

tert-Butyl 2-(1*H*-imidazol-1-yl)acetate

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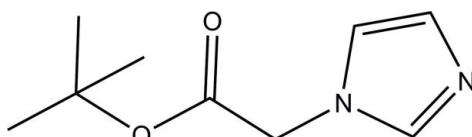
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Key indicators: single-crystal X-ray study; $T = 100\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$;
 R factor = 0.040; wR factor = 0.097; data-to-parameter ratio = 18.2.

In the title compound, $\text{C}_9\text{H}_{14}\text{N}_2\text{O}_2$, the imidazole ring and the acetate $\text{O}-\text{C}=\text{O}$ plane make a dihedral angle of $80.54(12)^\circ$. In the crystal, molecules are connected via pairs of $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds, forming centrosymmetric dimers.

Related literature

For related structures, see: Pak *et al.* (2003); Wang *et al.* (2010).

**Experimental***Crystal data*

$\text{C}_9\text{H}_{14}\text{N}_2\text{O}_2$	$V = 963.9(4)\text{ \AA}^3$
$M_r = 182.22$	$Z = 4$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
$a = 10.558(2)\text{ \AA}$	$\mu = 0.09\text{ mm}^{-1}$
$b = 9.287(2)\text{ \AA}$	$T = 100\text{ K}$
$c = 11.047(2)\text{ \AA}$	$0.33 \times 0.27 \times 0.05\text{ mm}$
$\beta = 117.157(4)^\circ$	

Data collection

Bruker APEXII CCD diffractometer
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
 $T_{\min} = 0.971$, $T_{\max} = 0.996$

6535 measured reflections
2208 independent reflections
1688 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.035$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.097$
 $S = 1.05$
2208 reflections

121 parameters
H-atom parameters constrained
 $\Delta\rho_{\max} = 0.23\text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.27\text{ e \AA}^{-3}$

Table 1
Hydrogen-bond geometry (\AA , $^\circ$).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
C4—H4B···O2 ⁱ	0.99	2.56	3.3768 (18)	140

Symmetry code: (i) $-x + 1, -y + 1, -z$.

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *SHELXL97* and *publCIF* (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: IS5051).

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supporting information

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tert-Butyl 2-(1*H*-imidazol-1-yl)acetate

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S1. Comment

The title compound was obtained through the condensation reaction of imidazole with *tert*-butyl chloroacetate. The imidazole ring and the plane passing through C5/O1/O2, make a dihedral angle of 80.54 (12) $^{\circ}$. This value is comparable to those calculated for some similar structures (Pak *et al.*, 2003; Wang *et al.*, 2010). In the crystal, each two molecules are connected *via* a pair of C—H \cdots O bonds around a center of inversion.

S2. Experimental

Sodium hydroxide (1.32 g, 0.033 mol) was added to a solution of imidazole (1.5 g, 0.022 mol) in DMF (20 ml), followed by addition of *tert*-butyl chloroacetate (3.15 ml, 0.022 mol). The mixture was refluxed for 1 h. The reaction mass was quenched with cold water (50 ml) and extracted by dichloromethane (3×25 ml). The combined organic layers were washed with cold water and brine and dried over anhydrous sodium sulfate. The solvent was evaporated under vacuum and the formed amorphous solid was stirred in *n*-hexane (30 ml) at room temperature. The solid was filtered, washed with hexane (2×20 ml), and recrystallized from ethyl acetate to afford off-white crystals of the title compound (melting point = 384–386 K).

S3. Refinement

H atoms were placed at calculated positions and refined in riding mode, with C—H distances of 0.95 (imidazole), 0.98 (methyl) and 0.99 (methylene) Å, and with $U_{\text{iso}}(\text{H})$ set to 1.2 (1.5 for methyl) $U_{\text{eq}}(\text{C})$.

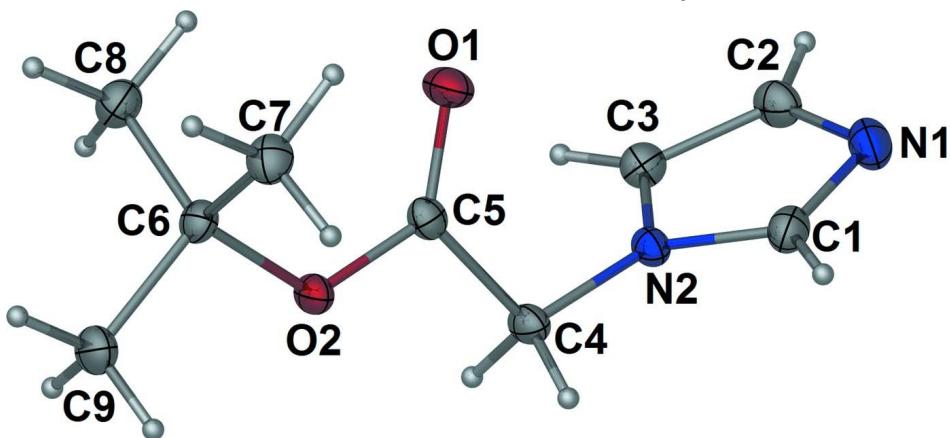


Figure 1

The molecular structure of the title compound showing displacement ellipsoids at the 50% probability level. Hydrogen atoms are drawn as spheres of arbitrary radius.

tert-Butyl 2-(1*H*-imidazol-1-yl)acetate*Crystal data*

$C_9H_{14}N_2O_2$
 $M_r = 182.22$
Monoclinic, $P2_1/n$
Hall symbol: -P 2yn
 $a = 10.558$ (2) Å
 $b = 9.287$ (2) Å
 $c = 11.047$ (2) Å
 $\beta = 117.157$ (4)°
 $V = 963.9$ (4) Å³
 $Z = 4$

$F(000) = 392$
 $D_x = 1.256$ Mg m⁻³
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 771 reflections
 $\theta = 3.0\text{--}29.0^\circ$
 $\mu = 0.09$ mm⁻¹
 $T = 100$ K
Plate, colorless
0.33 × 0.27 × 0.05 mm

Data collection

Bruker APEXII CCD
diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
 φ and ω scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)
 $T_{\min} = 0.971$, $T_{\max} = 0.996$

6535 measured reflections
2208 independent reflections
1688 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.035$
 $\theta_{\max} = 27.5^\circ$, $\theta_{\min} = 2.2^\circ$
 $h = -10 \rightarrow 13$
 $k = -12 \rightarrow 12$
 $l = -14 \rightarrow 11$

Refinement

Refinement on F^2
Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.097$
 $S = 1.05$
2208 reflections
121 parameters
0 restraints
Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map
Hydrogen site location: inferred from
neighbouring sites
H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0352P)^2 + 0.2973P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.23$ e Å⁻³
 $\Delta\rho_{\min} = -0.27$ e Å⁻³

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.58413 (11)	0.82030 (11)	0.21337 (11)	0.0244 (3)
O2	0.42983 (10)	0.63596 (10)	0.11631 (10)	0.0177 (2)
N1	1.01459 (13)	0.74927 (14)	0.30780 (13)	0.0250 (3)
N2	0.79333 (12)	0.67798 (12)	0.17766 (12)	0.0164 (3)

C1	0.91215 (15)	0.66189 (16)	0.29694 (15)	0.0210 (3)
H1	0.9207	0.5949	0.3653	0.025*
C2	0.95672 (16)	0.82623 (16)	0.18788 (16)	0.0231 (3)
H2	1.0055	0.8989	0.1650	0.028*
C3	0.82115 (15)	0.78441 (15)	0.10713 (15)	0.0197 (3)
H3	0.7584	0.8211	0.0198	0.024*
C4	0.65738 (15)	0.60761 (15)	0.13621 (15)	0.0188 (3)
H4A	0.6720	0.5166	0.1876	0.023*
H4B	0.6155	0.5835	0.0382	0.023*
C5	0.55466 (15)	0.70313 (15)	0.16122 (14)	0.0169 (3)
C6	0.30955 (14)	0.69930 (15)	0.13399 (14)	0.0178 (3)
C7	0.35240 (16)	0.71586 (17)	0.28396 (15)	0.0228 (3)
H7A	0.3916	0.6247	0.3308	0.034*
H7B	0.2688	0.7420	0.2957	0.034*
H7C	0.4245	0.7916	0.3225	0.034*
C8	0.26438 (16)	0.84140 (16)	0.05766 (16)	0.0230 (3)
H8A	0.3393	0.9133	0.1028	0.035*
H8B	0.1761	0.8747	0.0572	0.035*
H8C	0.2487	0.8278	-0.0362	0.035*
C9	0.19439 (15)	0.58562 (16)	0.06881 (15)	0.0230 (3)
H9A	0.1688	0.5775	-0.0281	0.034*
H9B	0.1102	0.6134	0.0790	0.034*
H9C	0.2299	0.4927	0.1135	0.034*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0231 (6)	0.0202 (5)	0.0320 (6)	-0.0045 (4)	0.0144 (5)	-0.0086 (4)
O2	0.0141 (5)	0.0182 (5)	0.0222 (5)	-0.0015 (4)	0.0096 (4)	-0.0035 (4)
N1	0.0185 (7)	0.0306 (7)	0.0238 (7)	-0.0012 (5)	0.0077 (6)	-0.0052 (6)
N2	0.0143 (6)	0.0174 (6)	0.0177 (6)	-0.0016 (5)	0.0075 (5)	-0.0027 (5)
C1	0.0195 (7)	0.0230 (7)	0.0192 (7)	0.0036 (6)	0.0076 (6)	0.0006 (6)
C2	0.0199 (7)	0.0233 (7)	0.0290 (8)	-0.0039 (6)	0.0138 (7)	-0.0026 (6)
C3	0.0213 (8)	0.0209 (7)	0.0191 (7)	0.0001 (6)	0.0111 (6)	0.0010 (6)
C4	0.0159 (7)	0.0177 (7)	0.0241 (8)	-0.0026 (5)	0.0102 (6)	-0.0037 (6)
C5	0.0163 (7)	0.0188 (7)	0.0148 (7)	-0.0015 (5)	0.0064 (6)	0.0003 (5)
C6	0.0147 (7)	0.0191 (7)	0.0210 (7)	0.0024 (5)	0.0094 (6)	-0.0002 (6)
C7	0.0218 (8)	0.0268 (8)	0.0216 (8)	0.0020 (6)	0.0116 (6)	-0.0015 (6)
C8	0.0216 (8)	0.0222 (7)	0.0259 (8)	0.0028 (6)	0.0114 (7)	0.0028 (6)
C9	0.0165 (8)	0.0241 (7)	0.0288 (8)	-0.0011 (6)	0.0107 (7)	-0.0011 (6)

Geometric parameters (\AA , $^\circ$)

O1—C5	1.2039 (17)	C4—H4B	0.9900
O2—C5	1.3325 (16)	C6—C7	1.513 (2)
O2—C6	1.4910 (16)	C6—C9	1.5207 (19)
N1—C1	1.3131 (19)	C6—C8	1.5209 (19)
N1—C2	1.379 (2)	C7—H7A	0.9800

N2—C1	1.3503 (18)	C7—H7B	0.9800
N2—C3	1.3709 (18)	C7—H7C	0.9800
N2—C4	1.4482 (17)	C8—H8A	0.9800
C1—H1	0.9500	C8—H8B	0.9800
C2—C3	1.353 (2)	C8—H8C	0.9800
C2—H2	0.9500	C9—H9A	0.9800
C3—H3	0.9500	C9—H9B	0.9800
C4—C5	1.5208 (19)	C9—H9C	0.9800
C4—H4A	0.9900		
C5—O2—C6	121.56 (10)	O2—C6—C9	102.05 (11)
C1—N1—C2	104.32 (12)	C7—C6—C9	111.34 (12)
C1—N2—C3	106.85 (12)	O2—C6—C8	109.88 (11)
C1—N2—C4	127.15 (12)	C7—C6—C8	112.35 (12)
C3—N2—C4	125.75 (12)	C9—C6—C8	111.06 (12)
N1—C1—N2	112.36 (13)	C6—C7—H7A	109.5
N1—C1—H1	123.8	C6—C7—H7B	109.5
N2—C1—H1	123.8	H7A—C7—H7B	109.5
C3—C2—N1	110.88 (13)	C6—C7—H7C	109.5
C3—C2—H2	124.6	H7A—C7—H7C	109.5
N1—C2—H2	124.6	H7B—C7—H7C	109.5
C2—C3—N2	105.60 (13)	C6—C8—H8A	109.5
C2—C3—H3	127.2	C6—C8—H8B	109.5
N2—C3—H3	127.2	H8A—C8—H8B	109.5
N2—C4—C5	111.43 (11)	C6—C8—H8C	109.5
N2—C4—H4A	109.3	H8A—C8—H8C	109.5
C5—C4—H4A	109.3	H8B—C8—H8C	109.5
N2—C4—H4B	109.3	C6—C9—H9A	109.5
C5—C4—H4B	109.3	C6—C9—H9B	109.5
H4A—C4—H4B	108.0	H9A—C9—H9B	109.5
O1—C5—O2	126.65 (13)	C6—C9—H9C	109.5
O1—C5—C4	124.35 (13)	H9A—C9—H9C	109.5
O2—C5—C4	108.99 (11)	H9B—C9—H9C	109.5
O2—C6—C7	109.68 (11)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
C4—H4B···O2 ⁱ	0.99	2.56	3.3768 (18)	140

Symmetry code: (i) $-x+1, -y+1, -z$.