

**(3*S*,4*Z*)-3-Chloro-1-methyl-4-[(2*E*)-(3-methylbenzylidene)hydrazinylidene]-3,4-dihydro-1*H*-2,1-benzothiazine 2,2-dioxide**

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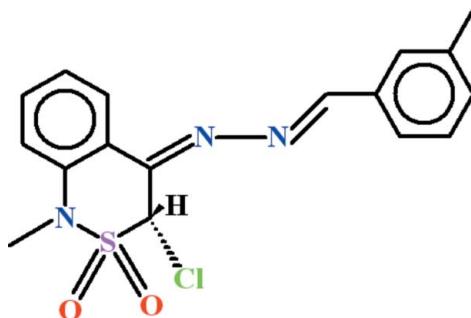
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Key indicators: single-crystal X-ray study;  $T = 296\text{ K}$ ; mean  $\sigma(\text{C}-\text{C}) = 0.004\text{ \AA}$ ;  $R$  factor = 0.036;  $wR$  factor = 0.088; data-to-parameter ratio = 19.5.

In the title compound,  $\text{C}_{17}\text{H}_{16}\text{ClN}_3\text{O}_2\text{S}$ , the dihedral angle between the benzene rings is  $7.75(13)^\circ$ . The thiazine ring adopts an envelope conformation with the S atom as the flap at a distance of  $0.813(2)\text{ \AA}$  from the plane through the other five atoms. In the crystal,  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds link the molecules into chains propagating in [100].

## Related literature

For related structures, see: Shafiq *et al.* (2011a,b,c). For further synthetic details, see: Shafiq *et al.* (2011d). For puckering parameters, see: Cremer & Pople (1975).



## Experimental

### Crystal data

$\text{C}_{17}\text{H}_{16}\text{ClN}_3\text{O}_2\text{S}$

$M_r = 361.84$

Orthorhombic,  $P2_12_12_1$

$a = 8.7734(2)\text{ \AA}$

$b = 11.1271(2)\text{ \AA}$

$c = 17.9423(3)\text{ \AA}$

$V = 1751.57(6)\text{ \AA}^3$   
 $Z = 4$   
Mo  $K\alpha$  radiation

$\mu = 0.35\text{ mm}^{-1}$   
 $T = 296\text{ K}$   
 $0.26 \times 0.18 \times 0.12\text{ mm}$

### Data collection

Bruker Kappa APEXII CCD diffractometer  
Absorption correction: multi-scan (*SADABS*; Bruker, 2005)  
 $T_{\min} = 0.930$ ,  $T_{\max} = 0.960$

17152 measured reflections  
4265 independent reflections  
3478 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.029$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.036$   
 $wR(F^2) = 0.088$   
 $S = 1.03$   
4265 reflections  
219 parameters  
H-atom parameters constrained

$\Delta\rho_{\max} = 0.24\text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.27\text{ e \AA}^{-3}$   
Absolute structure: Flack (1983),  
1788 Friedel pairs  
Flack parameter: 0.47 (6)

**Table 1**  
Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
C4—H4 $\cdots$ O2 <sup>i</sup>	0.93	2.57	3.469 (3)	163
C10—H10 $\cdots$ O1 <sup>ii</sup>	0.93	2.53	3.300 (3)	140

Symmetry codes: (i)  $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$ ; (ii)  $x + 1, y, z$ .

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2009); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PLATON*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6582).

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# supporting information

*Acta Cryst.* (2012). E68, o338 [doi:10.1107/S1600536811056315]

## (3*S*,4*Z*)-3-Chloro-1-methyl-4-[(2*E*)-(3-methylbenzylidene)hydrazinylidene]-3,4-dihydro-1*H*-2,1-benzothiazine 2,2-dioxide

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### S1. Comment

The title compound (**I**), (Fig. 1) has been synthesized in continuation of our studies of Schiff bases (Shafiq *et al.*, 2011*a,b,c*).

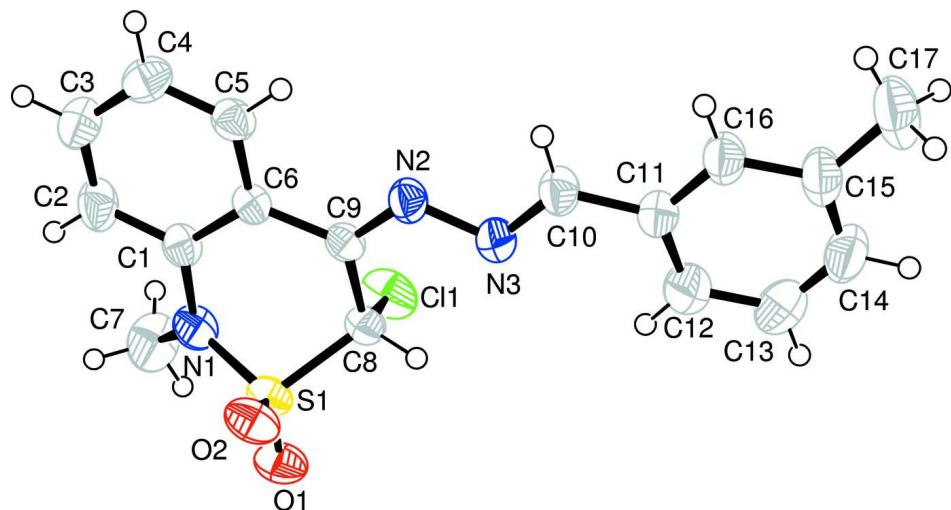
In (**I**), the benzene rings A (C1—C6) and B (C11—C16) are planar with r. m. s. deviation of 0.0033 and 0.0002 Å, respectively. The dihedral angle between A/B is 7.75 (13)°. The central group C (N2/N3/C10) is of course planar. The dihedral angle between A/C and B/C is 6.02 (19) and 5.11 (21)°, respectively. The thiazine ring D (C1/C6/C9/C8/S1/N1) is in the nvelope form, with the maximum puckering amplitude (Cremer & Pople, 1975), Q = 0.5707 (16) Å. The molecules form one-dimensional polymeric chains extending along the *a*-axis due to H-bonding of C—H···O type (Table 1).

### S2. Experimental

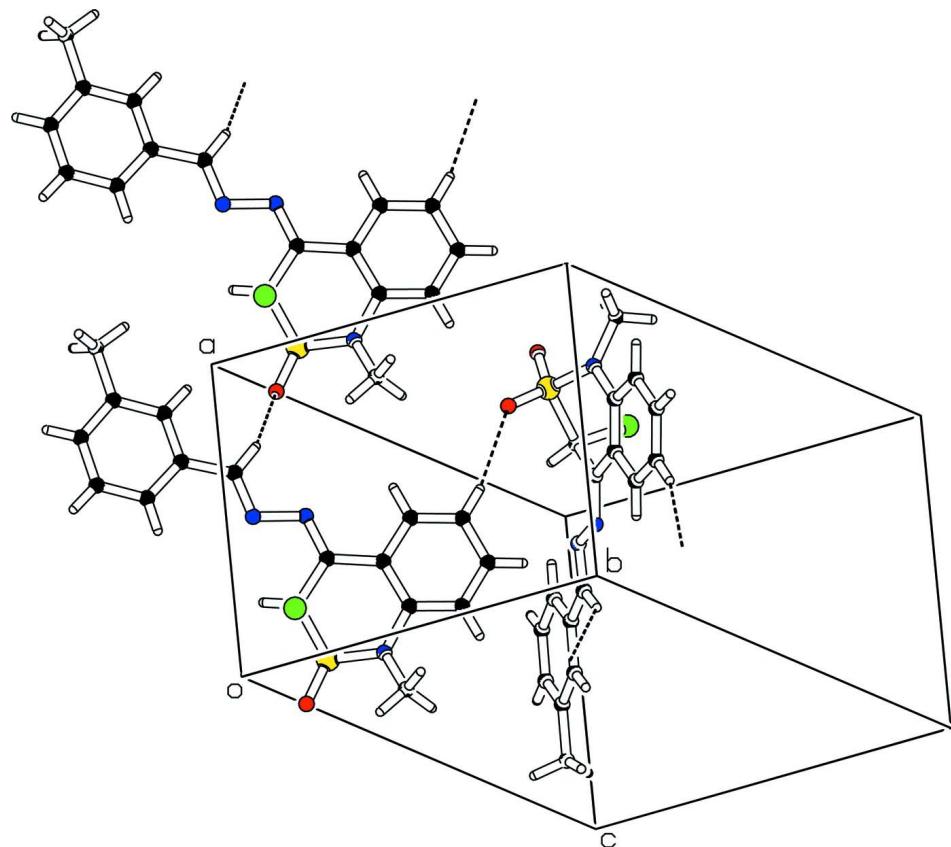
Schiff base derivative of (4*Z*)-4-hydrazinylidene-1-methyl-3,4-dihydro-1*H*-2,1-benzothiazine 2,2-dioxide and 3-methylbenzaldehyde was prepared using the method reported previously (Shafiq *et al.* 2011*d*). The chlorination of the schiff base was undertaken using *N*-chloro succinimide and dibenzoylperoxide (Shafiq *et al.*, 2011*a*). The crude product was re-crystallized from ethyl acetate to yield orange needles of (**I**).

### S3. Refinement

The H-atoms were positioned geometrically (C—H = 0.93–0.96 Å) and refined as riding with  $U_{\text{iso}}(\text{H}) = xU_{\text{eq}}(\text{C})$ , where x = 1.5 for methyl and x = 1.2 for aryl H-atoms.

**Figure 1**

View of the title compound with displacement ellipsoids drawn at the 50% probability level.

**Figure 2**

The partial packing showing chains extending along the [100] direction.

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*Crystal data*

C<sub>17</sub>H<sub>16</sub>ClN<sub>3</sub>O<sub>2</sub>S

M<sub>r</sub> = 361.84

Orthorhombic, P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>

Hall symbol: P 2ac 2ab

a = 8.7734 (2) Å

b = 11.1271 (2) Å

c = 17.9423 (3) Å

V = 1751.57 (6) Å<sup>3</sup>

Z = 4

F(000) = 752

D<sub>x</sub> = 1.372 Mg m<sup>-3</sup>

Mo Kα radiation, λ = 0.71073 Å

Cell parameters from 2106 reflections

θ = 1.4–25.3°

μ = 0.35 mm<sup>-1</sup>

T = 296 K

Needle, orange

0.26 × 0.18 × 0.12 mm

*Data collection*

Bruker Kappa APEXII CCD  
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Detector resolution: 7.50 pixels mm<sup>-1</sup>

ω scans

Absorption correction: multi-scan  
(SADABS; Bruker, 2005)

T<sub>min</sub> = 0.930, T<sub>max</sub> = 0.960

17152 measured reflections

4265 independent reflections

3478 reflections with I > 2σ(I)

R<sub>int</sub> = 0.029

θ<sub>max</sub> = 28.3°, θ<sub>min</sub> = 2.2°

h = -10→11

k = -14→14

l = -23→23

*Refinement*

Refinement on F<sup>2</sup>

Least-squares matrix: full

R[F<sup>2</sup> > 2σ(F<sup>2</sup>)] = 0.036

wR(F<sup>2</sup>) = 0.088

S = 1.03

4265 reflections

219 parameters

0 restraints

Primary atom site location: structure-invariant  
direct methods

Secondary atom site location: difference Fourier  
map

Hydrogen site location: inferred from  
neighbouring sites

H-atom parameters constrained

w = 1/[σ<sup>2</sup>(F<sub>o</sub><sup>2</sup>) + (0.041P)<sup>2</sup> + 0.2645P]  
where P = (F<sub>o</sub><sup>2</sup> + 2F<sub>c</sub><sup>2</sup>)/3

(Δ/σ)<sub>max</sub> < 0.001

Δρ<sub>max</sub> = 0.24 e Å<sup>-3</sup>

Δρ<sub>min</sub> = -0.27 e Å<sup>-3</sup>

Absolute structure: Flack (1983), **1788 Friedel  
pairs**

Absolute structure parameter: 0.47 (6)

*Special details*

**Geometry.** Bond distances, angles etc. have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

**Refinement.** Refinement of F<sup>2</sup> against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F<sup>2</sup>, conventional R-factors R are based on F, with F set to zero for negative F<sup>2</sup>. The threshold expression of F<sup>2</sup> > σ(F<sup>2</sup>) is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F<sup>2</sup> are statistically about twice as large as those based on F, and R-factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å<sup>2</sup>)*

	x	y	z	U <sub>iso</sub> * / U <sub>eq</sub>
C11	0.16886 (7)	0.16151 (6)	0.00310 (3)	0.0619 (2)
S1	0.10071 (6)	0.09643 (5)	0.15565 (3)	0.0411 (2)

O1	-0.04053 (18)	0.05118 (15)	0.12847 (9)	0.0559 (6)
O2	0.17040 (19)	0.04041 (14)	0.21814 (8)	0.0524 (5)
N1	0.0827 (2)	0.23974 (17)	0.16940 (11)	0.0453 (6)
N2	0.5175 (2)	0.13115 (15)	0.09335 (10)	0.0410 (5)
N3	0.5229 (2)	0.03374 (16)	0.04310 (10)	0.0467 (6)
C1	0.2182 (3)	0.30519 (17)	0.18673 (11)	0.0381 (6)
C2	0.2065 (3)	0.4082 (2)	0.22991 (13)	0.0523 (8)
C3	0.3334 (3)	0.4739 (2)	0.24869 (13)	0.0531 (8)
C4	0.4751 (3)	0.4373 (2)	0.22482 (12)	0.0499 (8)
C5	0.4885 (3)	0.3360 (2)	0.18144 (11)	0.0422 (7)
C6	0.3614 (2)	0.26764 (17)	0.16138 (10)	0.0338 (6)
C7	-0.0609 (3)	0.3020 (3)	0.15644 (19)	0.0690 (10)
C8	0.2429 (2)	0.09333 (19)	0.08520 (10)	0.0380 (6)
C9	0.3815 (2)	0.16115 (17)	0.11258 (10)	0.0339 (6)
C10	0.6581 (3)	0.01494 (19)	0.02078 (12)	0.0455 (7)
C11	0.6974 (3)	-0.07862 (19)	-0.03287 (12)	0.0464 (7)
C12	0.5872 (3)	-0.1500 (3)	-0.06694 (14)	0.0668 (10)
C13	0.6332 (4)	-0.2383 (3)	-0.11655 (18)	0.0808 (13)
C14	0.7839 (4)	-0.2554 (2)	-0.13190 (16)	0.0751 (12)
C15	0.8968 (3)	-0.1863 (2)	-0.09916 (14)	0.0588 (8)
C16	0.8497 (3)	-0.0975 (2)	-0.04930 (12)	0.0513 (8)
C17	1.0632 (4)	-0.2042 (3)	-0.11637 (18)	0.0803 (11)
H2	0.11120	0.43326	0.24642	0.0628*
H3	0.32356	0.54297	0.27748	0.0637*
H4	0.56141	0.48092	0.23799	0.0598*
H5	0.58447	0.31247	0.16504	0.0506*
H7A	-0.10558	0.32386	0.20337	0.1034*
H7B	-0.12931	0.25007	0.12985	0.1034*
H7C	-0.04256	0.37318	0.12755	0.1034*
H8	0.27104	0.00984	0.07453	0.0456*
H10	0.73577	0.06310	0.03951	0.0545*
H12	0.48435	-0.13858	-0.05656	0.0802*
H13	0.56053	-0.28641	-0.13963	0.0969*
H14	0.81172	-0.31540	-0.16535	0.0899*
H16	0.92277	-0.04956	-0.02640	0.0616*
H17A	1.09454	-0.28229	-0.09956	0.1203*
H17B	1.12213	-0.14373	-0.09134	0.1203*
H17C	1.07907	-0.19801	-0.16917	0.1203*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Cl1	0.0545 (4)	0.0921 (4)	0.0390 (3)	-0.0107 (3)	-0.0091 (3)	0.0114 (3)
S1	0.0304 (3)	0.0494 (3)	0.0434 (3)	-0.0036 (2)	0.0022 (2)	0.0065 (2)
O1	0.0360 (10)	0.0662 (10)	0.0655 (10)	-0.0139 (8)	-0.0005 (8)	0.0038 (8)
O2	0.0459 (10)	0.0616 (9)	0.0497 (8)	-0.0011 (8)	0.0023 (8)	0.0190 (7)
N1	0.0253 (10)	0.0544 (10)	0.0561 (11)	0.0053 (8)	0.0001 (8)	-0.0009 (8)
N2	0.0357 (10)	0.0437 (9)	0.0435 (9)	0.0038 (8)	0.0036 (8)	-0.0065 (7)

N3	0.0414 (12)	0.0459 (10)	0.0527 (10)	-0.0009 (9)	0.0088 (9)	-0.0098 (8)
C1	0.0322 (11)	0.0447 (11)	0.0374 (10)	0.0049 (9)	-0.0006 (8)	-0.0003 (8)
C2	0.0417 (13)	0.0592 (13)	0.0561 (13)	0.0133 (12)	0.0043 (11)	-0.0130 (12)
C3	0.0569 (17)	0.0487 (12)	0.0537 (12)	0.0045 (12)	-0.0028 (12)	-0.0115 (10)
C4	0.0488 (15)	0.0507 (12)	0.0501 (12)	-0.0065 (11)	-0.0107 (11)	-0.0038 (10)
C5	0.0331 (12)	0.0520 (12)	0.0414 (10)	-0.0031 (10)	0.0002 (9)	-0.0009 (9)
C6	0.0301 (11)	0.0399 (9)	0.0315 (9)	0.0022 (8)	-0.0007 (8)	0.0017 (8)
C7	0.0372 (15)	0.0779 (19)	0.092 (2)	0.0183 (12)	-0.0136 (14)	-0.0157 (17)
C8	0.0340 (11)	0.0412 (10)	0.0389 (10)	-0.0015 (9)	0.0017 (9)	0.0008 (9)
C9	0.0285 (11)	0.0386 (9)	0.0347 (9)	0.0022 (9)	0.0005 (8)	0.0035 (8)
C10	0.0416 (13)	0.0473 (12)	0.0475 (11)	0.0076 (10)	0.0017 (10)	-0.0071 (9)
C11	0.0489 (14)	0.0422 (11)	0.0481 (11)	0.0029 (11)	0.0066 (10)	-0.0052 (9)
C12	0.0661 (18)	0.0677 (16)	0.0667 (16)	-0.0120 (15)	0.0121 (14)	-0.0195 (14)
C13	0.093 (3)	0.0663 (17)	0.083 (2)	-0.0196 (17)	0.0097 (18)	-0.0307 (15)
C14	0.110 (3)	0.0500 (14)	0.0652 (16)	0.0004 (17)	0.0260 (17)	-0.0173 (12)
C15	0.0755 (18)	0.0493 (13)	0.0516 (12)	0.0193 (14)	0.0202 (14)	0.0017 (10)
C16	0.0553 (15)	0.0465 (12)	0.0521 (12)	0.0091 (12)	0.0076 (11)	-0.0046 (10)
C17	0.084 (2)	0.080 (2)	0.0770 (19)	0.0358 (17)	0.0277 (17)	0.0024 (16)

*Geometric parameters ( $\text{\AA}$ ,  $^\circ$ )*

C11—C8	1.7797 (19)	C12—C13	1.386 (4)
S1—O1	1.4237 (17)	C13—C14	1.364 (5)
S1—O2	1.4211 (16)	C14—C15	1.385 (4)
S1—N1	1.621 (2)	C15—C16	1.396 (3)
S1—C8	1.7763 (19)	C15—C17	1.505 (4)
N1—C1	1.428 (3)	C2—H2	0.9300
N1—C7	1.457 (3)	C3—H3	0.9300
N2—N3	1.411 (2)	C4—H4	0.9300
N2—C9	1.286 (2)	C5—H5	0.9300
N3—C10	1.269 (3)	C7—H7A	0.9600
C1—C2	1.387 (3)	C7—H7B	0.9600
C1—C6	1.400 (3)	C7—H7C	0.9600
C2—C3	1.374 (4)	C8—H8	0.9800
C3—C4	1.377 (4)	C10—H10	0.9300
C4—C5	1.375 (3)	C12—H12	0.9300
C5—C6	1.397 (3)	C13—H13	0.9300
C6—C9	1.484 (3)	C14—H14	0.9300
C8—C9	1.513 (3)	C16—H16	0.9300
C10—C11	1.459 (3)	C17—H17A	0.9600
C11—C12	1.393 (4)	C17—H17B	0.9600
C11—C16	1.384 (4)	C17—H17C	0.9600
O1—S1—O2	119.32 (10)	C16—C15—C17	120.8 (2)
O1—S1—N1	108.37 (10)	C11—C16—C15	122.0 (2)
O1—S1—C8	111.14 (9)	C1—C2—H2	119.00
O2—S1—N1	110.69 (10)	C3—C2—H2	119.00
O2—S1—C8	104.52 (9)	C2—C3—H3	120.00

N1—S1—C8	101.29 (10)	C4—C3—H3	120.00
S1—N1—C1	116.97 (14)	C3—C4—H4	120.00
S1—N1—C7	121.86 (17)	C5—C4—H4	120.00
C1—N1—C7	120.8 (2)	C4—C5—H5	119.00
N3—N2—C9	113.71 (16)	C6—C5—H5	119.00
N2—N3—C10	111.08 (17)	N1—C7—H7A	109.00
N1—C1—C2	118.8 (2)	N1—C7—H7B	110.00
N1—C1—C6	121.60 (17)	N1—C7—H7C	109.00
C2—C1—C6	119.6 (2)	H7A—C7—H7B	109.00
C1—C2—C3	121.1 (2)	H7A—C7—H7C	109.00
C2—C3—C4	119.9 (2)	H7B—C7—H7C	109.00
C3—C4—C5	119.7 (2)	C11—C8—H8	110.00
C4—C5—C6	121.6 (2)	S1—C8—H8	110.00
C1—C6—C5	118.04 (18)	C9—C8—H8	109.00
C1—C6—C9	122.46 (17)	N3—C10—H10	118.00
C5—C6—C9	119.47 (17)	C11—C10—H10	118.00
C11—C8—S1	108.93 (10)	C11—C12—H12	121.00
C11—C8—C9	110.45 (14)	C13—C12—H12	121.00
S1—C8—C9	108.89 (13)	C12—C13—H13	120.00
N2—C9—C6	118.41 (16)	C14—C13—H13	120.00
N2—C9—C8	121.94 (17)	C13—C14—H14	119.00
C6—C9—C8	119.63 (15)	C15—C14—H14	119.00
N3—C10—C11	123.1 (2)	C11—C16—H16	119.00
C10—C11—C12	122.2 (2)	C15—C16—H16	119.00
C10—C11—C16	118.5 (2)	C15—C17—H17A	109.00
C12—C11—C16	119.4 (2)	C15—C17—H17B	109.00
C11—C12—C13	119.0 (3)	C15—C17—H17C	109.00
C12—C13—C14	120.7 (3)	H17A—C17—H17B	109.00
C13—C14—C15	122.0 (3)	H17A—C17—H17C	109.00
C14—C15—C16	117.0 (2)	H17B—C17—H17C	109.00
C14—C15—C17	122.2 (2)		
O1—S1—N1—C1	-171.41 (15)	C1—C2—C3—C4	-0.3 (3)
O1—S1—N1—C7	1.7 (2)	C2—C3—C4—C5	0.9 (3)
O2—S1—N1—C1	55.99 (18)	C3—C4—C5—C6	-0.8 (3)
O2—S1—N1—C7	-130.9 (2)	C4—C5—C6—C1	0.1 (3)
C8—S1—N1—C1	-54.42 (17)	C4—C5—C6—C9	178.17 (19)
C8—S1—N1—C7	118.7 (2)	C1—C6—C9—N2	-178.30 (18)
O1—S1—C8—C11	49.97 (14)	C1—C6—C9—C8	3.4 (3)
O1—S1—C8—C9	170.49 (13)	C5—C6—C9—N2	3.7 (3)
O2—S1—C8—C11	179.94 (12)	C5—C6—C9—C8	-174.64 (17)
O2—S1—C8—C9	-59.54 (15)	C11—C8—C9—N2	-93.1 (2)
N1—S1—C8—C11	-64.97 (12)	C11—C8—C9—C6	85.20 (18)
N1—S1—C8—C9	55.54 (15)	S1—C8—C9—N2	147.38 (16)
S1—N1—C1—C2	-151.50 (17)	S1—C8—C9—C6	-34.4 (2)
S1—N1—C1—C6	28.4 (3)	N3—C10—C11—C12	3.8 (4)
C7—N1—C1—C2	35.3 (3)	N3—C10—C11—C16	-175.3 (2)
C7—N1—C1—C6	-144.8 (2)	C10—C11—C12—C13	-179.1 (2)

C9—N2—N3—C10	174.07 (18)	C16—C11—C12—C13	0.0 (4)
N3—N2—C9—C6	−176.04 (16)	C10—C11—C16—C15	179.1 (2)
N3—N2—C9—C8	2.2 (3)	C12—C11—C16—C15	0.0 (3)
N2—N3—C10—C11	−178.60 (19)	C11—C12—C13—C14	0.0 (4)
N1—C1—C2—C3	179.5 (2)	C12—C13—C14—C15	0.0 (5)
C6—C1—C2—C3	−0.4 (3)	C13—C14—C15—C16	0.0 (4)
N1—C1—C6—C5	−179.34 (18)	C13—C14—C15—C17	−179.5 (3)
N1—C1—C6—C9	2.6 (3)	C14—C15—C16—C11	0.0 (3)
C2—C1—C6—C5	0.5 (3)	C17—C15—C16—C11	179.6 (2)
C2—C1—C6—C9	−177.55 (19)		

*Hydrogen-bond geometry (Å, °)*

D—H···A	D—H	H···A	D···A	D—H···A
C4—H4···O2 <sup>i</sup>	0.93	2.57	3.469 (3)	163
C10—H10···O1 <sup>ii</sup>	0.93	2.53	3.300 (3)	140

Symmetry codes: (i)  $-x+1, y+1/2, -z+1/2$ ; (ii)  $x+1, y, z$ .