

5-Fluoro-3-phenyl-N'-(4-propylcyclohexylidene)-1*H*-indole-2-carbohydrazide

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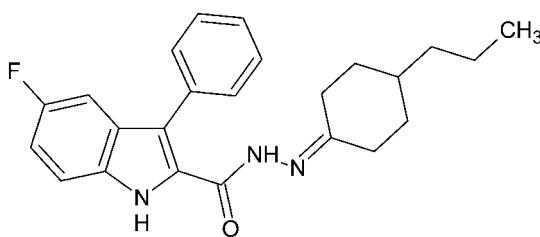
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Key indicators: single-crystal X-ray study; $T = 296\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.005\text{ \AA}$; R factor = 0.047; wR factor = 0.092; data-to-parameter ratio = 17.0.

In the title compound, $\text{C}_{24}\text{H}_{26}\text{FN}_3\text{O}$, the cyclohexane ring adopts a chair conformation; the propyl substituent is in an equatorial orientation and the bond-angle sum at the C atom bonded to the carbohydrazide N atom is 360.0° . The dihedral angle between the 1H -indole ring system and the phenyl ring is $82.77(13)^\circ$. A weak intramolecular $\text{C}-\text{H}\cdots\pi$ contact occurs. In the crystal, pairs of molecules related by a crystallographic twofold axis are linked by bifurcated $\text{N}-\text{H}\cdots(\text{O},\text{N})$ hydrogen bonds; a $\text{C}-\text{H}\cdots\text{O}$ interaction occurs between the same pair. The dimers are linked by $\text{C}-\text{H}\cdots\text{F}$ and $\text{C}-\text{H}\cdots\pi$ interactions, generating a three-dimensional network.

Related literature

For the design and synthesis of indolylhydrazones and their cyclization products, spirothiazolidinones, as potential anti-tuberculosis and anticancer agents, see: Akkurt *et al.* (2010, 2013); Cihan-Üstündağ & Çapan (2012). For puckering parameters, see: Cremer & Pople (1975).



Experimental

Crystal data

$\text{C}_{24}\text{H}_{26}\text{FN}_3\text{O}$
 $M_r = 391.48$

Tetragonal, $I\bar{4}$
 $a = 22.6986(11)\text{ \AA}$

$c = 8.4480(5)\text{ \AA}$
 $V = 4352.6(5)\text{ \AA}^3$
 $Z = 8$
Mo $K\alpha$ radiation

$\mu = 0.08\text{ mm}^{-1}$
 $T = 296\text{ K}$
 $0.63 \times 0.46 \times 0.28\text{ mm}$

Data collection

Stoe IPDS 2 diffractometer
Absorption correction: integration
(Stoe & Cie, 2002)
 $T_{\min} = 0.957$, $T_{\max} = 0.978$

15640 measured reflections
4531 independent reflections
3430 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.066$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.047$
 $wR(F^2) = 0.092$
 $S = 1.03$
4531 reflections
267 parameters
2 restraints

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.13\text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.11\text{ e \AA}^{-3}$

Table 1
Hydrogen-bond geometry (\AA , $^\circ$).

$Cg1$, $Cg2$ and $Cg3$ are the centroids of the 1H -pyrrole (N1/C1/C6/C7/C14), benzene (C1–C6) and phenyl (C8–C13) rings, respectively.

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
N1–H1 \cdots O1 ⁱ	0.86	2.50	3.048 (4)	123
N1–H1 \cdots N3 ⁱ	0.86	2.32	3.170 (3)	168
C2–H2 \cdots O1 ⁱ	0.93	2.51	3.112 (4)	123
C5–H5 \cdots F1 ⁱⁱ	0.93	2.52	3.383 (3)	154
C3–H3 \cdots Cg3 ⁱⁱⁱ	0.93	2.82	3.708 (3)	160
C11–H11 \cdots Cg1 ^{iv}	0.93	2.89	3.683 (3)	144
C17–H17A \cdots Cg2 ^v	0.97	2.81	3.571 (3)	136
C17–H17B \cdots Cg3	0.97	2.90	3.810 (4)	157

Symmetry codes: (i) $-x + \frac{3}{2}, -y + \frac{1}{2}, z + \frac{1}{2}$; (ii) $y + 1, -x + 1, -z + 1$; (iii) $-y + 1, x - 1, -z + 1$; (iv) $y + 1, -x + 1, -z$; (v) $x, y, z - 1$.

Data collection: *X-AREA* (Stoe & Cie, 2002); cell refinement: *X-AREA*; data reduction: *X-RED32* (Stoe & Cie, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012); software used to prepare material for publication: *WinGX* (Farrugia, 2012).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB7101).

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supporting information

Acta Cryst. (2013). E69, o1202 [doi:10.1107/S1600536813018333]

5-Fluoro-3-phenyl-N'-(4-propylcyclohexylidene)-1*H*-indole-2-carbohydrazide

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S1. Comment

Recently our work has been focused on the design and synthesis of novel indolylhydrazone and their cyclization products spirothiazolidinones as potential antituberculosis and anticancer agents (Akkurt *et al.*, 2010, 2013; Cihan-Üstündağ & Çapan, 2012). Within this context, we here report the synthesis and crystal structure of the title compound.

As shown in Fig. 1, the (C16–C21) cyclohexane ring of the title compound adopts a chair conformation with the puckering parameters of $Q(T) = 0.540$ (4) Å, $\theta = 176.6$ (4) ° and $\varphi = 129$ (12) °. The 1*H*-indole ring system makes a dihedral angle of 82.77 (13) ° with the C8–C13 phenyl ring. The N1–C14–C15–O1, N1–C14–C15–N2, C14–C15–N2–N3, C15–N2–N3–C16, N2–N3–C16–C17 and C19–C22–C23–C24 torsion angles are 11.3 (4), -167.4 (3), 174.2 (2), 174.8 (3), -1.1 (5) and 176.7 (5) °, respectively.

In the crystal, molecules are linked by N—H···O, N—H···N, C—H···O, C—H···F hydrogen bonds, forming layers parallel to the (110) plane (Table 1, Fig. 2). In addition, C—H···π interactions also contribute to the cohesion of the crystal packing.

S2. Experimental

A mixture of 5-fluoro-3-phenyl-1*H*-indole-2-carbohydrazide (0.005 mol) and 4-propyl cyclohexanone (0.007 mol) was refluxed in 15 ml ethanol for 5 h. The precipitate obtained was purified by recrystallization from an ethanol-water mixture to yield colourless prisms.

Yield: 78%, mp.: 446–447.5 K. IR(KBr): v_{max} 3366, 3243 (N—H), 1690 (C=O) cm⁻¹. ¹H-NMR (DMSO-d₆/500 MHz): δ 0.78–0.88 (m and t, 4H, J= 7.3 Hz, 4-CH₂CH₂CH₃, CH/CH₂-cyc.*), 1.04–1.07 (m, 1H, CH/CH₂-cyc.), 1.15 (br. d, 2H, J=6.3 Hz, 4-CH₂CH₂CH₃-cyc.), 1.27 (br. quin, 2H, 4-CH₂CH₂CH₃-cyc.), 1.47 (br. d, 2H, J=14.9 Hz, CH/CH₂-cyc.), 1.59 (d, 1H, J=12.7 Hz, CH/CH₂-cyc.), 1.70–1.88 (m, 2H, CH/CH₂-cyc.), 2.13 (s, 1H, CH/CH₂-cyc.), 2.30 (s, 1H, CH/CH₂-cyc.), 7.12 (br. t, 2H, J=8.8 Hz, H4, H6-ind.), 7.42–7.50 (m, 6H, H7, 3-C₆H₅-ind.), 9.44 (s, 1H, CONH), 12.02 (s, 1H, NH) p.p.m.. ¹³C-NMR (Proton decoupled, DMSO-d₆/100 MHz): δ 14.61 (4-CH₂CH₂CH₃-cyc.), 20.05 (4-CH₂CH₂CH₃-cyc.), 25.83 (CH₂-cyc.), 32.00 (CH₂-cyc.), 33.03 (CH₂-cyc.), 34.43 (CH₂-cyc.), 35.91 (CH-cyc.), 38.18 (4-CH₂CH₂CH₃-cyc.), 105.02 (d, J=23.7 Hz, C4-ind.), 113.17 (d, J=26.0 Hz, C6-ind.), 114.28 (C7-ind.), 117.96 (C3-ind.), 127.36 (d, C3a-ind.), 127.89 (3-C₆H₅(C4)-ind.), 129.46 (3-C₆H₅(C3,C5)-ind.), 129.99 (C2-ind.), 130.53 (3-C₆H₅(C2,C6)-ind.), 132.79** (C7a-ind.), 133.72** (3-C₆H₅(C1)-ind.), 158.03 (C=N), 158.08 (d, J=233.6 Hz, C5-ind.), 161.94 (C=O) p.p.m.. MS (APCI+) m/z(%) 392 (MH⁺, 90). Analysis calculated for C₂₄H₂₆FN₃O : C, 73.63; H, 6.69; N, 10.73%. Found: C, 73.43; H, 6.62; N, 10.51%. (*cyc.=cyclohexylidene, br.=broad, quin.=quintet, ind.=indole, ** interchangeable).

S3. Refinement

H atoms bonded to C atoms and the H atom (N1)H1 of the one of the two amide groups were positioned geometrically with C—H = 0.93 - 0.98 Å, and N—H = 0.86 Å and refined using a riding model with $U_{\text{iso}}(\text{H}) = 1.2$ or $1.5U_{\text{eq}}(\text{C}, \text{N})$. The

H atom (N2)H2A of the second groups were found in a difference Fourier map, restrained with N—H = 0.86 (2) Å and refined with $U_{\text{iso}} = 1.2U_{\text{eq}}(\text{N})$. The absolute structure was indeterminate in the present experiment.

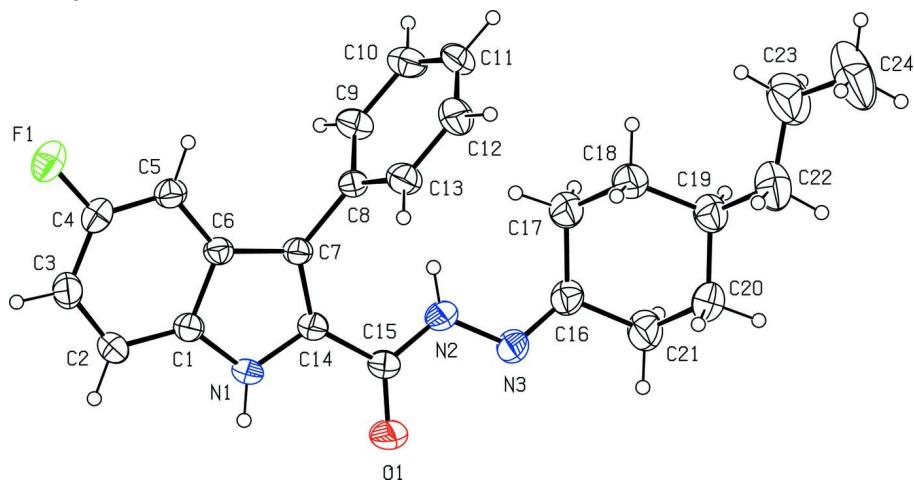


Figure 1

The molecular structure of (I) with displacement ellipsoids for non-H atoms drawn at the 30% probability level.

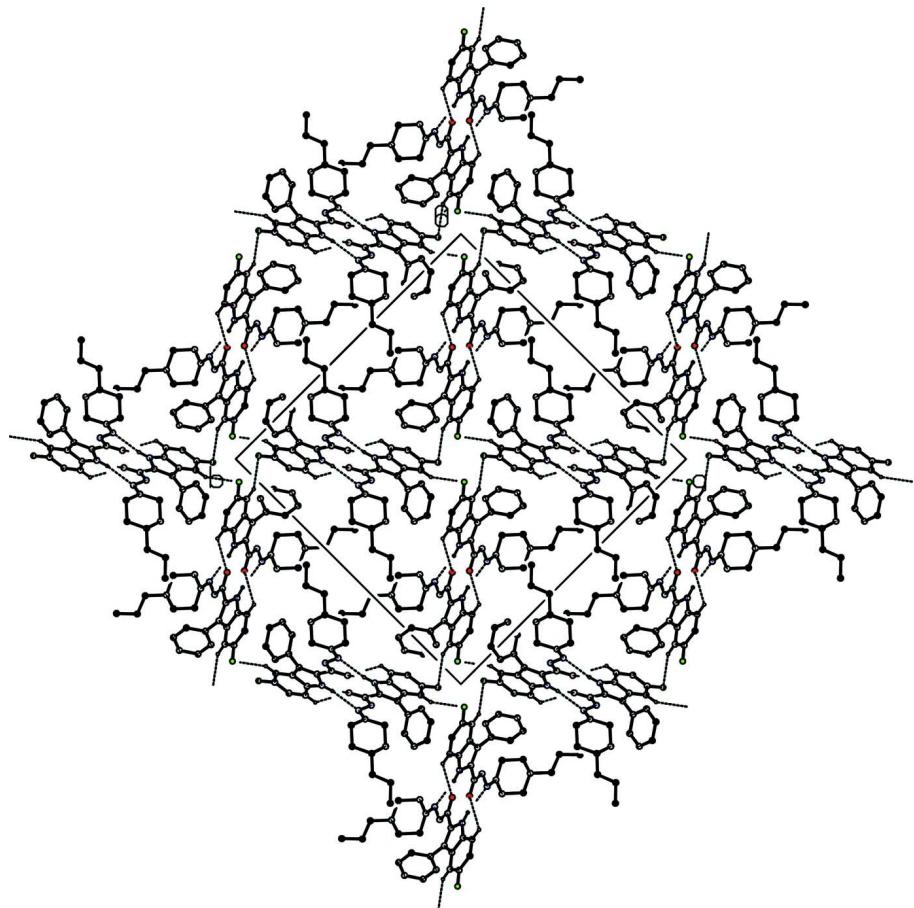


Figure 2

View of the packing and hydrogen bondings of the title compound, down the [001]-axis. H atoms not participating in hydrogen bonding have been omitted for clarity.

5-Fluoro-3-phenyl-N'-(4-propylcyclohexylidene)-1*H*-indole-2-carbohydrazide*Crystal data*

$C_{24}H_{26}FN_3O$
 $M_r = 391.48$
Tetragonal, $I\bar{4}$
Hall symbol: I -4
 $a = 22.6986 (11)$ Å
 $c = 8.4480 (5)$ Å
 $V = 4352.6 (5)$ Å³
 $Z = 8$
 $F(000) = 1664$

$D_x = 1.195$ Mg m⁻³
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 20884 reflections
 $\theta = 2.5\text{--}27.4^\circ$
 $\mu = 0.08$ mm⁻¹
 $T = 296$ K
Prism, colourless
 $0.63 \times 0.46 \times 0.28$ mm

Data collection

Stoe IPDS 2
diffractometer
Radiation source: sealed X-ray tube, 12 x 0.4
mm long-fine focus
Plane graphite monochromator
Detector resolution: 6.67 pixels mm⁻¹
 ω -scans
Absorption correction: integration
(Stoe & Cie, 2002)

$T_{\min} = 0.957$, $T_{\max} = 0.978$
15640 measured reflections
4531 independent reflections
3430 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.066$
 $\theta_{\max} = 26.5^\circ$, $\theta_{\min} = 2.5^\circ$
 $h = -28 \rightarrow 27$
 $k = -28 \rightarrow 28$
 $l = -10 \rightarrow 10$

Refinement

Refinement on F^2
Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.047$
 $wR(F^2) = 0.092$
 $S = 1.03$
4531 reflections
267 parameters
2 restraints

Hydrogen site location: mixed
H atoms treated by a mixture of independent
and constrained refinement
 $W = 1/[\Sigma^2(FO^2) + (0.0396P)^2 + 0.3772P]$
WHERE $P = (FO^2 + 2FC^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.13$ e Å⁻³
 $\Delta\rho_{\min} = -0.11$ e Å⁻³

Special details

Geometry. Bond distances, angles etc. have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted R -factors wR and all goodnesses of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The observed criterion of $F^2 > \sigma(F^2)$ is used only for calculating - R -factor-obs etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^* / U_{\text{eq}}$
F1	0.95641 (9)	0.05822 (8)	0.6774 (3)	0.0742 (7)
O1	0.77085 (10)	0.26762 (14)	0.0944 (4)	0.1013 (11)
N1	0.80598 (10)	0.18855 (10)	0.3230 (3)	0.0513 (8)
N2	0.84810 (11)	0.24602 (12)	-0.0590 (3)	0.0588 (9)
N3	0.83121 (11)	0.28546 (12)	-0.1750 (3)	0.0608 (9)
C1	0.83656 (11)	0.15385 (12)	0.4270 (4)	0.0476 (9)

C2	0.82036 (14)	0.13032 (13)	0.5733 (4)	0.0587 (10)
C3	0.86172 (14)	0.09835 (13)	0.6542 (4)	0.0605 (11)
C4	0.91761 (14)	0.09077 (12)	0.5911 (4)	0.0559 (10)
C5	0.93519 (12)	0.11312 (12)	0.4498 (4)	0.0507 (9)
C6	0.89358 (11)	0.14621 (11)	0.3643 (3)	0.0432 (8)
C7	0.89623 (11)	0.17781 (11)	0.2192 (3)	0.0428 (8)
C8	0.94986 (11)	0.18498 (11)	0.1185 (3)	0.0443 (9)
C9	0.96716 (13)	0.14270 (14)	0.0110 (4)	0.0584 (11)
C10	1.01697 (14)	0.15077 (16)	-0.0816 (4)	0.0665 (11)
C11	1.05025 (13)	0.20044 (16)	-0.0662 (4)	0.0667 (11)
C12	1.03435 (15)	0.24189 (17)	0.0407 (5)	0.0777 (14)
C13	0.98449 (13)	0.23467 (14)	0.1336 (4)	0.0634 (11)
C14	0.84140 (11)	0.20304 (11)	0.1986 (3)	0.0465 (9)
C15	0.81674 (12)	0.24211 (14)	0.0752 (4)	0.0566 (10)
C16	0.86492 (14)	0.29068 (14)	-0.2948 (4)	0.0592 (11)
C17	0.92192 (15)	0.26047 (14)	-0.3283 (4)	0.0687 (11)
C18	0.97104 (15)	0.30538 (16)	-0.3529 (4)	0.0723 (12)
C19	0.95600 (16)	0.35201 (15)	-0.4750 (4)	0.0673 (11)
C20	0.89727 (17)	0.38012 (17)	-0.4386 (5)	0.0780 (12)
C21	0.84788 (16)	0.33486 (19)	-0.4175 (5)	0.0830 (14)
C22	1.0046 (2)	0.3974 (2)	-0.4948 (5)	0.0937 (16)
C23	1.0627 (2)	0.3747 (3)	-0.5558 (8)	0.131 (3)
C24	1.1078 (3)	0.4215 (4)	-0.5818 (8)	0.169 (4)
H1	0.77000	0.19950	0.33440	0.0620*
H2	0.78280	0.13620	0.61440	0.0700*
H2A	0.8751 (11)	0.2227 (12)	-0.060 (3)	0.060 (9)*
H3	0.85230	0.08170	0.75150	0.0730*
H5	0.97300	0.10680	0.41120	0.0610*
H9	0.94520	0.10830	0.00050	0.0700*
H10	1.02770	0.12210	-0.15470	0.0800*
H11	1.08360	0.20580	-0.12860	0.0800*
H12	1.05710	0.27570	0.05200	0.0930*
H13	0.97430	0.26350	0.20680	0.0760*
H17A	0.91790	0.23630	-0.42240	0.0830*
H17B	0.93190	0.23480	-0.24040	0.0830*
H18A	0.97950	0.32450	-0.25280	0.0870*
H18B	1.00640	0.28490	-0.38620	0.0870*
H19	0.95180	0.33180	-0.57680	0.0810*
H20A	0.88690	0.40670	-0.52400	0.0930*
H20B	0.90080	0.40320	-0.34250	0.0930*
H21A	0.81190	0.35470	-0.38560	0.1000*
H21B	0.84050	0.31500	-0.51730	0.1000*
H22A	0.99070	0.42770	-0.56670	0.1130*
H22B	1.01140	0.41600	-0.39310	0.1130*
H23A	1.07820	0.34630	-0.48090	0.1570*
H23B	1.05600	0.35430	-0.65500	0.1570*
H24A	1.11610	0.44090	-0.48330	0.2030*
H24B	1.09310	0.44970	-0.65680	0.2030*

H24C	1.14330	0.40400	-0.62200	0.2030*
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Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
F1	0.0820 (13)	0.0652 (11)	0.0755 (13)	0.0059 (9)	-0.0167 (10)	0.0183 (10)
O1	0.0639 (15)	0.135 (2)	0.105 (2)	0.0507 (15)	0.0263 (15)	0.0539 (19)
N1	0.0347 (11)	0.0569 (14)	0.0623 (15)	0.0076 (10)	0.0116 (11)	0.0062 (12)
N2	0.0488 (14)	0.0670 (16)	0.0605 (16)	0.0165 (13)	-0.0026 (13)	0.0149 (14)
N3	0.0486 (13)	0.0740 (17)	0.0599 (16)	0.0073 (12)	-0.0061 (14)	0.0195 (14)
C1	0.0434 (14)	0.0435 (14)	0.0558 (17)	0.0010 (11)	0.0067 (14)	-0.0004 (14)
C2	0.0553 (17)	0.0578 (17)	0.0630 (19)	0.0022 (14)	0.0196 (16)	0.0060 (16)
C3	0.074 (2)	0.0535 (17)	0.0540 (18)	-0.0014 (15)	0.0104 (17)	0.0073 (15)
C4	0.0631 (19)	0.0438 (16)	0.0609 (19)	0.0019 (13)	-0.0076 (16)	0.0050 (15)
C5	0.0426 (14)	0.0480 (15)	0.0616 (18)	0.0028 (12)	-0.0002 (14)	0.0011 (14)
C6	0.0405 (14)	0.0390 (13)	0.0502 (15)	0.0013 (11)	0.0035 (12)	-0.0029 (12)
C7	0.0364 (13)	0.0437 (14)	0.0483 (16)	0.0026 (11)	0.0025 (12)	-0.0005 (12)
C8	0.0336 (13)	0.0507 (15)	0.0487 (17)	0.0053 (11)	-0.0003 (12)	0.0087 (13)
C9	0.0444 (16)	0.0658 (19)	0.065 (2)	-0.0009 (14)	0.0054 (15)	-0.0084 (16)
C10	0.0536 (18)	0.084 (2)	0.062 (2)	0.0113 (16)	0.0098 (16)	-0.0081 (18)
C11	0.0432 (16)	0.087 (2)	0.070 (2)	0.0038 (16)	0.0174 (16)	0.014 (2)
C12	0.060 (2)	0.066 (2)	0.107 (3)	-0.0134 (17)	0.020 (2)	0.013 (2)
C13	0.0551 (18)	0.0540 (17)	0.081 (2)	-0.0016 (14)	0.0174 (17)	-0.0034 (16)
C14	0.0390 (14)	0.0482 (15)	0.0523 (16)	0.0023 (12)	0.0038 (13)	0.0020 (13)
C15	0.0379 (14)	0.0648 (18)	0.067 (2)	0.0075 (13)	0.0065 (15)	0.0128 (16)
C16	0.0519 (17)	0.0660 (19)	0.0596 (19)	-0.0019 (15)	-0.0097 (16)	0.0107 (16)
C17	0.078 (2)	0.0631 (19)	0.065 (2)	0.0053 (16)	0.0121 (18)	0.0046 (18)
C18	0.060 (2)	0.088 (2)	0.069 (2)	0.0106 (17)	0.0040 (18)	0.015 (2)
C19	0.070 (2)	0.075 (2)	0.0570 (19)	-0.0035 (17)	0.0010 (16)	0.0060 (17)
C20	0.085 (2)	0.079 (2)	0.070 (2)	0.0109 (19)	0.002 (2)	0.024 (2)
C21	0.064 (2)	0.109 (3)	0.076 (2)	0.0032 (19)	-0.0066 (19)	0.034 (2)
C22	0.095 (3)	0.109 (3)	0.077 (2)	-0.022 (2)	-0.005 (2)	0.022 (2)
C23	0.084 (3)	0.169 (5)	0.139 (5)	-0.026 (3)	0.007 (3)	0.047 (4)
C24	0.113 (4)	0.272 (9)	0.122 (5)	-0.080 (5)	-0.013 (4)	0.074 (5)

Geometric parameters (\AA , ^\circ)

F1—C4	1.361 (4)	C19—C22	1.519 (6)
O1—C15	1.203 (4)	C19—C20	1.510 (5)
N1—C1	1.369 (4)	C20—C21	1.531 (6)
N1—C14	1.364 (4)	C22—C23	1.507 (7)
N2—N3	1.382 (4)	C23—C24	1.492 (10)
N2—C15	1.342 (4)	C2—H2	0.9300
N3—C16	1.274 (4)	C3—H3	0.9300
N1—H1	0.8600	C5—H5	0.9300
N2—H2A	0.81 (3)	C9—H9	0.9300
C1—C2	1.396 (5)	C10—H10	0.9300
C1—C6	1.409 (4)	C11—H11	0.9300

C2—C3	1.369 (4)	C12—H12	0.9300
C3—C4	1.387 (5)	C13—H13	0.9300
C4—C5	1.357 (5)	C17—H17A	0.9700
C5—C6	1.406 (4)	C17—H17B	0.9700
C6—C7	1.422 (4)	C18—H18A	0.9700
C7—C8	1.494 (4)	C18—H18B	0.9700
C7—C14	1.381 (4)	C19—H19	0.9800
C8—C9	1.378 (4)	C20—H20A	0.9700
C8—C13	1.381 (4)	C20—H20B	0.9700
C9—C10	1.387 (4)	C21—H21A	0.9700
C10—C11	1.363 (5)	C21—H21B	0.9700
C11—C12	1.353 (5)	C22—H22A	0.9700
C12—C13	1.387 (5)	C22—H22B	0.9700
C14—C15	1.479 (4)	C23—H23A	0.9700
C16—C21	1.493 (5)	C23—H23B	0.9700
C16—C17	1.491 (5)	C24—H24A	0.9600
C17—C18	1.525 (5)	C24—H24B	0.9600
C18—C19	1.517 (5)	C24—H24C	0.9600
C1—N1—C14	109.5 (2)	C4—C5—H5	122.00
N3—N2—C15	119.7 (3)	C6—C5—H5	121.00
N2—N3—C16	117.2 (3)	C8—C9—H9	120.00
C1—N1—H1	125.00	C10—C9—H9	120.00
C14—N1—H1	125.00	C9—C10—H10	120.00
N3—N2—H2A	128.8 (18)	C11—C10—H10	120.00
C15—N2—H2A	111.5 (18)	C10—C11—H11	120.00
N1—C1—C6	107.2 (3)	C12—C11—H11	120.00
C2—C1—C6	121.9 (3)	C11—C12—H12	120.00
N1—C1—C2	130.9 (3)	C13—C12—H12	120.00
C1—C2—C3	117.7 (3)	C8—C13—H13	120.00
C2—C3—C4	120.1 (3)	C12—C13—H13	120.00
F1—C4—C3	117.0 (3)	C16—C17—H17A	110.00
C3—C4—C5	124.1 (3)	C16—C17—H17B	109.00
F1—C4—C5	118.9 (3)	C18—C17—H17A	110.00
C4—C5—C6	117.0 (3)	C18—C17—H17B	110.00
C1—C6—C5	119.3 (3)	H17A—C17—H17B	108.00
C5—C6—C7	133.2 (2)	C17—C18—H18A	109.00
C1—C6—C7	107.5 (2)	C17—C18—H18B	109.00
C6—C7—C14	106.3 (2)	C19—C18—H18A	109.00
C8—C7—C14	128.1 (2)	C19—C18—H18B	109.00
C6—C7—C8	125.5 (2)	H18A—C18—H18B	108.00
C7—C8—C13	120.0 (2)	C18—C19—H19	107.00
C9—C8—C13	117.9 (3)	C20—C19—H19	107.00
C7—C8—C9	122.1 (2)	C22—C19—H19	107.00
C8—C9—C10	120.8 (3)	C19—C20—H20A	109.00
C9—C10—C11	120.5 (3)	C19—C20—H20B	109.00
C10—C11—C12	119.4 (3)	C21—C20—H20A	109.00
C11—C12—C13	120.9 (3)	C21—C20—H20B	109.00

C8—C13—C12	120.6 (3)	H20A—C20—H20B	108.00
N1—C14—C15	117.7 (2)	C16—C21—H21A	110.00
C7—C14—C15	132.7 (2)	C16—C21—H21B	110.00
N1—C14—C7	109.5 (2)	C20—C21—H21A	110.00
N2—C15—C14	115.8 (2)	C20—C21—H21B	110.00
O1—C15—N2	122.8 (3)	H21A—C21—H21B	108.00
O1—C15—C14	121.4 (3)	C19—C22—H22A	108.00
N3—C16—C21	117.3 (3)	C19—C22—H22B	108.00
C17—C16—C21	113.7 (3)	C23—C22—H22A	108.00
N3—C16—C17	129.0 (3)	C23—C22—H22B	108.00
C16—C17—C18	110.7 (3)	H22A—C22—H22B	107.00
C17—C18—C19	113.2 (3)	C22—C23—H23A	109.00
C18—C19—C20	110.8 (3)	C22—C23—H23B	109.00
C18—C19—C22	112.6 (3)	C24—C23—H23A	109.00
C20—C19—C22	112.2 (3)	C24—C23—H23B	109.00
C19—C20—C21	112.8 (3)	H23A—C23—H23B	108.00
C16—C21—C20	110.0 (3)	C23—C24—H24A	109.00
C19—C22—C23	116.2 (4)	C23—C24—H24B	109.00
C22—C23—C24	114.1 (6)	C23—C24—H24C	109.00
C1—C2—H2	121.00	H24A—C24—H24B	110.00
C3—C2—H2	121.00	H24A—C24—H24C	109.00
C2—C3—H3	120.00	H24B—C24—H24C	110.00
C4—C3—H3	120.00		
C1—N1—C14—C15	-178.2 (2)	C14—C7—C8—C13	79.6 (4)
C14—N1—C1—C2	176.9 (3)	C14—C7—C8—C9	-101.6 (3)
C14—N1—C1—C6	-0.2 (3)	C8—C7—C14—C15	2.2 (5)
C1—N1—C14—C7	0.0 (3)	C6—C7—C8—C13	-95.4 (3)
C15—N2—N3—C16	174.8 (3)	C9—C8—C13—C12	1.3 (5)
N3—N2—C15—O1	7.1 (5)	C7—C8—C13—C12	-179.9 (3)
N3—N2—C15—C14	-174.2 (2)	C7—C8—C9—C10	179.6 (3)
N2—N3—C16—C21	-177.8 (3)	C13—C8—C9—C10	-1.6 (4)
N2—N3—C16—C17	-1.1 (5)	C8—C9—C10—C11	1.0 (5)
C2—C1—C6—C7	-177.2 (3)	C9—C10—C11—C12	0.2 (5)
N1—C1—C2—C3	-177.8 (3)	C10—C11—C12—C13	-0.5 (5)
C6—C1—C2—C3	-1.0 (4)	C11—C12—C13—C8	-0.2 (5)
N1—C1—C6—C7	0.3 (3)	N1—C14—C15—N2	-167.4 (3)
C2—C1—C6—C5	1.1 (4)	C7—C14—C15—O1	-166.4 (3)
N1—C1—C6—C5	178.6 (2)	C7—C14—C15—N2	14.9 (5)
C1—C2—C3—C4	0.6 (4)	N1—C14—C15—O1	11.3 (4)
C2—C3—C4—F1	-179.6 (3)	N3—C16—C17—C18	-122.2 (4)
C2—C3—C4—C5	-0.3 (5)	C21—C16—C17—C18	54.6 (4)
C3—C4—C5—C6	0.4 (4)	N3—C16—C21—C20	121.6 (3)
F1—C4—C5—C6	179.7 (2)	C17—C16—C21—C20	-55.6 (4)
C4—C5—C6—C7	177.0 (3)	C16—C17—C18—C19	-52.3 (4)
C4—C5—C6—C1	-0.8 (4)	C17—C18—C19—C20	51.8 (4)
C5—C6—C7—C8	-2.3 (5)	C17—C18—C19—C22	178.4 (3)
C1—C6—C7—C14	-0.2 (3)	C18—C19—C20—C21	-52.9 (4)

C5—C6—C7—C14	−178.2 (3)	C22—C19—C20—C21	−179.7 (3)
C1—C6—C7—C8	175.7 (2)	C18—C19—C22—C23	63.2 (5)
C6—C7—C14—N1	0.1 (3)	C20—C19—C22—C23	−170.9 (4)
C6—C7—C8—C9	83.4 (4)	C19—C20—C21—C16	54.5 (4)
C6—C7—C14—C15	178.0 (3)	C19—C22—C23—C24	176.7 (5)
C8—C7—C14—N1	−175.6 (2)		

Hydrogen-bond geometry (Å, °)

Cg1, Cg2 and Cg3 are the centroids of the 1*H*-pyrrole (N1/C1/C6/C7/C14), benzene (C1—C6) and phenyl (C8—C13) rings, respectively.

D—H···A	D—H	H···A	D···A	D—H···A
N1—H1···O1 ⁱ	0.86	2.50	3.048 (4)	123
N1—H1···N3 ⁱ	0.86	2.32	3.170 (3)	168
C2—H2···O1 ⁱ	0.93	2.51	3.112 (4)	123
C5—H5···F1 ⁱⁱ	0.93	2.52	3.383 (3)	154
C3—H3···Cg3 ⁱⁱⁱ	0.93	2.82	3.708 (3)	160
C11—H11···Cg1 ^{iv}	0.93	2.89	3.683 (3)	144
C17—H17A···Cg2 ^v	0.97	2.81	3.571 (3)	136
C17—H17B···Cg3	0.97	2.90	3.810 (4)	157

Symmetry codes: (i) $-x+3/2, -y+1/2, z+1/2$; (ii) $y+1, -x+1, -z+1$; (iii) $-y+1, x-1, -z+1$; (iv) $y+1, -x+1, -z$; (v) $x, y, z-1$.