

[5,10,15,20-Tetrakis(4-methoxyphenyl)-porphyrinato]zinc dichloromethane disolvate

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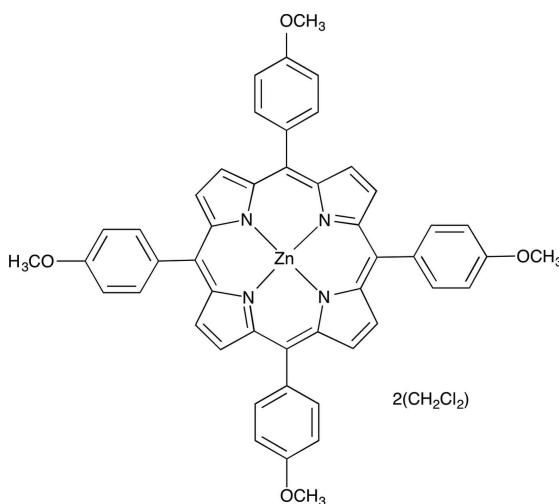
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Key indicators: single-crystal X-ray study; $T = 100\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$; R factor = 0.039; wR factor = 0.121; data-to-parameter ratio = 16.5.

In the title compound, $[\text{Zn}(\text{C}_{48}\text{H}_{36}\text{N}_4\text{O}_4)] \cdot 2\text{CH}_2\text{Cl}_2$, the Zn^{II} ion lies on an inversion center and is coordinated in an almost ideal square-planar geometry. The asymmetric unit also contains one dichloromethane solvent molecule. The unique methoxy-substituted benzene rings form dihedral angles of 59.38 (6) and 66.77 (6) $^\circ$ with the mean plane (r.m.s. deviation of fitted atoms = 0.0282 \AA) of the atoms in the porphyrin core. The packing is characterized by close contacts between the Zn^{II} ion and two symmetry-related molecules through the O atoms of a methoxyphenyl group [$\text{Zn}\cdots\text{O} = 2.694(2)\text{ \AA}$], forming a two-dimensional network parallel to (100).

Related literature

For related structures, see: Adilov & Thalladi (2007); Bhuyan & Sarkar (2011); Teo *et al.* (2003). For the synthesis, see: Adler *et al.* (1967). For van der Waals radii, see: Bondi (1964).



Experimental

Crystal data



$M_r = 968.03$

Monoclinic, $P2_1/c$

$a = 11.4189(9)\text{ \AA}$

$b = 10.6877(9)\text{ \AA}$

$c = 18.3778(15)\text{ \AA}$

$\beta = 106.022(1)^\circ$

$V = 2155.7(3)\text{ \AA}^3$

$Z = 2$

Mo $K\alpha$ radiation

$\mu = 0.87\text{ mm}^{-1}$

$T = 100\text{ K}$

$0.18 \times 0.16 \times 0.09\text{ mm}$

Data collection

Bruker APEXII CCD diffractometer

Absorption correction: multi-scan (*SADABS*; Bruker, 2007)

$T_{\min} = 0.861$, $T_{\max} = 0.926$

25964 measured reflections

4763 independent reflections

4022 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.036$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.039$

$wR(F^2) = 0.121$

$S = 1.06$

4763 reflections

288 parameters

H-atom parameters constrained

$\Delta\rho_{\max} = 0.52\text{ e \AA}^{-3}$

$\Delta\rho_{\min} = -0.78\text{ e \AA}^{-3}$

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

This research was funded in part by the National Science Foundation (CHE-0924153) and a chemistry department grant from the Welch Foundation (AD-0007). X-ray data were collected at the University of North Texas using a Bruker APEXII CCD diffractometer.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH5631).

References

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supporting information

Acta Cryst. (2013). E69, m471 [doi:10.1107/S1600536813019338]

[5,10,15,20-Tetrakis(4-methoxyphenyl)porphyrinato]zinc dichloromethane disolvate

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S1. Comment

While in pursuit of understanding zinc porphyrin coordination to alkyl alcohol solvents through the central zinc atom, we sought to crystallize zinc 5,10,15,20-tetrakis(*p*-methoxyphenyl)porphyrin with octanol. The resulting deep red crystals did not contain any octanol coordination, but contained a well ordered porphyrin structure with dichloromethane solvent molecules. The crystal structure of the title compound is presented herein.

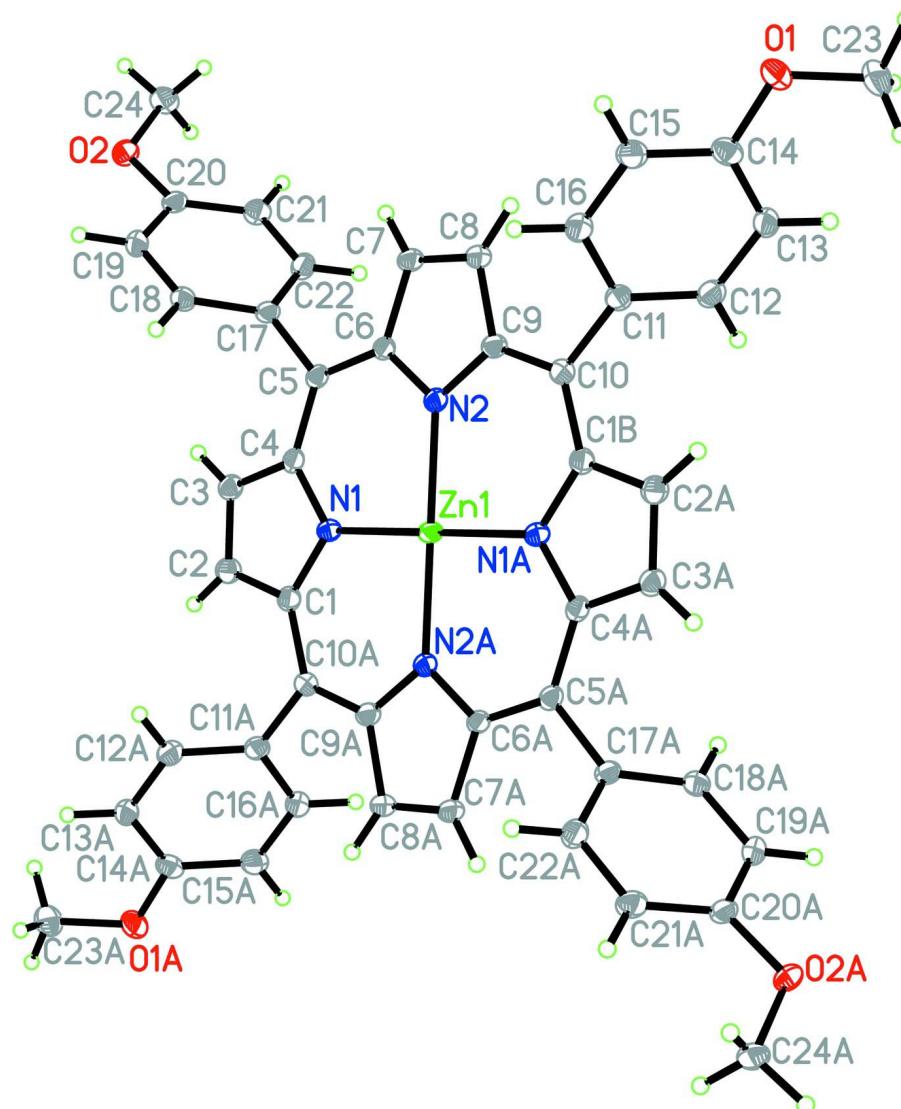
The title porphyrin (Fig. 1) has a zinc atom located on a center of inversion and hence the exact center of the mean-plane of the 24 other Non-H atoms with a r.m.s. deviation of the fitted atom = 0.0288 Å. The *p*-methoxy-substituted benzene rings form dihedral angles of 59.38 (6)° (C11-C16) and 66.77 (6)° (C17-C22) with the porphyrin mean plane. The crystal packing (Fig. 2) is characterized by close contacts between the Zn^{II} ion with two symmetry related molecules ($-x+1, y-1/2, -z+1/2$ and $x, -y+3/2, z+1/2$) through the oxygen atoms of a methoxy-phenyl group ($Zn \cdots O = 2.694$ (2) Å) forming a two-dimensional network parallel to (100). This distance is smaller than the sum of corresponding Van der Waals radii of atoms 2.910 Å (Bondi, 1964). There are some reports of Zn^{II} coordination bonds in porphyrins (Bhuyan & Sarkar, 2011; Adilov & Thalladi, 2007; Teo *et al.*, 2003). These reports indicate a bond distance shorter than what is found in the title compound.

S2. Experimental

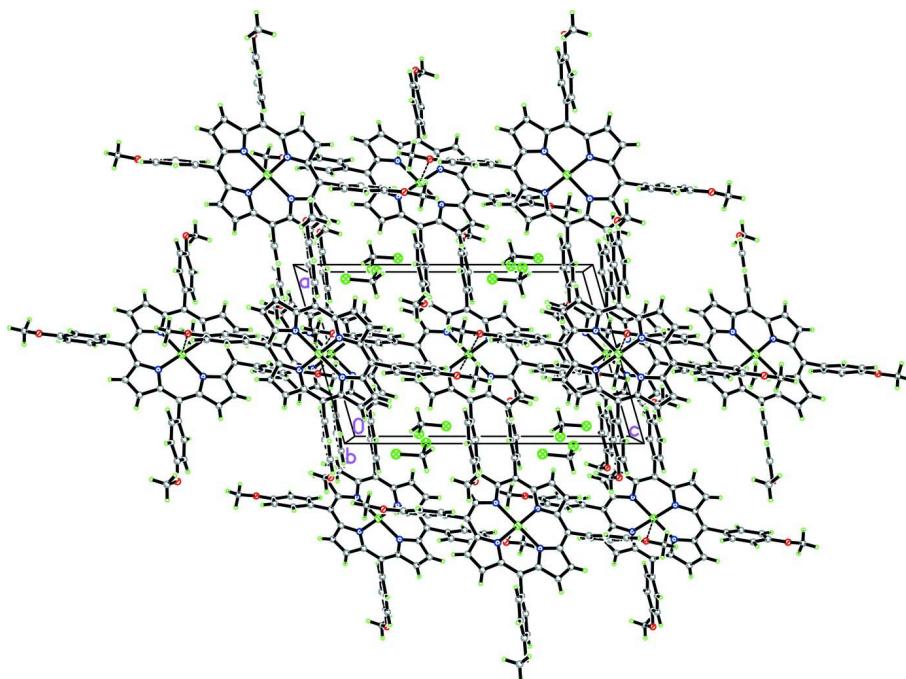
The synthesis of the title Zn complex was carried out according literature procedures (Adler *et al.*, 1967). Dark red crystals were grown by liquid diffusion of 1 ml octanol into a 3 ml dichloromethane solution containing 20 mg of zinc 5,10,15,20-tetrakis-(*p*-methoxyphenyl)porphyrin.

S3. Refinement

All H atoms attached to C atoms were placed in idealized positions (C—H = 0.95–0.99 Å) and allowed to ride on their parent atoms. All H atoms were constrained so that $U_{\text{iso}}(\text{H})$ were equal to 1.2Ueq or 1.5Ueq of their respective parent atoms.

**Figure 1**

The molecular structure of the title complex with 50% probability displacement ellipsoids. The solvent molecules are not shown [symmetry code: (A) $-x+1, -y+1, -z+1$].

**Figure 2**

Part of the crystal structure of the title compound viewed approximately along the b axis. The short $\text{Zn}\cdots\text{O}$ contacts are shown as dashed lines.

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Crystal data

$[\text{Zn}(\text{C}_{48}\text{H}_{36}\text{N}_4\text{O}_4)] \cdot 2\text{CH}_2\text{Cl}_2$

$M_r = 968.03$

Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

$a = 11.4189(9)$ Å

$b = 10.6877(9)$ Å

$c = 18.3778(15)$ Å

$\beta = 106.022(1)^\circ$

$V = 2155.7(3)$ Å 3

$Z = 2$

$F(000) = 996$

$D_x = 1.491$ Mg m $^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 8489 reflections

$\theta = 2.2\text{--}27.1^\circ$

$\mu = 0.87$ mm $^{-1}$

$T = 100$ K

Plate, red

$0.18 \times 0.16 \times 0.09$ mm

Data collection

Bruker APEXII CCD
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

ω scans

Absorption correction: multi-scan
(SADABS; Bruker, 2007)

$T_{\min} = 0.861$, $T_{\max} = 0.926$

25964 measured reflections

4763 independent reflections

4022 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.036$

$\theta_{\max} = 27.1^\circ$, $\theta_{\min} = 1.9^\circ$

$h = -14 \rightarrow 14$

$k = -13 \rightarrow 13$

$l = -23 \rightarrow 23$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.039$

$wR(F^2) = 0.121$

$S = 1.06$

4763 reflections

288 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.065P)^2 + 2.750P]$
where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} = 0.001$

$\Delta\rho_{\text{max}} = 0.52 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\text{min}} = -0.78 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

_iucr_refine_instructions_details

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0.0015 0.000 0.001 0.000 LAT T 1 SYMM -X, 0.5+Y, 0.5-Z SFAC C H N O Cl ZN UNIT 100 80 8 8 2 TEMP
-173.150 SIZE 0.09 0.157 0.178 acta L.S. 9 BOND $h htab FMAP 2 PLAN 20
WGHT 0.065000 2.750000 FVAR 0.12081 ZN1 6 0.500000 0.500000 0.500000 10.50000 0.01678 0.01721 = 0.01254
0.00333 0.00769 0.00349 O1 4 1.215897 0.070221 0.601887 11.00000 0.01859 0.02001 = 0.02872 0.00177 0.00721
0.00549 O2 4 0.617130 0.788402 0.058282 11.00000 0.02840 0.02118 = 0.01396 0.00475 0.01136 0.00312 N1 3
0.410408 0.615123 0.412964 11.00000 0.01692 0.01511 = 0.01311 0.00039 0.00676 0.00050 N2 3 0.630415 0.466937
0.445394 11.00000 0.01750 0.01366 = 0.01246 0.00085 0.00608 0.00049 C1 1 0.302121 0.675600 0.406600 11.00000
0.01653 0.01391 = 0.01524 0.00044 0.00463 - 0.00014 C2 1 0.266170 0.741753 0.335325 11.00000 0.01740 0.01863 =
0.01566 0.00210 0.00419 0.00153 AFIX 43 H2A 2 0.194325 0.790159 0.316778 11.00000 - 1.20000 AFIX 0 C3 1
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11.00000 0.01883 0.01725 = 0.01269 0.00138 0.00620 0.00189 C12 1 0.886843 0.127741 0.569024 11.00000 0.02077
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0 C13 1 0.996564 0.062037 0.584423 11.00000 0.02345 0.01568 = 0.01776 0.00253 0.00807 0.00209 AFIX 43 H13A 2
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0.02112 = 0.01535 - 0.00155 0.00576 0.00383 C15 1 1.100728 0.254478 0.575797 11.00000 0.01912 0.02027 = 0.02373
- 0.00157 0.00886 - 0.00081 AFIX 43 H15A 2 1.174018 0.298301 0.578027 11.00000 - 1.20000 AFIX 0 C16 1 0.991627
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0.00612 0.00109 C18 1 0.597241 0.768808 0.251103 11.00000 0.02074 0.01577 = 0.01549 - 0.00087 0.00781 0.00184
AFIX 43 H18A 2 0.603759 0.824277 0.292425 11.00000 - 1.20000 AFIX 0 C19 1 0.613282 0.814710 0.183941
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- 1.20000 AFIX 0 C20 1 0.603100 0.734439 0.122807 11.00000 0.01661 0.02078 = 0.01560 0.00552 0.00780 0.00353
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= 0.01723 0.00058 0.00701 - 0.00171 AFIX 43 H22A 2 0.551213 0.476471 0.203412 11.00000 - 1.20000 AFIX 0 C23 1
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1.196853 - 0.063083 0.672799 11.00000 - 1.50000 AFIX 0 C24 1 0.631393 0.705588 0.000316 11.00000 0.03584
0.02625 = 0.01679 0.00430 0.01573 0.00822 AFIX 137 H24A 2 0.649547 0.754219 - 0.040459 11.00000 - 1.50000
H24B 2 0.555899 0.658250 - 0.019910 11.00000 - 1.50000 H24C 2 0.698519 0.647479 0.021551 11.00000 - 1.50000
AFIX 0 C L1 5 1.016160 0.452210 0.737151 11.00000 0.04632 0.03578 = 0.03553 - 0.00369 0.02023 - 0.00035 C L2 5
1.070941 0.244096 0.845764 11.00000 0.05066 0.03732 = 0.04278 0.00165 0.02344 0.00160 C1A 1 1.082372 0.301804
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```

HKLF 4

REM 163p21c in P2(1)/c REM R1 = 0.0389 for 4022 $F_o > 4\text{sig}(F_o)$ and 0.0479 for all 4763 data REM 288 parameters refined using 0 restraints

END

WGHT 0.0533 3.1302 REM Highest difference peak 0.518, deepest hole -0.783, 1-sigma level 0.081 Q1 1 0.4013 0.6788
0.3211 11.00000 0.05 0.52 Q2 1 0.5621 0.6178 0.2947 11.00000 0.05 0.44 Q3 1 0.7446 0.3627 0.5013 11.00000 0.05
0.39 Q4 1 0.2799 0.7050 0.3700 11.00000 0.05 0.39 Q5 1 1.0627 0.0975 0.6026 11.00000 0.05 0.39 Q6 1 0.8242 0.2821
0.5447 11.00000 0.05 0.38 Q7 1 0.4999 0.5022 0.4666 11.00000 0.05 0.38 Q8 1 0.4430 0.6153 0.4402 11.00000 0.05
0.37 Q9 1 0.6311 0.5006 0.4190 11.00000 0.05 0.37 Q10 1 0.6781 0.4823 0.3609 11.00000 0.05 0.37 Q11 1 0.6142
0.7841 0.2163 11.00000 0.05 0.37 Q12 1 0.4338 0.5471 0.4547 11.00000 0.05 0.37 Q13 1 0.6085 0.7658 0.1538

Δ_{obs} 0.33, 0.36, 0.47, 0.41, 0.7640 0.3836 0.4414 11.00000 0.05 0.37 Q15 1 0.7809 0.4420 0.3877 11.00000 0.05 0.36

Q16 1 0.3624 0.6442 0.4135 11.00000 0.05 0.35 Q17 1 0.4265 0.6223 0.3814 11.00000 0.05 0.35 Q18 1 0.2755 0.6830
0.4343 11.00000 0.05 0.34 Q19 1 0.7735 0.4142 0.4510 11.00000 0.05 0.34 Q20 1 0.5817 0.5935 0.1662 11.00000 0.05
0.33

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Zn1	0.5000	0.5000	0.5000	0.01479 (12)
O1	1.21590 (16)	0.07022 (16)	0.60189 (10)	0.0223 (4)
O2	0.61713 (16)	0.78840 (16)	0.05828 (9)	0.0201 (4)
N1	0.41041 (17)	0.61512 (18)	0.41296 (10)	0.0145 (4)
N2	0.63042 (18)	0.46694 (18)	0.44539 (11)	0.0142 (4)
C1	0.3021 (2)	0.6756 (2)	0.40660 (13)	0.0152 (4)
C2	0.26662 (2)	0.7418 (2)	0.33533 (13)	0.0173 (5)
H2A	0.1943	0.7902	0.3168	0.021*
C3	0.3537 (2)	0.7220 (2)	0.29996 (13)	0.0167 (5)
H3A	0.3553	0.7543	0.2521	0.020*
C4	0.4446 (2)	0.6424 (2)	0.34878 (12)	0.0142 (4)
C5	0.5489 (2)	0.5967 (2)	0.33117 (12)	0.0144 (4)
C6	0.6332 (2)	0.5133 (2)	0.37583 (13)	0.0135 (4)
C7	0.7383 (2)	0.4640 (2)	0.35629 (13)	0.0162 (4)
H7A	0.7611	0.4809	0.3113	0.019*
C8	0.7980 (2)	0.3892 (2)	0.41388 (13)	0.0169 (5)
H8A	0.8705	0.3433	0.4169	0.020*
C9	0.7306 (2)	0.3921 (2)	0.47033 (13)	0.0150 (4)
C10	0.7649 (2)	0.3271 (2)	0.53954 (13)	0.0155 (4)
C11	0.8824 (2)	0.2566 (2)	0.55693 (13)	0.0159 (4)
C12	0.8868 (2)	0.1277 (2)	0.56902 (13)	0.0185 (5)
H12A	0.8136	0.0837	0.5667	0.022*
C13	0.9966 (2)	0.0620 (2)	0.58442 (13)	0.0185 (5)
H13A	0.9979	-0.0257	0.5927	0.022*
C14	1.1036 (2)	0.1257 (2)	0.58752 (13)	0.0185 (5)
C15	1.1007 (2)	0.2545 (2)	0.57580 (14)	0.0205 (5)
H15A	1.1740	0.2983	0.5780	0.025*
C16	0.9916 (2)	0.3189 (2)	0.56091 (14)	0.0186 (5)
H16A	0.9909	0.4068	0.5533	0.022*
C17	0.5717 (2)	0.6427 (2)	0.25917 (13)	0.0147 (4)
C18	0.5972 (2)	0.7688 (2)	0.25110 (13)	0.0168 (5)
H18A	0.6038	0.8243	0.2924	0.020*
C19	0.6133 (2)	0.8147 (2)	0.18394 (14)	0.0186 (5)
H19A	0.6312	0.9007	0.1796	0.022*
C20	0.6031 (2)	0.7344 (2)	0.12281 (13)	0.0170 (5)
C21	0.5822 (2)	0.6075 (2)	0.13054 (13)	0.0190 (5)
H21A	0.5789	0.5515	0.0898	0.023*
C22	0.5662 (2)	0.5631 (2)	0.19847 (13)	0.0179 (5)
H22A	0.5512	0.4765	0.2034	0.021*
C23	1.2243 (3)	-0.0568 (3)	0.62691 (16)	0.0270 (6)

H23A	1.3091	-0.0849	0.6380	0.041*
H23B	1.1728	-0.1095	0.5871	0.041*
H23C	1.1969	-0.0631	0.6728	0.041*
C24	0.6314 (3)	0.7056 (3)	0.00032 (14)	0.0246 (5)
H24A	0.6495	0.7542	-0.0405	0.037*
H24B	0.5559	0.6582	-0.0199	0.037*
H24C	0.6985	0.6475	0.0216	0.037*
Cl1	1.01616 (7)	0.45221 (7)	0.73715 (4)	0.03743 (19)
Cl2	1.07094 (8)	0.24410 (8)	0.84576 (5)	0.0415 (2)
C1A	1.0824 (3)	0.3018 (3)	0.75788 (18)	0.0368 (7)
H1AA	1.0412	0.2430	0.7173	0.044*
H1AB	1.1694	0.3061	0.7588	0.044*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Zn1	0.0168 (2)	0.0172 (2)	0.01254 (19)	0.00349 (14)	0.00769 (14)	0.00333 (14)
O1	0.0186 (8)	0.0200 (9)	0.0287 (10)	0.0055 (7)	0.0072 (7)	0.0018 (7)
O2	0.0284 (9)	0.0212 (9)	0.0140 (8)	0.0031 (7)	0.0114 (7)	0.0047 (7)
N1	0.0169 (9)	0.0151 (9)	0.0131 (9)	0.0005 (7)	0.0068 (7)	0.0004 (7)
N2	0.0175 (9)	0.0137 (9)	0.0125 (9)	0.0005 (7)	0.0061 (7)	0.0008 (7)
C1	0.0165 (11)	0.0139 (10)	0.0152 (11)	-0.0001 (8)	0.0046 (9)	0.0004 (8)
C2	0.0174 (11)	0.0186 (11)	0.0157 (11)	0.0015 (9)	0.0042 (9)	0.0021 (9)
C3	0.0196 (11)	0.0166 (11)	0.0143 (11)	0.0000 (9)	0.0057 (9)	0.0014 (9)
C4	0.0176 (11)	0.0121 (10)	0.0134 (10)	-0.0019 (8)	0.0051 (8)	0.0002 (8)
C5	0.0192 (11)	0.0127 (10)	0.0126 (10)	-0.0025 (8)	0.0067 (8)	-0.0011 (8)
C6	0.0172 (11)	0.0117 (10)	0.0131 (10)	-0.0017 (8)	0.0069 (8)	-0.0014 (8)
C7	0.0196 (11)	0.0166 (11)	0.0151 (11)	-0.0004 (9)	0.0093 (9)	-0.0010 (8)
C8	0.0187 (11)	0.0160 (11)	0.0191 (11)	0.0007 (9)	0.0103 (9)	-0.0001 (9)
C9	0.0166 (10)	0.0150 (10)	0.0149 (10)	-0.0011 (8)	0.0068 (8)	-0.0010 (9)
C10	0.0169 (11)	0.0143 (10)	0.0164 (11)	0.0004 (8)	0.0062 (9)	-0.0001 (8)
C11	0.0188 (11)	0.0173 (11)	0.0127 (10)	0.0019 (9)	0.0062 (9)	0.0014 (8)
C12	0.0208 (12)	0.0187 (12)	0.0178 (11)	-0.0012 (9)	0.0081 (9)	0.0012 (9)
C13	0.0234 (12)	0.0157 (11)	0.0178 (11)	0.0021 (9)	0.0081 (9)	0.0025 (9)
C14	0.0195 (11)	0.0211 (12)	0.0153 (11)	0.0038 (9)	0.0058 (9)	-0.0016 (9)
C15	0.0191 (11)	0.0203 (12)	0.0237 (12)	-0.0008 (9)	0.0089 (10)	-0.0016 (10)
C16	0.0213 (12)	0.0162 (11)	0.0198 (12)	0.0000 (9)	0.0083 (9)	0.0005 (9)
C17	0.0152 (10)	0.0169 (11)	0.0133 (10)	0.0011 (8)	0.0061 (8)	0.0021 (8)
C18	0.0207 (11)	0.0158 (11)	0.0155 (11)	0.0018 (9)	0.0078 (9)	-0.0009 (9)
C19	0.0233 (12)	0.0145 (11)	0.0199 (12)	0.0016 (9)	0.0091 (9)	0.0037 (9)
C20	0.0166 (11)	0.0208 (11)	0.0156 (11)	0.0035 (9)	0.0078 (9)	0.0055 (9)
C21	0.0243 (12)	0.0202 (12)	0.0144 (11)	-0.0009 (9)	0.0085 (9)	-0.0026 (9)
C22	0.0218 (12)	0.0157 (11)	0.0172 (11)	-0.0017 (9)	0.0070 (9)	0.0006 (9)
C23	0.0268 (13)	0.0246 (13)	0.0296 (14)	0.0106 (11)	0.0077 (11)	0.0082 (11)
C24	0.0358 (14)	0.0263 (13)	0.0168 (12)	0.0082 (11)	0.0157 (11)	0.0043 (10)
Cl1	0.0463 (4)	0.0358 (4)	0.0355 (4)	-0.0003 (3)	0.0202 (3)	-0.0037 (3)
Cl2	0.0507 (5)	0.0373 (4)	0.0428 (4)	0.0016 (3)	0.0234 (4)	0.0017 (3)
C1A	0.0347 (16)	0.0430 (18)	0.0354 (16)	0.0076 (13)	0.0140 (13)	-0.0017 (14)

Geometric parameters (\AA , $\text{^{\circ}}$)

Zn1—N2 ⁱ	2.0432 (19)	C11—C16	1.398 (3)
Zn1—N2	2.0432 (19)	C12—C13	1.395 (3)
Zn1—N1 ⁱ	2.0532 (19)	C12—H12A	0.9500
Zn1—N1	2.0532 (19)	C13—C14	1.387 (3)
O1—C14	1.371 (3)	C13—H13A	0.9500
O1—C23	1.428 (3)	C14—C15	1.392 (3)
O2—C20	1.367 (3)	C15—C16	1.383 (3)
O2—C24	1.428 (3)	C15—H15A	0.9500
N1—C1	1.371 (3)	C16—H16A	0.9500
N1—C4	1.372 (3)	C17—C22	1.391 (3)
N2—C9	1.367 (3)	C17—C18	1.396 (3)
N2—C6	1.380 (3)	C18—C19	1.387 (3)
C1—C10 ⁱ	1.409 (3)	C18—H18A	0.9500
C1—C2	1.445 (3)	C19—C20	1.393 (3)
C2—C3	1.351 (3)	C19—H19A	0.9500
C2—H2A	0.9500	C20—C21	1.392 (3)
C3—C4	1.446 (3)	C21—C22	1.394 (3)
C3—H3A	0.9500	C21—H21A	0.9500
C4—C5	1.405 (3)	C22—H22A	0.9500
C5—C6	1.400 (3)	C23—H23A	0.9800
C5—C17	1.500 (3)	C23—H23B	0.9800
C6—C7	1.443 (3)	C23—H23C	0.9800
C7—C8	1.351 (3)	C24—H24A	0.9800
C7—H7A	0.9500	C24—H24B	0.9800
C8—C9	1.453 (3)	C24—H24C	0.9800
C8—H8A	0.9500	C11—C1A	1.773 (3)
C9—C10	1.407 (3)	C12—C1A	1.767 (3)
C10—C1 ⁱ	1.409 (3)	C1A—H1AA	0.9900
C10—C11	1.495 (3)	C1A—H1AB	0.9900
C11—C12	1.393 (3)		
N2 ⁱ —Zn1—N2	180.00 (11)	C13—C12—H12A	119.3
N2 ⁱ —Zn1—N1 ⁱ	89.71 (8)	C14—C13—C12	119.6 (2)
N2—Zn1—N1 ⁱ	90.29 (8)	C14—C13—H13A	120.2
N2 ⁱ —Zn1—N1	90.29 (8)	C12—C13—H13A	120.2
N2—Zn1—N1	89.71 (8)	O1—C14—C13	124.2 (2)
N1 ⁱ —Zn1—N1	180.0	O1—C14—C15	116.0 (2)
C14—O1—C23	116.75 (19)	C13—C14—C15	119.7 (2)
C20—O2—C24	116.76 (19)	C16—C15—C14	120.3 (2)
C1—N1—C4	106.86 (18)	C16—C15—H15A	119.9
C1—N1—Zn1	126.34 (15)	C14—C15—H15A	119.9
C4—N1—Zn1	126.75 (15)	C15—C16—C11	121.0 (2)
C9—N2—C6	106.82 (18)	C15—C16—H16A	119.5
C9—N2—Zn1	126.33 (15)	C11—C16—H16A	119.5
C6—N2—Zn1	126.84 (15)	C22—C17—C18	118.1 (2)
N1—C1—C10 ⁱ	125.6 (2)	C22—C17—C5	121.8 (2)

N1—C1—C2	109.38 (19)	C18—C17—C5	120.2 (2)
C10 ⁱ —C1—C2	125.0 (2)	C19—C18—C17	121.3 (2)
C3—C2—C1	107.3 (2)	C19—C18—H18A	119.4
C3—C2—H2A	126.4	C17—C18—H18A	119.4
C1—C2—H2A	126.4	C18—C19—C20	119.9 (2)
C2—C3—C4	107.0 (2)	C18—C19—H19A	120.1
C2—C3—H3A	126.5	C20—C19—H19A	120.1
C4—C3—H3A	126.5	O2—C20—C21	124.3 (2)
N1—C4—C5	125.7 (2)	O2—C20—C19	115.9 (2)
N1—C4—C3	109.45 (19)	C21—C20—C19	119.8 (2)
C5—C4—C3	124.8 (2)	C20—C21—C22	119.5 (2)
C6—C5—C4	125.2 (2)	C20—C21—H21A	120.3
C6—C5—C17	117.8 (2)	C22—C21—H21A	120.3
C4—C5—C17	117.0 (2)	C17—C22—C21	121.5 (2)
N2—C6—C5	125.7 (2)	C17—C22—H22A	119.3
N2—C6—C7	109.45 (19)	C21—C22—H22A	119.3
C5—C6—C7	124.8 (2)	O1—C23—H23A	109.5
C8—C7—C6	107.2 (2)	O1—C23—H23B	109.5
C8—C7—H7A	126.4	H23A—C23—H23B	109.5
C6—C7—H7A	126.4	O1—C23—H23C	109.5
C7—C8—C9	107.1 (2)	H23A—C23—H23C	109.5
C7—C8—H8A	126.5	H23B—C23—H23C	109.5
C9—C8—H8A	126.5	O2—C24—H24A	109.5
N2—C9—C10	126.2 (2)	O2—C24—H24B	109.5
N2—C9—C8	109.44 (19)	H24A—C24—H24B	109.5
C10—C9—C8	124.3 (2)	O2—C24—H24C	109.5
C9—C10—C1 ⁱ	125.1 (2)	H24A—C24—H24C	109.5
C9—C10—C11	116.9 (2)	H24B—C24—H24C	109.5
C1 ⁱ —C10—C11	118.0 (2)	C12—C1A—Cl1	112.23 (17)
C12—C11—C16	118.1 (2)	C12—C1A—H1AA	109.2
C12—C11—C10	121.5 (2)	Cl1—C1A—H1AA	109.2
C16—C11—C10	120.4 (2)	C12—C1A—H1AB	109.2
C11—C12—C13	121.4 (2)	Cl1—C1A—H1AB	109.2
C11—C12—H12A	119.3	H1AA—C1A—H1AB	107.9

Symmetry code: (i) $-x+1, -y+1, -z+1$.