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7-Chloro-5-(chloromethyl)pyrazolo[1,5-*a*]pyrimidine-3-carbonitrile

Jingli Xu, Hang Liu, Guixia Li and Chuanmin Qi*

Key Laboratory of Radiopharmaceuticals, Ministry of Education, Department of Chemistry, Beijing Normal University, Xin Jie Kou Wai Street 19, 100875 Beijing, People's Republic of China

Correspondence e-mail: qicmin@sohu.com

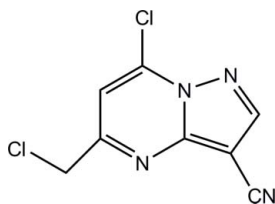
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 Key indicators: single-crystal X-ray study; $T = 301$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.038; wR factor = 0.104; data-to-parameter ratio = 16.6.

All non-H atoms of the title compound, $\text{C}_8\text{H}_4\text{Cl}_2\text{N}_4$, are essentially coplanar, with an r.m.s. deviation of 0.011 Å. In the crystal, weak $\text{C}-\text{H}\cdots\text{N}$ hydrogen bonds link the molecules into infinite sheets parallel to the bc plane.

Related literature

For details of the synthesis, see: Li *et al.* (2006). For applications of pyrazolo[1,5-*a*]pyrimidines as pharmacophores or building blocks in anti-tumor drug design, see: Li *et al.* (2006); Di Grandi *et al.* (2009); Powell *et al.* (2007); Gopalsamy *et al.* (2005).



Experimental

Crystal data

 $\text{C}_8\text{H}_4\text{Cl}_2\text{N}_4$
 $M_r = 227.05$

 Monoclinic, $P2_1/c$
 $a = 4.9817$ (4) Å

 $b = 18.4025$ (15) Å

 $c = 10.1526$ (9) Å

 $\beta = 95.924$ (1)°

 $V = 925.78$ (13) Å³
 $Z = 4$

 Mo $K\alpha$ radiation

 $\mu = 0.66$ mm⁻¹
 $T = 301$ K

 $0.60 \times 0.48 \times 0.20$ mm

Data collection

Bruker SMART APEX CCD area-detector diffractometer

Absorption correction: multi-scan (SADABS; Bruker, 2007)

 $T_{\min} = 0.693$, $T_{\max} = 0.879$

5429 measured reflections

2111 independent reflections

 1749 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.017$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.038$
 $wR(F^2) = 0.104$
 $S = 1.04$

2111 reflections

127 parameters

H-atom parameters constrained

 $\Delta\rho_{\max} = 0.46$ e Å⁻³
 $\Delta\rho_{\min} = -0.53$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C8}-\text{H8}\cdots\text{N2}^i$	0.93	2.50	3.337 (3)	150
$\text{C2}-\text{H2}\cdots\text{N2}^{ii}$	0.93	2.70	3.515 (3)	146

 Symmetry codes: (i) $-x + 2, -y + 1, -z$; (ii) $x - 1, -y + \frac{1}{2}, z + \frac{1}{2}$.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: IM2361).

References

- Bruker (1998). SMART. Bruker AXS Inc., Madison, Wisconsin, USA.
 Bruker (2005). SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
 Bruker (2007). SADABS. Bruker AXS Inc., Madison, Wisconsin, USA.
 Di Grandi, M. J., Berger, D. M., Hopper, D. W., Zhang, C., Dutia, M., Dunnick, A. L., Torres, N., Levin, J. I., Diamantidis, G., Zapf, C. W., Bloom, J. D., Dennis Powell, Y. B. H., Wojciechowicz, D., Collins, K. & Frommer, E. (2009). *Bioorg. Med. Chem. Lett.* **19**, 6957–6961.
 Gopalsamy, A., Yang, H., Ellingboe, J. W., Tsou, H.-R., Zhang, N., Honores, E., Powell, D., Miranda, M., McGinnis, J. P. & Rabindran, S. K. (2005). *Bioorg. Med. Chem. Lett.* **15**, 1591–1594.
 Li, J., Zhao, Y., Zhao, X., Yuan, X. & Gong, P. (2006). *Arch. Pharm. Chem. Life Sci.* **339**, 593–597.
 Powell, D., Gopalsamy, A., Wang, Y. D., Zhang, N., Miranda, M., McGinnis, J. P. & Rabindran, S. K. (2007). *Bioorg. Med. Chem. Lett.* **17**, 1641–1645.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.

supporting information

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7-Chloro-5-(chloromethyl)pyrazolo[1,5-*a*]pyrimidine-3-carbonitrile

Jingli Xu, Hang Liu, Guixia Li and Chuanmin Qi

S1. Comment

Pyrazolo[1,5-*a*]pyrimidines are widely applied as important pharmacophores or building blocks in anti-tumor drug design (Di Grandi *et al.*, 2009; Powell *et al.*, 2007; Gopalsamy *et al.*, 2005; Li *et al.*, 2006). Thus, the synthesis of the title compound may lead to the development of further pyrazolo[1,5-*a*]pyrimidine derivatives as new anti-tumor drugs. Here we report the crystal structure of the title compound.

The molecular structure of the title compound is shown in Fig. 1. The complete molecule is essentially planar, except the H atoms of the methylene group. Each molecule acts as a donor and a acceptor of weak intermolecular C—H \cdots N hydrogen-bond interactions linking the molecules into infinite sheets (Fig. 2).

S2. Experimental

The title compound can prepared by the reaction of 5-(chloromethyl)-7-hydroxypyrazolo[1,5-*a*]pyrimidine-3-carbonitrile with phosphorus oxychloride (Li *et al.*, 2006). Crystals suitable for X-ray diffraction were obtained by slow evaporation of a solution of the crude product in ethyl acetate at ambient temperature.

S3. Refinement

All H atoms attached to C atoms were fixed geometrically and treated as riding with C—H = 0.93 Å (CH) and C—H = 0.97 Å (CH₂) with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

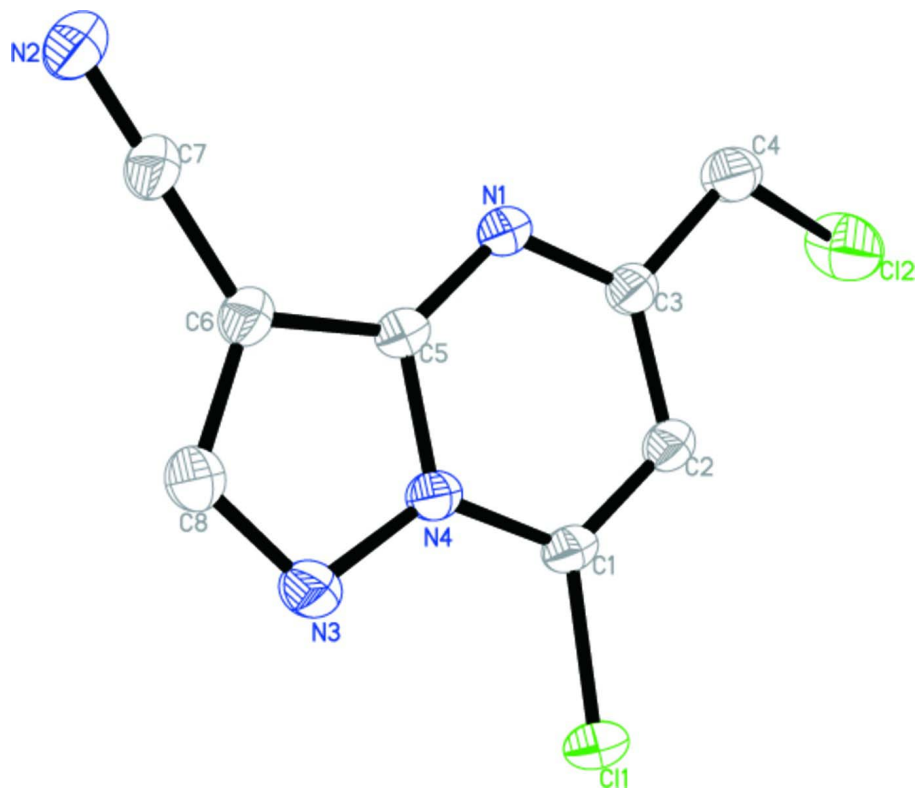


Figure 1

Molecular structure of the title compound showing displacement ellipsoids at the 45% probability level.

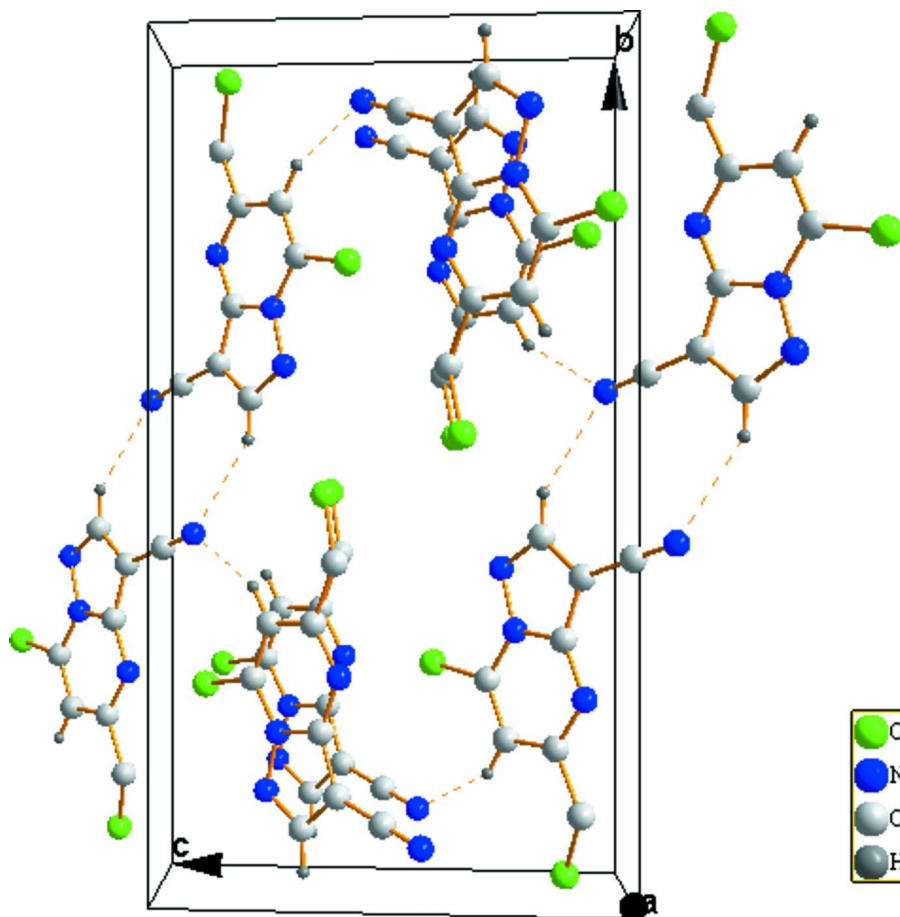


Figure 2

Packing diagram of the title compound, viewed along the a axis. Dashed lines indicate hydrogen bonds.

7-Chloro-5-(chloromethyl)pyrazolo[1,5-a]pyrimidine-3-carbonitrile

Crystal data

$C_8H_4Cl_2N_4$

$M_r = 227.05$

Monoclinic, $P2_1/c$

Hall symbol: $-P\ 2_1/c$

$a = 4.9817(4)\ \text{\AA}$

$b = 18.4025(15)\ \text{\AA}$

$c = 10.1526(9)\ \text{\AA}$

$\beta = 95.924(1)^\circ$

$V = 925.78(13)\ \text{\AA}^3$

$Z = 4$

$F(000) = 456$

$D_x = 1.629\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 2237 reflections

$\theta = 3.9\text{--}27.6^\circ$

$\mu = 0.66\ \text{mm}^{-1}$

$T = 301\ \text{K}$

Block, red

$0.60 \times 0.48 \times 0.20\ \text{mm}$

Data collection

Bruker SMART APEX CCD area-detector
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

ϕ and ω scans

Absorption correction: multi-scan
(*SADABS*; Bruker, 2007)

$T_{\min} = 0.693$, $T_{\max} = 0.879$

5429 measured reflections

2111 independent reflections

1749 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.017$
 $\theta_{\text{max}} = 27.6^\circ$, $\theta_{\text{min}} = 2.3^\circ$
 $h = -6 \rightarrow 6$

$k = -13 \rightarrow 23$
 $l = -12 \rightarrow 12$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.038$
 $wR(F^2) = 0.104$
 $S = 1.04$
 2111 reflections
 127 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.050P)^2 + 0.3703P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.46 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.53 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
C11	0.32935 (9)	0.26419 (3)	0.40816 (5)	0.04533 (17)
C12	0.73176 (14)	0.03358 (3)	0.13931 (8)	0.0784 (3)
N1	0.9223 (3)	0.23861 (8)	0.10690 (15)	0.0376 (3)
N2	1.2065 (4)	0.41634 (10)	-0.0585 (2)	0.0636 (5)
N3	0.5932 (4)	0.38077 (9)	0.26671 (18)	0.0503 (4)
N4	0.6551 (3)	0.31043 (8)	0.24035 (15)	0.0374 (3)
C1	0.5553 (4)	0.25052 (10)	0.29599 (18)	0.0363 (4)
C2	0.6391 (4)	0.18420 (10)	0.25839 (18)	0.0390 (4)
H2	0.5762	0.1420	0.2952	0.047*
C3	0.8249 (4)	0.18073 (10)	0.16175 (18)	0.0378 (4)
C4	0.9331 (5)	0.10975 (11)	0.1153 (2)	0.0554 (6)
H4A	1.1109	0.1018	0.1615	0.066*
H4B	0.9540	0.1138	0.0216	0.066*
C5	0.8380 (3)	0.30370 (10)	0.14617 (17)	0.0352 (4)
C6	0.8930 (4)	0.37484 (10)	0.11144 (19)	0.0423 (4)
C7	1.0677 (4)	0.39787 (10)	0.0175 (2)	0.0473 (5)
C8	0.7377 (5)	0.41835 (11)	0.1879 (2)	0.0520 (5)
H8	0.7362	0.4688	0.1837	0.062*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cl1	0.0409 (3)	0.0559 (3)	0.0418 (3)	-0.0012 (2)	0.01678 (19)	-0.0018 (2)
Cl2	0.0764 (4)	0.0456 (3)	0.1169 (6)	-0.0135 (3)	0.0270 (4)	-0.0184 (3)
N1	0.0376 (8)	0.0389 (8)	0.0378 (8)	0.0000 (6)	0.0108 (6)	0.0019 (6)
N2	0.0747 (13)	0.0502 (11)	0.0710 (13)	-0.0039 (10)	0.0321 (11)	0.0098 (10)
N3	0.0591 (11)	0.0374 (9)	0.0574 (10)	0.0019 (8)	0.0213 (8)	-0.0043 (8)
N4	0.0381 (8)	0.0379 (8)	0.0377 (8)	-0.0003 (6)	0.0110 (6)	-0.0009 (6)
C1	0.0323 (8)	0.0449 (10)	0.0328 (8)	-0.0023 (7)	0.0089 (7)	0.0012 (7)
C2	0.0403 (9)	0.0381 (10)	0.0398 (9)	-0.0032 (8)	0.0099 (8)	0.0035 (8)
C3	0.0403 (9)	0.0364 (9)	0.0376 (9)	-0.0009 (7)	0.0075 (7)	0.0007 (7)
C4	0.0654 (14)	0.0379 (11)	0.0676 (14)	0.0002 (10)	0.0292 (11)	-0.0003 (10)
C5	0.0333 (8)	0.0391 (9)	0.0341 (9)	-0.0023 (7)	0.0078 (7)	0.0007 (7)
C6	0.0458 (10)	0.0368 (10)	0.0458 (10)	-0.0045 (8)	0.0114 (8)	0.0033 (8)
C7	0.0549 (12)	0.0360 (10)	0.0527 (12)	-0.0045 (9)	0.0137 (10)	0.0047 (9)
C8	0.0621 (13)	0.0350 (10)	0.0612 (13)	-0.0022 (9)	0.0177 (11)	-0.0011 (9)

Geometric parameters (\AA , $^\circ$)

Cl1—C1	1.7006 (18)	C2—C3	1.418 (2)
Cl2—C4	1.755 (2)	C2—H2	0.9300
N1—C3	1.318 (2)	C3—C4	1.507 (3)
N1—C5	1.343 (2)	C4—H4A	0.9700
N2—C7	1.139 (3)	C4—H4B	0.9700
N3—C8	1.325 (3)	C5—C6	1.390 (3)
N3—N4	1.364 (2)	C6—C8	1.402 (3)
N4—C1	1.356 (2)	C6—C7	1.421 (3)
N4—C5	1.393 (2)	C8—H8	0.9300
C1—C2	1.357 (3)		
C3—N1—C5	117.09 (15)	C3—C4—H4A	108.6
C8—N3—N4	103.25 (16)	Cl2—C4—H4A	108.6
C1—N4—N3	126.15 (15)	C3—C4—H4B	108.6
C1—N4—C5	120.50 (15)	Cl2—C4—H4B	108.6
N3—N4—C5	113.35 (15)	H4A—C4—H4B	107.5
N4—C1—C2	118.49 (16)	N1—C5—C6	133.52 (16)
N4—C1—Cl1	117.07 (14)	N1—C5—N4	121.99 (15)
C2—C1—Cl1	124.44 (14)	C6—C5—N4	104.50 (15)
C1—C2—C3	118.48 (16)	C5—C6—C8	105.24 (17)
C1—C2—H2	120.8	C5—C6—C7	126.97 (18)
C3—C2—H2	120.8	C8—C6—C7	127.79 (19)
N1—C3—C2	123.45 (17)	N2—C7—C6	179.6 (3)
N1—C3—C4	114.14 (16)	N3—C8—C6	113.67 (19)
C2—C3—C4	122.40 (16)	N3—C8—H8	123.2
C3—C4—Cl2	114.86 (15)	C6—C8—H8	123.2
C8—N3—N4—C1	-179.43 (19)	C3—N1—C5—C6	-179.6 (2)

C8—N3—N4—C5	0.3 (2)	C3—N1—C5—N4	-0.2 (3)
N3—N4—C1—C2	-179.90 (18)	C1—N4—C5—N1	0.0 (3)
C5—N4—C1—C2	0.4 (3)	N3—N4—C5—N1	-179.71 (17)
N3—N4—C1—C11	0.6 (3)	C1—N4—C5—C6	179.58 (16)
C5—N4—C1—C11	-179.05 (13)	N3—N4—C5—C6	-0.1 (2)
N4—C1—C2—C3	-0.6 (3)	N1—C5—C6—C8	179.4 (2)
C11—C1—C2—C3	178.78 (14)	N4—C5—C6—C8	0.0 (2)
C5—N1—C3—C2	-0.1 (3)	N1—C5—C6—C7	0.0 (4)
C5—N1—C3—C4	-178.66 (18)	N4—C5—C6—C7	-179.5 (2)
C1—C2—C3—N1	0.5 (3)	N4—N3—C8—C6	-0.3 (3)
C1—C2—C3—C4	178.98 (19)	C5—C6—C8—N3	0.2 (3)
N1—C3—C4—C12	-159.03 (16)	C7—C6—C8—N3	179.7 (2)
C2—C3—C4—C12	22.4 (3)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C8—H8...N2 ⁱ	0.93	2.50	3.337 (3)	150
C2—H2...N2 ⁱⁱ	0.93	2.70	3.515 (3)	146

Symmetry codes: (i) $-x+2, -y+1, -z$; (ii) $x-1, -y+1/2, z+1/2$.