

Acta Crystallographica Section E

## Structure Reports

Online

ISSN 1600-5368

# 5-Anilino-3-benzylsulfanyl-6-(3-chloroanilino)-1-phenyl-1H-pyrazolo[3,4-d]pyrimidin-4(5H)-one

 Hong-Qing Wang,<sup>a\*</sup> Fang-Fang He,<sup>a</sup> Xiao-Feng Wang,<sup>a</sup> Wei-Ping Zhou<sup>b</sup> and Zhao-Jie Liu<sup>c</sup>

<sup>a</sup>School of Chemistry and Chemical Engineering, University of South China, Hengyang 421001, People's Republic of China, <sup>b</sup>College of Mathematics and Physical, University of South China, Hunan 421001, People's Republic of China, and <sup>c</sup>Institute of Organic Synthesis, Central China Normal University, Wuhan 430079, People's Republic of China

Correspondence e-mail: hqwang2009cn@yahoo.com.cn

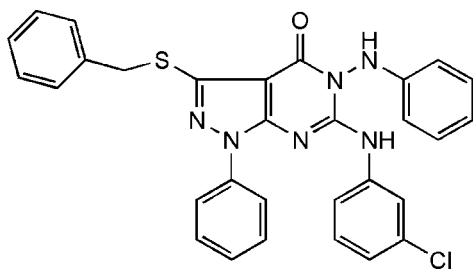
Received 5 May 2012; accepted 1 September 2012

 Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å;  $R$  factor = 0.046;  $wR$  factor = 0.114; data-to-parameter ratio = 15.1.

In the title compound,  $\text{C}_{30}\text{H}_{23}\text{ClN}_6\text{OS}$ , the benzyl, the 3-chloroanilino, the phenyl and the anilino groups form dihedral angles of 85.95 (6), 29.63 (7), 28.55 (1) and 87.48 (6)°, respectively, with the pyrazolo[3,4-*d*]pyrimidine unit [maximum deviation = 0.052 (2) Å]. An intramolecular N—H···N hydrogen bond occurs. The crystal structure features N—H···O hydrogen bonds.

## Related literature

For similar compounds, see: Wang *et al.* (2004, 2008). For their applications, see: Bendich *et al.* (1954); Ballell *et al.* (2007); Holla *et al.* (2006). For standard bond lengths, see: Allen *et al.* (1987).



## Experimental

### Crystal data

$\text{C}_{30}\text{H}_{23}\text{ClN}_6\text{OS}$   
 $M_r = 551.05$   
 Triclinic,  $P\bar{1}$   
 $a = 10.881$  (5) Å

$b = 10.948$  (4) Å  
 $c = 12.496$  (5) Å  
 $\alpha = 103.824$  (7)°  
 $\beta = 109.725$  (7)°

$\gamma = 93.377$  (7)°  
 $V = 1344.7$  (9) Å<sup>3</sup>  
 $Z = 2$   
 Mo  $K\alpha$  radiation

$\mu = 0.26$  mm<sup>-1</sup>  
 $T = 293$  K  
 $0.42 \times 0.40 \times 0.36$  mm

### Data collection

Bruker SMART CCD area-detector diffractometer  
 Absorption correction: multi-scan (SADABS; Bruker 2000)  
 $T_{\min} = 0.771$ ,  $T_{\max} = 1.000$   
 7799 measured reflections  
 5440 independent reflections  
 3418 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.022$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.046$   
 $wR(F^2) = 0.114$   
 $S = 1.01$   
 5440 reflections  
 360 parameters

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\max} = 0.40$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.40$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N5}-\text{H5}\cdots\text{N6}$	0.79 (2)	2.19 (2)	2.605 (3)	114 (2)
$\text{N6}-\text{H6}\cdots\text{O1}^{\dagger}$	0.81 (2)	2.12 (2)	2.904 (3)	162 (2)

 Symmetry code: (i)  $-x + 2, -y + 1, -z$ .

Data collection: SMART (Bruker, 2000); cell refinement: SAINT (Bruker, 2000); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL and Mercury (Macrae *et al.*, 2008).

We gratefully acknowledge the service provided by the X-ray facility at the Department of Chemistry and Chemical Engineering, Hengyang Normal University. Financial support of this work by the National Science Foundation (No. 11175080) and by the Natural Science Fund of Hunan Province (No. 10 J J6025) is gratefully acknowledged.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: RN2104).

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## supporting information

*Acta Cryst.* (2012). E68, o2895 [https://doi.org/10.1107/S1600536812037683]

## 5-Anilino-3-benzylsulfanyl-6-(3-chloroanilino)-1-phenyl-1*H*-pyrazolo[3,4-*d*]pyrimidin-4(5*H*)-one

Hong-Qing Wang, Fang-Fang He, Xiao-Feng Wang, Wei-Ping Zhou and Zhao-Jie Liu

### S1. Comment

Pyrazolo[3,4-*d*]pyrimidine derivatives are important heterocycles as purine analogs (Bendich *et al.*, 1954). They are usually used due to their potency as antagonists, antifungal agents (Ballell *et al.*, 2007; Holla *et al.*, 2006) and agrochemicals (Wang *et al.*, 2008). We have reported the synthesis and structure of pyrazolo[3,4-*d*]pyrimidin-4(5*H*)-one derivatives (Wang *et al.*, 2004).

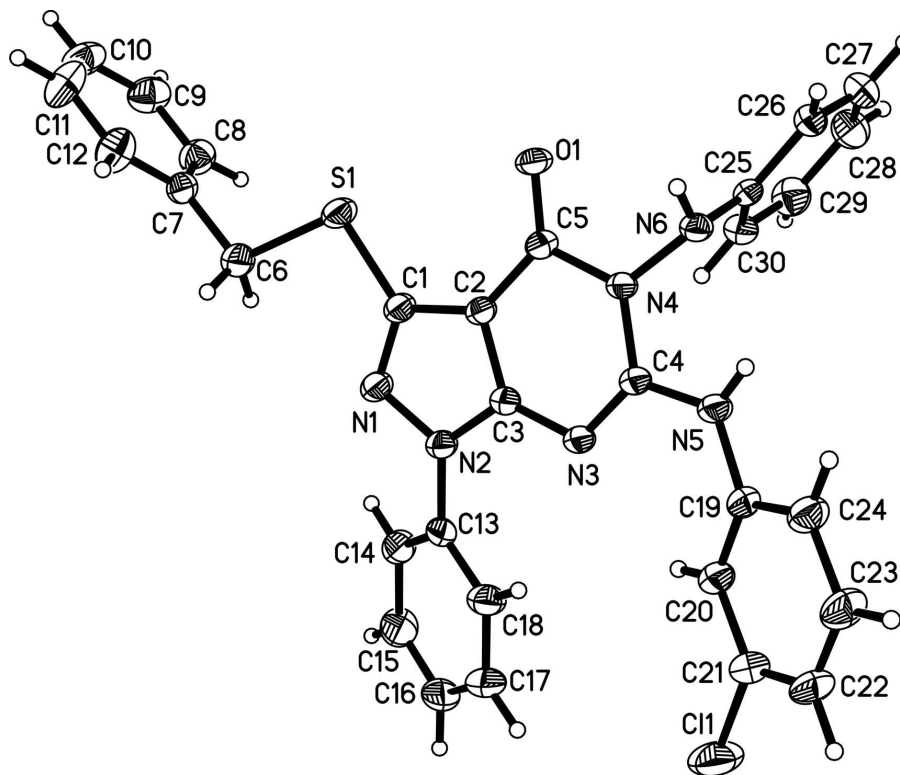
The molecule and packing diagrams of the title compound were shown in Figure 1 and 2, respectively. As shown in Fig. 1, the N1=C1 and N3=C4 bonds are 1.318 (3) and 1.304 (3) Å, respectively, slightly shorter than the normal C=N bond (1.329 and 1.336 Å; Allen *et al.*, 1987) but close to 1.314 (3) and 1.302 (3) Å in our previous report (Wang *et al.*, 2004), and the C2=C3 distance is 1.388 (3) Å, longer than the typical C=C (1.384 Å; Allen *et al.*, 1987). In the molecule, the N–C bonds are remarkably shorter than the normal N–C single bond (1.469 Å) and longer than the C=N double bond. The packing was stabilized by the intermolecular N–H⋯O hydrogen bond.

### S2. Experimental

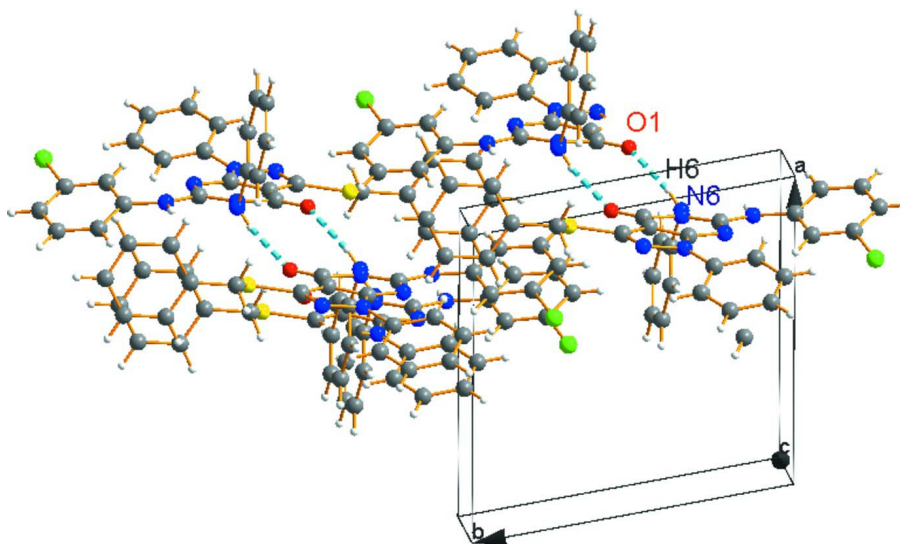
To a solution of iminophosphorane (2 mmol) in dry dichloromethane (20 ml) aryl isocyanate (2 mmol) was added under nitrogen atmosphere at room temperature. After the reaction mixture was stirred for 1.5 h, 0.231 g (2.0 mmol) *m*-chlorophenyl isocyanate was added, and the resulting mixture was stirred for an additional 30 min. Then the solvent was removed under reduced pressure, and 25 ml anhydrous ethanol and 1.5 ml of sodium ethoxide (3 mol/L) in ethanol were added to the mixture. After 3 h of stirring at room temperature, the solution was concentrated under reduced pressure and successively cooled. The crude product was collected by filtration. After recrystallization from DMF/petroleum ether or column chromatography on silica gel, white crystals were obtained.

### S3. Refinement

H atoms were included in their idealized positions, with C–H = 0.96 Å, and refined using a riding model with  $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$  for other H atoms.

**Figure 1**

View of the title compound, with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

**Figure 2**

Hydrogen bonding pattern found in the title compound

5-Anilino-3-benzylsulfanyl-6-(3-chloroanilino)-1-phenyl-1*H*-pyrazolo[3,4-*d*]pyrimidin-4(5*H*)-one

## Crystal data

C<sub>30</sub>H<sub>23</sub>ClN<sub>6</sub>OS $M_r = 551.05$ Triclinic,  $P\bar{1}$  $a = 10.881 (5) \text{ \AA}$  $b = 10.948 (4) \text{ \AA}$  $c = 12.496 (5) \text{ \AA}$  $\alpha = 103.824 (7)^\circ$  $\beta = 109.725 (7)^\circ$  $\gamma = 93.377 (7)^\circ$  $V = 1344.7 (9) \text{ \AA}^3$  $Z = 2$  $F(000) = 572$  $D_x = 1.361 \text{ Mg m}^{-3}$ Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$ 

Cell parameters from 809 reflections

 $\theta = 2.6\text{--}25.6^\circ$  $\mu = 0.26 \text{ mm}^{-1}$  $T = 293 \text{ K}$ 

Block, white

 $0.42 \times 0.40 \times 0.36 \text{ mm}$ 

## Data collection

Bruker SMART CCD area-detector  
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

phi and  $\omega$  scansAbsorption correction: multi-scan  
(*SADABS*; Bruker 2000) $T_{\min} = 0.771$ ,  $T_{\max} = 1.000$ 

7799 measured reflections

5440 independent reflections

3418 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.022$  $\theta_{\max} = 26.4^\circ$ ,  $\theta_{\min} = 1.9^\circ$  $h = -13 \rightarrow 9$  $k = -13 \rightarrow 13$  $l = -9 \rightarrow 15$ 

## Refinement

Refinement on  $F^2$ 

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.046$  $wR(F^2) = 0.114$  $S = 1.01$ 

5440 reflections

360 parameters

0 restraints

Primary atom site location: structure-invariant  
direct methodsSecondary atom site location: difference Fourier  
mapHydrogen site location: inferred from  
neighbouring sitesH atoms treated by a mixture of independent  
and constrained refinement $w = 1/[\sigma^2(F_o^2) + (0.0437P)^2 + 0.3561P]$ where  $P = (F_o^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\max} = 0.001$  $\Delta\rho_{\max} = 0.40 \text{ e \AA}^{-3}$  $\Delta\rho_{\min} = -0.40 \text{ e \AA}^{-3}$ 

## Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Cl1	0.58922 (10)	-0.29645 (7)	-0.00499 (9)	0.0950 (3)
S1	0.84719 (7)	0.64314 (5)	0.33369 (6)	0.0557 (2)
N1	0.7368 (2)	0.41189 (16)	0.32750 (16)	0.0462 (5)

N2	0.71298 (19)	0.28629 (16)	0.25760 (16)	0.0422 (5)
N3	0.75420 (19)	0.16942 (15)	0.08617 (15)	0.0395 (4)
N4	0.84282 (18)	0.30693 (15)	0.00045 (15)	0.0359 (4)
N5	0.8107 (2)	0.08909 (17)	-0.07736 (18)	0.0448 (5)
H5	0.844 (2)	0.108 (2)	-0.119 (2)	0.045 (7)*
N6	0.8716 (2)	0.31275 (17)	-0.10020 (17)	0.0382 (5)
H6	0.944 (2)	0.354 (2)	-0.0795 (18)	0.034 (6)*
O1	0.89538 (16)	0.52346 (13)	0.07639 (13)	0.0454 (4)
C1	0.7947 (2)	0.47893 (19)	0.27824 (19)	0.0417 (5)
C2	0.8102 (2)	0.40058 (18)	0.17656 (18)	0.0374 (5)
C3	0.7579 (2)	0.27820 (19)	0.16756 (18)	0.0371 (5)
C4	0.8006 (2)	0.18745 (18)	0.00674 (18)	0.0359 (5)
C5	0.8535 (2)	0.42237 (19)	0.08658 (18)	0.0360 (5)
C6	0.8389 (3)	0.6721 (2)	0.4802 (2)	0.0571 (7)
H6A	0.7510	0.6403	0.4744	0.068*
H6B	0.9018	0.6285	0.5272	0.068*
C7	0.8708 (2)	0.8136 (2)	0.5377 (2)	0.0438 (6)
C8	0.7750 (3)	0.8901 (2)	0.5120 (2)	0.0568 (7)
H8	0.6906	0.8537	0.4584	0.068*
C9	0.8016 (3)	1.0192 (3)	0.5641 (3)	0.0661 (8)
H9	0.7353	1.0691	0.5462	0.079*
C10	0.9247 (3)	1.0737 (3)	0.6416 (3)	0.0691 (8)
H10	0.9423	1.1608	0.6779	0.083*
C11	1.0216 (3)	1.0016 (3)	0.6661 (3)	0.0770 (9)
H11	1.1063	1.0395	0.7181	0.092*
C12	0.9953 (3)	0.8704 (3)	0.6138 (2)	0.0655 (8)
H12	1.0626	0.8214	0.6307	0.079*
C13	0.6338 (2)	0.1923 (2)	0.27856 (19)	0.0421 (5)
C14	0.5503 (3)	0.2301 (2)	0.3367 (2)	0.0567 (7)
H14	0.5454	0.3160	0.3624	0.068*
C15	0.4731 (3)	0.1394 (3)	0.3569 (2)	0.0700 (8)
H15	0.4168	0.1650	0.3969	0.084*
C16	0.4787 (3)	0.0121 (3)	0.3187 (2)	0.0657 (8)
H16	0.4275	-0.0482	0.3335	0.079*
C17	0.5606 (3)	-0.0250 (2)	0.2585 (2)	0.0613 (7)
H17	0.5632	-0.1112	0.2306	0.074*
C18	0.6397 (3)	0.0644 (2)	0.2385 (2)	0.0537 (6)
H18	0.6962	0.0386	0.1986	0.064*
C19	0.7924 (2)	-0.04243 (19)	-0.08586 (19)	0.0414 (5)
C20	0.7077 (3)	-0.0968 (2)	-0.0437 (2)	0.0488 (6)
H20	0.6589	-0.0476	-0.0069	0.059*
C21	0.6971 (3)	-0.2271 (2)	-0.0578 (2)	0.0554 (7)
C22	0.7651 (3)	-0.3030 (2)	-0.1140 (2)	0.0629 (8)
H22	0.7567	-0.3900	-0.1219	0.075*
C23	0.8462 (3)	-0.2473 (2)	-0.1585 (2)	0.0628 (7)
H23	0.8912	-0.2978	-0.1989	0.075*
C24	0.8620 (3)	-0.1173 (2)	-0.1443 (2)	0.0532 (6)
H24	0.9185	-0.0804	-0.1734	0.064*

C25	0.7690 (2)	0.34949 (17)	-0.18724 (18)	0.0367 (5)
C26	0.7993 (3)	0.3826 (2)	-0.2774 (2)	0.0466 (6)
H26	0.8847	0.3840	-0.2777	0.056*
C27	0.7010 (3)	0.4132 (2)	-0.3663 (2)	0.0621 (8)
H27	0.7206	0.4345	-0.4271	0.075*
C28	0.5748 (3)	0.4127 (3)	-0.3663 (2)	0.0661 (8)
H28	0.5097	0.4340	-0.4264	0.079*
C29	0.5450 (3)	0.3804 (2)	-0.2768 (2)	0.0599 (7)
H29	0.4597	0.3804	-0.2763	0.072*
C30	0.6417 (2)	0.3480 (2)	-0.1879 (2)	0.0454 (6)
H30	0.6210	0.3250	-0.1283	0.054*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Cl1	0.1326 (8)	0.0500 (4)	0.1256 (7)	-0.0044 (4)	0.0769 (6)	0.0270 (5)
S1	0.0814 (5)	0.0344 (3)	0.0551 (4)	-0.0089 (3)	0.0419 (4)	-0.0015 (3)
N1	0.0592 (13)	0.0348 (10)	0.0453 (11)	-0.0037 (9)	0.0270 (10)	0.0030 (9)
N2	0.0562 (13)	0.0326 (10)	0.0416 (11)	-0.0012 (9)	0.0257 (10)	0.0072 (8)
N3	0.0549 (12)	0.0273 (9)	0.0405 (10)	0.0031 (8)	0.0231 (9)	0.0092 (8)
N4	0.0485 (12)	0.0268 (9)	0.0376 (10)	0.0031 (8)	0.0220 (9)	0.0095 (8)
N5	0.0684 (15)	0.0268 (9)	0.0482 (12)	0.0058 (9)	0.0333 (11)	0.0095 (9)
N6	0.0462 (13)	0.0320 (10)	0.0425 (11)	0.0015 (9)	0.0247 (10)	0.0097 (8)
O1	0.0586 (11)	0.0292 (8)	0.0553 (10)	-0.0024 (7)	0.0322 (8)	0.0093 (7)
C1	0.0506 (15)	0.0322 (11)	0.0431 (13)	-0.0009 (10)	0.0233 (11)	0.0042 (10)
C2	0.0441 (14)	0.0295 (11)	0.0390 (12)	-0.0009 (9)	0.0190 (10)	0.0060 (9)
C3	0.0446 (14)	0.0326 (11)	0.0362 (12)	0.0020 (10)	0.0175 (10)	0.0099 (9)
C4	0.0435 (14)	0.0272 (10)	0.0385 (12)	0.0043 (9)	0.0163 (10)	0.0100 (9)
C5	0.0364 (13)	0.0297 (11)	0.0419 (12)	0.0019 (9)	0.0160 (10)	0.0078 (9)
C6	0.081 (2)	0.0429 (14)	0.0478 (15)	-0.0009 (13)	0.0318 (14)	0.0046 (11)
C7	0.0543 (16)	0.0395 (12)	0.0403 (13)	0.0015 (11)	0.0266 (12)	0.0035 (10)
C8	0.0528 (17)	0.0561 (16)	0.0603 (16)	0.0026 (13)	0.0232 (13)	0.0114 (13)
C9	0.074 (2)	0.0498 (16)	0.085 (2)	0.0168 (15)	0.0418 (18)	0.0178 (15)
C10	0.087 (2)	0.0399 (14)	0.087 (2)	-0.0006 (16)	0.0523 (19)	0.0005 (14)
C11	0.061 (2)	0.0630 (19)	0.083 (2)	-0.0084 (16)	0.0222 (17)	-0.0132 (16)
C12	0.0561 (18)	0.0586 (17)	0.0715 (19)	0.0141 (14)	0.0196 (15)	0.0030 (14)
C13	0.0484 (15)	0.0421 (12)	0.0358 (12)	-0.0045 (11)	0.0161 (11)	0.0122 (10)
C14	0.0707 (19)	0.0505 (15)	0.0511 (15)	-0.0058 (13)	0.0332 (14)	0.0051 (12)
C15	0.076 (2)	0.077 (2)	0.0628 (18)	-0.0123 (16)	0.0422 (16)	0.0110 (15)
C16	0.074 (2)	0.0634 (18)	0.0602 (17)	-0.0176 (15)	0.0256 (15)	0.0226 (14)
C17	0.077 (2)	0.0452 (14)	0.0671 (18)	-0.0042 (14)	0.0309 (16)	0.0218 (13)
C18	0.0669 (18)	0.0450 (14)	0.0590 (16)	0.0046 (12)	0.0314 (14)	0.0204 (12)
C19	0.0525 (15)	0.0284 (11)	0.0404 (13)	0.0054 (10)	0.0145 (11)	0.0079 (10)
C20	0.0653 (17)	0.0320 (12)	0.0514 (14)	0.0056 (11)	0.0260 (13)	0.0089 (11)
C21	0.0706 (19)	0.0365 (13)	0.0586 (16)	-0.0001 (12)	0.0230 (14)	0.0150 (12)
C22	0.086 (2)	0.0288 (12)	0.0727 (19)	0.0135 (13)	0.0263 (17)	0.0151 (13)
C23	0.075 (2)	0.0366 (13)	0.0789 (19)	0.0184 (13)	0.0331 (16)	0.0098 (13)
C24	0.0639 (18)	0.0365 (13)	0.0624 (16)	0.0082 (12)	0.0292 (14)	0.0103 (12)

C25	0.0520 (15)	0.0214 (10)	0.0363 (12)	0.0018 (9)	0.0186 (11)	0.0046 (9)
C26	0.0625 (17)	0.0385 (12)	0.0442 (14)	0.0022 (11)	0.0292 (13)	0.0083 (11)
C27	0.088 (2)	0.0571 (16)	0.0436 (15)	0.0016 (15)	0.0250 (15)	0.0192 (13)
C28	0.075 (2)	0.0660 (18)	0.0525 (17)	0.0097 (16)	0.0115 (15)	0.0254 (14)
C29	0.0521 (17)	0.0617 (17)	0.0612 (17)	0.0069 (13)	0.0148 (14)	0.0173 (14)
C30	0.0495 (16)	0.0426 (13)	0.0473 (14)	0.0013 (11)	0.0214 (12)	0.0140 (11)

*Geometric parameters (Å, °)*

C11—C21	1.746 (3)	C12—H12	0.9300
S1—C1	1.748 (2)	C13—C14	1.367 (3)
S1—C6	1.816 (2)	C13—C18	1.383 (3)
N1—C1	1.318 (3)	C14—C15	1.386 (3)
N1—N2	1.397 (2)	C14—H14	0.9300
N2—C3	1.357 (3)	C15—C16	1.373 (4)
N2—C13	1.430 (3)	C15—H15	0.9300
N3—C4	1.304 (3)	C16—C17	1.369 (4)
N3—C3	1.357 (3)	C16—H16	0.9300
N4—C4	1.389 (2)	C17—C18	1.387 (3)
N4—N6	1.409 (2)	C17—H17	0.9300
N4—C5	1.420 (3)	C18—H18	0.9300
N5—C4	1.351 (3)	C19—C20	1.376 (3)
N5—C19	1.415 (3)	C19—C24	1.392 (3)
N5—H5	0.79 (2)	C20—C21	1.388 (3)
N6—C25	1.426 (3)	C20—H20	0.9300
N6—H6	0.81 (2)	C21—C22	1.369 (4)
O1—C5	1.223 (2)	C22—C23	1.375 (4)
C1—C2	1.421 (3)	C22—H22	0.9300
C2—C3	1.388 (3)	C23—C24	1.384 (3)
C2—C5	1.420 (3)	C23—H23	0.9300
C6—C7	1.508 (3)	C24—H24	0.9300
C6—H6A	0.9700	C25—C30	1.381 (3)
C6—H6B	0.9700	C25—C26	1.393 (3)
C7—C12	1.369 (4)	C26—C27	1.382 (4)
C7—C8	1.377 (3)	C26—H26	0.9300
C8—C9	1.375 (4)	C27—C28	1.372 (4)
C8—H8	0.9300	C27—H27	0.9300
C9—C10	1.357 (4)	C28—C29	1.379 (4)
C9—H9	0.9300	C28—H28	0.9300
C10—C11	1.351 (4)	C29—C30	1.382 (3)
C10—H10	0.9300	C29—H29	0.9300
C11—C12	1.396 (4)	C30—H30	0.9300
C11—H11	0.9300		
C1—S1—C6	101.17 (11)	C14—C13—C18	120.3 (2)
C1—N1—N2	105.43 (17)	C14—C13—N2	119.3 (2)
C3—N2—N1	110.80 (16)	C18—C13—N2	120.4 (2)
C3—N2—C13	129.75 (18)	C13—C14—C15	119.5 (2)

N1—N2—C13	118.85 (17)	C13—C14—H14	120.2
C4—N3—C3	113.84 (17)	C15—C14—H14	120.2
C4—N4—N6	117.28 (16)	C16—C15—C14	120.9 (3)
C4—N4—C5	123.97 (17)	C16—C15—H15	119.6
N6—N4—C5	118.71 (15)	C14—C15—H15	119.6
C4—N5—C19	128.2 (2)	C17—C16—C15	119.1 (2)
C4—N5—H5	115.2 (17)	C17—C16—H16	120.4
C19—N5—H5	115.4 (17)	C15—C16—H16	120.4
N4—N6—C25	114.04 (18)	C16—C17—C18	120.8 (3)
N4—N6—H6	109.4 (15)	C16—C17—H17	119.6
C25—N6—H6	114.9 (15)	C18—C17—H17	119.6
N1—C1—C2	111.45 (18)	C13—C18—C17	119.3 (2)
N1—C1—S1	123.62 (16)	C13—C18—H18	120.4
C2—C1—S1	124.93 (16)	C17—C18—H18	120.4
C3—C2—C5	119.40 (19)	C20—C19—C24	120.6 (2)
C3—C2—C1	105.10 (18)	C20—C19—N5	123.2 (2)
C5—C2—C1	135.25 (19)	C24—C19—N5	116.2 (2)
N3—C3—N2	125.70 (18)	C19—C20—C21	118.1 (2)
N3—C3—C2	127.08 (19)	C19—C20—H20	120.9
N2—C3—C2	107.21 (18)	C21—C20—H20	120.9
N3—C4—N5	121.72 (19)	C22—C21—C20	122.6 (2)
N3—C4—N4	123.69 (19)	C22—C21—C11	119.16 (19)
N5—C4—N4	114.59 (19)	C20—C21—C11	118.2 (2)
O1—C5—C2	128.6 (2)	C21—C22—C23	118.3 (2)
O1—C5—N4	119.93 (19)	C21—C22—H22	120.9
C2—C5—N4	111.51 (17)	C23—C22—H22	120.9
C7—C6—S1	108.09 (16)	C22—C23—C24	121.1 (2)
C7—C6—H6A	110.1	C22—C23—H23	119.4
S1—C6—H6A	110.1	C24—C23—H23	119.4
C7—C6—H6B	110.1	C23—C24—C19	119.2 (2)
S1—C6—H6B	110.1	C23—C24—H24	120.4
H6A—C6—H6B	108.4	C19—C24—H24	120.4
C12—C7—C8	118.1 (2)	C30—C25—C26	119.7 (2)
C12—C7—C6	121.6 (2)	C30—C25—N6	122.8 (2)
C8—C7—C6	120.3 (2)	C26—C25—N6	117.5 (2)
C9—C8—C7	121.3 (3)	C27—C26—C25	119.2 (2)
C9—C8—H8	119.3	C27—C26—H26	120.4
C7—C8—H8	119.3	C25—C26—H26	120.4
C10—C9—C8	119.9 (3)	C28—C27—C26	121.0 (2)
C10—C9—H9	120.1	C28—C27—H27	119.5
C8—C9—H9	120.1	C26—C27—H27	119.5
C11—C10—C9	120.1 (3)	C27—C28—C29	119.8 (3)
C11—C10—H10	120.0	C27—C28—H28	120.1
C9—C10—H10	120.0	C29—C28—H28	120.1
C10—C11—C12	120.3 (3)	C28—C29—C30	120.0 (3)
C10—C11—H11	119.9	C28—C29—H29	120.0
C12—C11—H11	119.9	C30—C29—H29	120.0
C7—C12—C11	120.3 (3)	C25—C30—C29	120.3 (2)



C7—C12—H12	119.9	C25—C30—H30	119.8
C11—C12—H12	119.9	C29—C30—H30	119.8
C1—N1—N2—C3	0.7 (3)	C6—C7—C8—C9	179.4 (2)
C1—N1—N2—C13	-171.3 (2)	C7—C8—C9—C10	0.7 (4)
C4—N4—N6—C25	100.4 (2)	C8—C9—C10—C11	1.1 (4)
C5—N4—N6—C25	-77.3 (2)	C9—C10—C11—C12	-1.3 (5)
N2—N1—C1—C2	0.0 (3)	C8—C7—C12—C11	2.0 (4)
N2—N1—C1—S1	-179.17 (17)	C6—C7—C12—C11	-179.6 (2)
C6—S1—C1—N1	13.9 (2)	C10—C11—C12—C7	-0.3 (5)
C6—S1—C1—C2	-165.1 (2)	C3—N2—C13—C14	-149.5 (2)
N1—C1—C2—C3	-0.6 (3)	N1—N2—C13—C14	20.7 (3)
S1—C1—C2—C3	178.52 (17)	C3—N2—C13—C18	29.3 (4)
N1—C1—C2—C5	173.3 (2)	N1—N2—C13—C18	-160.5 (2)
S1—C1—C2—C5	-7.5 (4)	C18—C13—C14—C15	1.0 (4)
C4—N3—C3—N2	178.3 (2)	N2—C13—C14—C15	179.8 (2)
C4—N3—C3—C2	-3.2 (3)	C13—C14—C15—C16	-0.4 (4)
N1—N2—C3—N3	177.6 (2)	C14—C15—C16—C17	-0.9 (4)
C13—N2—C3—N3	-11.5 (4)	C15—C16—C17—C18	1.6 (4)
N1—N2—C3—C2	-1.1 (3)	C14—C13—C18—C17	-0.3 (4)
C13—N2—C3—C2	169.7 (2)	N2—C13—C18—C17	-179.1 (2)
C5—C2—C3—N3	7.2 (4)	C16—C17—C18—C13	-1.0 (4)
C1—C2—C3—N3	-177.7 (2)	C4—N5—C19—C20	30.2 (4)
C5—C2—C3—N2	-174.1 (2)	C4—N5—C19—C24	-152.2 (2)
C1—C2—C3—N2	1.0 (3)	C24—C19—C20—C21	2.0 (4)
C3—N3—C4—N5	176.6 (2)	N5—C19—C20—C21	179.5 (2)
C3—N3—C4—N4	-3.8 (3)	C19—C20—C21—C22	-1.5 (4)
C19—N5—C4—N3	-9.2 (4)	C19—C20—C21—C11	-179.75 (19)
C19—N5—C4—N4	171.1 (2)	C20—C21—C22—C23	-0.4 (4)
N6—N4—C4—N3	-170.6 (2)	C11—C21—C22—C23	177.8 (2)
C5—N4—C4—N3	6.9 (3)	C21—C22—C23—C24	1.8 (4)
N6—N4—C4—N5	9.1 (3)	C22—C23—C24—C19	-1.3 (4)
C5—N4—C4—N5	-173.4 (2)	C20—C19—C24—C23	-0.7 (4)
C3—C2—C5—O1	175.9 (2)	N5—C19—C24—C23	-178.3 (2)
C1—C2—C5—O1	2.6 (5)	N4—N6—C25—C30	-15.6 (3)
C3—C2—C5—N4	-3.7 (3)	N4—N6—C25—C26	167.26 (17)
C1—C2—C5—N4	-177.0 (2)	C30—C25—C26—C27	-0.1 (3)
C4—N4—C5—O1	177.8 (2)	N6—C25—C26—C27	177.1 (2)
N6—N4—C5—O1	-4.7 (3)	C25—C26—C27—C28	0.6 (4)
C4—N4—C5—C2	-2.6 (3)	C26—C27—C28—C29	-0.4 (4)
N6—N4—C5—C2	174.89 (19)	C27—C28—C29—C30	-0.4 (4)
C1—S1—C6—C7	-175.34 (19)	C26—C25—C30—C29	-0.7 (3)
S1—C6—C7—C12	-96.8 (3)	N6—C25—C30—C29	-177.8 (2)
S1—C6—C7—C8	81.5 (3)	C28—C29—C30—C25	0.9 (4)
C12—C7—C8—C9	-2.2 (4)		

*Hydrogen-bond geometry (Å, °)*

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
N5—H5 $\cdots$ N6	0.79 (2)	2.19 (2)	2.605 (3)	114 (2)
N6—H6 $\cdots$ O1 <sup>i</sup>	0.81 (2)	2.12 (2)	2.904 (3)	162 (2)

Symmetry code: (i)  $-x+2, -y+1, -z$ .