

2-[Anilino(phenyl)methyl]cycloheptanone

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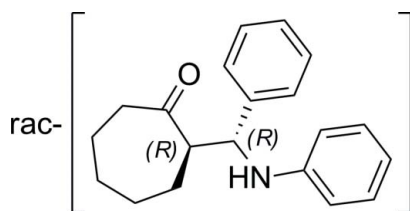
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Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.048; wR factor = 0.136; data-to-parameter ratio = 17.0.

In the title compound, $\text{C}_{20}\text{H}_{23}\text{NO}$, the cycloheptanone ring adopts a twist-chair conformation, with the aminomethyl substituent in an equatorial position. The relative configuration of the two stereocenters is R,R . In the crystal, molecules are linked by $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds into chains along [100].

Related literature

For the synthesis of title compound and related compounds, see: Eftekhari-Sis *et al.* (2013). For the biological activity of β -amino ketones, see: Arend *et al.* (1998); Jadhav *et al.* (2008); Kalluraya *et al.* (2001). For information on the Mannich reaction, see, for example: Eftekhari-Sis *et al.* (2006); Azizi *et al.* (2006); Cordova (2004). For the crystal structures of related compounds, see: Eftekhari-Sis *et al.* (2012); Yuan *et al.* (2007); Fun *et al.* (2009). For puckering parameters, see: Evans & Boeyens (1989).



Experimental

Crystal data

$\text{C}_{20}\text{H}_{23}\text{NO}$
 $M_r = 293.39$
 Monoclinic, $P2_1/c$

$a = 5.7534$ (4) Å
 $b = 16.1336$ (8) Å
 $c = 18.1980$ (13) Å

$\beta = 99.371$ (6)°
 $V = 1666.65$ (19) Å³
 $Z = 4$
 Mo $K\alpha$ radiation

$\mu = 0.07$ mm⁻¹
 $T = 296$ K
 $0.72 \times 0.56 \times 0.27$ mm

Data collection

Stoe IPDS 2 diffractometer
 Absorption correction: integration
 (X -RED32; Stoe & Cie, 2002)
 $T_{\min} = 0.965$, $T_{\max} = 0.985$

10757 measured reflections
 3449 independent reflections
 2170 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.041$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.048$
 $wR(F^2) = 0.136$
 $S = 1.01$
 3449 reflections
 203 parameters

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.26$ e Å⁻³
 $\Delta\rho_{\min} = -0.13$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H1A}\cdots\text{O1}^i$	0.879 (19)	2.23 (2)	3.065 (2)	158.4 (16)

Symmetry code: (i) $x + 1, y, z$.

Data collection: X -AREA (Stoe & Cie, 2002); cell refinement: X -AREA; data reduction: X -RED32 (Stoe & Cie, 2002); program(s) used to solve structure: $SHELXS97$ (Sheldrick, 2008); program(s) used to refine structure: $SHELXL97$ (Sheldrick, 2008); molecular graphics: $ORTEP-3$ for Windows (Farrugia, 2012); software used to prepare material for publication: $WinGX$ (Farrugia, 2012).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: FY2069).

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supporting information

Acta Cryst. (2013). E69, o109 [https://doi.org/10.1107/S1600536812050659]

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S1. Comment

Mannich reaction (Eftekhari-Sis *et al.*, 2006; Azizi *et al.*, 2006; Cordova, 2004) is one of the most important basic reactions in organic chemistry for its use in natural product and pharmaceutical syntheses due to formation of C—C and C—N bonds, simultaneously, to furnish β -amino ketones, which exhibit biological activity such as anti-inflammatory (Jadhav *et al.*, 2008) and antimicrobial (Kalluraya *et al.*, 2001) activities. We have synthesized the title compound and report its structure here, Fig 1. The cycloheptanone ring adopts twist-chair conformation, with puckering parameters of $Q(2) = 0.539(2) \text{ \AA}$, $\Phi(2) = 41.7(2)^\circ$ and $Q(3) = 0.644(2) \text{ \AA}$, $\Phi(3) = 319.2(2)^\circ$ (Evans & Boeyens, 1989). The amino-methyl moiety is positioned equatorially on the ring at C1.

S2. Experimental

The title compound was obtained by adding of 0.04 g of Laponite-HPMC nano composite (Eftekhari-Sis *et al.*, 2013) to a mixture of 0.5 mmol of benzaldehyde, 0.5 mmol of aniline and 3 equiv. of cycloheptanone and stirring at room temperature for 24 h. After completion of the reaction, 5 ml EtOH was added and the catalyst was removed by filtration. The filtrate was concentrated under reduced pressure. The obtained crude product was recrystallized from EtOH to afford the title compound in 60% yield. Colorless crystals suitable for crystal structure determination were grown from 96% EtOH.

S3. Refinement

Carbon bound H atoms were positioned geometrically, with C—H=0.93, 0.97, and 0.98 \AA for aromatic, methylene and methine H, and constrained to ride on their parent atoms, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. The nitrogen H atoms were located from the difference Fourier map and allowed to refine freely.

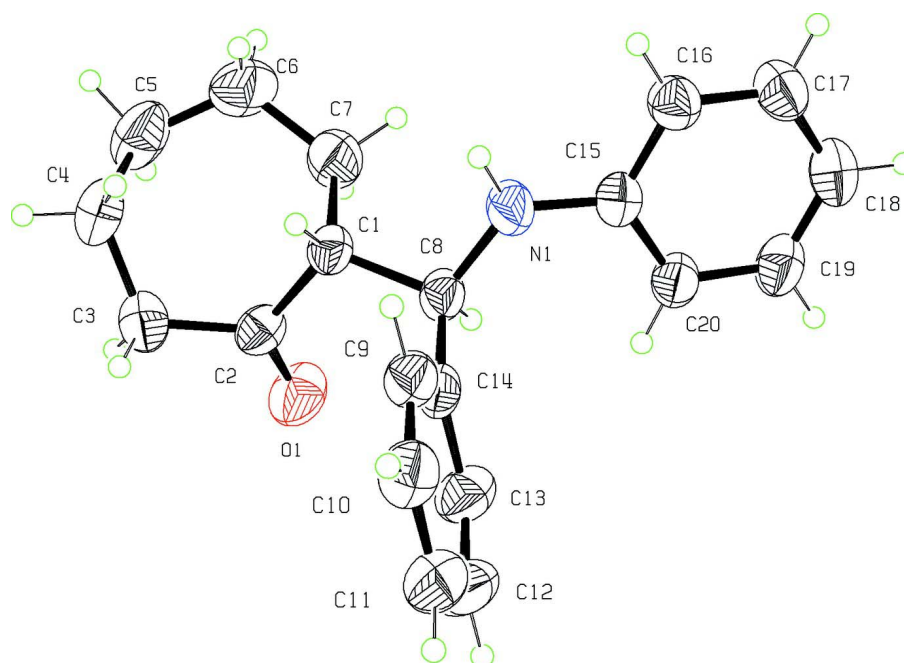


Figure 1

The structure of title compound, showing 40% probability displacement ellipsoids and the atom numbering scheme.

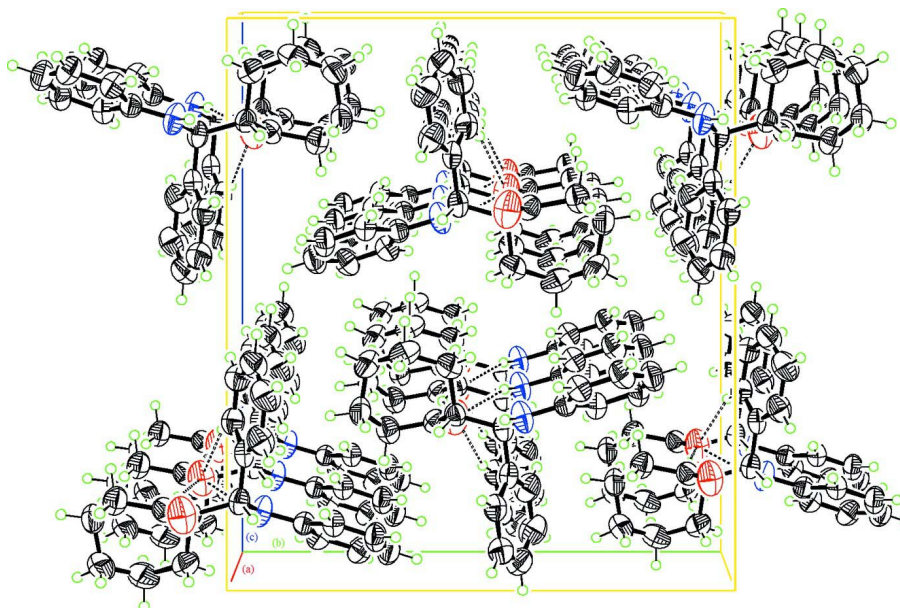


Figure 2

The unit cell contents of title compound.

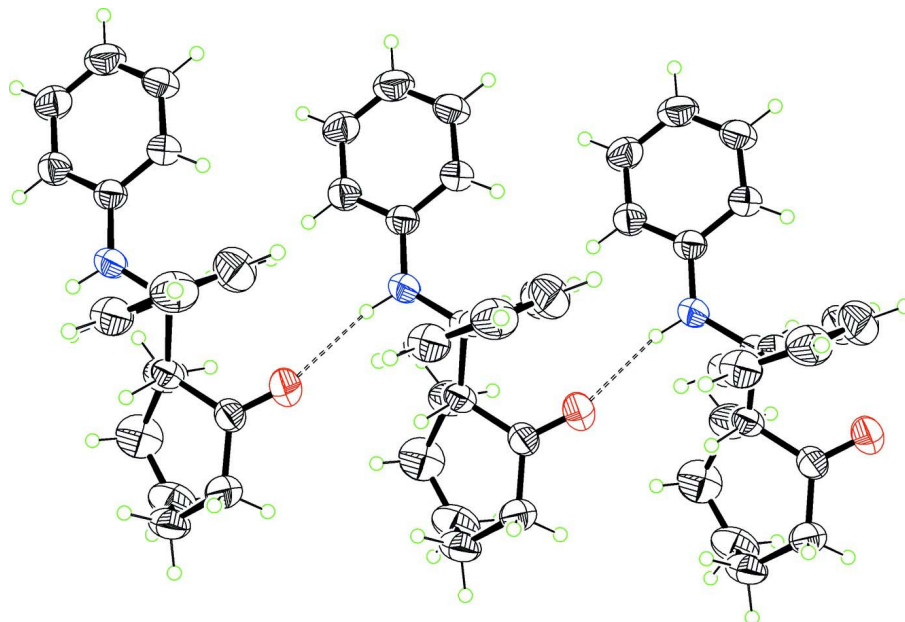


Figure 3

The linking of molecules by N—H···O hydrogen bonds into infinite one-dimensional chains. Hydrogen bonds are shown as dashed lines.

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Crystal data

$C_{20}H_{23}NO$

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Monoclinic, $P2_1/c$

Hall symbol: $-P\ 2ybc$

$a = 5.7534$ (4) Å

$b = 16.1336$ (8) Å

$c = 18.1980$ (13) Å

$\beta = 99.371$ (6)°

$V = 1666.65$ (19) Å³

$Z = 4$

$F(000) = 632$

$D_x = 1.169$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 10757 reflections

$\theta = 1.7$ – 28.0 °

$\mu = 0.07$ mm⁻¹

$T = 296$ K

Prism, colourless

$0.72 \times 0.56 \times 0.27$ mm

Data collection

Stoe IPDS 2

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

rotation method scans

Absorption correction: integration

(*X-RED32*; Stoe & Cie, 2002)

$T_{\min} = 0.965$, $T_{\max} = 0.985$

10757 measured reflections

3449 independent reflections

2170 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.041$

$\theta_{\max} = 26.5$ °, $\theta_{\min} = 1.7$ °

$h = -7 \rightarrow 7$

$k = -20 \rightarrow 20$

$l = -22 \rightarrow 18$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.048$

$wR(F^2) = 0.136$

$S = 1.01$

3449 reflections

203 parameters

0 restraints

Primary atom site location: structure-invariant direct methods
 Secondary atom site location: difference Fourier map
 Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$$w = 1/[\sigma^2(F_o^2) + (0.0722P)^2]$$

where $P = (F_o^2 + 2F_c^2)/3$

$$(\Delta/\sigma)_{\max} < 0.001$$

$$\Delta\rho_{\max} = 0.26 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.13 \text{ e } \text{\AA}^{-3}$$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.4026 (3)	0.45073 (9)	0.32943 (9)	0.0508 (4)
H1	0.5213	0.4211	0.3065	0.061*
C2	0.1676 (3)	0.41004 (10)	0.30531 (10)	0.0545 (4)
C3	0.1590 (3)	0.31909 (11)	0.28983 (11)	0.0680 (5)
H3A	0.1590	0.3111	0.2370	0.082*
H3B	0.0108	0.2978	0.3008	0.082*
C4	0.3563 (4)	0.26786 (11)	0.33231 (12)	0.0750 (6)
H4A	0.3205	0.2097	0.3229	0.090*
H4B	0.5002	0.2801	0.3131	0.090*
C5	0.3998 (5)	0.28183 (13)	0.41489 (13)	0.0919 (7)
H5A	0.2488	0.2880	0.4315	0.110*
H5B	0.4747	0.2328	0.4388	0.110*
C6	0.5488 (5)	0.35543 (15)	0.44101 (12)	0.0922 (7)
H6A	0.5733	0.3557	0.4950	0.111*
H6B	0.7017	0.3476	0.4261	0.111*
C7	0.4594 (4)	0.43901 (13)	0.41458 (11)	0.0786 (6)
H7A	0.5761	0.4800	0.4346	0.094*
H7B	0.3177	0.4505	0.4353	0.094*
C8	0.4030 (3)	0.54291 (9)	0.30610 (10)	0.0533 (4)
H8	0.2812	0.5716	0.3283	0.064*
C9	0.5035 (3)	0.52774 (10)	0.17700 (11)	0.0606 (5)
H9	0.6439	0.5026	0.1982	0.073*
C10	0.4556 (4)	0.53841 (12)	0.10111 (12)	0.0726 (5)
H10	0.5642	0.5210	0.0717	0.087*
C11	0.2490 (4)	0.57456 (13)	0.06854 (13)	0.0801 (6)
H11	0.2159	0.5810	0.0171	0.096*
C12	0.0910 (4)	0.60127 (14)	0.11243 (14)	0.0850 (7)
H12	-0.0489	0.6264	0.0907	0.102*
C13	0.1397 (3)	0.59089 (12)	0.18865 (12)	0.0713 (5)

H13	0.0314	0.6093	0.2178	0.086*
C14	0.3459 (3)	0.55379 (9)	0.22262 (10)	0.0530 (4)
C15	0.6636 (3)	0.65940 (9)	0.36011 (9)	0.0525 (4)
C16	0.8846 (3)	0.68313 (11)	0.39812 (10)	0.0604 (4)
H16	1.0051	0.6441	0.4065	0.073*
C17	0.9275 (4)	0.76247 (12)	0.42320 (12)	0.0706 (5)
H17	1.0764	0.7766	0.4481	0.085*
C18	0.7531 (4)	0.82152 (12)	0.41206 (12)	0.0760 (6)
H18	0.7823	0.8754	0.4293	0.091*
C19	0.5343 (4)	0.79933 (11)	0.37481 (12)	0.0688 (5)
H19	0.4152	0.8389	0.3671	0.083*
C20	0.4877 (3)	0.71963 (10)	0.34864 (10)	0.0587 (4)
H20	0.3387	0.7062	0.3233	0.070*
N1	0.6284 (3)	0.57845 (9)	0.33722 (10)	0.0661 (5)
O1	-0.0127 (2)	0.44979 (8)	0.30255 (10)	0.0854 (5)
H1A	0.751 (3)	0.5457 (11)	0.3392 (10)	0.064 (5)*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0554 (9)	0.0428 (8)	0.0545 (10)	0.0063 (7)	0.0095 (7)	-0.0041 (7)
C2	0.0518 (9)	0.0529 (9)	0.0604 (11)	0.0067 (8)	0.0142 (8)	0.0069 (8)
C3	0.0697 (11)	0.0561 (10)	0.0797 (13)	-0.0070 (9)	0.0166 (9)	-0.0084 (9)
C4	0.0957 (15)	0.0449 (9)	0.0878 (15)	0.0060 (10)	0.0253 (11)	0.0033 (9)
C5	0.132 (2)	0.0664 (13)	0.0802 (16)	0.0179 (13)	0.0271 (14)	0.0195 (11)
C6	0.1148 (18)	0.0980 (17)	0.0584 (13)	0.0108 (15)	-0.0023 (12)	0.0184 (11)
C7	0.1024 (16)	0.0695 (12)	0.0605 (12)	0.0015 (11)	0.0029 (11)	-0.0085 (10)
C8	0.0482 (9)	0.0410 (8)	0.0709 (11)	0.0037 (7)	0.0104 (8)	-0.0043 (7)
C9	0.0577 (10)	0.0510 (9)	0.0719 (13)	0.0038 (8)	0.0067 (9)	-0.0009 (8)
C10	0.0826 (14)	0.0633 (12)	0.0738 (14)	0.0020 (10)	0.0182 (11)	-0.0008 (9)
C11	0.0962 (16)	0.0693 (12)	0.0719 (13)	0.0007 (12)	0.0054 (12)	0.0129 (10)
C12	0.0760 (14)	0.0830 (14)	0.0903 (17)	0.0155 (11)	-0.0031 (12)	0.0220 (12)
C13	0.0606 (11)	0.0674 (11)	0.0857 (15)	0.0125 (9)	0.0110 (10)	0.0134 (10)
C14	0.0509 (9)	0.0348 (7)	0.0730 (12)	-0.0021 (7)	0.0092 (8)	0.0022 (7)
C15	0.0585 (10)	0.0456 (8)	0.0551 (10)	-0.0011 (7)	0.0142 (7)	-0.0021 (7)
C16	0.0588 (10)	0.0578 (10)	0.0639 (11)	0.0009 (8)	0.0076 (8)	-0.0071 (8)
C17	0.0700 (12)	0.0648 (11)	0.0757 (13)	-0.0106 (10)	0.0076 (10)	-0.0152 (9)
C18	0.0901 (14)	0.0522 (10)	0.0873 (15)	-0.0114 (10)	0.0198 (11)	-0.0138 (10)
C19	0.0754 (12)	0.0462 (9)	0.0876 (14)	0.0064 (9)	0.0215 (10)	0.0000 (9)
C20	0.0591 (10)	0.0469 (9)	0.0707 (12)	0.0001 (8)	0.0122 (8)	0.0007 (8)
N1	0.0520 (9)	0.0463 (8)	0.0961 (13)	0.0057 (7)	0.0010 (8)	-0.0161 (7)
O1	0.0558 (8)	0.0696 (9)	0.1328 (14)	0.0105 (7)	0.0216 (8)	0.0106 (8)

Geometric parameters (Å, °)

C1—C2	1.502 (2)	C9—C14	1.390 (2)
C1—C7	1.542 (2)	C9—H9	0.9300
C1—C8	1.547 (2)	C10—C11	1.369 (3)

C1—H1	0.9800	C10—H10	0.9300
C2—O1	1.214 (2)	C11—C12	1.373 (3)
C2—C3	1.493 (2)	C11—H11	0.9300
C3—C4	1.511 (3)	C12—C13	1.380 (3)
C3—H3A	0.9700	C12—H12	0.9300
C3—H3B	0.9700	C13—C14	1.382 (2)
C4—C5	1.500 (3)	C13—H13	0.9300
C4—H4A	0.9700	C15—N1	1.376 (2)
C4—H4B	0.9700	C15—C20	1.394 (2)
C5—C6	1.496 (3)	C15—C16	1.399 (2)
C5—H5A	0.9700	C16—C17	1.368 (2)
C5—H5B	0.9700	C16—H16	0.9300
C6—C7	1.495 (3)	C17—C18	1.375 (3)
C6—H6A	0.9700	C17—H17	0.9300
C6—H6B	0.9700	C18—C19	1.376 (3)
C7—H7A	0.9700	C18—H18	0.9300
C7—H7B	0.9700	C19—C20	1.382 (2)
C8—N1	1.446 (2)	C19—H19	0.9300
C8—C14	1.511 (2)	C20—H20	0.9300
C8—H8	0.9800	N1—H1A	0.879 (19)
C9—C10	1.374 (3)		
C2—C1—C7	105.86 (14)	N1—C8—H8	107.9
C2—C1—C8	112.43 (13)	C14—C8—H8	107.9
C7—C1—C8	112.58 (14)	C1—C8—H8	107.9
C2—C1—H1	108.6	C10—C9—C14	121.25 (17)
C7—C1—H1	108.6	C10—C9—H9	119.4
C8—C1—H1	108.6	C14—C9—H9	119.4
O1—C2—C3	120.63 (16)	C11—C10—C9	120.4 (2)
O1—C2—C1	120.23 (15)	C11—C10—H10	119.8
C3—C2—C1	119.02 (14)	C9—C10—H10	119.8
C2—C3—C4	116.21 (16)	C10—C11—C12	119.5 (2)
C2—C3—H3A	108.2	C10—C11—H11	120.3
C4—C3—H3A	108.2	C12—C11—H11	120.3
C2—C3—H3B	108.2	C11—C12—C13	120.1 (2)
C4—C3—H3B	108.2	C11—C12—H12	120.0
H3A—C3—H3B	107.4	C13—C12—H12	120.0
C5—C4—C3	114.78 (17)	C12—C13—C14	121.4 (2)
C5—C4—H4A	108.6	C12—C13—H13	119.3
C3—C4—H4A	108.6	C14—C13—H13	119.3
C5—C4—H4B	108.6	C13—C14—C9	117.37 (18)
C3—C4—H4B	108.6	C13—C14—C8	122.03 (16)
H4A—C4—H4B	107.5	C9—C14—C8	120.59 (15)
C6—C5—C4	115.53 (18)	N1—C15—C20	123.29 (16)
C6—C5—H5A	108.4	N1—C15—C16	119.13 (15)
C4—C5—H5A	108.4	C20—C15—C16	117.57 (15)
C6—C5—H5B	108.4	C17—C16—C15	121.42 (17)
C4—C5—H5B	108.4	C17—C16—H16	119.3

H5A—C5—H5B	107.5	C15—C16—H16	119.3
C7—C6—C5	117.68 (19)	C16—C17—C18	120.77 (18)
C7—C6—H6A	107.9	C16—C17—H17	119.6
C5—C6—H6A	107.9	C18—C17—H17	119.6
C7—C6—H6B	107.9	C17—C18—C19	118.66 (17)
C5—C6—H6B	107.9	C17—C18—H18	120.7
H6A—C6—H6B	107.2	C19—C18—H18	120.7
C6—C7—C1	116.06 (17)	C18—C19—C20	121.50 (18)
C6—C7—H7A	108.3	C18—C19—H19	119.3
C1—C7—H7A	108.3	C20—C19—H19	119.3
C6—C7—H7B	108.3	C19—C20—C15	120.08 (17)
C1—C7—H7B	108.3	C19—C20—H20	120.0
H7A—C7—H7B	107.4	C15—C20—H20	120.0
N1—C8—C14	112.45 (14)	C15—N1—C8	124.99 (15)
N1—C8—C1	108.37 (13)	C15—N1—H1A	118.8 (12)
C14—C8—C1	112.20 (13)	C8—N1—H1A	116.2 (11)
C7—C1—C2—O1	90.59 (19)	C12—C13—C14—C9	0.4 (3)
C8—C1—C2—O1	-32.7 (2)	C12—C13—C14—C8	179.25 (17)
C7—C1—C2—C3	-85.34 (18)	C10—C9—C14—C13	-0.1 (2)
C8—C1—C2—C3	151.35 (16)	C10—C9—C14—C8	-178.96 (15)
O1—C2—C3—C4	-147.48 (19)	N1—C8—C14—C13	-125.11 (17)
C1—C2—C3—C4	28.4 (2)	C1—C8—C14—C13	112.42 (17)
C2—C3—C4—C5	50.6 (2)	N1—C8—C14—C9	53.71 (18)
C3—C4—C5—C6	-82.4 (3)	C1—C8—C14—C9	-68.77 (18)
C4—C5—C6—C7	61.6 (3)	N1—C15—C16—C17	-178.84 (17)
C5—C6—C7—C1	-56.2 (3)	C20—C15—C16—C17	0.0 (3)
C2—C1—C7—C6	78.4 (2)	C15—C16—C17—C18	0.3 (3)
C8—C1—C7—C6	-158.42 (19)	C16—C17—C18—C19	-0.3 (3)
C2—C1—C8—N1	175.01 (14)	C17—C18—C19—C20	-0.1 (3)
C7—C1—C8—N1	55.55 (19)	C18—C19—C20—C15	0.4 (3)
C2—C1—C8—C14	-60.23 (18)	N1—C15—C20—C19	178.42 (18)
C7—C1—C8—C14	-179.69 (15)	C16—C15—C20—C19	-0.4 (3)
C14—C9—C10—C11	-0.6 (3)	C20—C15—N1—C8	-7.0 (3)
C9—C10—C11—C12	1.0 (3)	C16—C15—N1—C8	171.72 (16)
C10—C11—C12—C13	-0.7 (3)	C14—C8—N1—C15	87.9 (2)
C11—C12—C13—C14	0.0 (3)	C1—C8—N1—C15	-147.48 (17)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N1—H1A...O1 ⁱ	0.879 (19)	2.23 (2)	3.065 (2)	158.4 (16)

Symmetry code: (i) *x*+1, *y*, *z*.