

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

2-[4-(Trifluoromethyl)phenylsulfanyl]-benzoic acid

Thammarse S. Yamuna,^a Jerry P. Jasinski,^{b*} Brian J. Anderson,^b H.S. Yathirajan^a and Manpreet Kaur^a^aDepartment of Studies in Chemistry, University of Mysore, Manasagangotri, Mysore 570 006, India, and ^bDepartment of Chemistry, Keene State College, 229 Main Street, Keene, NH 03435-2001, USA

Correspondence e-mail: jjasinski@keene.edu

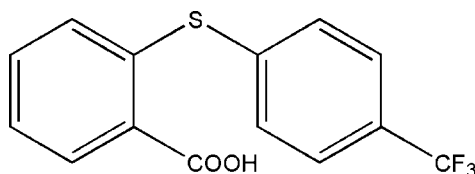
Received 13 October 2013; accepted 20 October 2013

Key indicators: single-crystal X-ray study; $T = 173$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; disorder in main residue; R factor = 0.054; wR factor = 0.155; data-to-parameter ratio = 12.5.

In the title compound, $\text{C}_{14}\text{H}_9\text{F}_3\text{O}_2\text{S}$, the dihedral angle between the mean planes of the benzene rings is $88.7(2)^\circ$. The carboxylic acid group is twisted by $13.6(7)^\circ$ from the mean plane of its attached aromatic ring. One of the F atoms of the trifluoromethyl group is disordered over two sites in a 0.61(7):0.39(7) ratio. In the crystal, inversion dimers linked by pairs of $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds generate $R_2^2(8)$ loops. Weak $\text{C}-\text{H}\cdots\text{F}$ interactions are also observed.

Related literature

For background to the neuroleptic agent flupentixol (systematic name: (*EZ*)-2-[4-[3-[2-(trifluoromethyl)thioxanthen-9-ylidene]propyl]piperazin-1-yl]ethanol), see: Young *et al.* (1976). For related structures, see: Post *et al.* (1975*a,b*); Siddegowda *et al.* (2011*a,b*).



Experimental

Crystal data

 $\text{C}_{14}\text{H}_9\text{F}_3\text{O}_2\text{S}$
 $M_r = 298.28$
 Triclinic, $P\bar{1}$
 $a = 7.3071(5)$ Å
 $b = 8.0790(7)$ Å
 $c = 11.3878(11)$ Å

 $\alpha = 82.678(8)^\circ$
 $\beta = 83.642(7)^\circ$
 $\gamma = 72.309(7)^\circ$
 $V = 633.41(10)$ Å³
 $Z = 2$

 Cu $K\alpha$ radiation
 $\mu = 2.63$ mm⁻¹
 $T = 173$ K
 $0.24 \times 0.22 \times 0.12$ mm

Data collection

 Agilent Gemini EOS diffractometer
 Absorption correction: multi-scan
 (*CrysAlis PRO* and *CrysAlis RED*; Agilent, 2012)
 $T_{\min} = 0.848$, $T_{\max} = 1.000$

 3701 measured reflections
 2419 independent reflections
 2049 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.038$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.054$
 $wR(F^2) = 0.155$
 $S = 1.04$
 2419 reflections

 193 parameters
 H-atom parameters constrained
 $\Delta\rho_{\max} = 0.71$ e Å⁻³
 $\Delta\rho_{\min} = -0.38$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{O2}-\text{H2}\cdots\text{O1}^{\text{i}}$	0.82	1.86	2.677 (3)	175
$\text{C6}-\text{H6}\cdots\text{F3}^{\text{ii}}$	0.93	2.59	3.319 (10)	136
$\text{C6}-\text{H6}\cdots\text{F3A}^{\text{ii}}$	0.93	2.50	3.294 (16)	144

Symmetry codes: (i) $-x, -y + 1, -z + 1$; (ii) $-x + 2, -y + 1, -z + 2$.

Data collection: *CrysAlis PRO* (Agilent, 2012); cell refinement: *CrysAlis PRO* (Agilent, 2012); data reduction: *CrysAlis RED*; program(s) used to solve structure: *SUPERFLIP* (Palatinus & Chapuis, 2007); program(s) used to refine structure: *SHELXL2012* (Sheldrick, 2008); molecular graphics: *OLEX2* (Dolomanov *et al.*, 2009); software used to prepare material for publication: *OLEX2*.

TSY thanks University of Mysore for research facilities and is also grateful to the Principal, Maharani's Science College for Women, Mysore, for giving permission to undertake research. JPJ acknowledges the NSF-MRI program (grant No. CHE-1039027) for funds to purchase the X-ray diffractometer.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB7150).

References

- Agilent (2012). *CrysAlis PRO* and *CrysAlis RED*. Agilent Technologies, Yarnton, England.
- Dolomanov, O. V., Bourhis, L. J., Gildea, R. J., Howard, J. A. K. & Puschmann, H. (2009). *J. Appl. Cryst.* **42**, 339–341.
- Palatinus, L. & Chapuis, G. (2007). *J. Appl. Cryst.* **40**, 786–790.
- Post, M. L., Kennard, O. & Horn, A. S. (1975*a*). *Acta Cryst.* **B31**, 2724–2726.
- Post, M. L., Kennard, O., Sheldrick, G. M. & Horn, A. S. (1975*b*). *Acta Cryst.* **B31**, 2366–2368.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Siddegowda, M. S., Butcher, R. J., Akkurt, M., Yathirajan, H. S. & Narayana, B. (2011*a*). *Acta Cryst.* **E67**, o2079–o2080.
- Siddegowda, M. S., Butcher, R. J., Akkurt, M., Yathirajan, H. S. & Ramesh, A. R. (2011*b*). *Acta Cryst.* **E67**, o2017–o2018.
- Young, J. P. R., Hughes, W. C. & Lader, M. H. (1976). *Br. Med. J.* **1**, 1116–1118.

supporting information

Acta Cryst. (2013). E69, o1704 [doi:10.1107/S1600536813028778]

2-[4-(Trifluoromethyl)phenylsulfanyl]benzoic acid

Thammarse S. Yamuna, Jerry P. Jasinski, Brian J. Anderson, H.S. Yathirajan and Manpreet Kaur

S1. Comment

The title compound, (I), is a starting material for the synthesis of flupentixol, a well-documented neuroleptic (Young *et al.*, 1976). The crystal structures of α -flupentixol (Post *et al.*, 1975*a*) and β -flupentixol (Post *et al.*, 1975*b*) have been reported. As part of our ongoing studies in this area (Siddegowda *et al.*, 2011*a,b*), we now describe the crystal structure of (I).

The dihedral angle between the mean planes of the phenyl rings is 88.7 (2)°. The carboxylic acid group (C2/C1/O2/O1) is twisted by 13.6 (7)° from the mean plane of the adjacent benzene ring (C2–C7). Disorder was modeled over two sets of sites for one fluorine atom (F3), of the trifluoromethyl group with an occupancy ratio of 0.61 (7) : 0.39 (7). In the crystal, O—H···O hydrogen bonds (Table 1) link the molecules into dimers with R₂²[8] graph-set motifs (Fig. 2). Weak C—H···F interactions are also observed.

S2. Experimental

The title compound was obtained as a gift sample from R. L. Fine chemicals, Bengaluru. It was dissolved in 15 ml of mixture of acetonitrile and dimethyl sulphoxide (1:2), stirred for 10 minutes at room temperature. After few days, irregular colourless crystals were formed by slow evaporation of the solvent mixture (m.p: 413-418 K).

S3. Refinement

All of the H atoms were placed in their calculated positions and then refined using the riding model with Atom—H lengths of 0.93 Å (CH) or 0.82 Å (OH). Isotropic displacement parameters for these atoms were set to 1.2 (CH) or 1.5 (OH) times U_{eq} of the parent atom. Disorder was modeled over two sets of sites for one fluorine atom (F3) in the trifluoromethyl group with an occupancy ratio of 0.61 (7) : 0.39 (7). Idealised tetrahedral OH refined as rotating group.

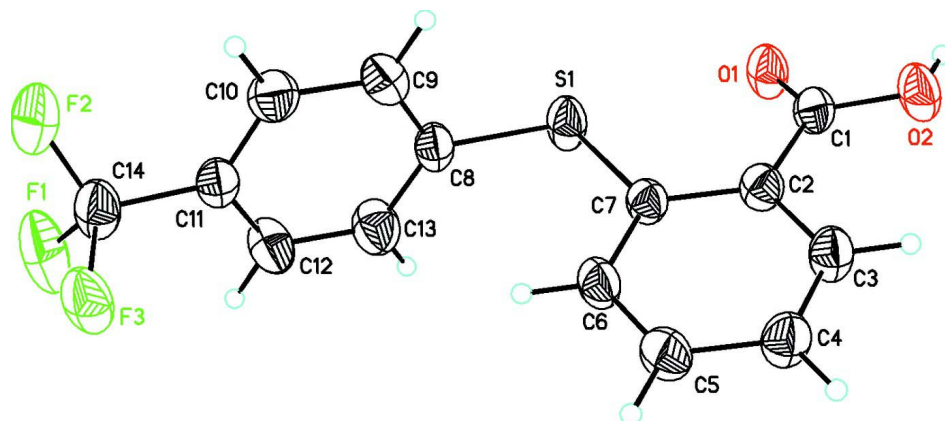
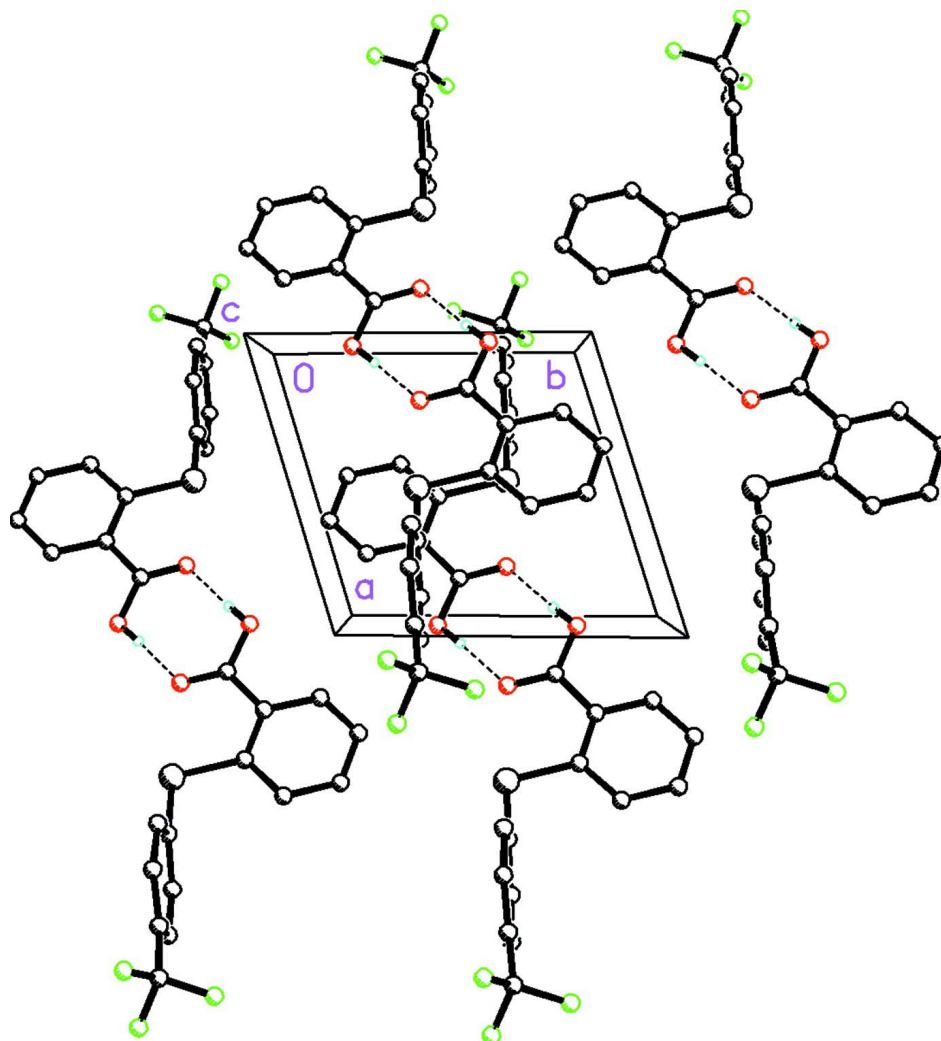


Figure 1

ORTEP drawing of (I) showing 50% probability displacement ellipsoids. Disordered atom, F3A (0.39 (7) occupancy) has been removed for clarity.

**Figure 2**

Molecular packing for (I) viewed along the *c* axis. Dashed lines indicate O—H...O intermolecular hydrogen bonds which link the molecules into dimers with $R_2^2[8]$ graph-set motifs and influence the crystal packing. Disordered atom, F3A (0.39 (7) occupancy) and H atoms not involved in hydrogen bonding have been removed for clarity.

2-[4-(Trifluoromethyl)phenylsulfanyl]benzoic acid

Crystal data

$C_{14}H_9F_3O_2S$

$M_r = 298.28$

Triclinic, $P\bar{1}$

$a = 7.3071$ (5) Å

$b = 8.0790$ (7) Å

$c = 11.3878$ (11) Å

$\alpha = 82.678$ (8)°

$\beta = 83.642$ (7)°

$\gamma = 72.309$ (7)°

$V = 633.41$ (10) Å³

$Z = 2$

$F(000) = 304$

$D_x = 1.564$ Mg m⁻³

Cu $K\alpha$ radiation, $\lambda = 1.54184$ Å

Cell parameters from 1645 reflections

$\theta = 5.8$ – 72.1 °

$\mu = 2.63$ mm⁻¹

$T = 173$ K

Irregular, colourless

$0.24 \times 0.22 \times 0.12$ mm

Data collection

Agilent Gemini EOS diffractometer	3701 measured reflections
Radiation source: Enhance (Cu) X-ray Source	2419 independent reflections
Detector resolution: 16.0416 pixels mm ⁻¹	2049 reflections with $I > 2\sigma(I)$
ω scans	$R_{\text{int}} = 0.038$
Absorption correction: multi-scan (<i>CrysAlis PRO</i> and <i>CrysAlis RED</i> ; Agilent, 2012)	$\theta_{\text{max}} = 72.2^\circ$, $\theta_{\text{min}} = 5.8^\circ$
$T_{\text{min}} = 0.848$, $T_{\text{max}} = 1.000$	$h = -8 \rightarrow 6$
	$k = -9 \rightarrow 6$
	$l = -13 \rightarrow 14$

Refinement

Refinement on F^2	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.054$	$w = 1/[\sigma^2(F_o^2) + (0.0889P)^2 + 0.2312P]$
$wR(F^2) = 0.155$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.04$	$(\Delta/\sigma)_{\text{max}} < 0.001$
2419 reflections	$\Delta\rho_{\text{max}} = 0.71 \text{ e } \text{\AA}^{-3}$
193 parameters	$\Delta\rho_{\text{min}} = -0.38 \text{ e } \text{\AA}^{-3}$
0 restraints	Extinction correction: <i>SHELXL2012</i> (Sheldrick, 2008), $F_c^* = kFc^*[1 + 0.001x \text{Fc}^2\lambda^3/\sin(2\theta)]^{-1/4}$
Primary atom site location: structure-invariant direct methods	Extinction coefficient: 0.0044 (14)

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
S1	0.51402 (10)	0.35821 (8)	0.70404 (6)	0.0398 (3)	
F1	1.2832 (3)	0.1041 (3)	1.0280 (2)	0.0764 (7)	
F2	1.0648 (3)	0.1392 (4)	1.16595 (19)	0.0849 (8)	
F3	1.1664 (14)	0.3549 (10)	1.0613 (12)	0.054 (2)	0.61 (7)
F3A	1.094 (10)	0.3613 (16)	1.106 (6)	0.100 (14)	0.39 (7)
O1	0.2047 (3)	0.4458 (2)	0.56512 (17)	0.0405 (5)	
O2	-0.0035 (3)	0.7122 (2)	0.54064 (19)	0.0427 (5)	
H2	-0.0646	0.6592	0.5121	0.064*	
C1	0.1620 (4)	0.6021 (3)	0.5733 (2)	0.0332 (5)	
C2	0.2907 (4)	0.6861 (3)	0.6177 (2)	0.0316 (5)	
C3	0.2487 (4)	0.8678 (3)	0.5973 (2)	0.0376 (6)	
H3	0.1387	0.9315	0.5595	0.045*	
C4	0.3668 (4)	0.9550 (3)	0.6319 (2)	0.0396 (6)	
H4	0.3375	1.0757	0.6172	0.048*	
C5	0.5291 (4)	0.8595 (4)	0.6888 (2)	0.0400 (6)	
H5	0.6098	0.9166	0.7125	0.048*	
C6	0.5730 (4)	0.6794 (3)	0.7109 (2)	0.0362 (6)	
H6	0.6825	0.6174	0.7497	0.043*	

C7	0.4555 (3)	0.5897 (3)	0.6759 (2)	0.0303 (5)
C8	0.6908 (4)	0.3131 (3)	0.8091 (2)	0.0328 (5)
C9	0.6355 (4)	0.3133 (3)	0.9291 (2)	0.0378 (6)
H9	0.5055	0.3396	0.9548	0.045*
C10	0.7722 (4)	0.2748 (4)	1.0114 (2)	0.0405 (6)
H10	0.7343	0.2740	1.0921	0.049*
C11	0.9646 (4)	0.2375 (3)	0.9731 (2)	0.0335 (6)
C12	1.0216 (4)	0.2340 (4)	0.8540 (3)	0.0447 (7)
H12	1.1518	0.2070	0.8288	0.054*
C13	0.8857 (4)	0.2704 (4)	0.7719 (3)	0.0461 (7)
H13	0.9246	0.2663	0.6915	0.055*
C14	1.1122 (4)	0.2104 (4)	1.0605 (3)	0.0450 (7)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0441 (4)	0.0305 (4)	0.0495 (4)	-0.0120 (3)	-0.0242 (3)	-0.0013 (3)
F1	0.0508 (12)	0.0796 (15)	0.0942 (17)	0.0002 (10)	-0.0385 (11)	-0.0105 (12)
F2	0.0770 (15)	0.134 (2)	0.0508 (12)	-0.0459 (15)	-0.0335 (10)	0.0265 (12)
F3	0.057 (5)	0.0356 (19)	0.078 (4)	-0.0181 (19)	-0.034 (4)	-0.002 (2)
F3A	0.15 (3)	0.034 (3)	0.14 (2)	-0.024 (7)	-0.11 (2)	-0.002 (6)
O1	0.0417 (10)	0.0349 (10)	0.0504 (11)	-0.0126 (8)	-0.0228 (8)	-0.0039 (8)
O2	0.0356 (10)	0.0404 (10)	0.0559 (12)	-0.0100 (8)	-0.0219 (8)	-0.0055 (9)
C1	0.0336 (13)	0.0385 (14)	0.0295 (12)	-0.0123 (10)	-0.0104 (10)	0.0000 (10)
C2	0.0343 (13)	0.0335 (12)	0.0293 (12)	-0.0112 (10)	-0.0075 (9)	-0.0039 (9)
C3	0.0381 (13)	0.0339 (13)	0.0410 (14)	-0.0075 (11)	-0.0138 (11)	-0.0026 (10)
C4	0.0456 (15)	0.0300 (13)	0.0460 (15)	-0.0124 (11)	-0.0143 (12)	-0.0016 (11)
C5	0.0415 (15)	0.0410 (15)	0.0452 (15)	-0.0190 (12)	-0.0141 (12)	-0.0055 (11)
C6	0.0331 (13)	0.0373 (14)	0.0412 (14)	-0.0117 (11)	-0.0143 (11)	-0.0019 (11)
C7	0.0308 (12)	0.0316 (12)	0.0302 (12)	-0.0101 (10)	-0.0078 (9)	-0.0023 (9)
C8	0.0339 (12)	0.0267 (12)	0.0393 (13)	-0.0082 (9)	-0.0143 (10)	-0.0009 (10)
C9	0.0317 (13)	0.0391 (14)	0.0441 (15)	-0.0103 (11)	-0.0066 (11)	-0.0067 (11)
C10	0.0466 (15)	0.0464 (16)	0.0333 (13)	-0.0192 (12)	-0.0066 (11)	-0.0041 (11)
C11	0.0387 (13)	0.0219 (11)	0.0424 (14)	-0.0100 (9)	-0.0145 (11)	-0.0005 (9)
C12	0.0320 (13)	0.0542 (17)	0.0466 (16)	-0.0077 (12)	-0.0078 (11)	-0.0086 (13)
C13	0.0395 (15)	0.0595 (18)	0.0365 (14)	-0.0091 (13)	-0.0037 (11)	-0.0071 (13)
C14	0.0550 (18)	0.0317 (13)	0.0538 (17)	-0.0154 (12)	-0.0258 (14)	0.0006 (12)

Geometric parameters (Å, °)

S1—C7	1.782 (2)	C5—H5	0.9300
S1—C8	1.783 (2)	C5—C6	1.387 (4)
F1—C14	1.328 (4)	C6—H6	0.9300
F2—C14	1.322 (4)	C6—C7	1.396 (3)
F3—C14	1.343 (8)	C8—C9	1.381 (4)
F3A—C14	1.35 (2)	C8—C13	1.390 (4)
O1—C1	1.218 (3)	C9—H9	0.9300
O2—H2	0.8200	C9—C10	1.384 (4)

O2—C1	1.324 (3)	C10—H10	0.9300
C1—C2	1.477 (3)	C10—C11	1.378 (4)
C2—C3	1.400 (4)	C11—C12	1.375 (4)
C2—C7	1.404 (3)	C11—C14	1.496 (4)
C3—H3	0.9300	C12—H12	0.9300
C3—C4	1.383 (4)	C12—C13	1.380 (4)
C4—H4	0.9300	C13—H13	0.9300
C4—C5	1.381 (4)		
C7—S1—C8	101.59 (11)	C13—C8—S1	120.6 (2)
C1—O2—H2	109.5	C8—C9—H9	119.7
O1—C1—O2	122.6 (2)	C8—C9—C10	120.5 (2)
O1—C1—C2	123.3 (2)	C10—C9—H9	119.7
O2—C1—C2	114.1 (2)	C9—C10—H10	120.2
C3—C2—C1	118.3 (2)	C11—C10—C9	119.7 (2)
C3—C2—C7	119.4 (2)	C11—C10—H10	120.2
C7—C2—C1	122.3 (2)	C10—C11—C14	120.1 (2)
C2—C3—H3	119.2	C12—C11—C10	120.3 (2)
C4—C3—C2	121.6 (2)	C12—C11—C14	119.5 (3)
C4—C3—H3	119.2	C11—C12—H12	120.0
C3—C4—H4	120.6	C11—C12—C13	120.0 (3)
C5—C4—C3	118.8 (2)	C13—C12—H12	120.0
C5—C4—H4	120.6	C8—C13—H13	119.9
C4—C5—H5	119.7	C12—C13—C8	120.2 (3)
C4—C5—C6	120.7 (2)	C12—C13—H13	119.9
C6—C5—H5	119.7	F1—C14—F3	97.8 (5)
C5—C6—H6	119.5	F1—C14—F3A	122 (3)
C5—C6—C7	121.1 (2)	F1—C14—C11	113.7 (2)
C7—C6—H6	119.5	F2—C14—F1	104.4 (2)
C2—C7—S1	120.83 (18)	F2—C14—F3	115.3 (6)
C6—C7—S1	120.71 (19)	F2—C14—F3A	91 (3)
C6—C7—C2	118.5 (2)	F2—C14—C11	113.2 (2)
C9—C8—S1	120.1 (2)	F3—C14—C11	111.3 (4)
C9—C8—C13	119.2 (2)	F3A—C14—C11	110.3 (8)
S1—C8—C9—C10	178.6 (2)	C8—S1—C7—C2	-165.8 (2)
S1—C8—C13—C12	-179.4 (2)	C8—S1—C7—C6	14.5 (2)
O1—C1—C2—C3	165.0 (3)	C8—C9—C10—C11	0.6 (4)
O1—C1—C2—C7	-13.0 (4)	C9—C8—C13—C12	-2.0 (4)
O2—C1—C2—C3	-13.7 (3)	C9—C10—C11—C12	-1.7 (4)
O2—C1—C2—C7	168.3 (2)	C9—C10—C11—C14	174.9 (2)
C1—C2—C3—C4	-177.4 (2)	C10—C11—C12—C13	1.0 (4)
C1—C2—C7—S1	-2.1 (3)	C10—C11—C14—F1	151.3 (3)
C1—C2—C7—C6	177.6 (2)	C10—C11—C14—F2	32.4 (4)
C2—C3—C4—C5	-0.5 (4)	C10—C11—C14—F3	-99.4 (7)
C3—C2—C7—S1	179.85 (19)	C10—C11—C14—F3A	-68 (4)
C3—C2—C7—C6	-0.4 (4)	C11—C12—C13—C8	0.9 (5)
C3—C4—C5—C6	0.0 (4)	C12—C11—C14—F1	-32.1 (4)

C4—C5—C6—C7	0.3 (4)	C12—C11—C14—F2	-151.0 (3)
C5—C6—C7—S1	179.6 (2)	C12—C11—C14—F3	77.2 (7)
C5—C6—C7—C2	-0.1 (4)	C12—C11—C14—F3A	109 (4)
C7—S1—C8—C9	89.1 (2)	C13—C8—C9—C10	1.2 (4)
C7—S1—C8—C13	-93.5 (2)	C14—C11—C12—C13	-175.6 (3)
C7—C2—C3—C4	0.7 (4)		

Hydrogen-bond geometry (Å, °)

<i>D—H...A</i>	<i>D—H</i>	<i>H...A</i>	<i>D...A</i>	<i>D—H...A</i>
O2—H2...O1 ⁱ	0.82	1.86	2.677 (3)	175
C6—H6...F3 ⁱⁱ	0.93	2.59	3.319 (10)	136
C6—H6...F3A ⁱⁱ	0.93	2.50	3.294 (16)	144

Symmetry codes: (i) $-x, -y+1, -z+1$; (ii) $-x+2, -y+1, -z+2$.