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Methyl *N*-hydroxy-*N*-(2-methylphenyl)-carbamate

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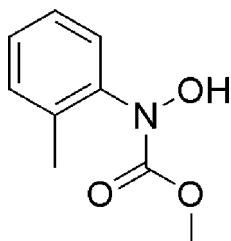
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Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.006$ Å; R factor = 0.063; wR factor = 0.161; data-to-parameter ratio = 15.6.

There are three independent molecules in the asymmetric unit of the title compound, $\text{C}_9\text{H}_{11}\text{NO}_3$, which are connected by $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds, forming an $R_3^2(15)$ ring. The dihedral angles between the planes of the benzene and amide groups are $75.16(3)$, $71.47(3)$ and $70.56(3)^\circ$. The hydroxy O atom lies $0.912(3)$, $1.172(2)$ and $1.339(2)$ Å from the mean plane of the corresponding benzene ring in the three molecules.

Related literature

The title compound is an intermediate in the synthesis of the strobilurin fungicide pyraclostrobin. For general background, see: Hou *et al.* (2002); Yang *et al.* (2012); Tao *et al.* (2009). For related structures, see: Mercader *et al.* (2011). For graph-set notation, see: Bernstein *et al.* (1995).



Experimental

Crystal data

$\text{C}_9\text{H}_{11}\text{NO}_3$
 $M_r = 181.19$
Monoclinic, $P2_1/n$
 $a = 7.6418(3)$ Å

$b = 20.8825(9)$ Å
 $c = 18.0412(9)$ Å
 $\beta = 94.485(1)^\circ$
 $V = 2870.2(2)$ Å³

$Z = 12$
Mo $K\alpha$ radiation
 $\mu = 0.10$ mm⁻¹

$T = 296$ K
 $0.54 \times 0.37 \times 0.18$ mm

Data collection

Rigaku R-Axis RAPID/ZJUG diffractometer
Absorption correction: multi-scan (ABSCOR; Higashi, 1995)
 $T_{\min} = 0.946$, $T_{\max} = 0.983$

24402 measured reflections
5643 independent reflections
3163 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.051$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.063$
 $wR(F^2) = 0.161$
 $S = 1.01$
5643 reflections

362 parameters
H-atom parameters constrained
 $\Delta\rho_{\max} = 0.41$ e Å⁻³
 $\Delta\rho_{\min} = -0.18$ e Å⁻³

Table 1
Hydrogen-bond geometry (Å, °).

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|---|-------|-------------|-------------|---------------|
| $\text{O1A}-\text{H1A}\cdots\text{O2B}$ | 0.82 | 1.94 | 2.719 (3) | 157 |
| $\text{O1B}-\text{H1B}\cdots\text{O2C}$ | 0.82 | 1.94 | 2.716 (3) | 157 |
| $\text{O1C}-\text{H1C}\cdots\text{O2A}$ | 0.82 | 1.99 | 2.757 (3) | 156 |

Data collection: *PROCESS-AUTO* (Rigaku, 2006); cell refinement: *PROCESS-AUTO*; data reduction: *CrystalStructure* (Rigaku, 2007); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012); software used to prepare material for publication: *WinGX* (Farrugia, 2012).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: PK2462).

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supporting information

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Methyl *N*-hydroxy-*N*-(2-methylphenyl)carbamate**Binbin Zhang, Yifeng Wang, Kun Dong and Danqian Xu****S1. Comment**

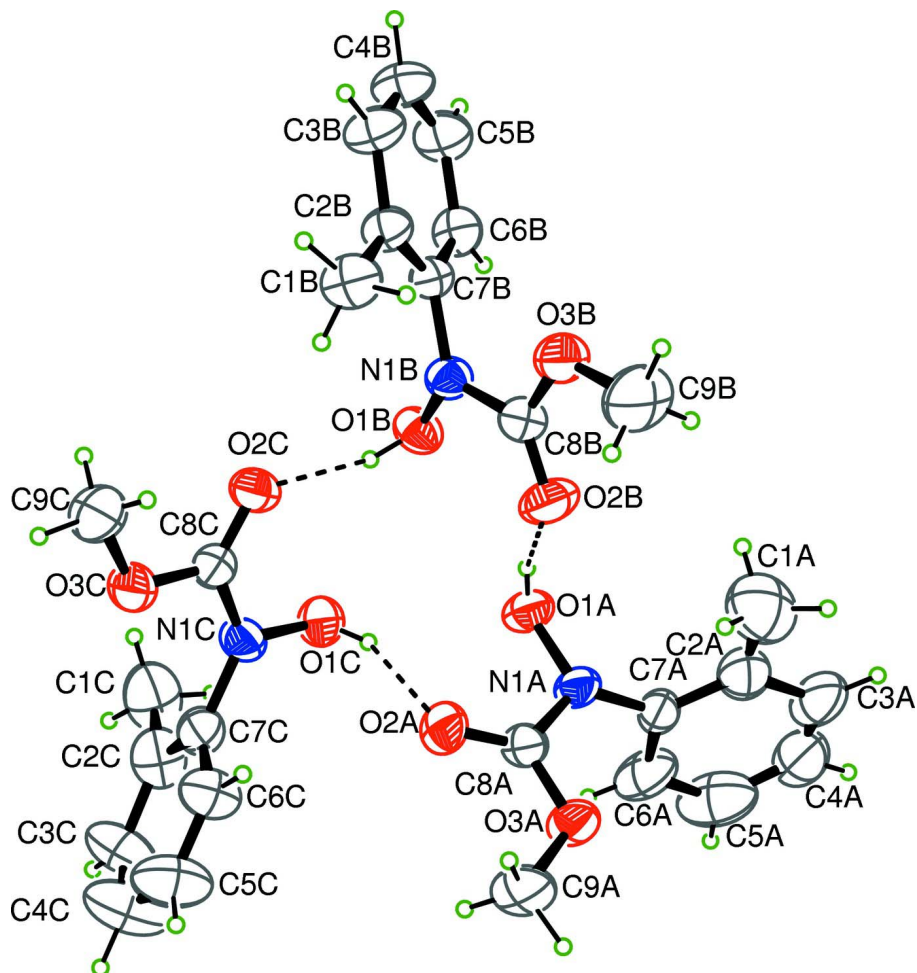
N-aryl hydroxylamines are a significant class of compounds that are key building blocks in the synthesis of natural products and biologically active compounds. The title compound, which was readily synthesized from (*N*)-(2-methylphenyl)hydroxylamine, act as an intermediate for the synthesis of Strobilurin fungicide Pyraclostrobin. In this article, the crystal structure of the title compound methyl(*N*)-hydroxy-2- methylphenylcarbamate is described (Fig. 1). There are three independent molecules in the asymmetric unit, which are connected by intermolecular O—H···O hydrogen bonds to construct a large ring involving 15 atoms with graph set notation $R^3_3(15)$ (Fig. 2). In each molecule, the dihedral angles of the plane of the phenyl ring and the plane of the amide moiety are 75.16 (3)°, 71.47 (3)°, 70.56 (3)° respectively, while the phenyl rings of the three molecules make dihedral angles of 79.87 (3)°, 71.01 (3)°, 55.86 (3)° with each other. Each hydroxyl O atom lies 0.912 (3) Å, 1.172 (2) Å and 1.339 (2) Å from the mean plane of the corresponding phenyl ring.

S2. Experimental

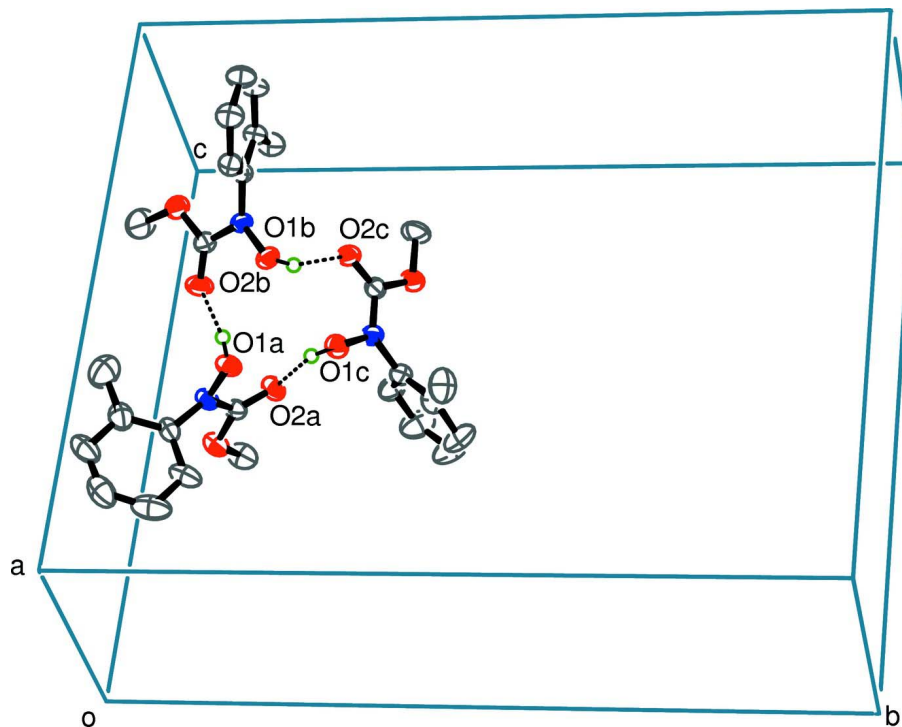
To a solution of (*N*)-(2-methylphenyl)hydroxylamine (0.022 mol) in CH₂Cl₂ (20 ml), sodium bicarbonate (0.033 mol) was added and methyl chloroformate(0.024 mol) was added dropwise, and the mixture was stirred at 0° C for 2 h (monitored by HPLC). Then the reaction mixture was filtered and distilled under vacuum, and the residue was recrystallized from petroleum ether to give the title compound. Single crystals were obtained by slow evaporation of a CH₂Cl₂ and cyclohexane solution.

S3. Refinement

H atoms were placed in calculated positions with O—H = 0.82 Å, C—H = 0.96 Å (*sp*), C—H = 0.93 Å (aromatic). All H atoms included in the final cycles of refinement using a riding model, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}$ or $1.5U_{\text{eq}}$ (*sp*³) of the carrier atoms.

**Figure 1**

The structure of the title compound. Displacement ellipsoids are drawn at the 50% probability level.

**Figure 2**

Placement of the hydrogen-bonded trimer in the unit cell.

Methyl *N*-hydroxy-*N*-(2-methylphenyl)carbamate

Crystal data

$C_9H_{11}NO_3$

$M_r = 181.19$

Monoclinic, $P2_1/n$

Hall symbol: $-P\ 2_1n$

$a = 7.6418\ (3)\ \text{\AA}$

$b = 20.8825\ (9)\ \text{\AA}$

$c = 18.0412\ (9)\ \text{\AA}$

$\beta = 94.485\ (1)^\circ$

$V = 2870.2\ (2)\ \text{\AA}^3$

$Z = 12$

$F(000) = 1152$

$D_x = 1.258\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 13549 reflections

$\theta = 3.0\text{--}27.4^\circ$

$\mu = 0.10\ \text{mm}^{-1}$

$T = 296\ \text{K}$

Needle, colorless

$0.54 \times 0.37 \times 0.18\ \text{mm}$

Data collection

Rigaku R-AXIS RAPID/ZJUG
diffractometer

Radiation source: rotating anode

Graphite monochromator

Detector resolution: $10.00\ \text{pixels mm}^{-1}$

ω scans

Absorption correction: multi-scan
(*ABSCOR*; Higashi, 1995)

$T_{\min} = 0.946$, $T_{\max} = 0.983$

24402 measured reflections

5643 independent reflections

3163 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.051$

$\theta_{\max} = 26.0^\circ$, $\theta_{\min} = 3.0^\circ$

$h = -8 \rightarrow 9$

$k = -25 \rightarrow 24$

$l = -22 \rightarrow 22$

*Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.063$ $wR(F^2) = 0.161$ $S = 1.01$

5643 reflections

362 parameters

0 restraints

Primary atom site location: structure-invariant
direct methodsSecondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0464P)^2 + 2.225P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} < 0.001$ $\Delta\rho_{\max} = 0.41 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{\min} = -0.18 \text{ e } \text{\AA}^{-3}$ Extinction correction: *SHELXL97* (Sheldrick,
2008), $F_c^* = kF_c[1 + 0.001x F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.0071 (7)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|------|------------|--------------|--------------|----------------------------------|
| C1A | 0.7015 (7) | 0.0013 (2) | 0.4220 (3) | 0.1281 (17) |
| H1A1 | 0.8160 | 0.0147 | 0.4417 | 0.192* |
| H1A2 | 0.6990 | -0.0445 | 0.4173 | 0.192* |
| H1A3 | 0.6157 | 0.0146 | 0.4550 | 0.192* |
| C2A | 0.6622 (5) | 0.02987 (19) | 0.3507 (3) | 0.0855 (11) |
| C3A | 0.6870 (5) | -0.0069 (3) | 0.2864 (3) | 0.1027 (15) |
| H3A | 0.7291 | -0.0486 | 0.2901 | 0.123* |
| C4A | 0.6481 (5) | 0.0204 (3) | 0.2205 (4) | 0.1099 (17) |
| H4A | 0.6686 | -0.0035 | 0.1785 | 0.132* |
| C5A | 0.5789 (6) | 0.0819 (3) | 0.2091 (2) | 0.1206 (18) |
| H5A | 0.5485 | 0.0977 | 0.1616 | 0.145* |
| C6A | 0.5584 (5) | 0.1173 (2) | 0.2709 (2) | 0.0930 (13) |
| H6A | 0.5189 | 0.1593 | 0.2662 | 0.112* |
| C7A | 0.5973 (3) | 0.09029 (14) | 0.34306 (18) | 0.0586 (8) |
| C8A | 0.4200 (4) | 0.15525 (14) | 0.42474 (15) | 0.0522 (7) |
| C9A | 0.1141 (4) | 0.1418 (2) | 0.4157 (2) | 0.0876 (12) |
| H9A1 | 0.1087 | 0.1362 | 0.4683 | 0.131* |
| H9A2 | 0.0244 | 0.1164 | 0.3896 | 0.131* |
| H9A3 | 0.0962 | 0.1861 | 0.4032 | 0.131* |
| N1A | 0.5733 (3) | 0.13103 (12) | 0.40482 (13) | 0.0576 (6) |
| O1A | 0.7214 (2) | 0.16837 (11) | 0.42788 (11) | 0.0633 (6) |
| H1A | 0.7555 | 0.1589 | 0.4707 | 0.095* |
| O2A | 0.4081 (3) | 0.20012 (10) | 0.46662 (12) | 0.0667 (6) |

| | | | | |
|------|------------|--------------|--------------|-------------|
| O3A | 0.2843 (2) | 0.12164 (11) | 0.39471 (12) | 0.0685 (6) |
| C1B | 0.9234 (4) | 0.19851 (18) | 0.78477 (18) | 0.0734 (9) |
| H1B1 | 0.8450 | 0.2246 | 0.7535 | 0.110* |
| H1B2 | 0.9386 | 0.2171 | 0.8335 | 0.110* |
| H1B3 | 0.8751 | 0.1563 | 0.7881 | 0.110* |
| C2B | 1.0991 (4) | 0.19463 (14) | 0.75196 (16) | 0.0552 (7) |
| C3B | 1.2551 (4) | 0.19938 (17) | 0.79671 (18) | 0.0725 (9) |
| H3B | 1.2507 | 0.2035 | 0.8479 | 0.087* |
| C4B | 1.4155 (4) | 0.19809 (18) | 0.7672 (2) | 0.0772 (10) |
| H4B | 1.5179 | 0.2008 | 0.7984 | 0.093* |
| C5B | 1.4247 (4) | 0.19279 (17) | 0.6920 (2) | 0.0735 (9) |
| H5B | 1.5333 | 0.1930 | 0.6721 | 0.088* |
| C6B | 1.2733 (4) | 0.18723 (14) | 0.64585 (17) | 0.0593 (8) |
| H6B | 1.2791 | 0.1835 | 0.5947 | 0.071* |
| C7B | 1.1122 (3) | 0.18725 (13) | 0.67609 (15) | 0.0478 (6) |
| C8B | 0.8634 (4) | 0.12549 (15) | 0.61676 (18) | 0.0586 (7) |
| C9B | 0.7983 (6) | 0.02666 (19) | 0.6715 (3) | 0.1248 (18) |
| H9B1 | 0.6769 | 0.0378 | 0.6732 | 0.187* |
| H9B2 | 0.8351 | 0.0009 | 0.7139 | 0.187* |
| H9B3 | 0.8136 | 0.0030 | 0.6268 | 0.187* |
| N1B | 0.9562 (3) | 0.18018 (11) | 0.62738 (12) | 0.0525 (6) |
| O1B | 0.9443 (3) | 0.22088 (9) | 0.56502 (10) | 0.0586 (5) |
| H1B | 0.8743 | 0.2497 | 0.5714 | 0.088* |
| O2B | 0.7543 (3) | 0.11544 (11) | 0.56597 (13) | 0.0803 (7) |
| O3B | 0.9030 (3) | 0.08435 (10) | 0.67221 (13) | 0.0744 (6) |
| C1C | 0.5743 (5) | 0.43964 (19) | 0.4292 (3) | 0.1070 (14) |
| H1C1 | 0.6516 | 0.4438 | 0.4736 | 0.161* |
| H1C2 | 0.5595 | 0.4807 | 0.4056 | 0.161* |
| H1C3 | 0.6238 | 0.4101 | 0.3958 | 0.161* |
| C2C | 0.4036 (5) | 0.41579 (18) | 0.4486 (2) | 0.0846 (11) |
| C3C | 0.2437 (6) | 0.4432 (2) | 0.4177 (2) | 0.1088 (15) |
| H3C | 0.2472 | 0.4771 | 0.3844 | 0.131* |
| C4C | 0.0889 (7) | 0.4210 (3) | 0.4356 (3) | 0.1309 (19) |
| H4C | -0.0135 | 0.4403 | 0.4152 | 0.157* |
| C5C | 0.0768 (5) | 0.3698 (3) | 0.4841 (3) | 0.1148 (16) |
| H5C | -0.0324 | 0.3549 | 0.4956 | 0.138* |
| C6C | 0.2236 (4) | 0.34211 (18) | 0.5139 (2) | 0.0800 (10) |
| H6C | 0.2175 | 0.3073 | 0.5457 | 0.096* |
| C9C | 0.5530 (5) | 0.3753 (2) | 0.71892 (19) | 0.0984 (13) |
| H9C1 | 0.5448 | 0.3326 | 0.7379 | 0.148* |
| H9C2 | 0.4735 | 0.4027 | 0.7427 | 0.148* |
| H9C3 | 0.6708 | 0.3908 | 0.7288 | 0.148* |
| C7C | 0.3899 (4) | 0.36664 (15) | 0.49641 (17) | 0.0646 (8) |
| C8C | 0.6045 (4) | 0.33602 (14) | 0.60086 (17) | 0.0547 (7) |
| N1C | 0.5431 (3) | 0.33508 (12) | 0.52944 (13) | 0.0592 (6) |
| O1C | 0.6448 (3) | 0.29974 (10) | 0.48250 (11) | 0.0627 (6) |
| H1C | 0.6012 | 0.2642 | 0.4754 | 0.094* |
| O2C | 0.7308 (3) | 0.30629 (11) | 0.62704 (11) | 0.0682 (6) |

O3C 0.5079 (3) 0.37516 (11) 0.64001 (12) 0.0741 (6)

Atomic displacement parameters (Å²)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|-------------|-------------|-------------|--------------|--------------|--------------|
| C1A | 0.134 (4) | 0.114 (4) | 0.136 (4) | 0.002 (3) | 0.007 (3) | 0.027 (3) |
| C2A | 0.063 (2) | 0.081 (3) | 0.112 (3) | -0.0024 (19) | 0.002 (2) | -0.017 (2) |
| C3A | 0.072 (3) | 0.126 (4) | 0.109 (4) | 0.005 (2) | 0.004 (2) | -0.063 (3) |
| C4A | 0.065 (3) | 0.123 (4) | 0.143 (5) | 0.003 (3) | 0.014 (3) | -0.062 (4) |
| C5A | 0.090 (3) | 0.206 (6) | 0.066 (3) | -0.009 (4) | 0.006 (2) | -0.016 (3) |
| C6A | 0.079 (2) | 0.137 (4) | 0.063 (2) | -0.012 (2) | 0.0030 (19) | -0.037 (2) |
| C7A | 0.0411 (15) | 0.0571 (18) | 0.078 (2) | -0.0017 (14) | 0.0078 (14) | -0.0214 (16) |
| C8A | 0.0461 (16) | 0.0638 (18) | 0.0470 (16) | 0.0028 (14) | 0.0054 (13) | -0.0060 (14) |
| C9A | 0.0410 (17) | 0.133 (3) | 0.091 (3) | 0.0032 (19) | 0.0174 (17) | -0.028 (2) |
| N1A | 0.0386 (12) | 0.0714 (16) | 0.0622 (15) | 0.0011 (12) | 0.0000 (11) | -0.0223 (13) |
| O1A | 0.0450 (11) | 0.0824 (15) | 0.0615 (13) | -0.0052 (10) | -0.0012 (9) | -0.0166 (11) |
| O2A | 0.0605 (13) | 0.0747 (14) | 0.0665 (14) | 0.0030 (11) | 0.0150 (10) | -0.0227 (12) |
| O3A | 0.0395 (10) | 0.0911 (15) | 0.0756 (14) | -0.0023 (11) | 0.0092 (10) | -0.0285 (12) |
| C1B | 0.0615 (19) | 0.101 (3) | 0.059 (2) | -0.0030 (18) | 0.0166 (16) | -0.0048 (18) |
| C2B | 0.0498 (16) | 0.0672 (18) | 0.0488 (17) | -0.0035 (14) | 0.0040 (13) | -0.0018 (14) |
| C3B | 0.064 (2) | 0.102 (3) | 0.0503 (19) | -0.0097 (19) | -0.0019 (15) | -0.0053 (18) |
| C4B | 0.0501 (18) | 0.105 (3) | 0.075 (2) | -0.0117 (18) | -0.0063 (17) | -0.002 (2) |
| C5B | 0.0491 (18) | 0.093 (3) | 0.080 (3) | -0.0057 (17) | 0.0119 (17) | 0.000 (2) |
| C6B | 0.0597 (18) | 0.0678 (19) | 0.0516 (18) | 0.0000 (15) | 0.0127 (15) | -0.0013 (15) |
| C7B | 0.0470 (15) | 0.0499 (15) | 0.0460 (16) | -0.0019 (12) | 0.0005 (12) | 0.0006 (12) |
| C8B | 0.0542 (17) | 0.0606 (19) | 0.060 (2) | 0.0039 (15) | -0.0015 (15) | -0.0007 (16) |
| C9B | 0.133 (4) | 0.078 (3) | 0.157 (4) | -0.046 (3) | -0.031 (3) | 0.035 (3) |
| N1B | 0.0566 (14) | 0.0528 (14) | 0.0468 (14) | -0.0014 (11) | -0.0043 (11) | 0.0066 (11) |
| O1B | 0.0638 (13) | 0.0631 (13) | 0.0490 (12) | 0.0123 (10) | 0.0050 (9) | 0.0096 (10) |
| O2B | 0.0765 (15) | 0.0873 (17) | 0.0722 (15) | -0.0163 (13) | -0.0246 (13) | -0.0017 (13) |
| O3B | 0.0747 (15) | 0.0585 (13) | 0.0865 (16) | -0.0123 (11) | -0.0162 (12) | 0.0143 (12) |
| C1C | 0.098 (3) | 0.085 (3) | 0.142 (4) | -0.010 (2) | 0.040 (3) | 0.004 (3) |
| C2C | 0.086 (3) | 0.079 (2) | 0.091 (3) | -0.004 (2) | 0.024 (2) | 0.004 (2) |
| C3C | 0.087 (3) | 0.129 (4) | 0.109 (3) | 0.042 (3) | -0.001 (3) | 0.036 (3) |
| C4C | 0.085 (3) | 0.170 (5) | 0.135 (4) | 0.029 (3) | -0.006 (3) | 0.061 (4) |
| C5C | 0.063 (2) | 0.152 (4) | 0.127 (4) | -0.007 (3) | -0.004 (2) | 0.031 (3) |
| C6C | 0.063 (2) | 0.089 (3) | 0.087 (3) | 0.0022 (19) | 0.0027 (19) | 0.016 (2) |
| C9C | 0.109 (3) | 0.127 (3) | 0.057 (2) | 0.030 (3) | -0.008 (2) | -0.035 (2) |
| C7C | 0.073 (2) | 0.0644 (19) | 0.0556 (19) | 0.0068 (17) | -0.0026 (16) | -0.0058 (16) |
| C8C | 0.0504 (17) | 0.0554 (17) | 0.0577 (19) | 0.0015 (14) | 0.0013 (14) | -0.0084 (15) |
| N1C | 0.0570 (14) | 0.0695 (16) | 0.0505 (15) | 0.0180 (12) | -0.0002 (12) | -0.0072 (12) |
| O1C | 0.0628 (13) | 0.0682 (13) | 0.0579 (13) | 0.0055 (10) | 0.0108 (10) | -0.0100 (11) |
| O2C | 0.0620 (13) | 0.0795 (14) | 0.0607 (14) | 0.0194 (12) | -0.0091 (10) | -0.0087 (11) |
| O3C | 0.0716 (14) | 0.0880 (16) | 0.0612 (14) | 0.0252 (12) | -0.0039 (11) | -0.0230 (12) |

Geometric parameters (Å, °)

| | | | |
|---------------|-----------|--------------|-----------|
| C1A—C2A | 1.429 (6) | C6B—C7B | 1.385 (4) |
| C1A—H1A1 | 0.9600 | C6B—H6B | 0.9300 |
| C1A—H1A2 | 0.9600 | C7B—N1B | 1.432 (3) |
| C1A—H1A3 | 0.9600 | C8B—O2B | 1.208 (3) |
| C2A—C7A | 1.359 (5) | C8B—O3B | 1.335 (3) |
| C2A—C3A | 1.416 (5) | C8B—N1B | 1.350 (4) |
| C3A—C4A | 1.330 (6) | C9B—O3B | 1.445 (4) |
| C3A—H3A | 0.9300 | C9B—H9B1 | 0.9600 |
| C4A—C5A | 1.398 (7) | C9B—H9B2 | 0.9600 |
| C4A—H4A | 0.9300 | C9B—H9B3 | 0.9600 |
| C5A—C6A | 1.357 (6) | N1B—O1B | 1.407 (3) |
| C5A—H5A | 0.9300 | O1B—H1B | 0.8200 |
| C6A—C7A | 1.429 (5) | C1C—C2C | 1.464 (5) |
| C6A—H6A | 0.9300 | C1C—H1C1 | 0.9600 |
| C7A—N1A | 1.425 (4) | C1C—H1C2 | 0.9600 |
| C8A—O2A | 1.212 (3) | C1C—H1C3 | 0.9600 |
| C8A—O3A | 1.332 (3) | C2C—C7C | 1.351 (5) |
| C8A—N1A | 1.350 (3) | C2C—C3C | 1.423 (5) |
| C9A—O3A | 1.445 (3) | C3C—C4C | 1.333 (6) |
| C9A—H9A1 | 0.9600 | C3C—H3C | 0.9300 |
| C9A—H9A2 | 0.9600 | C4C—C5C | 1.389 (6) |
| C9A—H9A3 | 0.9600 | C4C—H4C | 0.9300 |
| N1A—O1A | 1.410 (3) | C5C—C6C | 1.337 (5) |
| O1A—H1A | 0.8200 | C5C—H5C | 0.9300 |
| C1B—C2B | 1.511 (4) | C6C—C7C | 1.428 (5) |
| C1B—H1B1 | 0.9600 | C6C—H6C | 0.9300 |
| C1B—H1B2 | 0.9600 | C9C—O3C | 1.438 (4) |
| C1B—H1B3 | 0.9600 | C9C—H9C1 | 0.9600 |
| C2B—C7B | 1.389 (4) | C9C—H9C2 | 0.9600 |
| C2B—C3B | 1.390 (4) | C9C—H9C3 | 0.9600 |
| C3B—C4B | 1.374 (4) | C7C—N1C | 1.432 (4) |
| C3B—H3B | 0.9300 | C8C—O2C | 1.212 (3) |
| C4B—C5B | 1.369 (5) | C8C—N1C | 1.337 (4) |
| C4B—H4B | 0.9300 | C8C—O3C | 1.339 (3) |
| C5B—C6B | 1.377 (4) | N1C—O1C | 1.403 (3) |
| C5B—H5B | 0.9300 | O1C—H1C | 0.8200 |
| C2A—C1A—H1A1 | 109.5 | C5B—C6B—H6B | 120.2 |
| C2A—C1A—H1A2 | 109.5 | C7B—C6B—H6B | 120.2 |
| H1A1—C1A—H1A2 | 109.5 | C6B—C7B—C2B | 121.5 (3) |
| C2A—C1A—H1A3 | 109.5 | C6B—C7B—N1B | 118.8 (3) |
| H1A1—C1A—H1A3 | 109.5 | C2B—C7B—N1B | 119.7 (2) |
| H1A2—C1A—H1A3 | 109.5 | O2B—C8B—O3B | 124.0 (3) |
| C7A—C2A—C3A | 119.4 (4) | O2B—C8B—N1B | 125.3 (3) |
| C7A—C2A—C1A | 121.9 (4) | O3B—C8B—N1B | 110.6 (3) |
| C3A—C2A—C1A | 118.7 (4) | O3B—C9B—H9B1 | 109.5 |

| | | | |
|---------------|-----------|---------------|-----------|
| C4A—C3A—C2A | 117.7 (5) | O3B—C9B—H9B2 | 109.5 |
| C4A—C3A—H3A | 121.1 | H9B1—C9B—H9B2 | 109.5 |
| C2A—C3A—H3A | 121.1 | O3B—C9B—H9B3 | 109.5 |
| C3A—C4A—C5A | 125.5 (5) | H9B1—C9B—H9B3 | 109.5 |
| C3A—C4A—H4A | 117.3 | H9B2—C9B—H9B3 | 109.5 |
| C5A—C4A—H4A | 117.3 | C8B—N1B—O1B | 113.4 (2) |
| C6A—C5A—C4A | 116.5 (5) | C8B—N1B—C7B | 125.2 (2) |
| C6A—C5A—H5A | 121.8 | O1B—N1B—C7B | 115.3 (2) |
| C4A—C5A—H5A | 121.8 | N1B—O1B—H1B | 109.5 |
| C5A—C6A—C7A | 120.3 (5) | C8B—O3B—C9B | 115.9 (3) |
| C5A—C6A—H6A | 119.8 | C2C—C1C—H1C1 | 109.5 |
| C7A—C6A—H6A | 119.8 | C2C—C1C—H1C2 | 109.5 |
| C2A—C7A—N1A | 122.9 (3) | H1C1—C1C—H1C2 | 109.5 |
| C2A—C7A—C6A | 120.5 (3) | C2C—C1C—H1C3 | 109.5 |
| N1A—C7A—C6A | 116.5 (3) | H1C1—C1C—H1C3 | 109.5 |
| O2A—C8A—O3A | 124.5 (2) | H1C2—C1C—H1C3 | 109.5 |
| O2A—C8A—N1A | 124.4 (3) | C7C—C2C—C3C | 116.6 (4) |
| O3A—C8A—N1A | 111.0 (2) | C7C—C2C—C1C | 121.7 (4) |
| O3A—C9A—H9A1 | 109.5 | C3C—C2C—C1C | 121.6 (4) |
| O3A—C9A—H9A2 | 109.5 | C4C—C3C—C2C | 121.1 (4) |
| H9A1—C9A—H9A2 | 109.5 | C4C—C3C—H3C | 119.4 |
| O3A—C9A—H9A3 | 109.5 | C2C—C3C—H3C | 119.4 |
| H9A1—C9A—H9A3 | 109.5 | C3C—C4C—C5C | 121.6 (4) |
| H9A2—C9A—H9A3 | 109.5 | C3C—C4C—H4C | 119.2 |
| C8A—N1A—O1A | 114.0 (2) | C5C—C4C—H4C | 119.2 |
| C8A—N1A—C7A | 126.8 (2) | C6C—C5C—C4C | 119.4 (4) |
| O1A—N1A—C7A | 114.3 (2) | C6C—C5C—H5C | 120.3 |
| N1A—O1A—H1A | 109.5 | C4C—C5C—H5C | 120.3 |
| C8A—O3A—C9A | 115.3 (2) | C5C—C6C—C7C | 119.3 (4) |
| C2B—C1B—H1B1 | 109.5 | C5C—C6C—H6C | 120.4 |
| C2B—C1B—H1B2 | 109.5 | C7C—C6C—H6C | 120.4 |
| H1B1—C1B—H1B2 | 109.5 | O3C—C9C—H9C1 | 109.5 |
| C2B—C1B—H1B3 | 109.5 | O3C—C9C—H9C2 | 109.5 |
| H1B1—C1B—H1B3 | 109.5 | H9C1—C9C—H9C2 | 109.5 |
| H1B2—C1B—H1B3 | 109.5 | O3C—C9C—H9C3 | 109.5 |
| C7B—C2B—C3B | 117.1 (3) | H9C1—C9C—H9C3 | 109.5 |
| C7B—C2B—C1B | 121.8 (3) | H9C2—C9C—H9C3 | 109.5 |
| C3B—C2B—C1B | 121.1 (3) | C2C—C7C—C6C | 121.9 (3) |
| C4B—C3B—C2B | 121.6 (3) | C2C—C7C—N1C | 120.9 (3) |
| C4B—C3B—H3B | 119.2 | C6C—C7C—N1C | 117.1 (3) |
| C2B—C3B—H3B | 119.2 | O2C—C8C—N1C | 125.3 (3) |
| C5B—C4B—C3B | 120.1 (3) | O2C—C8C—O3C | 124.2 (3) |
| C5B—C4B—H4B | 119.9 | N1C—C8C—O3C | 110.5 (3) |
| C3B—C4B—H4B | 119.9 | C8C—N1C—O1C | 114.8 (2) |
| C4B—C5B—C6B | 120.0 (3) | C8C—N1C—C7C | 127.5 (2) |
| C4B—C5B—H5B | 120.0 | O1C—N1C—C7C | 117.7 (2) |
| C6B—C5B—H5B | 120.0 | N1C—O1C—H1C | 109.5 |
| C5B—C6B—C7B | 119.5 (3) | C8C—O3C—C9C | 115.1 (3) |

| | | | |
|-----------------|------------|-----------------|------------|
| C7A—C2A—C3A—C4A | 1.2 (6) | O2B—C8B—N1B—O1B | -13.3 (4) |
| C1A—C2A—C3A—C4A | 179.4 (4) | O3B—C8B—N1B—O1B | 169.4 (2) |
| C2A—C3A—C4A—C5A | -2.2 (7) | O2B—C8B—N1B—C7B | -164.4 (3) |
| C3A—C4A—C5A—C6A | 3.4 (7) | O3B—C8B—N1B—C7B | 18.4 (4) |
| C4A—C5A—C6A—C7A | -3.4 (6) | C6B—C7B—N1B—C8B | 101.2 (3) |
| C3A—C2A—C7A—N1A | -178.1 (3) | C2B—C7B—N1B—C8B | -79.5 (4) |
| C1A—C2A—C7A—N1A | 3.8 (5) | C6B—C7B—N1B—O1B | -49.4 (3) |
| C3A—C2A—C7A—C6A | -1.5 (5) | C2B—C7B—N1B—O1B | 129.9 (3) |
| C1A—C2A—C7A—C6A | -179.6 (4) | O2B—C8B—O3B—C9B | -3.8 (5) |
| C5A—C6A—C7A—C2A | 2.8 (5) | N1B—C8B—O3B—C9B | 173.5 (3) |
| C5A—C6A—C7A—N1A | 179.5 (3) | C7C—C2C—C3C—C4C | 0.1 (7) |
| O2A—C8A—N1A—O1A | -9.2 (4) | C1C—C2C—C3C—C4C | 179.9 (5) |
| O3A—C8A—N1A—O1A | 173.4 (2) | C2C—C3C—C4C—C5C | -1.1 (9) |
| O2A—C8A—N1A—C7A | -162.9 (3) | C3C—C4C—C5C—C6C | 0.4 (9) |
| O3A—C8A—N1A—C7A | 19.7 (4) | C4C—C5C—C6C—C7C | 1.2 (7) |
| C2A—C7A—N1A—C8A | -117.7 (4) | C3C—C2C—C7C—C6C | 1.5 (5) |
| C6A—C7A—N1A—C8A | 65.6 (4) | C1C—C2C—C7C—C6C | -178.3 (4) |
| C2A—C7A—N1A—O1A | 88.7 (4) | C3C—C2C—C7C—N1C | 178.9 (3) |
| C6A—C7A—N1A—O1A | -88.0 (3) | C1C—C2C—C7C—N1C | -1.0 (5) |
| O2A—C8A—O3A—C9A | -0.5 (4) | C5C—C6C—C7C—C2C | -2.2 (6) |
| N1A—C8A—O3A—C9A | 177.0 (3) | C5C—C6C—C7C—N1C | -179.6 (4) |
| C7B—C2B—C3B—C4B | 1.2 (5) | O2C—C8C—N1C—O1C | -4.7 (4) |
| C1B—C2B—C3B—C4B | -177.8 (3) | O3C—C8C—N1C—O1C | 174.7 (2) |
| C2B—C3B—C4B—C5B | 0.8 (6) | O2C—C8C—N1C—C7C | 176.6 (3) |
| C3B—C4B—C5B—C6B | -1.5 (6) | O3C—C8C—N1C—C7C | -4.0 (4) |
| C4B—C5B—C6B—C7B | 0.2 (5) | C2C—C7C—N1C—C8C | 111.8 (4) |
| C5B—C6B—C7B—C2B | 1.9 (4) | C6C—C7C—N1C—C8C | -70.7 (4) |
| C5B—C6B—C7B—N1B | -178.8 (3) | C2C—C7C—N1C—O1C | -66.8 (4) |
| C3B—C2B—C7B—C6B | -2.6 (4) | C6C—C7C—N1C—O1C | 110.6 (3) |
| C1B—C2B—C7B—C6B | 176.4 (3) | O2C—C8C—O3C—C9C | -6.0 (5) |
| C3B—C2B—C7B—N1B | 178.2 (3) | N1C—C8C—O3C—C9C | 174.6 (3) |
| C1B—C2B—C7B—N1B | -2.9 (4) | | |

Hydrogen-bond geometry (Å, °)

| <i>D</i> —H... <i>A</i> | <i>D</i> —H | H... <i>A</i> | <i>D</i> ... <i>A</i> | <i>D</i> —H... <i>A</i> |
|-------------------------|-------------|---------------|-----------------------|-------------------------|
| O1A—H1A...O2B | 0.82 | 1.94 | 2.719 (3) | 157 |
| O1B—H1B...O2C | 0.82 | 1.94 | 2.716 (3) | 157 |
| O1C—H1C...O2A | 0.82 | 1.99 | 2.757 (3) | 156 |