

# {[(1Z)-3-Chloro-1*H*-isoindol-1-ylidene]-methyl}dimethylamine

Zhao-Yun Wang\* and Chang-Jiang Yu

School of Chemistry and Material Science, Anhui Normal University, Wuhu, People's Republic of China  
Correspondence e-mail: shine\_zywang@yahoo.com.cn

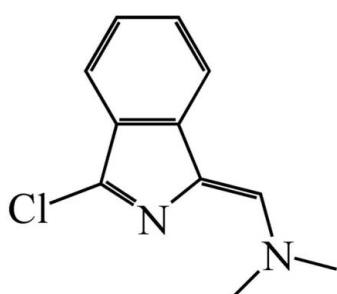
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Key indicators: single-crystal X-ray study;  $T = 293\text{ K}$ ; mean  $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$ ;  $R$  factor = 0.032;  $wR$  factor = 0.092; data-to-parameter ratio = 14.1.

The asymmetric unit of the title compound,  $\text{C}_{11}\text{H}_{11}\text{ClN}_2$ , contains two almost-planar independent molecules: the isoindole and dimethylaminomethylene mean planes in the two molecules form dihedral angles of 5.45 (8) and 1.34 (8) $^\circ$ . The crystal packing exhibits no short intermolecular contacts, except for a relatively short  $\text{Cl}\cdots\text{Cl}$  distance of 3.4907 (7)  $\text{\AA}$ .

## Related literature

For applications of related isoindole derivatives, see: Jiao *et al.* (2010). For details of the synthesis, see: von Doheneck *et al.* (1969). For the crystal structure of the related compound 4,5,6,7-tetrafluoro-1-(*N,N*-dimethylaminomethylene)-1*H*-isoindole, see: Uno *et al.* (2007).



## Experimental

### Crystal data

$\text{C}_{11}\text{H}_{11}\text{ClN}_2$	$V = 2062.6\text{ (3) \AA}^3$
$M_r = 206.67$	$Z = 8$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 12.1588\text{ (9) \AA}$	$\mu = 0.33\text{ mm}^{-1}$
$b = 10.4087\text{ (8) \AA}$	$T = 293\text{ K}$
$c = 16.3165\text{ (12) \AA}$	$0.30 \times 0.30 \times 0.20\text{ mm}$
$\beta = 92.720\text{ (1)}^\circ$	

### Data collection

Bruker SMART APEX CCD diffractometer	14393 measured reflections
Absorption correction: multi-scan ( <i>SADABS</i> ; Bruker, 2002)	3628 independent reflections
$T_{\min} = 0.908$ , $T_{\max} = 0.937$	3175 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.023$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.032$	257 parameters
$wR(F^2) = 0.092$	H-atom parameters constrained
$S = 1.02$	$\Delta\rho_{\max} = 0.22\text{ e \AA}^{-3}$
3628 reflections	$\Delta\rho_{\min} = -0.25\text{ e \AA}^{-3}$

Data collection: *SMART* (Bruker, 2002); cell refinement: *SAINT* (Bruker, 2002); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CV5388).

## References

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# supporting information

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## {[(1Z)-3-Chloro-1H-isoindol-1-ylidene]methyl}dimethylamine

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### S1. Comment

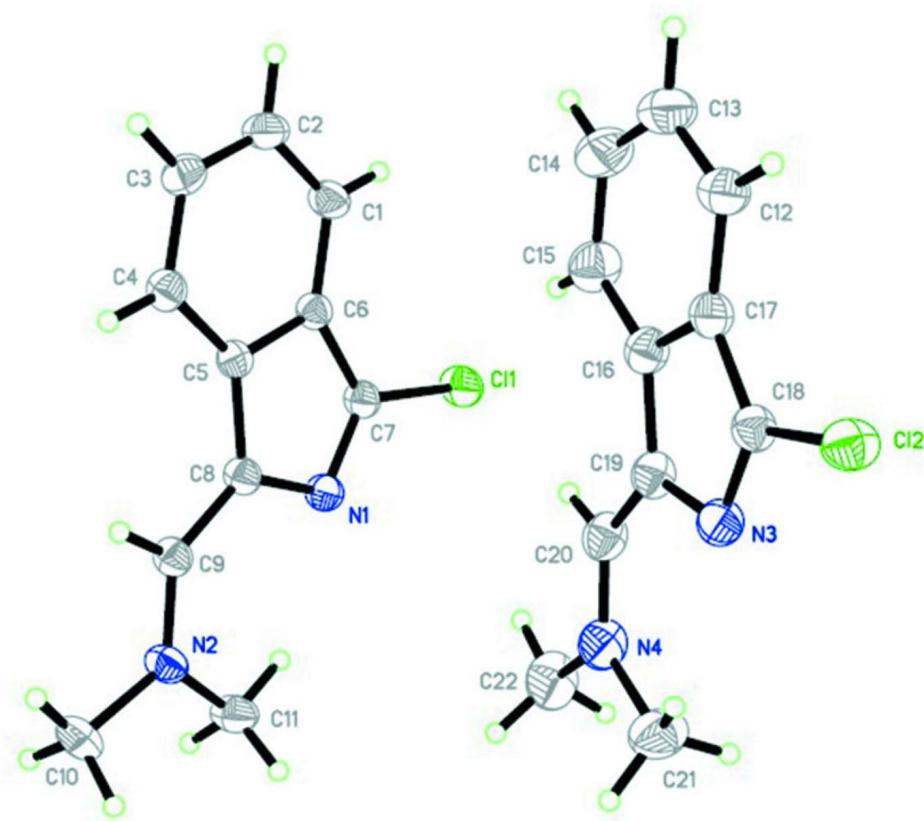
The title compound, (I), was obtained as an intermediate in our ongoing search (Jiao *et al.*, 2010) for new synthetic routes to boron-dipyrromethene (BODIPY). In (I) (Fig. 1), the bond lengths and angles are normal and correspond to those observed in the related 4,5,6,7-tetrafluoro-1-(N,N-dimethylaminomethylene)-1H-isoindole (Uno *et al.*, 2007). In the crystal structure, there are no short intermolecular contacts, except of relatively short Cl···Cl distance of 3.4907 (7) Å.

### S2. Experimental

The title compound was prepared following the known procedure (von Doheneck *et al.*, 1969). To DMF (2.4 ml, 30 mmol) in 7 ml CH<sub>2</sub>Cl<sub>2</sub> was added POCl<sub>3</sub> (2.0 ml, 20 mmol), the reaction mixture was stirred under ice-cold condition for 0.5 h. Then isoindolin-1-one (1 g, 7.5 mmol) in 10 ml CH<sub>2</sub>Cl<sub>2</sub> was added. After stirring at 333 K for 3 h, the reaction was monitored by TLC, adjusted pH-value to 8 with saturated potassium carbonate, poured into water (50 ml) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (3×30 ml). Organic layers were combined, dried over Na<sub>2</sub>SO<sub>4</sub>, and evaporated to dryness under vacuum. The desired compound was obtained as powder in 30% (280 mg) from column chromatography.

### S3. Refinement

All hydrogen atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with C—H = 0.93 - 0.97 Å and  $U_{\text{iso}}(\text{H}) = 1.2 - 1.5 U_{\text{eq}}$ .

**Figure 1**

Two independent molecules of (I) showing the atomic numbering and 30% probability displacement ellipsoids.

### {[(1*Z*)-3-Chloro-1*H*-isoindol-1-ylidene]methyl}dimethylamine

#### Crystal data

$C_{11}H_{11}ClN_2$   
 $M_r = 206.67$   
Monoclinic,  $P2_1/c$   
Hall symbol: -P 2ybc  
 $a = 12.1588 (9)$  Å  
 $b = 10.4087 (8)$  Å  
 $c = 16.3165 (12)$  Å  
 $\beta = 92.720 (1)^\circ$   
 $V = 2062.6 (3)$  Å<sup>3</sup>  
 $Z = 8$

$F(000) = 864$   
 $D_x = 1.331 \text{ Mg m}^{-3}$   
Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å  
Cell parameters from 9633 reflections  
 $\theta = 2.3\text{--}27.6^\circ$   
 $\mu = 0.33 \text{ mm}^{-1}$   
 $T = 293 \text{ K}$   
Block, colourless  
 $0.30 \times 0.30 \times 0.20 \text{ mm}$

#### Data collection

Bruker SMART APEX CCD  
diffractometer

Radiation source: fine-focus sealed tube  
Graphite monochromator

phi and  $\omega$  scans  
Absorption correction: multi-scan  
(SADABS; Bruker, 2002)  
 $T_{\min} = 0.908$ ,  $T_{\max} = 0.937$

14393 measured reflections  
3628 independent reflections  
3175 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.023$   
 $\theta_{\max} = 25.0^\circ$ ,  $\theta_{\min} = 1.7^\circ$   
 $h = -12 \rightarrow 14$   
 $k = -12 \rightarrow 12$   
 $l = -19 \rightarrow 19$

*Refinement*Refinement on  $F^2$ 

Least-squares matrix: full

$$R[F^2 > 2\sigma(F^2)] = 0.032$$

$$wR(F^2) = 0.092$$

$$S = 1.02$$

3628 reflections

257 parameters

0 restraints

Primary atom site location: structure-invariant  
direct methodsSecondary atom site location: difference Fourier  
mapHydrogen site location: inferred from  
neighbouring sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0482P)^2 + 0.6549P]$$
$$\text{where } P = (F_o^2 + 2F_c^2)/3$$

$$(\Delta/\sigma)_{\max} = 0.001$$

$$\Delta\rho_{\max} = 0.22 \text{ e \AA}^{-3}$$

$$\Delta\rho_{\min} = -0.25 \text{ e \AA}^{-3}$$

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.10344 (14)	0.95021 (15)	0.17664 (10)	0.0446 (4)
H1	0.0995	1.0308	0.1520	0.054*
C2	0.09492 (15)	0.93661 (16)	0.25997 (10)	0.0485 (4)
H2	0.0855	1.0090	0.2923	0.058*
C3	0.10025 (14)	0.81523 (16)	0.29649 (10)	0.0462 (4)
H3	0.0948	0.8086	0.3530	0.055*
C4	0.11330 (13)	0.70544 (15)	0.25124 (9)	0.0413 (4)
H4	0.1158	0.6253	0.2764	0.050*
C5	0.12269 (12)	0.71688 (14)	0.16626 (9)	0.0351 (3)
C6	0.11816 (12)	0.83963 (14)	0.13021 (9)	0.0360 (3)
C7	0.13229 (13)	0.81558 (14)	0.04517 (9)	0.0379 (3)
C8	0.13736 (12)	0.62763 (14)	0.09977 (9)	0.0363 (3)
C9	0.14062 (12)	0.49581 (14)	0.10851 (9)	0.0382 (3)
H9	0.1383	0.4670	0.1624	0.046*
C10	0.14543 (16)	0.26882 (15)	0.07727 (12)	0.0540 (4)
H10A	0.1493	0.2628	0.1360	0.081*
H10B	0.2076	0.2257	0.0557	0.081*
H10C	0.0787	0.2292	0.0560	0.081*
C11	0.14694 (16)	0.42637 (17)	-0.03484 (10)	0.0521 (4)
H11A	0.1402	0.5168	-0.0454	0.078*
H11B	0.0862	0.3819	-0.0618	0.078*
H11C	0.2147	0.3955	-0.0554	0.078*
C12	0.59631 (16)	1.03344 (17)	0.23675 (11)	0.0538 (4)
H12	0.6607	1.0464	0.2688	0.065*
C13	0.50202 (19)	1.1010 (2)	0.25156 (13)	0.0660 (5)

H13	0.5023	1.1603	0.2942	0.079*
C14	0.40623 (19)	1.0810 (2)	0.20311 (14)	0.0709 (6)
H14	0.3431	1.1266	0.2149	0.085*
C15	0.40161 (16)	0.9966 (2)	0.13874 (13)	0.0624 (5)
H15	0.3371	0.9864	0.1065	0.075*
C16	0.49652 (14)	0.92593 (16)	0.12242 (10)	0.0478 (4)
C17	0.59264 (14)	0.94515 (15)	0.17246 (10)	0.0448 (4)
C18	0.66982 (13)	0.85782 (16)	0.13977 (10)	0.0434 (4)
C19	0.52280 (14)	0.82944 (16)	0.06357 (10)	0.0471 (4)
C20	0.45005 (15)	0.78325 (18)	0.00284 (11)	0.0528 (4)
H20	0.3807	0.8210	0.0022	0.063*
C21	0.56360 (18)	0.6267 (2)	-0.06552 (14)	0.0758 (6)
H21A	0.6006	0.6639	-0.1105	0.114*
H21B	0.5476	0.5381	-0.0772	0.114*
H21C	0.6100	0.6327	-0.0164	0.114*
C22	0.37124 (18)	0.6668 (2)	-0.11346 (14)	0.0765 (7)
H22A	0.3079	0.7170	-0.1011	0.115*
H22B	0.3533	0.5771	-0.1107	0.115*
H22C	0.3931	0.6873	-0.1677	0.115*
C11	0.13909 (4)	0.93494 (4)	-0.02823 (2)	0.05378 (15)
Cl2	0.80400 (4)	0.83930 (5)	0.17864 (3)	0.05637 (15)
N1	0.14301 (11)	0.69544 (12)	0.02584 (8)	0.0394 (3)
N2	0.14649 (11)	0.40346 (12)	0.05304 (8)	0.0425 (3)
N3	0.63250 (11)	0.79025 (14)	0.07737 (8)	0.0470 (3)
N4	0.46144 (13)	0.69555 (17)	-0.05409 (10)	0.0597 (4)

*Atomic displacement parameters ( $\text{\AA}^2$ )*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0570 (10)	0.0325 (8)	0.0444 (9)	0.0018 (7)	0.0031 (7)	0.0002 (6)
C2	0.0598 (11)	0.0440 (9)	0.0418 (9)	0.0026 (8)	0.0044 (8)	-0.0093 (7)
C3	0.0534 (10)	0.0532 (10)	0.0322 (8)	-0.0002 (8)	0.0028 (7)	-0.0016 (7)
C4	0.0485 (9)	0.0403 (8)	0.0351 (8)	-0.0013 (7)	0.0000 (7)	0.0056 (6)
C5	0.0352 (8)	0.0344 (7)	0.0356 (8)	-0.0012 (6)	0.0001 (6)	0.0017 (6)
C6	0.0387 (8)	0.0338 (7)	0.0353 (8)	-0.0008 (6)	0.0011 (6)	0.0020 (6)
C7	0.0461 (9)	0.0338 (8)	0.0339 (8)	0.0001 (6)	0.0020 (6)	0.0053 (6)
C8	0.0426 (8)	0.0330 (7)	0.0330 (7)	0.0004 (6)	-0.0005 (6)	0.0029 (6)
C9	0.0422 (8)	0.0359 (8)	0.0363 (8)	0.0003 (6)	-0.0007 (6)	0.0032 (6)
C10	0.0635 (11)	0.0305 (8)	0.0678 (12)	0.0005 (8)	-0.0006 (9)	0.0020 (8)
C11	0.0700 (12)	0.0432 (9)	0.0431 (9)	0.0055 (8)	0.0020 (8)	-0.0062 (7)
C12	0.0610 (11)	0.0499 (10)	0.0507 (10)	0.0000 (8)	0.0022 (8)	-0.0042 (8)
C13	0.0795 (14)	0.0571 (12)	0.0621 (12)	0.0092 (10)	0.0101 (10)	-0.0101 (9)
C14	0.0653 (13)	0.0690 (13)	0.0794 (15)	0.0198 (10)	0.0122 (11)	-0.0055 (11)
C15	0.0512 (11)	0.0667 (12)	0.0692 (12)	0.0101 (9)	0.0008 (9)	-0.0007 (10)
C16	0.0481 (9)	0.0486 (9)	0.0468 (9)	0.0005 (7)	0.0038 (7)	0.0047 (7)
C17	0.0501 (9)	0.0416 (9)	0.0429 (9)	-0.0018 (7)	0.0047 (7)	0.0043 (7)
C18	0.0429 (9)	0.0456 (9)	0.0415 (9)	-0.0034 (7)	0.0004 (7)	0.0006 (7)
C19	0.0431 (9)	0.0531 (10)	0.0450 (9)	-0.0012 (7)	0.0005 (7)	0.0016 (7)

C20	0.0452 (10)	0.0637 (11)	0.0493 (10)	-0.0022 (8)	0.0007 (8)	0.0013 (9)
C21	0.0614 (13)	0.0909 (16)	0.0746 (14)	-0.0004 (11)	-0.0014 (11)	-0.0300 (13)
C22	0.0618 (13)	0.1050 (19)	0.0612 (13)	-0.0195 (12)	-0.0116 (10)	-0.0111 (12)
Cl1	0.0843 (3)	0.0378 (2)	0.0396 (2)	0.00008 (19)	0.0071 (2)	0.01022 (16)
Cl2	0.0481 (3)	0.0647 (3)	0.0553 (3)	0.0043 (2)	-0.00815 (19)	-0.0132 (2)
N1	0.0484 (7)	0.0355 (7)	0.0342 (7)	0.0003 (6)	0.0030 (6)	0.0010 (5)
N2	0.0530 (8)	0.0308 (6)	0.0436 (7)	0.0030 (6)	-0.0001 (6)	0.0003 (6)
N3	0.0453 (8)	0.0512 (8)	0.0445 (8)	-0.0018 (6)	0.0009 (6)	-0.0026 (6)
N4	0.0486 (9)	0.0767 (11)	0.0533 (9)	-0.0088 (8)	-0.0045 (7)	-0.0110 (8)

Geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

C1—C2	1.376 (2)	C12—C13	1.376 (3)
C1—C6	1.394 (2)	C12—C17	1.394 (2)
C1—H1	0.9300	C12—H12	0.9300
C2—C3	1.397 (2)	C13—C14	1.392 (3)
C2—H2	0.9300	C13—H13	0.9300
C3—C4	1.374 (2)	C14—C15	1.368 (3)
C3—H3	0.9300	C14—H14	0.9300
C4—C5	1.402 (2)	C15—C16	1.405 (3)
C4—H4	0.9300	C15—H15	0.9300
C5—C6	1.406 (2)	C16—C17	1.408 (2)
C5—C8	1.446 (2)	C16—C19	1.436 (2)
C6—C7	1.428 (2)	C17—C18	1.428 (2)
C7—N1	1.2977 (19)	C18—N3	1.301 (2)
C7—Cl1	1.7303 (14)	C18—Cl2	1.7328 (16)
C8—C9	1.380 (2)	C19—C20	1.383 (2)
C8—N1	1.4021 (19)	C19—N3	1.403 (2)
C9—N2	1.325 (2)	C20—N4	1.314 (2)
C9—H9	0.9300	C20—H20	0.9300
C10—N2	1.456 (2)	C21—N4	1.454 (3)
C10—H10A	0.9600	C21—H21A	0.9600
C10—H10B	0.9600	C21—H21B	0.9600
C10—H10C	0.9600	C21—H21C	0.9600
C11—N2	1.454 (2)	C22—N4	1.460 (2)
C11—H11A	0.9600	C22—H22A	0.9600
C11—H11B	0.9600	C22—H22B	0.9600
C11—H11C	0.9600	C22—H22C	0.9600
C2—C1—C6	118.01 (14)	C12—C13—H13	119.8
C2—C1—H1	121.0	C14—C13—H13	119.8
C6—C1—H1	121.0	C15—C14—C13	122.32 (19)
C1—C2—C3	120.69 (15)	C15—C14—H14	118.8
C1—C2—H2	119.7	C13—C14—H14	118.8
C3—C2—H2	119.7	C14—C15—C16	118.53 (19)
C4—C3—C2	121.82 (15)	C14—C15—H15	120.7
C4—C3—H3	119.1	C16—C15—H15	120.7
C2—C3—H3	119.1	C15—C16—C17	118.84 (17)

C3—C4—C5	118.54 (14)	C15—C16—C19	134.76 (17)
C3—C4—H4	120.7	C17—C16—C19	106.39 (15)
C5—C4—H4	120.7	C12—C17—C16	121.75 (16)
C4—C5—C6	119.17 (14)	C12—C17—C18	134.65 (17)
C4—C5—C8	134.94 (14)	C16—C17—C18	103.60 (14)
C6—C5—C8	105.89 (13)	N3—C18—C17	115.17 (15)
C1—C6—C5	121.76 (14)	N3—C18—Cl2	120.90 (13)
C1—C6—C7	134.18 (14)	C17—C18—Cl2	123.93 (13)
C5—C6—C7	104.06 (13)	C20—C19—N3	125.80 (16)
N1—C7—C6	115.03 (13)	C20—C19—C16	124.55 (16)
N1—C7—Cl1	121.00 (12)	N3—C19—C16	109.65 (14)
C6—C7—Cl1	123.95 (11)	N4—C20—C19	131.49 (18)
C9—C8—N1	125.96 (14)	N4—C20—H20	114.3
C9—C8—C5	124.43 (13)	C19—C20—H20	114.3
N1—C8—C5	109.56 (12)	N4—C21—H21A	109.5
N2—C9—C8	130.78 (14)	N4—C21—H21B	109.5
N2—C9—H9	114.6	H21A—C21—H21B	109.5
C8—C9—H9	114.6	N4—C21—H21C	109.5
N2—C10—H10A	109.5	H21A—C21—H21C	109.5
N2—C10—H10B	109.5	H21B—C21—H21C	109.5
H10A—C10—H10B	109.5	N4—C22—H22A	109.5
N2—C10—H10C	109.5	N4—C22—H22B	109.5
H10A—C10—H10C	109.5	H22A—C22—H22B	109.5
H10B—C10—H10C	109.5	N4—C22—H22C	109.5
N2—C11—H11A	109.5	H22A—C22—H22C	109.5
N2—C11—H11B	109.5	H22B—C22—H22C	109.5
H11A—C11—H11B	109.5	C7—N1—C8	105.46 (12)
N2—C11—H11C	109.5	C9—N2—C11	123.90 (13)
H11A—C11—H11C	109.5	C9—N2—C10	120.76 (14)
H11B—C11—H11C	109.5	C11—N2—C10	115.21 (13)
C13—C12—C17	118.16 (18)	C18—N3—C19	105.19 (14)
C13—C12—H12	120.9	C20—N4—C21	123.54 (16)
C17—C12—H12	120.9	C20—N4—C22	120.70 (18)
C12—C13—C14	120.38 (19)	C21—N4—C22	115.68 (17)
C6—C1—C2—C3	-0.4 (3)	C13—C12—C17—C18	-178.64 (19)
C1—C2—C3—C4	-0.5 (3)	C15—C16—C17—C12	-0.9 (3)
C2—C3—C4—C5	0.7 (3)	C19—C16—C17—C12	179.98 (15)
C3—C4—C5—C6	-0.2 (2)	C15—C16—C17—C18	178.96 (16)
C3—C4—C5—C8	179.67 (17)	C19—C16—C17—C18	-0.17 (17)
C2—C1—C6—C5	1.0 (2)	C12—C17—C18—N3	180.00 (18)
C2—C1—C6—C7	-178.50 (17)	C16—C17—C18—N3	0.18 (19)
C4—C5—C6—C1	-0.7 (2)	C12—C17—C18—Cl2	0.3 (3)
C8—C5—C6—C1	179.44 (14)	C16—C17—C18—Cl2	-179.51 (12)
C4—C5—C6—C7	178.93 (14)	C15—C16—C19—C20	1.2 (3)
C8—C5—C6—C7	-0.97 (16)	C17—C16—C19—C20	-179.85 (17)
C1—C6—C7—N1	-179.57 (17)	C15—C16—C19—N3	-178.8 (2)
C5—C6—C7—N1	0.91 (19)	C17—C16—C19—N3	0.13 (19)

C1—C6—C7—Cl1	2.2 (3)	N3—C19—C20—N4	0.8 (3)
C5—C6—C7—Cl1	−177.28 (12)	C16—C19—C20—N4	−179.20 (19)
C4—C5—C8—C9	3.3 (3)	C6—C7—N1—C8	−0.42 (19)
C6—C5—C8—C9	−176.81 (15)	Cl1—C7—N1—C8	177.84 (11)
C4—C5—C8—N1	−179.07 (16)	C9—C8—N1—C7	177.31 (16)
C6—C5—C8—N1	0.80 (17)	C5—C8—N1—C7	−0.25 (17)
N1—C8—C9—N2	−2.0 (3)	C8—C9—N2—C11	−2.3 (3)
C5—C8—C9—N2	175.21 (15)	C8—C9—N2—C10	−178.06 (16)
C17—C12—C13—C14	−0.1 (3)	C17—C18—N3—C19	−0.11 (19)
C12—C13—C14—C15	−1.2 (4)	Cl2—C18—N3—C19	179.60 (12)
C13—C14—C15—C16	1.4 (3)	C20—C19—N3—C18	179.96 (17)
C14—C15—C16—C17	−0.4 (3)	C16—C19—N3—C18	−0.02 (19)
C14—C15—C16—C19	178.4 (2)	C19—C20—N4—C21	−1.5 (3)
C13—C12—C17—C16	1.2 (3)	C19—C20—N4—C22	−178.1 (2)