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(1R*,21S*,22R*,24S*)-Methyl ethyl 2-[23-hydroxy-22,24-diphenyl-8,11,14trioxa-25-azatetracyclo[19.3.1.0^{2,7}.-0^{15,20}]pentacosa-2,4,6,15(20),16,18hexaen-25-yl]but-2-enedioate

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Key indicators: single-crystal X-ray study; T = 100 K; mean σ (C–C) = 0.002 Å; disorder in main residue; R factor = 0.048; wR factor = 0.136; data-to-parameter ratio = 21.8.

The title compound, $C_{40}H_{41}NO_8$, is a product of the reduction of the cyclic carbonyl group of the γ -piperidone subunit of the aza-14-crown-4 ether with subsequent re-esterification of its dimethyl butenoate substituent into a monoethyl monomethyl group. The azacrown macrocycle exhibits a bowl conformation with a dihedral angle of $70.82 (5)^{\circ}$ between the benzene rings fused to it. The piperidine ring adopts a chair conformation and the methyl ethyl ethylenedicarboxylate fragment has a cis conformation, with a dihedral angle of $66.51 (7)^{\circ}$ between the two carboxylate groups. The ethyl group is disordered over two sites with occupancies of 0.70 (1):0.30 (1). In the crystal, molecules form inversion dimers, via pairs of O-H···O hydrogen bonds, that stack along the *a* axis.

Related literature

For the synthesis of azacrown ethers of this type, see: Levov et al. (2006, 2008); Anh et al. (2008); Hieu et al. (2011); Khieu et al. (2011). For the structures of related compounds, see: Anh et al. (2012a,b); Hieu et al. (2012).



Experimental

Crystal data C40H41NO8 $M_r = 663.74$ Monoclinic, $P2_1/n$ a = 11.6594 (4) Å b = 19.3088 (6) Å c = 15.8522 (5) Å $\beta = 108.887 (1)^{\circ}$

Data collection

Bruker APEXII CCD diffractometer Absorption correction: multi-scan

(SADABS; Bruker, 2003) $T_{\min} = 0.984, T_{\max} = 0.989$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.048$
$wR(F^2) = 0.136$
S = 1.00
9846 reflections
452 parameters
4 restraints

Z = 4Mo $K\alpha$ radiation $\mu = 0.09 \text{ mm}^{-3}$ T = 100 K $0.18 \times 0.15 \times 0.12 \text{ mm}$

 $V = 3376.64 (19) \text{ Å}^3$

44045 measured reflections 9846 independent reflections 6852 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.041$

H atoms treated by a mixture of
independent and constrained
refinement
$\Delta \rho_{\rm max} = 0.42 \ {\rm e} \ {\rm \AA}^{-3}$
$\Delta \rho_{\rm min} = -0.48 \text{ e } \text{\AA}^{-3}$

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$O23-H23O\cdots O43^{i}$	0.82 (2)	2.39 (2)	3.1109 (14)	148 (2)
Symmetry code: (i) -r +	1 - v + 1 - z	+ 1		

Symmetry code: (i) -x + 1, -y + 1, -z + 1.

Data collection: APEX2 (Bruker, 2005); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 2008); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: RK2405).

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(1*R**,21*S**,22*R**,24*S**)-Methyl ethyl 2-[23-hydroxy-22,24-diphenyl-8,11,14trioxa-25-azatetracyclo[19.3.1.0^{2,7}.0^{15,20}]pentacosa-2,4,6,15(20),16,18hexaen-25-yl]but-2-enedioate

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S1. Comment

Recently we have developed effective methods of synthesis of azacrown ethers including piperidine (Levov *et al.*, 2006, 2008; Anh *et al.*, 2008), perhydropyrimidine (Hieu *et al.*, 2011) and perhydrotriazine subunits (Khieu *et al.*, 2011). Currently we study their structures and properties systematically (Anh *et al.*, 2012*a,b*; Hieu *et al.*, 2012). In attempt to reduce the cyclic carbonyl group of the γ -piperidone subunit into the carbinol one of the initial bis(benzo)-(β , β' -diphenyl- γ -piperidono)aza-14-crown-4 ether containing *N*-(dimethyl)maleinate fragment, we found that the expected reduction was accompanied by re-esterification of one methoxy group of the dimethyl butenoate substituent into the ethoxy one (Fig. 1). The structure of the resulting compound - the higher sterically hindered product (**I**) was unambiguously established by X-ray diffraction analysis.

The title compound **I**, $C_{40}H_{41}NO_8$, comprises the aza-14-crown-4-ether skeletal moiety and adopts a bowl conformation (Fig. 2). The configuration of the C7-O8-C9-C10-O11-C12-C13-O14-C15 polyether chain is t-g⁽⁻⁾-t-t-g⁽⁺⁾-t (t = *trans*, 180°; g = *gauche*, ±60°). The piperidine ring of the bicyclic fragment have a *chair* conformation. The dihedral angle between the planes of the benzene rings fused to the aza-14-crown-4-ether moiety is 70.82 (5)°. The phenyl rings at the C22 and C24 carbon atoms occupy the sterically favorable equatorial positions, and are rotated to each other by 65.00 (6)°. Contrary to that, the hydroxyl group at the C23 carbon atom occupies the axial position. The methyl ethyl ethylenedicarboxylate fragment has a *cis* configuration with the dihedral angle of 66.51 (7)° between the two carboxylate groups. The ethyl group is disordered over two sites with the occupancies of 0.70 (1):0.30 (1). The volume of the internal cavity of macrocycle **I** is approximately equal to 61Å³.

The molecule of **I** possesses four asymmetric centers at the C1, C21, C22 and C24 carbon atoms and can have potentially numerous diastereomers. The crystal of **I** is racemic and consists of enantiomeric pairs with the following relative configuration of the centers: rac-1R*,21S*,22R*,24S*.

In the crystal, the molecules of I form centrosymmetrical dimers by the intermolecular O23–H23···O43ⁱ hydrogen bonds (Fig. 3, Table 1). The crystal packing of the dimers is stacking along the *a* axis (Fig. 3). Symmetry code: (i) -x+1, -y+1, -z+1.

S2. Experimental

A powder of NaBH₄ (1.14 g, 30 mmol) was added to a suspension of azacrown ether (6.47 g, 10 mmol) in ethanol (50 ml). The mixture was stirred for 30 min at r.t. and then boiled for 1 h. After the solvent evaporation, the residue was washed with hot water (30 ml) and purified by re-crystallization from ethanol to give 2.27 g of colourless crystals of **I**.

Yield is 34%. M.p. = 526-528 K. IR (KBr), ν/cm^{-1} : 3436, 1714. ¹H NMR (CDCl₃, 400 MHz, 300 K): $\delta = 0,82$ (t, 3H, ³J = 5.5, OCH₂CH₃), 3.48 (s, 3H, OCH₃), 3.67 (q, 2H, ³J = 5.5, CH₂CH₃), 3.85-4.20 (m, 11H, 2×OCH₂CH₂O and H22, H23, H24), 4.37 (d, 2H, ³J = 10.5, H1, H21), 5.02 (s, 1H, OH), 6.51 (m, 3H, H_{arom}), 6.63 (s, 1H, C=CHCOO), 6.45-6.67 (m, 3H, H_{arom}), 6.88-7.15 (m, 12H, H_{arom}). Mass-spectrum (LCMS), m/*z*: 664 [*M*+1]⁺. Anal. Calcd. for C₄₀H₄₁NO₈: C, 72.38; H, 6.23; N 2.11. Found: C, 72.32; H, 6.19; N, 2.08.

S3. Refinement

The 4 distance restraints were used to fit the ideal conformations for both orientations of the disordered ethyl group. The C–C distances were fixed at 1.500 (3)Å (C41–C42, C41–C42') (two restraints). The corresponding O…C distances (O41…C42, O41…C42') were fixed at 2.420 (3)Å (two restraints). Moreover, it was taken into account that the anisotropic displacement parameters for the C42 and C42' carbon atoms of the ethyl group are equal (one restraint).

The hydrogen atoms were placed in calculated positions with C–H = 0.95-1.00Å and refined in the riding model with fixed isotropic displacement parameters [$U_{iso}(H) = 1.5U_{eq}(C)$ for the methyl groups and $U_{iso}(H) = 1.2U_{eq}(C)$ for the other groups].



Figure 1

The reaction of reduction and subsequent re-esterification of initial dimethyl 2-[bis(benzo)-(β , β' -diphenyl- γ -piperidono)aza-14-crown-4-ether]butenoate.



Figure 2

Molecular structure of **I**. Displacement ellipsoids are shown at the 50% probability level. H atoms are presented as small spheres of arbitraryradius. The alternative minor position of the disordered ethyl group is not depicted.



Figure 3

The projection of the crystal structure of I along the *a* axis demonstrating the packing of the centrosymmetrical dimers. Dashed lines indicate the intermolecular $O-H\cdots O$ hydrogen bonds.

(1*R**,21*S**,22*R**,24*S**)-Methyl ethyl 2-[23-hydroxy-22,24-diphenyl-8,11,14-trioxa-25azatetracyclo[19.3.1.0^{2,7}.0^{15,20}]pentacosa-2,4,6,15(20),16,18-hexaen-25-yl]but-2-enedioate

F(000) = 1408

 $\theta = 2.5 - 30.4^{\circ}$

 $\mu = 0.09 \text{ mm}^{-1}$

Prism, colourless

 $0.18 \times 0.15 \times 0.12$ mm

T = 100 K

 $D_{\rm x} = 1.306 {\rm Mg} {\rm m}^{-3}$

Melting point = 526–528 K Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 9431 reflections

Crystal data

C₄₀H₄₁NO₈ $M_r = 663.74$ Monoclinic, $P2_1/n$ Hall symbol: -P 2yn a = 11.6594 (4) Å b = 19.3088 (6) Å c = 15.8522 (5) Å $\beta = 108.887$ (1)° V = 3376.64 (19) Å³ Z = 4

Data collection

Bruker APEXII CCD	44045 measured reflections
diffractometer	9846 independent reflections
Radiation source: fine-focus sealed tube	6852 reflections with $I > 2\sigma(I)$
Graphite monochromator	$R_{\rm int} = 0.041$
φ and ω scans	$\theta_{\text{max}} = 30.0^{\circ}, \theta_{\text{min}} = 1.7^{\circ}$
Absorption correction: multi-scan	$h = -16 \rightarrow 16$
(SADABS; Bruker, 2003)	$k = -27 \rightarrow 27$
$T_{\min} = 0.984, \ T_{\max} = 0.989$	$l = -22 \rightarrow 22$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier
Least-squares matrix: full	map
$R[F^2 > 2\sigma(F^2)] = 0.048$	Hydrogen site location: difference Fourier map
$wR(F^2) = 0.136$	H atoms treated by a mixture of independent
S = 1.00	and constrained refinement
9846 reflections	$w = 1/[\sigma^2(F_o^2) + (0.0675P)^2 + 0.867P]$
452 parameters	where $P = (F_0^2 + 2F_c^2)/3$
4 restraints	$(\Delta/\sigma)_{\rm max} < 0.001$
Primary atom site location: structure-invariant	$\Delta \rho_{\rm max} = 0.42 \text{ e } \text{\AA}^{-3}$
direct methods	$\Delta \rho_{\rm min} = -0.48 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All s.u.'s (except the s.u. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2 , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (A^2)

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	Occ. (<1)
C1	0.49983 (11)	0.57458 (7)	0.29788 (8)	0.0224 (2)	
H1	0.5405	0.5351	0.3371	0.027*	
C2	0.53553 (12)	0.57269 (7)	0.21364 (9)	0.0252 (3)	

C3	0.61542 (13)	0.52223 (9)	0.20458 (10)	0.0342 (3)
Н3	0.6468	0.4898	0.2514	0.041*
C4	0.65123 (15)	0.51760 (11)	0.12888 (12)	0.0442 (4)
H4	0.7066	0.4827	0.1247	0.053*
C5	0.60556 (16)	0.56398 (11)	0.06046 (11)	0.0459 (4)
Н5	0.6289	0.5610	0.0085	0.055*
C6	0.52562 (15)	0.61508 (10)	0.06706 (10)	0.0404 (4)
H6	0.4945	0.6470	0.0196	0.049*
C7	0.49050 (13)	0.61998 (8)	0.14288 (9)	0.0298 (3)
08	0.41266 (10)	0.66954 (6)	0.15329 (7)	0.0355(2)
C9	0 35409 (15)	0 71378 (10)	0.07948(11)	0.0403(4)
H9A	0 3138	0.6859	0.0256	0.048*
H9B	0 4141	0 7447	0.0665	0.048*
C10	0.26254(15)	0 75544 (9)	0.10525(12)	0.0404(4)
H10A	0.3006	0 7767	0.1646	0.048*
H10R	0.2304	0.7929	0.0614	0.048*
011	0.16716 (10)	0.7929	0.0014 0.10756 (7)	0.040
C12	0.10710(10)	0.71000(0) 0.73823(0)	0.10750(7)	0.0379(3)
	0.09270(10)	0.73823 (9)	0.13413(11) 0.1147	0.0388 (4)
1112A 1112D	0.0314	0.7090	0.1147	0.047*
П12Б	0.1420 0.02124 (14)	0.7051	0.2003 0.18425 (10)	0.047°
	0.03124 (14)	0.07938 (9)	0.16433 (10)	0.0344(3)
ПІЗА 1112D	-0.0320	0.09/3	0.2072	0.041*
ПІЗБ	-0.0000	0.04//	0.1341 0.25242(7)	0.041°
014	0.12256 (9)	0.64436 (5)	0.25342 (7)	0.0313(2)
CI5	0.09638 (12)	0.58011 (7)	0.2/916 (9)	0.0258 (3)
C16	-0.02065 (13)	0.55346 (9)	0.25550 (10)	0.0333 (3)
H16	-0.0872	0.5801	0.2198	0.040*
C17	-0.03998 (14)	0.48800 (10)	0.28412 (11)	0.0393 (4)
H17	-0.1200	0.4700	0.2681	0.047*
C18	0.05563 (14)	0.44881 (9)	0.33557 (11)	0.0366 (3)
H18	0.0419	0.4041	0.3554	0.044*
C19	0.17243 (13)	0.47531 (8)	0.35823 (9)	0.0282 (3)
H19	0.2383	0.4479	0.3932	0.034*
C20	0.19565 (12)	0.54073 (7)	0.33120 (8)	0.0231 (3)
C21	0.32667 (11)	0.56407 (7)	0.35540 (8)	0.0208 (2)
H21	0.3779	0.5290	0.3974	0.025*
C22	0.35632 (12)	0.63557 (7)	0.40031 (8)	0.0221 (2)
H22	0.3200	0.6712	0.3534	0.027*
C23	0.49512 (12)	0.64603 (7)	0.43145 (8)	0.0232 (3)
H23	0.5143	0.6929	0.4594	0.028*
O23	0.55753 (9)	0.59506 (6)	0.49501 (6)	0.0281 (2)
H23O	0.5347 (18)	0.5986 (10)	0.5383 (14)	0.045 (5)*
C24	0.53900 (12)	0.64288 (7)	0.35017 (8)	0.0233 (3)
H24	0.4960	0.6809	0.3094	0.028*
N25	0.36706 (9)	0.56630 (6)	0.27489 (7)	0.0216 (2)
C26	0.30571 (12)	0.64843 (8)	0.47576 (9)	0.0271 (3)
C27	0.27685 (14)	0.71585 (9)	0.49246 (12)	0.0411 (4)
H27	0.2883	0.7525	0.4560	0.049*

C28	0 23144 (16)	0.73048(13)	0 56181 (14)	0.0586 (6)	
H28	0.23144 (10)	0.75048 (15)	0.5734	0.0380 (0)	
C29	0.2143 0.21149 (18)	0.7770	0.61346 (13)	0.070 0.0632(7)	
H20	0.1788	0.6871	0.6500	0.0052 (7)	
C30	0.1788 0.23885 (18)	0.0071 0.61087 (14)	0.05774(12)	0.0576 (6)	
H30	0.2242	0.5743	0.53774 (12)	0.0570 (0)	
C31	0.2242 0.28807 (15)	0.5745	0.0330	0.009 0.0387 (4)	
U21	0.2007 (15)	0.59595 (10)	0.53049 (10)	0.0387 (4)	
C32	0.5097 0.67345 (12)	0.5497	0.3220	0.040°	
C32	0.07343(12) 0.76284(12)	0.03300(8)	0.30900(9)	0.0271(3)	
033	0.70284 (15)	0.01124 (9)	0.42003 (10)	0.0328 (3)	
H33	0.7403	0.3/1/	0.4472	0.039^{*}	
C34	0.88485 (14)	0.62371 (10)	0.43350 (11)	0.0418 (4)	
H34	0.9446	0.5927	0.4686	0.050*	
C35	0.91940 (16)	0.68080 (12)	0.39551 (12)	0.0501 (5)	
H35	1.0027	0.6896	0.4046	0.060*	
C36	0.83257 (17)	0.72469 (11)	0.34460 (14)	0.0534 (5)	
H36	0.8558	0.7640	0.3179	0.064*	
C37	0.71051 (15)	0.71244 (9)	0.33146 (12)	0.0410 (4)	
H37	0.6514	0.7435	0.2957	0.049*	
C38	0.31867 (12)	0.51263 (7)	0.21213 (8)	0.0228 (2)	
C39	0.22377 (12)	0.52554 (8)	0.14012 (9)	0.0277 (3)	
H39	0.1989	0.5724	0.1289	0.033*	
C40	0.15353 (13)	0.47283 (9)	0.07592 (10)	0.0331 (3)	
O40	0.13153 (11)	0.47494 (7)	-0.00366 (7)	0.0477 (3)	
O41	0.11288 (10)	0.42495 (6)	0.11879 (8)	0.0414 (3)	
C41	0.02979 (17)	0.37424 (10)	0.06340 (12)	0.0507 (5)	
H41A	0.0732	0.3440	0.0333	0.061*	0.70
H41B	-0.0366	0.3980	0.0171	0.061*	0.70
H41C	0.0753	0.3329	0.0553	0.061*	0.30
H41D	-0.0125	0.3943	0.0039	0.061*	0.30
C42	-0.0209 (4)	0.33185 (17)	0.1222 (2)	0.0801 (13)	0.70
H42A	0.0446	0.3055	0.1645	0.120*	0.70
H42B	-0.0820	0.2998	0.0855	0.120*	0.70
H42C	-0.0584	0.3625	0.1550	0.120*	0.70
C42′	-0.0642(6)	0.3559 (5)	0.1058 (5)	0.0801 (13)	0.30
H42D	-0.1233	0.3936	0.0962	0.120*	0.30
H42E	-0.0249	0.3488	0.1699	0.120*	0.30
H42F	-0.1057	0 3132	0.0790	0.120*	0.30
C43	0 37713 (12)	0.44258(7)	0.23154(9)	0.0253(3)	0.00
043	0.42283(11)	0.41895 (6)	0 30529 (7)	0.0355(2)	
044	0.37842(10)	0.41068(5)	0.15658(7)	0.0331(2)	
C44	0.37642(10) 0.4458(2)	0.41000(5) 0.34731(10)	0.15050(7) 0.16785(13)	0.0531(2) 0.0523(5)	
Стт Н <i>44</i> Δ	0.4741	0.3403	0.1166	0.078*	
H44R	0.3037	0.3085	0.1721	0.078*	
	0.5957	0.3003	0.1721	0.078*	
11440	0.3137	0.3490	0.2223	0.0/0	

Atomic displacement parameters $(Å^2)$

	U^{11}	U^{22}	<i>U</i> ³³	U^{12}	U^{13}	U ²³
C1	0.0179 (5)	0.0297 (6)	0.0198 (6)	-0.0009 (5)	0.0065 (5)	0.0019 (5)
C2	0.0191 (6)	0.0362 (7)	0.0211 (6)	-0.0037 (5)	0.0075 (5)	-0.0021 (5)
C3	0.0250 (7)	0.0477 (9)	0.0305 (7)	0.0035 (6)	0.0100 (6)	-0.0024 (6)
C4	0.0306 (8)	0.0663 (12)	0.0391 (9)	0.0067 (8)	0.0162 (7)	-0.0098 (8)
C5	0.0351 (8)	0.0795 (13)	0.0285 (8)	-0.0013 (8)	0.0178 (7)	-0.0057 (8)
C6	0.0368 (8)	0.0633 (11)	0.0249 (7)	-0.0022 (8)	0.0150 (6)	0.0038 (7)
C7	0.0258 (7)	0.0423 (8)	0.0225 (6)	-0.0036 (6)	0.0093 (5)	0.0006 (6)
08	0.0399 (6)	0.0437 (6)	0.0248 (5)	0.0099 (5)	0.0131 (4)	0.0109 (4)
C9	0.0382 (8)	0.0535 (10)	0.0289 (8)	0.0028 (7)	0.0103 (6)	0.0185 (7)
C10	0.0393 (9)	0.0396 (9)	0.0391 (9)	0.0022 (7)	0.0084 (7)	0.0158 (7)
O11	0.0379 (6)	0.0427 (6)	0.0349 (6)	-0.0014 (5)	0.0141 (5)	0.0001 (5)
C12	0.0434 (9)	0.0385 (8)	0.0346 (8)	0.0119 (7)	0.0128 (7)	0.0074 (7)
C13	0.0283 (7)	0.0439 (9)	0.0287 (7)	0.0120 (6)	0.0062 (6)	0.0068 (6)
O14	0.0245 (5)	0.0342 (5)	0.0304 (5)	0.0024 (4)	0.0022 (4)	0.0063 (4)
C15	0.0219 (6)	0.0337 (7)	0.0225 (6)	0.0005 (5)	0.0079 (5)	-0.0014 (5)
C16	0.0194 (6)	0.0489 (9)	0.0306 (7)	0.0014 (6)	0.0066 (5)	0.0007 (6)
C17	0.0218 (7)	0.0586 (10)	0.0371 (8)	-0.0110 (7)	0.0089 (6)	0.0012 (7)
C18	0.0312 (8)	0.0433 (9)	0.0349 (8)	-0.0113 (6)	0.0104 (6)	0.0044 (7)
C19	0.0244 (6)	0.0345 (7)	0.0259 (6)	-0.0030 (5)	0.0084 (5)	0.0030 (5)
C20	0.0199 (6)	0.0313 (7)	0.0189 (6)	-0.0015 (5)	0.0075 (5)	-0.0013 (5)
C21	0.0191 (6)	0.0260 (6)	0.0174 (5)	0.0001 (5)	0.0062 (4)	0.0012 (5)
C22	0.0206 (6)	0.0260 (6)	0.0194 (6)	0.0007 (5)	0.0060 (5)	0.0008 (5)
C23	0.0211 (6)	0.0277 (6)	0.0200 (6)	-0.0023 (5)	0.0056 (5)	0.0011 (5)
O23	0.0226 (5)	0.0426 (6)	0.0190 (4)	0.0026 (4)	0.0068 (4)	0.0057 (4)
C24	0.0206 (6)	0.0294 (6)	0.0198 (6)	-0.0033 (5)	0.0063 (5)	0.0016 (5)
N25	0.0179 (5)	0.0291 (6)	0.0183 (5)	-0.0019 (4)	0.0064 (4)	-0.0014 (4)
C26	0.0181 (6)	0.0390 (8)	0.0226 (6)	-0.0005 (5)	0.0046 (5)	-0.0060(5)
C27	0.0292 (8)	0.0450 (9)	0.0487 (9)	0.0000(7)	0.0121 (7)	-0.0143 (8)
C28	0.0341 (9)	0.0770 (14)	0.0624 (13)	0.0076 (9)	0.0125 (9)	-0.0398 (11)
C29	0.0378 (10)	0.120 (2)	0.0341 (9)	0.0092 (11)	0.0150 (8)	-0.0224 (11)
C30	0.0445 (10)	0.1052 (18)	0.0279 (8)	0.0089 (11)	0.0183 (8)	0.0054 (10)
C31	0.0378 (8)	0.0556 (10)	0.0256 (7)	0.0054 (7)	0.0141 (6)	0.0045 (7)
C32	0.0242 (6)	0.0374 (7)	0.0201 (6)	-0.0084 (5)	0.0077 (5)	-0.0040 (5)
C33	0.0238 (7)	0.0462 (9)	0.0284 (7)	-0.0037 (6)	0.0085 (6)	-0.0019 (6)
C34	0.0243 (7)	0.0674 (12)	0.0326 (8)	-0.0023 (7)	0.0079 (6)	-0.0099 (8)
C35	0.0290 (8)	0.0812 (14)	0.0439 (10)	-0.0225 (9)	0.0169 (7)	-0.0183 (9)
C36	0.0415 (10)	0.0658 (13)	0.0567 (11)	-0.0256 (9)	0.0212 (9)	0.0018 (10)
C37	0.0351 (8)	0.0476 (10)	0.0412 (9)	-0.0124 (7)	0.0135 (7)	0.0057 (7)
C38	0.0211 (6)	0.0288 (6)	0.0201 (6)	-0.0008(5)	0.0087 (5)	-0.0018 (5)
C39	0.0243 (6)	0.0361 (7)	0.0222 (6)	0.0027 (5)	0.0068 (5)	-0.0035 (5)
C40	0.0239 (7)	0.0449 (9)	0.0267 (7)	0.0066 (6)	0.0029 (5)	-0.0089 (6)
O40	0.0460 (7)	0.0662 (8)	0.0248 (6)	0.0053 (6)	0.0029 (5)	-0.0115 (5)
O41	0.0369 (6)	0.0493 (7)	0.0343 (6)	-0.0101 (5)	0.0064 (5)	-0.0150 (5)
C41	0.0451 (10)	0.0455 (10)	0.0510 (11)	-0.0062 (8)	0.0007 (8)	-0.0205 (8)
C42	0.105 (3)	0.058 (2)	0.069(2)	-0.025(2)	0.018 (2)	-0.0062(18)

C42′	0.105 (3)	0.058 (2)	0.069 (2)	-0.025 (2)	0.018 (2)	-0.0062 (18)
C43	0.0236 (6)	0.0290 (7)	0.0240 (6)	-0.0014 (5)	0.0088 (5)	-0.0024 (5)
O43	0.0450 (6)	0.0329 (6)	0.0271 (5)	0.0062 (5)	0.0096 (5)	0.0039 (4)
O44	0.0392 (6)	0.0345 (6)	0.0277 (5)	0.0094 (5)	0.0137 (4)	-0.0024 (4)
C44	0.0783 (14)	0.0416 (9)	0.0440 (10)	0.0261 (9)	0.0296 (10)	0.0034 (8)

Geometric parameters (Å, °)

C1—N25	1.4797 (16)	C24—C32	1.5166 (18)
C1—C2	1.5225 (17)	C24—H24	1.0000
C1—C24	1.5459 (18)	N25—C38	1.4208 (17)
C1—H1	1.0000	C26—C27	1.391 (2)
C2—C3	1.387 (2)	C26—C31	1.392 (2)
C2—C7	1.409 (2)	C27—C28	1.395 (3)
C3—C4	1.395 (2)	С27—Н27	0.9500
С3—Н3	0.9500	C28—C29	1.377 (3)
C4—C5	1.374 (3)	C28—H28	0.9500
C4—H4	0.9500	C29—C30	1.367 (3)
C5—C6	1.384 (3)	С29—Н29	0.9500
С5—Н5	0.9500	C30—C31	1.395 (2)
C6—C7	1.3928 (19)	С30—Н30	0.9500
С6—Н6	0.9500	С31—Н31	0.9500
C7—O8	1.3651 (19)	C32—C37	1.387 (2)
O8—C9	1.4310 (18)	C32—C33	1.389 (2)
C9—C10	1.495 (2)	C33—C34	1.391 (2)
С9—Н9А	0.9900	С33—Н33	0.9500
С9—Н9В	0.9900	C34—C35	1.377 (3)
C10—O11	1.419 (2)	С34—Н34	0.9500
C10—H10A	0.9900	C35—C36	1.366 (3)
C10—H10B	0.9900	С35—Н35	0.9500
O11—C12	1.4127 (19)	C36—C37	1.390 (2)
C12—C13	1.500 (2)	С36—Н36	0.9500
C12—H12A	0.9900	С37—Н37	0.9500
C12—H12B	0.9900	C38—C39	1.3311 (19)
C13—O14	1.4283 (17)	C38—C43	1.5010 (19)
С13—Н13А	0.9900	C39—C40	1.484 (2)
C13—H13B	0.9900	С39—Н39	0.9500
O14—C15	1.3710 (17)	C40—O40	1.2040 (18)
C15—C16	1.3914 (19)	C40—O41	1.323 (2)
C15—C20	1.4071 (19)	O41—C41	1.455 (3)
C16—C17	1.386 (2)	C41—C42	1.498 (3)
C16—H16	0.9500	C41—C42′	1.501 (3)
C17—C18	1.375 (2)	C41—H41A	0.9900
С17—Н17	0.9500	C41—H41B	0.9900
C18—C19	1.389 (2)	C41—H41C	0.9901
C18—H18	0.9500	C41—H41D	0.9901
C19—C20	1.3885 (19)	C42—H42A	0.9800
С19—Н19	0.9500	C42—H42B	0.9800

C20—C21	1.5180 (17)	C42—H42C	0.9800
C21—N25	1.4972 (15)	C42'—H42D	0.9800
C21—C22	1.5405 (18)	C42′—H42E	0.9800
C21—H21	1.0000	C42′—H42F	0.9800
$C^{22} - C^{26}$	1 5157 (18)	$C_{43} - O_{43}$	1 2065 (17)
C_{22} C_{23}	1 5448 (18)	$C_{43} = 0.44$	1 3429 (16)
C22—C23	1.0000	$C_{43} = C_{44}$	1.3429(10) 1.4227(10)
C22—I122	1.0000		0.0900
$C_{23} = 0_{23}$	1.42/8 (10)	C44—H44A	0.9800
C23—C24	1.5337 (17)	C44—H44B	0.9800
С23—Н23	1.0000	C44—H44C	0.9800
O23—H23O	0.81 (2)		
N25—C1—C2	110.04 (10)	С23—О23—Н23О	107.8 (14)
N25—C1—C24	109.16 (10)	C32—C24—C23	115.23 (11)
C2-C1-C24	111.93 (10)	C32—C24—C1	110.76 (11)
N25—C1—H1	108.5	C23—C24—C1	111.30 (10)
C2—C1—H1	108.5	C32—C24—H24	106.3
C24—C1—H1	108.5	C23—C24—H24	106.3
C3—C2—C7	117.54 (13)	C1—C24—H24	106.3
C_{3} C_{2} C_{1}	119 41 (13)	C_{38} N25 $-C_{1}$	11345(10)
C_{7} C_{7} C_{7} C_{1}	123.05(12)	C_{38} N25 C_{21}	113.13(10) 114.38(10)
C_{1} C_{2} C_{3} C_{4}	123.03(12) 122.13(15)	C_{1} N25 C21	117.50(10)
$C_2 = C_3 = C_4$	122.13 (13)	$C_1 = N_2 = C_2 I$	112.09(10) 117.07(14)
$C_2 = C_3 = H_3$	118.9	$C_2 = C_2 = C_3 $	117.97 (14)
C4—C3—H3	118.9	$C_2/-C_{26}-C_{22}$	118.90 (14)
C5—C4—C3	119.32 (16)	C31—C26—C22	123.12 (13)
С5—С4—Н4	120.3	C26—C27—C28	121.09 (19)
C3—C4—H4	120.3	С26—С27—Н27	119.5
C4—C5—C6	120.22 (14)	C28—C27—H27	119.5
С4—С5—Н5	119.9	C29—C28—C27	119.88 (19)
С6—С5—Н5	119.9	C29—C28—H28	120.1
C5—C6—C7	120.45 (16)	C27—C28—H28	120.1
С5—С6—Н6	119.8	C30—C29—C28	119.79 (17)
С7—С6—Н6	119.8	С30—С29—Н29	120.1
08-07-06	123.00 (14)	C28—C29—H29	120.1
08-07-02	116 67 (12)	C_{29} C_{30} C_{31}	120.8(2)
C_{6} C_{7} C_{2}	120.33(14)	C_{29} C_{30} H_{30}	119.6
C7 C8 C9	120.55(14) 118 65 (12)	$C_{23} = C_{30} = H_{30}$	119.6
$C_{1} = C_{2}$	116.05(12) 106.06(12)	$C_{26} = C_{21} = C_{20}$	119.0
08 - 09 - 010	100.90 (12)	$C_{20} = C_{31} = C_{30}$	120.45 (16)
Clo Co HoA	110.3	C20—C31—H31	119.8
С10—С9—Н9А	110.3	C30—C31—H31	119.8
08—С9—Н9В	110.3	C37—C32—C33	117.44 (14)
С10—С9—Н9В	110.3	C37—C32—C24	119.15 (14)
H9A—C9—H9B	108.6	C33—C32—C24	123.33 (13)
O11—C10—C9	108.30 (14)	C32—C33—C34	121.15 (15)
O11-C10-H10A	110.0	С32—С33—Н33	119.4
С9—С10—Н10А	110.0	С34—С33—Н33	119.4
O11—C10—H10B	110.0	C35—C34—C33	120.29 (17)
C9—C10—H10B	110.0	С35—С34—Н34	119.9

H10A—C10—H10B	108.4	С33—С34—Н34	119.9
C12—O11—C10	113.76 (13)	C36—C35—C34	119.29 (15)
O11—C12—C13	108.69 (13)	С36—С35—Н35	120.4
O11—C12—H12A	110.0	С34—С35—Н35	120.4
C13—C12—H12A	110.0	C35—C36—C37	120.65 (17)
O11—C12—H12B	110.0	С35—С36—Н36	119.7
C13—C12—H12B	110.0	С37—С36—Н36	119.7
H12A—C12—H12B	108.3	C32—C37—C36	121.17 (17)
O14—C13—C12	106.82 (13)	С32—С37—Н37	119.4
O14—C13—H13A	110.4	С36—С37—Н37	119.4
С12—С13—Н13А	110.4	C39—C38—N25	119.70 (12)
O14—C13—H13B	110.4	C39—C38—C43	122.73 (12)
С12—С13—Н13В	110.4	N25—C38—C43	117.56 (11)
H13A—C13—H13B	108.6	C38—C39—C40	125.50 (14)
C15—O14—C13	118.45 (11)	С38—С39—Н39	117.3
O14—C15—C16	123.21 (13)	С40—С39—Н39	117.3
O14—C15—C20	116.32 (12)	O40—C40—O41	124.87 (15)
C16—C15—C20	120.47 (13)	O40—C40—C39	125.34 (16)
C17—C16—C15	119.88 (14)	O41—C40—C39	109.69 (12)
C17—C16—H16	120.1	C40—O41—C41	116.08 (12)
C15—C16—H16	120.1	O41—C41—C42	108.18 (15)
C18—C17—C16	120.63 (14)	O41—C41—C42′	109.19 (19)
C18—C17—H17	119.7	O41—C41—H41A	110.1
C16—C17—H17	119.7	C42—C41—H41A	110.1
C17—C18—C19	119.29 (15)	O41—C41—H41B	110.1
C17—C18—H18	120.4	C42—C41—H41B	110.1
C19—C18—H18	120.4	H41A—C41—H41B	108.4
C20—C19—C18	121.89 (14)	O41—C41—H41C	109.9
С20—С19—Н19	119.1	C42′—C41—H41C	111.5
C18—C19—H19	119.1	O41—C41—H41D	109.9
C19—C20—C15	117.84 (12)	C42′—C41—H41D	108.0
C19—C20—C21	118.23 (12)	H41C—C41—H41D	108.3
C15—C20—C21	123.86 (12)	C41—C42—H42A	109.5
N25-C21-C20	111.20 (10)	C41—C42—H42B	109.5
N25—C21—C22	106.42 (10)	C41—C42—H42C	109.5
C20—C21—C22	116.09 (11)	C41—C42′—H42D	109.5
N25—C21—H21	107.6	C41—C42′—H42E	109.5
C20—C21—H21	107.6	H42D—C42′—H42E	109.5
C22—C21—H21	107.6	C41—C42′—H42F	109.5
C26—C22—C21	115.15 (11)	H42D—C42′—H42F	109.5
C26—C22—C23	111.20 (10)	H42E—C42′—H42F	109.5
C21—C22—C23	108.66 (10)	O43—C43—O44	123.79 (13)
C26—C22—H22	107.2	O43—C43—C38	124.62 (12)
C21—C22—H22	107.2	O44—C43—C38	111.50 (11)
C23—C22—H22	107.2	C43—O44—C44	116.36 (12)
O23—C23—C24	109.46 (11)	O44—C44—H44A	109.5
O23—C23—C22	112.17 (10)	O44—C44—H44B	109.5
C24—C23—C22	109.04 (10)	H44A—C44—H44B	109.5

O23—C23—H23	108.7	O44—C44—H44C	109.5
С24—С23—Н23	108.7	H44A—C44—H44C	109.5
С22—С23—Н23	108.7	H44B—C44—H44C	109.5
N25—C1—C2—C3	-120.41 (14)	C2-C1-C24-C32	-54.16 (14)
C24—C1—C2—C3	118.02 (14)	N25—C1—C24—C23	54.20 (13)
N25—C1—C2—C7	59.18 (17)	C2-C1-C24-C23	176.27 (11)
C24—C1—C2—C7	-62.39 (16)	C2-C1-N25-C38	45.07 (14)
C7—C2—C3—C4	0.1 (2)	C24—C1—N25—C38	168.27 (10)
C1—C2—C3—C4	179.73 (14)	C2-C1-N25-C21	177.07 (10)
C2—C3—C4—C5	-0.5 (3)	C24—C1—N25—C21	-59.73 (13)
C3—C4—C5—C6	0.5 (3)	C20-C21-N25-C38	-36.60 (15)
C4—C5—C6—C7	-0.1 (3)	C22—C21—N25—C38	-163.90 (10)
C5—C6—C7—O8	179.55 (16)	C20-C21-N25-C1	-168.13 (11)
C5—C6—C7—C2	-0.3 (2)	C22—C21—N25—C1	64.56 (13)
C3—C2—C7—O8	-179.56 (13)	C21—C22—C26—C27	-150.42 (13)
C1—C2—C7—O8	0.8 (2)	C23—C22—C26—C27	85.44 (15)
C3—C2—C7—C6	0.3 (2)	C21—C22—C26—C31	30.52 (18)
C1—C2—C7—C6	-179.30(14)	C23—C22—C26—C31	-93.62 (16)
C6-C7-O8-C9	7.0 (2)	C31—C26—C27—C28	-0.1(2)
C2-C7-O8-C9	-173.11 (13)	C22—C26—C27—C28	-179.20(14)
C7—O8—C9—C10	172.22 (13)	C26—C27—C28—C29	-1.8(3)
08-C9-C10-011	-70.31(17)	C27-C28-C29-C30	1.4 (3)
C9-C10-O11-C12	163.33 (13)	$C_{28} - C_{29} - C_{30} - C_{31}$	0.7(3)
C10-011-C12-C13	-157.25(13)	C_{27} C_{26} C_{31} C_{30}	2, 2, (2)
011 - C12 - C13 - 014	70 49 (16)	C^{2} C^{2} C^{2} C^{2} C^{3} C^{3	-17871(15)
C12-C13-O14-C15	-165.89(12)	C^{29} C^{30} C^{31} C^{26}	-2.6(3)
$C_{13} - O_{14} - C_{15} - C_{16}$	-13.83(19)	C^{23} C^{24} C^{32} C^{37}	-11739(15)
$C_{13} - O_{14} - C_{15} - C_{20}$	165 50 (12)	C1 - C24 - C32 - C37	115 16 (15)
014-C15-C16-C17	-179.98(14)	C^{23} C^{24} C^{22} C^{23} C^{23}	66.03 (18)
C_{20} C_{15} C_{16} C_{17}	0.7(2)	C1 - C24 - C32 - C33	-61.42(17)
C_{15} C_{16} C_{17} C_{18}	-0.3(2)	C_{37} C_{32} C_{33} C_{34}	0.6(2)
C_{16} C_{17} C_{18} C_{19}	-0.4(3)	C_{24} C_{32} C_{33} C_{34}	177 24 (14)
C_{17} C_{18} C_{19} C_{20}	0.7(2)	C_{32} C_{33} C_{34} C_{35}	0.0(2)
C18 - C19 - C20 - C15	-0.2(2)	C_{33} C_{34} C_{35} C_{35} C_{36}	-0.5(3)
C18 - C19 - C20 - C21	-177 17 (13)	$C_{34} - C_{35} - C_{36} - C_{37}$	0.4(3)
014 - C15 - C20 - C19	-179.83(12)	C_{33} C_{32} C_{37} C_{36}	-0.7(2)
$C_{16} - C_{15} - C_{20} - C_{19}$	-0.48(19)	C_{24} C_{32} C_{37} C_{36}	-177.45(16)
014-C15-C20-C21	-3.06(18)	$C_{24} = C_{32} = C_{37} = C_{30}$	0.2(3)
$C_{16} - C_{15} - C_{20} - C_{21}$	176 28 (12)	$C_{1} = N_{2} = C_{3} = C_{3} = C_{3}$	-12949(13)
$C_{10} = C_{10} = C_{20} = C_{21}$	170.20(12) 108.80(13)	$C_1 = 1125 = C_3 = C_3$	129.49(13)
$C_{15} = C_{20} = C_{21} = N_{25}$	-67.95(16)	$C_{21} = N_{25} = C_{38} = C_{39}$	51 47 (15)
$C_{10} - C_{20} - C_{21} - C_{23}$	-120 36 (13)	$C_1 = 125 = C_{30} = C_{43}$ $C_{21} = N_25 = C_{38} = C_{43}$	-79.71(13)
$C_{13} - C_{20} - C_{21} - C_{22}$	127.30(13)	121 - 1123 - 113 - 123	-171 04 (12)
N25 C21 C22 C26	33.07(17) 171.20(10)	1123 - C30 - C39 - C40	1/1.04(13)
1123 - 021 - 022 - 020	1/1.20 (10)	$C_{+3} = C_{50} = C_{59} = C_{40}$	0.0(2) -127.60(17)
120-21-22-20	+0.07(13) -62.22(12)	$C_{30} = C_{39} = C_{40} = O_{40}$	-127.00(17)
1N23 - 0.21 - 0.22 - 0.23	-05.55(12)	0.30 - 0.39 - 0.41 - 0.41	2.0.(2)
$U_2 U = U_2 I = U_2 Z = U_2 J$	1/2.34(10)	040-040-041-041	-3.9(2)

C26—C22—C23—O23 C21—C22—C23—O23 C26—C22—C23—C24 C21—C22—C23—C24 O23—C23—C24—C32 C22—C23—C24—C32 O23—C23—C24—C1 C22—C23—C24—C1 C22—C23—C24—C1	66.75 (14) -61.00 (13) -171.84 (11) 60.42 (13) -59.67 (15) 177.29 (11) 67.51 (13) -55.53 (14) 176 (22 (10))	C39—C40—O41—C41 C40—O41—C41—C42 C40—O41—C41—C42 C39—C38—C43—O43 N25—C38—C43—O43 C39—C38—C43—O44 N25—C38—C43—O44 N25—C38—C43—O44 O43—C43—O44—C44	172.73 (13) -171.3 (2) -143.9 (5) -146.67 (15) 32.35 (19) 36.78 (18) -144.20 (11) -4.3 (2)
N25—C1—C24—C32	-176.23 (10)	C38—C43—O44—C44	172.24 (14)

Hydrogen-bond geometry (Å, °)

D—H···A	<i>D</i> —Н	H···A	D····A	D—H…A
O23—H23 <i>O</i> ···O43 ⁱ	0.82 (2)	2.39 (2)	3.1109 (14)	148 (2)

Symmetry code: (i) -x+1, -y+1, -z+1.