

Acta Crystallographica Section E

## Structure Reports

Online

ISSN 1600-5368

## (2*E*,4*E*)-Ethyl 5-(phenylsulfonyl)penta-2,4-dienoate

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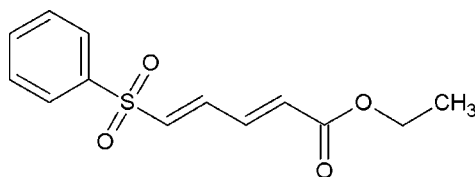
Received 23 February 2012; accepted 6 March 2012

 Key indicators: single-crystal X-ray study;  $T = 298$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.043;  $wR$  factor = 0.147; data-to-parameter ratio = 18.8.

In the title compound,  $\text{C}_{13}\text{H}_{14}\text{O}_4\text{S}$ , both  $\text{C}=\text{C}$  double bonds adopt an *E* conformation. In the crystal, molecules are linked into centrosymmetric  $R_2^2(14)$  dimers *via* pairs of  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds.

### Related literature

For the biological activity of phenyl sulfonyl-containing compounds see: De-Benedetti *et al.* (1985). For a related structure, see: Li (2011).



### Experimental

#### Crystal data

 $\text{C}_{13}\text{H}_{14}\text{O}_4\text{S}$   
 $M_r = 266.30$   
 Triclinic,  $P\bar{1}$ 
 $a = 6.2525$  (3) Å  
 $b = 7.8889$  (4) Å  
 $c = 14.5049$  (7) Å

 $\alpha = 82.828$  (3)°  
 $\beta = 87.261$  (2)°  
 $\gamma = 72.426$  (2)°  
 $V = 676.69$  (6) Å<sup>3</sup>  
 $Z = 2$ 

 Mo  $K\alpha$  radiation  
 $\mu = 0.24$  mm<sup>-1</sup>  
 $T = 298$  K  
 $0.45 \times 0.38 \times 0.15$  mm

#### Data collection

 Bruker APEXII KappaCCD  
 diffractometer  
 Absorption correction: multi-scan  
 (SADABS; Bruker, 2008)  
 $T_{\min} = 0.899$ ,  $T_{\max} = 0.965$ 

 8948 measured reflections  
 3084 independent reflections  
 2621 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.020$ 

#### Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.043$   
 $wR(F^2) = 0.147$   
 $S = 1.87$   
 3084 reflections

 164 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.29$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.30$  e Å<sup>-3</sup>

**Table 1**  
 Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C7}-\text{H7}\cdots\text{O3}^i$	0.93	2.32	3.212 (2)	161

 Symmetry code: (i)  $-x + 2, -y, -z + 1$ .

Data collection: APEX2 (Bruker, 2008); cell refinement: SAINT (Bruker, 2008); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: SHELXL97 and PLATON (Spek, 2009).

SA thanks the UGC, India, for financial support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5831).

### References

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## supporting information

*Acta Cryst.* (2012). E68, o1093 [https://doi.org/10.1107/S1600536812009907]

**(2E,4E)-Ethyl 5-(phenylsulfonyl)penta-2,4-dienoate**

**Ulaganathan Sankar, V. Sabari, G. Suresh, Ramakrishnan Uma and S. Aravindhan**

**S1. Comment**

Phenyl sulfonyl containing compounds show a wide range of biological properties (De-Benedetti *et al.*, 1985).

Fig. 1. shows a displacement ellipsoid plot of the title compound. Both C=C double bonds display an E configuration. The title molecule exhibits structural similarities with the already reported related structure (Li, 2011). The dihedral angle between two planes (C5—C6—S1—O1) and (C1—C6—S1—O2) is 37.32 (6)°. The crystal packing is stabilized by C—H···O intermolecular interactions. The molecules are linked into centrosymmetric  $R_2^2(14)$  dimers via C7—H7···O3 hydrogen bonds (Table 1). The packing of the compound is shown in (Fig. 2).

**S2. Experimental**

LIHMDS (8 ml, 8.4 mmol, 1.06 molar solution in THF) was added drop wise to a 0 °C cooled solution of bisphenyl sulfonyl methane (1 g, 3.4 mmol) in dried THF (15 ml) under argon atmosphere. The reaction mixture was stirred at same temp for 1 h, and then trans ethyl 4-bromo coronate (0.71 g, 3.7 mmol) in dry THF (5 ml) was added dropwise and the reaction mixture was allowed to come to RT and it was stirred under argon atmosphere for 16 h. The reaction mixture was quenched by adding saturated  $\text{NH}_4\text{Cl}$  (20 ml) and then extracted with ethyl acetate (2x20 ml) and washed with water (2x20 ml) and sat brine (20 ml). Then, the organic layer was dried over  $\text{MgSO}_4$ . Evaporation of the solvent under vacuum furnished the crude product, the residue was chromatographed (25% ethyl acetate in hexanes) to give analytically pure (2E, 4E)-ethyl 5-(phenyl sulfonyl)penta-2,4-dienote (0.673 g. yield 75%) as a colorless solid.

**S3. Refinement**

Hydrogen atoms were placed in calculated positions with C—H ranging from 0.93 Å to 0.97 Å and refined using a the riding model with fixed isotropic displacement parameters:  $U_{\text{iso}}(\text{H}) = 1.5 U_{\text{eq}}(\text{C})$  for the methyl group and  $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$  for other groups.

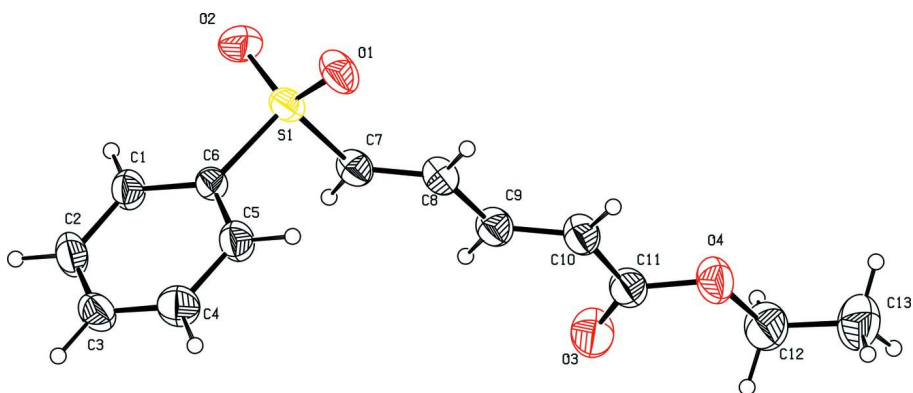


Figure 1

The molecular structure of the title compound with the atom numbering scheme. Displacement ellipsoids are shown at 30% probability level. The H atoms are presented as a small spheres of arbitrary radius. Related atoms have symmetry code: (i)  $-x + 2, -y, -z + 1$ .

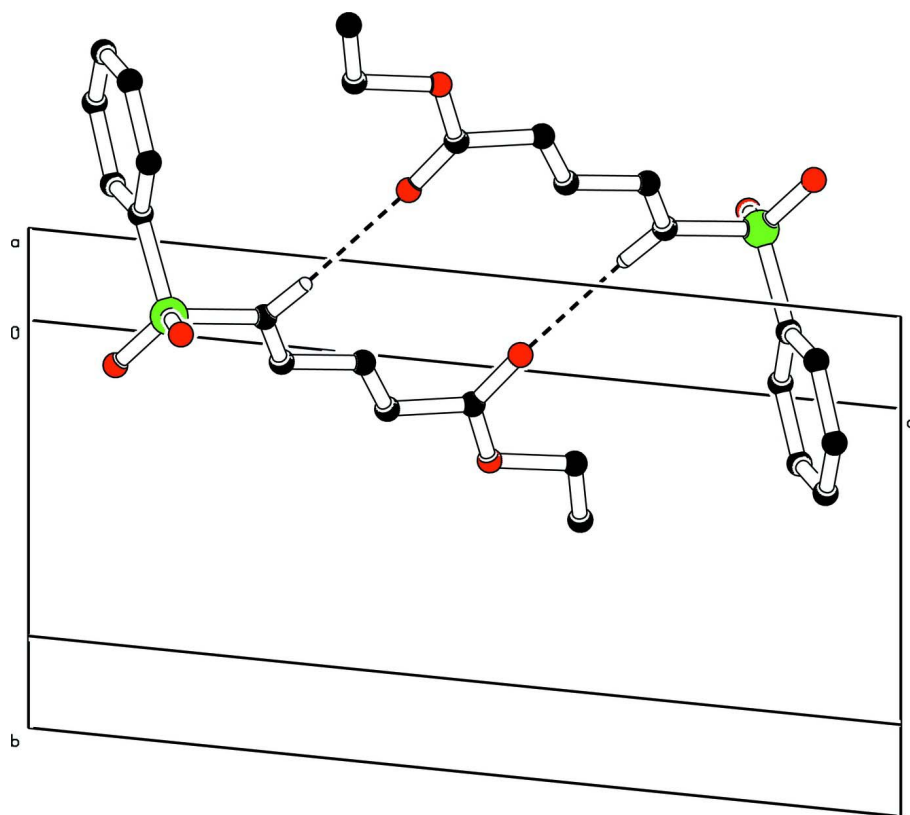


Figure 2

A view of the crystal packing. H atoms not involved in hydrogen bonding (dashed lines) have been omitted for clarity.

**(2*E*,4*E*)-Ethyl 5-(phenylsulfonyl)penta-2,4-dienoate**

*Crystal data*

$C_{13}H_{14}O_4S$   
 $M_r = 266.30$

Triclinic,  $P\bar{1}$   
 $a = 6.2525 (3) \text{ \AA}$

$b = 7.8889 (4) \text{ \AA}$   
 $c = 14.5049 (7) \text{ \AA}$   
 $\alpha = 82.828 (3)^\circ$   
 $\beta = 87.261 (2)^\circ$   
 $\gamma = 72.426 (2)^\circ$   
 $V = 676.69 (6) \text{ \AA}^3$   
 $Z = 2$   
 $F(000) = 280$

$D_x = 1.307 \text{ Mg m}^{-3}$   
 Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$   
 Cell parameters from 5946 reflections  
 $\theta = 2.7\text{--}28.3^\circ$   
 $\mu = 0.24 \text{ mm}^{-1}$   
 $T = 298 \text{ K}$   
 Triclinic, colourless  
 $0.45 \times 0.38 \times 0.15 \text{ mm}$

*Data collection*

Bruker APEXII KappaCCD  
 diffractometer  
 Radiation source: fine-focus sealed tube  
 Graphite monochromator  
 Detector resolution: 15.9948 pixels  $\text{mm}^{-1}$   
 $\omega$  and  $\varphi$  scans  
 Absorption correction: multi-scan  
 (SADABS; Bruker, 2008)  
 $T_{\min} = 0.899$ ,  $T_{\max} = 0.965$

8948 measured reflections  
 3084 independent reflections  
 2621 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.020$   
 $\theta_{\max} = 28.4^\circ$ ,  $\theta_{\min} = 2.7^\circ$   
 $h = -8 \rightarrow 8$   
 $k = -10 \rightarrow 9$   
 $l = -19 \rightarrow 19$

*Refinement*

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.043$   
 $wR(F^2) = 0.147$   
 $S = 1.87$   
 3084 reflections  
 164 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods  
 Secondary atom site location: difference Fourier  
 map

Hydrogen site location: inferred from  
 neighbouring sites  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0551P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.29 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\min} = -0.30 \text{ e \AA}^{-3}$   
 Extinction correction: SHELXL97 (Sheldrick,  
 2008)  
 Extinction coefficient: 0.0173 (18)

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	1.5671 (3)	-0.0639 (2)	0.13787 (12)	0.0577 (4)
H1	1.6549	-0.0001	0.1586	0.069*
C2	1.6638 (3)	-0.2359 (3)	0.11608 (12)	0.0670 (5)
H2	1.8179	-0.2881	0.1219	0.080*
C3	1.5361 (3)	-0.3300 (2)	0.08618 (12)	0.0634 (5)
H3	1.6035	-0.4465	0.0726	0.076*

C4	1.3103 (3)	-0.2555 (3)	0.07586 (14)	0.0714 (5)
H4	1.2246	-0.3206	0.0549	0.086*
C5	1.2088 (3)	-0.0820 (2)	0.09680 (12)	0.0612 (4)
H5	1.0550	-0.0299	0.0895	0.073*
C6	1.3387 (2)	0.0124 (2)	0.12854 (9)	0.0446 (3)
C7	1.0978 (3)	0.1812 (2)	0.27041 (10)	0.0534 (4)
H7	1.1952	0.1132	0.3169	0.064*
C8	0.8815 (3)	0.2378 (2)	0.29034 (11)	0.0545 (4)
H8	0.7812	0.3023	0.2441	0.065*
C9	0.7980 (3)	0.2008 (2)	0.38347 (11)	0.0585 (4)
H9	0.9013	0.1296	0.4271	0.070*
C10	0.5861 (3)	0.2604 (2)	0.41125 (11)	0.0603 (4)
H10	0.4768	0.3307	0.3697	0.072*
C11	0.5253 (3)	0.2140 (3)	0.50916 (12)	0.0616 (4)
C12	0.2457 (4)	0.2713 (3)	0.62613 (15)	0.0902 (7)
H12A	0.2830	0.1440	0.6466	0.108*
H12B	0.3229	0.3242	0.6655	0.108*
C13	0.0051 (4)	0.3538 (4)	0.63268 (17)	0.1022 (8)
H13A	-0.0313	0.4779	0.6084	0.153*
H13B	-0.0420	0.3436	0.6966	0.153*
H13C	-0.0704	0.2944	0.5974	0.153*
O1	1.0292 (2)	0.32081 (16)	0.09966 (8)	0.0719 (4)
O2	1.3810 (2)	0.30846 (17)	0.17469 (9)	0.0749 (4)
O3	0.6500 (2)	0.1095 (2)	0.56403 (9)	0.0889 (5)
O4	0.3154 (2)	0.3006 (2)	0.52876 (8)	0.0778 (4)
S1	1.20983 (6)	0.22766 (5)	0.16117 (3)	0.05254 (19)

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0426 (8)	0.0583 (10)	0.0721 (10)	-0.0123 (7)	0.0000 (7)	-0.0144 (8)
C2	0.0466 (9)	0.0661 (12)	0.0762 (11)	0.0015 (8)	0.0056 (8)	-0.0106 (9)
C3	0.0706 (11)	0.0474 (9)	0.0635 (9)	-0.0046 (8)	0.0109 (8)	-0.0111 (7)
C4	0.0729 (12)	0.0618 (12)	0.0873 (12)	-0.0260 (10)	0.0008 (10)	-0.0236 (10)
C5	0.0462 (8)	0.0578 (10)	0.0782 (11)	-0.0109 (8)	-0.0021 (8)	-0.0140 (8)
C6	0.0428 (7)	0.0419 (8)	0.0454 (7)	-0.0087 (6)	0.0009 (6)	-0.0020 (6)
C7	0.0576 (9)	0.0475 (9)	0.0494 (8)	-0.0075 (7)	-0.0029 (7)	-0.0041 (6)
C8	0.0565 (9)	0.0496 (9)	0.0516 (8)	-0.0064 (7)	-0.0025 (7)	-0.0074 (7)
C9	0.0590 (10)	0.0562 (10)	0.0548 (9)	-0.0099 (8)	-0.0007 (7)	-0.0047 (7)
C10	0.0572 (10)	0.0603 (11)	0.0552 (9)	-0.0060 (8)	-0.0003 (7)	-0.0046 (8)
C11	0.0595 (10)	0.0586 (11)	0.0615 (10)	-0.0110 (8)	0.0010 (8)	-0.0047 (8)
C12	0.0948 (15)	0.0925 (16)	0.0674 (11)	-0.0119 (13)	0.0204 (11)	0.0019 (11)
C13	0.0908 (16)	0.122 (2)	0.0886 (15)	-0.0233 (15)	0.0273 (13)	-0.0240 (15)
O1	0.0726 (8)	0.0597 (8)	0.0600 (7)	0.0108 (6)	-0.0020 (6)	0.0051 (5)
O2	0.0824 (9)	0.0565 (8)	0.0942 (9)	-0.0331 (7)	0.0119 (7)	-0.0141 (7)
O3	0.0740 (9)	0.1009 (12)	0.0691 (8)	-0.0048 (8)	-0.0020 (7)	0.0225 (7)
O4	0.0705 (8)	0.0792 (10)	0.0640 (7)	0.0013 (7)	0.0155 (6)	0.0001 (7)
S1	0.0557 (3)	0.0399 (3)	0.0553 (3)	-0.00635 (19)	0.00194 (18)	-0.00117 (17)

*Geometric parameters (Å, °)*

C1—C2	1.378 (2)	C8—H8	0.9300
C1—C6	1.378 (2)	C9—C10	1.326 (2)
C1—H1	0.9300	C9—H9	0.9300
C2—C3	1.359 (2)	C10—C11	1.483 (2)
C2—H2	0.9300	C10—H10	0.9300
C3—C4	1.363 (3)	C11—O3	1.195 (2)
C3—H3	0.9300	C11—O4	1.319 (2)
C4—C5	1.389 (3)	C12—C13	1.451 (3)
C4—H4	0.9300	C12—O4	1.469 (2)
C5—C6	1.383 (2)	C12—H12A	0.9700
C5—H5	0.9300	C12—H12B	0.9700
C6—S1	1.7595 (15)	C13—H13A	0.9600
C7—C8	1.320 (2)	C13—H13B	0.9600
C7—S1	1.7434 (15)	C13—H13C	0.9600
C7—H7	0.9300	O1—S1	1.4294 (12)
C8—C9	1.452 (2)	O2—S1	1.4328 (13)
C2—C1—C6	119.12 (15)	C8—C9—H9	117.4
C2—C1—H1	120.4	C9—C10—C11	119.36 (17)
C6—C1—H1	120.4	C9—C10—H10	120.3
C3—C2—C1	120.67 (15)	C11—C10—H10	120.3
C3—C2—H2	119.7	O3—C11—O4	123.62 (16)
C1—C2—H2	119.7	O3—C11—C10	124.36 (17)
C2—C3—C4	120.76 (17)	O4—C11—C10	112.01 (15)
C2—C3—H3	119.6	C13—C12—O4	108.24 (19)
C4—C3—H3	119.6	C13—C12—H12A	110.1
C3—C4—C5	119.78 (17)	O4—C12—H12A	110.1
C3—C4—H4	120.1	C13—C12—H12B	110.1
C5—C4—H4	120.1	O4—C12—H12B	110.1
C6—C5—C4	119.32 (16)	H12A—C12—H12B	108.4
C6—C5—H5	120.3	C12—C13—H13A	109.5
C4—C5—H5	120.3	C12—C13—H13B	109.5
C1—C6—C5	120.34 (15)	H13A—C13—H13B	109.5
C1—C6—S1	119.90 (12)	C12—C13—H13C	109.5
C5—C6—S1	119.71 (11)	H13A—C13—H13C	109.5
C8—C7—S1	123.19 (13)	H13B—C13—H13C	109.5
C8—C7—H7	118.4	C11—O4—C12	115.52 (15)
S1—C7—H7	118.4	O1—S1—O2	119.36 (8)
C7—C8—C9	120.95 (15)	O1—S1—C7	108.53 (8)
C7—C8—H8	119.5	O2—S1—C7	107.09 (8)
C9—C8—H8	119.5	O1—S1—C6	109.53 (7)
C10—C9—C8	125.22 (16)	O2—S1—C6	108.57 (7)
C10—C9—H9	117.4	C7—S1—C6	102.40 (7)
C6—C1—C2—C3	−0.3 (3)	O3—C11—O4—C12	4.5 (3)
C1—C2—C3—C4	0.9 (3)	C10—C11—O4—C12	−176.12 (16)

C2—C3—C4—C5	-0.5 (3)	C13—C12—O4—C11	-171.7 (2)
C3—C4—C5—C6	-0.4 (3)	C8—C7—S1—O1	-4.46 (18)
C2—C1—C6—C5	-0.7 (2)	C8—C7—S1—O2	125.64 (16)
C2—C1—C6—S1	176.98 (12)	C8—C7—S1—C6	-120.23 (16)
C4—C5—C6—C1	1.0 (2)	C1—C6—S1—O1	144.75 (14)
C4—C5—C6—S1	-176.61 (14)	C5—C6—S1—O1	-37.59 (14)
S1—C7—C8—C9	-177.71 (12)	C1—C6—S1—O2	12.83 (15)
C7—C8—C9—C10	176.02 (18)	C5—C6—S1—O2	-169.51 (12)
C8—C9—C10—C11	-179.12 (15)	C1—C6—S1—C7	-100.21 (14)
C9—C10—C11—O3	-8.3 (3)	C5—C6—S1—C7	77.46 (14)
C9—C10—C11—O4	172.26 (17)		

*Hydrogen-bond geometry (Å, °)*

<i>D—H...A</i>	<i>D—H</i>	<i>H...A</i>	<i>D...A</i>	<i>D—H...A</i>
C7—H7...O3 <sup>i</sup>	0.93	2.32	3.212 (2)	161

Symmetry code: (i)  $-x+2, -y, -z+1$ .