.377 (6) |
| Co—Cl2 | 2.2676 (8) | C22—H22 | 0.9500 |
| N1—C1 | 1.274 (4) | C23—C24 | 1.373 (5) |
| N1—C10 | 1.482 (3) | C23—H23 | 0.9500 |
| N2—C4 | 1.484 (3) | C24—C25 | 1.384 (4) |
| N2—C3 | 1.511 (3) | C24—H24 | 0.9500 |
| N2—H2N | 0.877 (10) | C25—H25 | 0.9500 |
| N3—C6 | 1.276 (3) | C26—C27 | 1.519 (4) |
| N3—C5 | 1.493 (3) | C26—H26A | 0.9900 |
| N4—C9 | 1.494 (3) | C26—H26B | 0.9900 |
| N4—C8 | 1.512 (3) | C27—C28 | 1.393 (4) |
| N4—H4N | 0.878 (10) | C27—C32 | 1.396 (4) |
| C1—C2 | 1.506 (4) | C28—C29 | 1.390 (5) |
| C1—C11 | 1.530 (4) | C28—H28 | 0.9500 |
| C2—C3 | 1.529 (4) | C29—C30 | 1.381 (5) |
| C2—H2A | 0.9900 | C29—H29 | 0.9500 |
| C2—H2B | 0.9900 | C30—C31 | 1.377 (5) |
| C3—C18 | 1.517 (4) | C30—H30 | 0.9500 |
| C3—C19 | 1.552 (3) | C31—C32 | 1.390 (4) |
| C4—C5 | 1.511 (4) | C31—H31 | 0.9500 |
| C4—H4A | 0.9900 | C32—H32 | 0.9500 |

| | | | |
|-----------|------------|---------------|------------|
| C4—H4B | 0.9900 | C33—H33A | 0.9800 |
| C5—H5A | 0.9900 | C33—H33B | 0.9800 |
| C5—H5B | 0.9900 | C33—H33C | 0.9800 |
| C6—C7 | 1.506 (4) | C13—O3 | 1.421 (2) |
| C6—C26 | 1.522 (4) | C13—O2 | 1.425 (2) |
| C7—C8 | 1.535 (4) | C13—O4 | 1.441 (3) |
| C7—H7A | 0.9900 | C13—O1 | 1.442 (2) |
| C7—H7B | 0.9900 | C34A—C35A | 1.486 (17) |
| C8—C33 | 1.531 (4) | C34A—H34A | 0.9900 |
| C8—C34B | 1.36 (3) | C34A—H34B | 0.9900 |
| C8—C34A | 1.629 (10) | C35A—C36A | 1.3900 |
| C9—C10 | 1.515 (4) | C35A—C40A | 1.3900 |
| C9—H9A | 0.9900 | C36A—C37A | 1.3900 |
| C9—H9B | 0.9900 | C36A—H36A | 0.9500 |
| C10—H10A | 0.9900 | C37A—C38A | 1.3900 |
| C10—H10B | 0.9900 | C37A—H37A | 0.9500 |
| C11—C12 | 1.505 (4) | C38A—C39A | 1.3900 |
| C11—H11A | 0.9900 | C38A—H38A | 0.9500 |
| C11—H11B | 0.9900 | C39A—C40A | 1.3900 |
| C12—C13 | 1.375 (4) | C39A—H39A | 0.9500 |
| C12—C17 | 1.379 (5) | C40A—H40A | 0.9500 |
| C13—C14 | 1.380 (5) | C34B—C35B | 1.60 (4) |
| C13—H13 | 0.9500 | C34B—H34C | 0.9900 |
| C14—C15 | 1.365 (6) | C34B—H34D | 0.9900 |
| C14—H14 | 0.9500 | C35B—C36B | 1.3900 |
| C15—C16 | 1.374 (6) | C35B—C40B | 1.3900 |
| C15—H15 | 0.9500 | C36B—C37B | 1.3900 |
| C16—C17 | 1.393 (6) | C36B—H36B | 0.9500 |
| C16—H16 | 0.9500 | C37B—C38B | 1.3900 |
| C17—H17 | 0.9500 | C37B—H37B | 0.9500 |
| C18—H18A | 0.9800 | C38B—C39B | 1.3900 |
| C18—H18B | 0.9800 | C38B—H38B | 0.9500 |
| C18—H18C | 0.9800 | C39B—C40B | 1.3900 |
| C19—C20 | 1.514 (4) | C39B—H39B | 0.9500 |
| C19—H19A | 0.9900 | C40B—H40B | 0.9500 |
| C19—H19B | 0.9900 | | |
| N1—Co—N3 | 178.83 (9) | C3—C18—H18A | 109.5 |
| N1—Co—N4 | 84.22 (9) | C3—C18—H18B | 109.5 |
| N3—Co—N4 | 95.83 (9) | H18A—C18—H18B | 109.5 |
| N1—Co—N2 | 96.07 (9) | C3—C18—H18C | 109.5 |
| N3—Co—N2 | 83.80 (9) | H18A—C18—H18C | 109.5 |
| N4—Co—N2 | 176.17 (9) | H18B—C18—H18C | 109.5 |
| N1—Co—C11 | 89.85 (7) | C20—C19—C3 | 115.3 (2) |
| N3—Co—C11 | 91.32 (7) | C20—C19—H19A | 108.4 |
| N4—Co—C11 | 91.81 (7) | C3—C19—H19A | 108.4 |
| N2—Co—C11 | 92.00 (7) | C20—C19—H19B | 108.4 |
| N1—Co—Cl2 | 90.11 (7) | C3—C19—H19B | 108.4 |

| | | | |
|------------|-------------|---------------|-----------|
| N3—Co—Cl2 | 88.72 (7) | H19A—C19—H19B | 107.5 |
| N4—Co—Cl2 | 87.68 (7) | C21—C20—C25 | 118.6 (3) |
| N2—Co—Cl2 | 88.50 (7) | C21—C20—C19 | 120.0 (3) |
| Cl1—Co—Cl2 | 179.50 (3) | C25—C20—C19 | 121.3 (3) |
| C1—N1—C10 | 119.7 (2) | C20—C21—C22 | 120.8 (3) |
| C1—N1—Co | 126.26 (18) | C20—C21—H21 | 119.6 |
| C10—N1—Co | 113.59 (17) | C22—C21—H21 | 119.6 |
| C4—N2—C3 | 117.1 (2) | C23—C22—C21 | 119.7 (3) |
| C4—N2—Co | 106.90 (16) | C23—C22—H22 | 120.2 |
| C3—N2—Co | 119.99 (15) | C21—C22—H22 | 120.2 |
| C4—N2—H2N | 103 (2) | C24—C23—C22 | 120.0 (3) |
| C3—N2—H2N | 107 (2) | C24—C23—H23 | 120.0 |
| Co—N2—H2N | 100 (2) | C22—C23—H23 | 120.0 |
| C6—N3—C5 | 120.0 (2) | C23—C24—C25 | 120.4 (3) |
| C6—N3—Co | 126.05 (18) | C23—C24—H24 | 119.8 |
| C5—N3—Co | 112.83 (16) | C25—C24—H24 | 119.8 |
| C9—N4—C8 | 116.6 (2) | C24—C25—C20 | 120.4 (3) |
| C9—N4—Co | 107.52 (16) | C24—C25—H25 | 119.8 |
| C8—N4—Co | 120.35 (16) | C20—C25—H25 | 119.8 |
| C9—N4—H4N | 105 (2) | C27—C26—C6 | 109.7 (2) |
| C8—N4—H4N | 105 (2) | C27—C26—H26A | 109.7 |
| Co—N4—H4N | 100 (2) | C6—C26—H26A | 109.7 |
| N1—C1—C2 | 120.8 (2) | C27—C26—H26B | 109.7 |
| N1—C1—C11 | 121.6 (2) | C6—C26—H26B | 109.7 |
| C2—C1—C11 | 117.5 (2) | H26A—C26—H26B | 108.2 |
| C1—C2—C3 | 116.3 (2) | C28—C27—C32 | 118.8 (3) |
| C1—C2—H2A | 108.2 | C28—C27—C26 | 121.1 (3) |
| C3—C2—H2A | 108.2 | C32—C27—C26 | 120.1 (3) |
| C1—C2—H2B | 108.2 | C29—C28—C27 | 120.4 (3) |
| C3—C2—H2B | 108.2 | C29—C28—H28 | 119.8 |
| H2A—C2—H2B | 107.4 | C27—C28—H28 | 119.8 |
| N2—C3—C18 | 112.3 (2) | C30—C29—C28 | 120.2 (3) |
| N2—C3—C2 | 105.6 (2) | C30—C29—H29 | 119.9 |
| C18—C3—C2 | 111.0 (2) | C28—C29—H29 | 119.9 |
| N2—C3—C19 | 107.4 (2) | C31—C30—C29 | 119.9 (3) |
| C18—C3—C19 | 110.3 (2) | C31—C30—H30 | 120.0 |
| C2—C3—C19 | 109.9 (2) | C29—C30—H30 | 120.0 |
| N2—C4—C5 | 106.9 (2) | C30—C31—C32 | 120.4 (3) |
| N2—C4—H4A | 110.3 | C30—C31—H31 | 119.8 |
| C5—C4—H4A | 110.3 | C32—C31—H31 | 119.8 |
| N2—C4—H4B | 110.3 | C31—C32—C27 | 120.3 (3) |
| C5—C4—H4B | 110.3 | C31—C32—H32 | 119.9 |
| H4A—C4—H4B | 108.6 | C27—C32—H32 | 119.9 |
| N3—C5—C4 | 109.4 (2) | C8—C33—H33A | 109.5 |
| N3—C5—H5A | 109.8 | C8—C33—H33B | 109.5 |
| C4—C5—H5A | 109.8 | H33A—C33—H33B | 109.5 |
| N3—C5—H5B | 109.8 | C8—C33—H33C | 109.5 |
| C4—C5—H5B | 109.8 | H33A—C33—H33C | 109.5 |

| | | | |
|---------------|------------|----------------|-------------|
| H5A—C5—H5B | 108.2 | H33B—C33—H33C | 109.5 |
| N3—C6—C7 | 121.2 (2) | O3—C13—O2 | 109.85 (16) |
| N3—C6—C26 | 124.3 (2) | O3—C13—O4 | 110.38 (19) |
| C7—C6—C26 | 114.3 (2) | O2—C13—O4 | 109.29 (17) |
| C6—C7—C8 | 118.6 (2) | O3—C13—O1 | 108.94 (15) |
| C6—C7—H7A | 107.7 | O2—C13—O1 | 109.77 (15) |
| C8—C7—H7A | 107.7 | O4—C13—O1 | 108.59 (14) |
| C6—C7—H7B | 107.7 | C35A—C34A—C8 | 113.0 (10) |
| C8—C7—H7B | 107.7 | C35A—C34A—H34A | 109.0 |
| H7A—C7—H7B | 107.1 | C8—C34A—H34A | 109.0 |
| N4—C8—C33 | 108.7 (2) | C35A—C34A—H34B | 109.0 |
| N4—C8—C7 | 106.2 (2) | C8—C34A—H34B | 109.0 |
| C33—C8—C7 | 107.4 (2) | H34A—C34A—H34B | 107.8 |
| N4—C8—C34B | 111.2 (18) | C36A—C35A—C40A | 120.0 |
| C33—C8—C34B | 107.7 (16) | C36A—C35A—C34A | 118.2 (7) |
| C7—C8—C34B | 115.3 (17) | C40A—C35A—C34A | 121.8 (7) |
| N4—C8—C34A | 110.5 (7) | C35A—C36A—C37A | 120.0 |
| C33—C8—C34A | 112.0 (6) | C35A—C36A—H36A | 120.0 |
| C7—C8—C34A | 111.8 (6) | C37A—C36A—H36A | 120.0 |
| N4—C9—C10 | 107.3 (2) | C38A—C37A—C36A | 120.0 |
| N4—C9—H9A | 110.3 | C38A—C37A—H37A | 120.0 |
| C10—C9—H9A | 110.3 | C36A—C37A—H37A | 120.0 |
| N4—C9—H9B | 110.3 | C37A—C38A—C39A | 120.0 |
| C10—C9—H9B | 110.3 | C37A—C38A—H38A | 120.0 |
| H9A—C9—H9B | 108.5 | C39A—C38A—H38A | 120.0 |
| N1—C10—C9 | 110.0 (2) | C38A—C39A—C40A | 120.0 |
| N1—C10—H10A | 109.7 | C38A—C39A—H39A | 120.0 |
| C9—C10—H10A | 109.7 | C40A—C39A—H39A | 120.0 |
| N1—C10—H10B | 109.7 | C39A—C40A—C35A | 120.0 |
| C9—C10—H10B | 109.7 | C39A—C40A—H40A | 120.0 |
| H10A—C10—H10B | 108.2 | C35A—C40A—H40A | 120.0 |
| C12—C11—C1 | 114.6 (2) | C35B—C34B—C8 | 121 (3) |
| C12—C11—H11A | 108.6 | C35B—C34B—H34C | 107.1 |
| C1—C11—H11A | 108.6 | C8—C34B—H34C | 107.1 |
| C12—C11—H11B | 108.6 | C35B—C34B—H34D | 107.1 |
| C1—C11—H11B | 108.6 | C8—C34B—H34D | 107.1 |
| H11A—C11—H11B | 107.6 | H34C—C34B—H34D | 106.8 |
| C13—C12—C17 | 118.5 (3) | C36B—C35B—C40B | 120.0 |
| C13—C12—C11 | 121.5 (3) | C36B—C35B—C34B | 121.1 (14) |
| C17—C12—C11 | 119.9 (3) | C40B—C35B—C34B | 118.5 (14) |
| C12—C13—C14 | 121.7 (4) | C37B—C36B—C35B | 120.0 |
| C12—C13—H13 | 119.1 | C37B—C36B—H36B | 120.0 |
| C14—C13—H13 | 119.1 | C35B—C36B—H36B | 120.0 |
| C15—C14—C13 | 119.4 (4) | C36B—C37B—C38B | 120.0 |
| C15—C14—H14 | 120.3 | C36B—C37B—H37B | 120.0 |
| C13—C14—H14 | 120.3 | C38B—C37B—H37B | 120.0 |
| C14—C15—C16 | 120.1 (4) | C39B—C38B—C37B | 120.0 |
| C14—C15—H15 | 119.9 | C39B—C38B—H38B | 120.0 |

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| C16—C15—H15 | 119.9 | C37B—C38B—H38B | 120.0 |
| C15—C16—C17 | 120.2 (4) | C38B—C39B—C40B | 120.0 |
| C15—C16—H16 | 119.9 | C38B—C39B—H39B | 120.0 |
| C17—C16—H16 | 119.9 | C40B—C39B—H39B | 120.0 |
| C12—C17—C16 | 120.0 (4) | C39B—C40B—C35B | 120.0 |
| C12—C17—H17 | 120.0 | C39B—C40B—H40B | 120.0 |
| C16—C17—H17 | 120.0 | C35B—C40B—H40B | 120.0 |
| | | | |
| N4—Co—N1—C1 | 159.7 (2) | C6—C7—C8—C34A | 54.4 (7) |
| N2—Co—N1—C1 | −16.4 (2) | C8—N4—C9—C10 | 176.6 (2) |
| C1—Co—N1—C1 | −108.4 (2) | Co—N4—C9—C10 | −44.8 (2) |
| C12—Co—N1—C1 | 72.1 (2) | C1—N1—C10—C9 | 177.0 (2) |
| N4—Co—N1—C10 | −12.12 (17) | Co—N1—C10—C9 | −10.5 (3) |
| N2—Co—N1—C10 | 171.72 (17) | N4—C9—C10—N1 | 35.8 (3) |
| C1—Co—N1—C10 | 79.72 (17) | N1—C1—C11—C12 | −146.9 (3) |
| C12—Co—N1—C10 | −99.78 (17) | C2—C1—C11—C12 | 36.6 (4) |
| N1—Co—N2—C4 | −147.55 (17) | C1—C11—C12—C13 | −100.9 (4) |
| N3—Co—N2—C4 | 33.62 (16) | C1—C11—C12—C17 | 76.4 (4) |
| C1—Co—N2—C4 | −57.49 (16) | C17—C12—C13—C14 | −0.8 (5) |
| C12—Co—N2—C4 | 122.50 (16) | C11—C12—C13—C14 | 176.6 (3) |
| N1—Co—N2—C3 | −11.10 (19) | C12—C13—C14—C15 | 0.0 (5) |
| N3—Co—N2—C3 | 170.07 (19) | C13—C14—C15—C16 | 0.3 (6) |
| C1—Co—N2—C3 | 78.97 (18) | C14—C15—C16—C17 | 0.1 (7) |
| C12—Co—N2—C3 | −101.05 (18) | C13—C12—C17—C16 | 1.2 (5) |
| N4—Co—N3—C6 | −20.2 (2) | C11—C12—C17—C16 | −176.2 (4) |
| N2—Co—N3—C6 | 156.0 (2) | C15—C16—C17—C12 | −0.9 (7) |
| C1—Co—N3—C6 | −112.1 (2) | N2—C3—C19—C20 | −173.9 (2) |
| C12—Co—N3—C6 | 67.4 (2) | C18—C3—C19—C20 | 63.4 (3) |
| N4—Co—N3—C5 | 172.08 (17) | C2—C3—C19—C20 | −59.5 (3) |
| N2—Co—N3—C5 | −11.75 (17) | C3—C19—C20—C21 | 99.1 (3) |
| C1—Co—N3—C5 | 80.12 (16) | C3—C19—C20—C25 | −83.0 (3) |
| C12—Co—N3—C5 | −100.38 (16) | C25—C20—C21—C22 | 0.7 (4) |
| N1—Co—N4—C9 | 32.08 (16) | C19—C20—C21—C22 | 178.6 (3) |
| N3—Co—N4—C9 | −149.10 (16) | C20—C21—C22—C23 | −0.2 (5) |
| C1—Co—N4—C9 | −57.59 (16) | C21—C22—C23—C24 | 0.0 (5) |
| C12—Co—N4—C9 | 122.43 (16) | C22—C23—C24—C25 | −0.5 (5) |
| N1—Co—N4—C8 | 168.8 (2) | C23—C24—C25—C20 | 1.0 (5) |
| N3—Co—N4—C8 | −12.35 (19) | C21—C20—C25—C24 | −1.1 (4) |
| C1—Co—N4—C8 | 79.16 (18) | C19—C20—C25—C24 | −179.0 (3) |
| C12—Co—N4—C8 | −100.83 (18) | N3—C6—C26—C27 | 114.5 (3) |
| C10—N1—C1—C2 | 172.4 (2) | C7—C6—C26—C27 | −60.0 (3) |
| Co—N1—C1—C2 | 1.0 (4) | C6—C26—C27—C28 | 117.1 (3) |
| C10—N1—C1—C11 | −4.0 (4) | C6—C26—C27—C32 | −62.3 (3) |
| Co—N1—C1—C11 | −175.4 (2) | C32—C27—C28—C29 | −0.6 (4) |
| N1—C1—C2—C3 | 47.3 (4) | C26—C27—C28—C29 | 179.9 (3) |
| C11—C1—C2—C3 | −136.2 (3) | C27—C28—C29—C30 | 0.4 (5) |
| C4—N2—C3—C18 | 59.2 (3) | C28—C29—C30—C31 | 0.2 (5) |
| Co—N2—C3—C18 | −73.0 (2) | C29—C30—C31—C32 | −0.6 (4) |

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| C4—N2—C3—C2 | −179.6 (2) | C30—C31—C32—C27 | 0.4 (4) |
| Co—N2—C3—C2 | 48.2 (2) | C28—C27—C32—C31 | 0.2 (4) |
| C4—N2—C3—C19 | −62.3 (3) | C26—C27—C32—C31 | 179.7 (2) |
| Co—N2—C3—C19 | 165.48 (17) | N4—C8—C34A—C35A | −175.9 (8) |
| C1—C2—C3—N2 | −70.3 (3) | C33—C8—C34A—C35A | −54.6 (11) |
| C1—C2—C3—C18 | 51.6 (3) | C7—C8—C34A—C35A | 66.0 (11) |
| C1—C2—C3—C19 | 174.0 (2) | C8—C34A—C35A—C36A | 95.3 (10) |
| C3—N2—C4—C5 | 174.1 (2) | C8—C34A—C35A—C40A | −87.4 (9) |
| Co—N2—C4—C5 | −48.0 (2) | C40A—C35A—C36A—C37A | 0.0 |
| C6—N3—C5—C4 | 178.9 (2) | C34A—C35A—C36A—C37A | 177.3 (6) |
| Co—N3—C5—C4 | −12.5 (3) | C35A—C36A—C37A—C38A | 0.0 |
| N2—C4—C5—N3 | 39.2 (3) | C36A—C37A—C38A—C39A | 0.0 |
| C5—N3—C6—C7 | 177.1 (2) | C37A—C38A—C39A—C40A | 0.0 |
| Co—N3—C6—C7 | 10.1 (4) | C38A—C39A—C40A—C35A | 0.0 |
| C5—N3—C6—C26 | 3.0 (4) | C36A—C35A—C40A—C39A | 0.0 |
| Co—N3—C6—C26 | −163.99 (19) | C34A—C35A—C40A—C39A | −177.2 (6) |
| N3—C6—C7—C8 | 37.7 (4) | N4—C8—C34B—C35B | 165 (2) |
| C26—C6—C7—C8 | −147.7 (2) | C33—C8—C34B—C35B | −76 (3) |
| C9—N4—C8—C33 | −62.9 (3) | C7—C8—C34B—C35B | 43 (3) |
| Co—N4—C8—C33 | 164.00 (18) | C8—C34B—C35B—C36B | 79 (3) |
| C9—N4—C8—C7 | −178.2 (2) | C8—C34B—C35B—C40B | −95 (3) |
| Co—N4—C8—C7 | 48.7 (3) | C40B—C35B—C36B—C37B | 0.0 |
| C9—N4—C8—C34B | 55.5 (17) | C34B—C35B—C36B—C37B | −173.2 (16) |
| Co—N4—C8—C34B | −77.6 (17) | C35B—C36B—C37B—C38B | 0.0 |
| C9—N4—C8—C34A | 60.3 (7) | C36B—C37B—C38B—C39B | 0.0 |
| Co—N4—C8—C34A | −72.8 (6) | C37B—C38B—C39B—C40B | 0.0 |
| C6—C7—C8—N4 | −66.1 (3) | C38B—C39B—C40B—C35B | 0.0 |
| C6—C7—C8—C33 | 177.7 (2) | C36B—C35B—C40B—C39B | 0.0 |
| C6—C7—C8—C34B | 57.6 (19) | C34B—C35B—C40B—C39B | 173.3 (15) |

Hydrogen-bond geometry (Å, °)

| D—H···A | D—H | H···A | D···A | D—H···A |
|------------------------------|----------|----------|-----------|---------|
| N2—H2n···O1 | 0.88 (3) | 2.24 (3) | 3.063 (4) | 155 (2) |
| N4—H4n···Cl2 ⁱ | 0.88 (3) | 2.64 (2) | 3.432 (2) | 150 (3) |
| C10—H10a···O3 ⁱⁱ | 0.99 | 2.50 | 3.437 (4) | 159 |
| C19—H19b···O1 ⁱⁱⁱ | 0.99 | 2.54 | 3.409 (4) | 147 |
| C38a—H38a···O4 ^{iv} | 0.95 | 2.57 | 3.480 (3) | 160 |

Symmetry codes: (i) $-x, -y+1, -z+1$; (ii) $x-1, y, z$; (iii) $-x+1, -y, -z+1$; (iv) $x-1, y, z+1$.