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Trinuclear nickel coordination complexes of phenanthrene-9,10-dione dioxime

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A trinuclear nickel complex of phenanthrene-9,10-dione dioxime (H_2 pqd), namelv bis[μ_2 -9,10-bis(oxidoimino)phenanthrene]bis[μ_2 -10-(oxidoimino)phenanthrene-9-one oxime](phenanthrene-9,10-dione dioxime)trinickel(II) toluene disolvate, $[Ni_3(C_{14}H_8N_2O_2)_2(C_{14}H_9N_2O_2)_2(C_{14}H_{10}N_2O_2)]\cdot 2C_7H_8$, has been isolated and its crystal structure determined. This complex features three independent Ni^{II} atoms that are arranged in a triangular fashion along with five supporting ligands. There are two square-planar Ni^{II} atoms and a third pseudooctahedral Ni^{II} atom. While the square-planar Ni^{II} atoms are stacked, there are no ligand bridges between them. Each square-planar Ni^{II} atom, however, bridges with the pseudo-octahedral Ni^{II} atom through Ni-N-O-Ni and Ni-O-Ni bonds. A fluoridoboration reaction of the proton-bridged species gave the analogous complex bis(μ_2 -bis{[10-(oxidoimino)-9,10-dihydrophenanthren-9-ylidene]amino}difluoridoborato)(phenanthrene-9,10-dione dioxime)trinickel(II) dichloromethane trisolvate, $[Ni_3(C_{28}H_{16}BF_2N_4O_4)_2(C_{14}H_{10}N_2O_2)]$. 3CH₂Cl₂, which shows the same binding structure, but features a widened Ni-Ni interaction between the square-planar Ni^{II} atoms. The proton-bridged complex completes the macrocyclic coordination around the square-planar Ni^{II} atoms by means of an O-H···O hydrogen bond. Both compounds feature O- $H \cdot \cdot \cdot N$ hydrogen bonds between the oxime and the N atoms attached to squareplanar nickel atom. The nickel units show no direct interaction with their nearest neighbors in the extended lattice. Two π -stacking interactions between adjacent molecules are found: one with a centroid-centroid distance of 3.886(2) A and the other with a centroid–centroid distance of 4.256(3) A. In the latter case, although not aromatic, the distance to the centroid of the central phenanthrene ring is shorter, with a distance of 3.528 (3) A. Toluene molecules occupy the solvent channels that are oriented along the c axis. In the fluoridoboronate structure, the solvent (DCM) was too badly disordered to be modelled, so its contribution was removed using SQUEEZE [Spek (2015). Acta Cryst. C71, 9-18].

1. Chemical context

Oxime functional groups can coordinate to transition metal ions in a variety of ways, due to the presence of both nitrogen and oxygen donors. On account of the multitude of possible coordinations, these ligands, and particularly α - β dioximes, have the capability of forming bridging multinuclear complexes with many transition metals, including nickel (Chaudhuri, 2003). From the standpoint of single-molecule magnets, these multi-nuclear complexes play an important role due to their ability to facilitate spin-frustration in magnetic transition-metal clusters (Aromí & Brechin, 2006). Other nickel polynuclear compounds supported by oxime ligands have been reported (Jiang *et al.*, 2005; Biswas *et al.*, 2009).





Phenanthrenequinone dioxime (pqdH₂) is an α - β dioxime ligand that incorporates a constrained ring system. Similar to other dioximes, however, it exists as three separate stereoisomers (*E*-*E*, *E*-*Z*, and *Z*-*Z*), as confirmed by liquid chromatography – mass spectrometry. Interestingly, although this compound was synthesized over 100 years ago (Schmidt & Söll, 1907), no coordination complexes of this ligand have been structurally characterized to date.





2. Structural commentary

Fig. 1 shows the structure of $[Ni_3(H_2pqd)(Hpqd)_2(pqd)_2]$, (1). This complex consists of three Ni^{II} atoms in a triangular arrangement, two of which are in a square-planar coordination environment, while the third is in a pseudo-octahedral coordination environment. The square-planar Ni^{II} atoms (Ni1 and Ni2) consist of one *N*,*N*-coordinating and one *N*,*O*-coordinating ligand. These ligands form bridges with the pseudo-octahedral Ni^{II} atom (Ni3) by means of their oxime O atoms.



Figure 1

Displacement ellipsoid plot at the 50% probability level for $[Ni_3(H_2pqd)(Hpqd)_2(pqd)_2]$. H atoms (with the exception of hydrogenbonded atoms) and solvent molecules have been omitted for clarity.

This arrangement permits the formation of Ni-N-O-Niand Ni-O-Ni bridges between each square-planar Ni^{II} atom and the pseudo-octahedral Ni^{II} atom.

The structural features of the core ligation sphere warrant special attention. The Ni_{sp}-Ni_{sp} distance is 3.3657 (9) Å, a distance that precludes the presence of any metal-metal bonding. However, the distances between each of these nickel moieties and the pseudo-octahedral Ni^{II} atom are nearly identical [Ni_{sp}-Ni_{oct} = 3.2697 (7), 3.2674 (7) Å]. The pseudo-octahedral nickel geometry deviates significantly from a perfect octahedral symmetry [O2-Ni3-O8 = 160.00 (9), O4-Ni3-N10 = 164.8 (1), N9-Ni3-O6 = 165.5 (1)°].

Fig. 2 shows the complete coordination geometry of the compound $[Ni_3(pqdH_2)(pqdBF_2)_2(pqd)_2]$, (2). The physical arrangement of the ligation sphere directly mimics that of (1). In this case, however, the steric bulk of the BF₂ groups forces an expansion of the stacked square-planar nickel units, resulting in an $Ni_{sp}-Ni_{sp}$ distance of 3.592 (1) Å. The distances between these units and the pseudo-octahedral Ni^{II} atom, however, remain similar $[Ni_{sp}-Ni_{oct} = 3.274 (1), 3.255 (1) Å]$. Overall, the entire structure retains all the other structural features that are present in the proton-bridged compound.

The phenanthrene backbones show pronounced twisting between their aromatic rings, which precludes conjugation across this unit. For the proton-bridged complex, the angle between mean planes within a single phenanthrene backbone ranges from 9.24 (19)° (between C45–C50 and C51–C56) to 15.44 (13)° (between C59–C64 and C65–C70). For the

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Displacement ellipsoid plot at the 50% probability level of $[Ni_3(H_2pqd)(BF_2pqd)_2(pqd)_2]$. H atoms (with the exception of hydrogen-bonded atoms) and solvent molecules have been omitted for clarity.

BF₂-bridged complex, there is a wider range of angles, with 5.2 (4)° (between C31–C36 and C37–C42) being the smallest, and 17.5 (3)° (between C59–C64 and C65–C70) being the largest.

Table 1	
Hydrogen-bond geometry (Å, $^{\circ}$) for (1).	

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
$O1 - H1A \cdots O5$	1.29 (6)	1.13 (6)	2.405 (4)	169 (6)
$O3 - H2A \cdots O7$	1.17 (5)	1.23 (5)	2.402 (4)	176 (5)
$O9 - H3A \cdots N4$	0.86 (6)	1.82 (6)	2.672 (4)	175 (7)
$O10 - H4A \cdots N6$	1.04 (5)	1.63 (5)	2.657 (4)	168 (4)

Table 2

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Hydrogen-bond geometry (A, \circ) for	(2).
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$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
$O9-H1A\cdots N2$	0.85	1.96	2.771 (6)	158
$O10-H2A\cdots N8$	1.06	1.77	2.765 (6)	155

3. Supramolecular features

The proton-bridged complex completes the macrocyclic coordination around the square-planar Ni^{II} atoms by means of hydrogen bonds. Furthermore, the ligand that coordinates the pseudo-octahedral Ni^{II} atom features hydrogen-bonding interactions (Table 1) between the oxime hydroxy groups and the ligands of the square-planar Ni^{II} atoms. The nickel units show no direct interaction with their nearest neighbors in the extended lattice. Some π -stacking between adjacent molecules is, however, evident (Fig. 3). Two interactions were found, one with a centroid-centroid distance of 3.886 (2) Å (symmetry code: 1 - x, $-\frac{1}{2} + y$, $\frac{3}{2} - z$) and the other with a centroidcentroid distance of 4.256 (3) Å (symmetry code: -x, -y, 2-z). In the latter case, although not aromatic, the distance to the centroid of the central ring of phenanthrene is shorter, with a distance of 3.528 (3) Å. Toluene molecules occupy the solvent channels that are oriented along the c axis.



Figure 3

Packing diagram of [Ni₃(H₂pqd)(Hpqd)₂(pqd)₂], viewed approximately down the *c*-axis direction.



Figure 4

Packing diagram of $Ni_3(H_2pqd)(BF_2pqd)_2(pqd)_2$, viewed approximately down the *c*-axis direction. Voids presented in brown were calculated in *Mercury* (Macrae *et al.*, 2006) using a probe radius of 1.2 Å.

The BF₂-bridged complex completes the macrocyclic coordination around the square-planar $\mathrm{Ni}^{\mathrm{II}}$ atoms by means of covalent O-B-O bonds. However, the hydrogen-bonding interactions (Table 2) that lock the pseudo-octahedral $\mathrm{Ni}^{\mathrm{II}}$ atom remain in place. The nickel units show no direct inter-

action with their nearest neighbors in the extended lattice. A solvent channel oriented along the c axis is also evident (Fig. 4). However, the extreme disorder of the solvent does not permit the determination of a suitable model.

Table 3 Experimental details.

	(1)	(2)
Crystal data		
Chemical formula	$[Ni_3(C_{14}H_8N_2O_2)_2(C_{14}H_9N_2O_2)_2-(C_{14}H_{10}N_2O_2)] \cdot 2C_7H_8$	$\begin{array}{l} [Ni_{3}(C_{28}H_{16}BF_{2}N_{4}O_{4})_{2}(C_{14}H_{10}N_{2}O_{2})] \\ \cdot \\ 3CH_{2}Cl_{2} \end{array}$
$M_{\rm r}$	1545.55	1456.88
Crystal system, space group	Monoclinic, $P2_1/c$	Monoclinic, $P2_1/c$
Temperature (K)	100	100
a, b, c (Å)	15.973 (3), 18.639 (3), 22.785 (4)	15.6414 (8), 30.8358 (11), 14.7380 (8)
β (°)	101.757 (4)	112.411 (6)
$V(A^3)$	6641.1 (19)	6571.5 (6)
Z	4	4
Radiation type	Μο Κα	Cu Κα
$\mu (\text{mm}^{-1})$	0.92	1.67
Crystal size (mm)	$0.16 \times 0.11 \times 0.11$	$0.29 \times 0.07 \times 0.04$
Data collection		
Diffractometer	Rigaku Saturn724+	Agilent SuperNova Dual Source diffractometer with an Atlas detector
Absorption correction	Multi-scan (ABSCOR; Higashi, 1995)	Multi-scan (CrysAlis PRO; Agilent, 2014)
T_{\min}, T_{\max}	0.799, 1.000	0.303, 1.000
No. of measured, independent and observed	79592, 15260, 13993	25337, 13041, 8782
$[I > 2\sigma(I)]$ reflections		
R _{int}	0.054	0.060
$(\sin \theta / \lambda)_{\max} (\text{\AA}^{-1})$	0.650	0.626
Refinement		
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.067, 0.151, 1.21	0.077, 0.226, 1.03
No. of reflections	15260	13041
No. of parameters	1046	896
No. of restraints	291	0
H-atom treatment	H atoms treated by a mixture of independent and constrained refinement	H atoms treated by a mixture of independent and constrained refinement
$\Delta \rho_{\rm max}, \Delta \rho_{\rm min} \ (e \ {\rm \AA}^{-3})$	0.56, -0.68	1.16, -0.83

Computer programs: CrystalClear (Rigaku Inc, 2008), CrysAlis PRO (Agilent, 2014), SIR2004 (Burla et al., 2005), OLEX2 (Dolomanov et al., 2009), SHELXL2013 (Sheldrick, 2015), Mercury (Macrae et al., 2006), publCIF (Westrip, 2010) and WinGX (Farrugia, 1999).

4. Synthesis and crystallization

The parent ligand, $pqdH_2$ (0.75 g; 3.1 mmol), was dissolved in 100 ml of ethanol, to which nickel(II) acetate (0.33 g, 1.3 mmol) was added. A red precipitate began to form after approximately 30 min. The solution was then allowed to stir for 1 h, followed by cooling in a freezer and filtration of the crude product (yield: 272 mg, 0.2 mmol, 32%). The resulting product was dissolved in DMF solution and layered with toluene, resulting in the formation of crystals of $[Ni_3(H_2pqd)(Hpqd)_2(pqd)_2]$ after a period of 3–4 d. The crystals grew as red blocks with an asymmetric unit consisting of a complete $[Ni_3(H_2pqd)(Hpqd)_2(pqd)_2]$ molecule and two toluene solvent molecules.

The foregoing complex is stable enough to undergo a fluoridoboration reaction with boron trifluoride, thereby affording the compound $[Ni_3(H_2pqd)(BF_2pqd)_2(pqd)_2]$. $[Ni_3(H_2pqd)(Hpqd)_2(pqd)_2]$ was diluted in diethyl ether, thereby creating a slurry. One ml of 1.0 molar BF₃–OEt₂ (in ether) was then added and the mixture was allowed to react overnight. The resulting precipitate was then filtered off and washed thoroughly with EtOH and Et₂O. The resulting precipitate was then dissolved in dichloromethane (DCM) and filtered through Celite (yield: 43 mg, 30 µmol, 79%). Subsequently, a crop of red block-shaped crystals was grown by solvent evaporation over a period of one day.

5. Refinement

Crystal data, data collection and structure refinement details are summarized in Table 3. In proton-bridged structure (1), atoms H1A, H2A and H4A were found by assignment of difference map peaks and refined isotropically without geometrical constraints. The proton H3A was initially placed with the *SHELXL* HFIX 147 command (refinement on rotation) on O9, but was refined freely. Four distinct hydrogenbonding interactions were evident in the trinuclear cluster. Finally, there were two O-H-O interactions between an oxime and oximato of each [Ni(Hpqd)(pqd)]⁻ unit that could not be resolved due to rapid conversion to [Ni(pqd)(Hpqd)]⁻. All the restraints that are reported were included for the modelling of the disordered toluene solvent molecules.

In the case of BF₂-bridged structure (2), atoms H1A and H2A were affixed to O9 and O10, respectively. They were then refined isotropically without rotational constraints. The SQUEEZE routine (Spek, 2015) as implemented in *PLATON* (Spek, 2009) was used to remove the electron density of three solvent DCM molecules per unit cell (calculated: 134 e^- ; 593 Å³).

Acknowledgements

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Computing details

Data collection: *CrystalClear* (Rigaku Inc, 2008) for (1); *CrysAlis PRO* (Agilent, 2014) for (2). Cell refinement: *CrystalClear* (Rigaku Inc, 2008) for (1); *CrysAlis PRO* (Agilent, 2014) for (2). Data reduction: *CrystalClear* (Rigaku Inc, 2008) for (1); *CrysAlis PRO* (Agilent, 2014) for (2). Program(s) used to solve structure: *SIR2004* (Burla *et al.*, 2005) for (1); *OLEX2* (Dolomanov *et al.*, 2009) for (2). For both compounds, program(s) used to refine structure: *SHELXL2013* (Sheldrick, 2015); molecular graphics: *Mercury* (Macrae *et al.*, 2006); software used to prepare material for publication: *publCIF* (Westrip, 2010) and *WinGX* (Farrugia, 1999).

(1) Bis[μ_2 -9,10-bis(oxidoimino)phenanthrene]bis[μ_2 -10-(oxidoimino)phenanthrene-9-one oxime] (phenanthrene-9,10-dione dioxime)trinickel(II) toluene disolvate

Crystal data

$[Ni_3(C_{14}H_8N_2O_2)_2(C_{14}H_9N_2O_2)_2(C_{14}H_{10}N_2O_2)]$ $\cdot 2C_7H_8$
$M_r = 1545.55$
Monoclinic, $P2_1/c$
a = 15.973 (3) Å
b = 18.639 (3) Å
c = 22.785 (4) Å
$\beta = 101.757 \ (4)^{\circ}$
$V = 6641.1 (19) \text{ Å}^3$
Z = 4

Data collection

Rigaku Saturn724+ (2x2 bin mode) diffractometer Radiation source: Sealed Tube Detector resolution: 28.5714 pixels mm⁻¹ dtprofit.ref scans Absorption correction: multi-scan (ABSCOR; Higashi, 1995) $T_{min} = 0.799, T_{max} = 1.000$

Refinement

Refinement on F^2 Hydrogen siteLeast-squares matrix: fullH atoms treate $R[F^2 > 2\sigma(F^2)] = 0.067$ and constra $wR(F^2) = 0.151$ $w = 1/[\sigma^2(F_o^2)]$ S = 1.21where P = 015260 reflections $(\Delta/\sigma)_{max} = 0.00$ 1046 parameters $\Delta\rho_{max} = 0.56$ e291 restraints $\Delta\rho_{min} = -0.68$

F(000) = 3192 $D_x = 1.546 \text{ Mg m}^{-3}$ Mo K\alpha radiation, $\lambda = 0.71069 \text{ Å}$ Cell parameters from 20424 reflections $\theta = 1.3-31.1^{\circ}$ $\mu = 0.92 \text{ mm}^{-1}$ T = 100 KBlock, red $0.16 \times 0.11 \times 0.11 \text{ mm}$

79592 measured reflections 15260 independent reflections 13993 reflections with $I > 2\sigma(I)$ $R_{int} = 0.054$ $\theta_{max} = 27.5^{\circ}, \ \theta_{min} = 1.8^{\circ}$ $h = -20 \rightarrow 20$ $k = -24 \rightarrow 24$ $l = -29 \rightarrow 29$

Hydrogen site location: mixed H atoms treated by a mixture of independent and constrained refinement $w = 1/[\sigma^2(F_o^2) + (0.0495P)^2 + 7.4732P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} = 0.001$ $\Delta\rho_{max} = 0.56 \text{ e } \text{Å}^{-3}$ $\Delta\rho_{min} = -0.68 \text{ e } \text{Å}^{-3}$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	Occ. (<1)
C1	0.3573 (2)	0.26380 (17)	0.77353 (15)	0.0300 (7)	
C2	0.4143 (2)	0.20289 (17)	0.79218 (14)	0.0287 (6)	
C3	0.5058 (2)	0.21463 (17)	0.81436 (14)	0.0289 (7)	
C4	0.5622 (2)	0.15956 (19)	0.83987 (15)	0.0341 (7)	
H4	0.5433	0.1123	0.8376	0.041*	
C5	0.6447 (2)	0.1749 (2)	0.86803 (16)	0.0392 (8)	
H5	0.6811	0.1381	0.8850	0.047*	
C6	0.6738 (2)	0.2450 (2)	0.87123 (17)	0.0423 (9)	
H6	0.7292	0.2554	0.8913	0.051*	
C7	0.6209 (2)	0.2991 (2)	0.84477 (17)	0.0412 (8)	
H7	0.6418	0.3458	0.8465	0.049*	
C8	0.5364 (2)	0.28614 (18)	0.81521 (15)	0.0318 (7)	
C9	0.4809 (2)	0.34500 (17)	0.78633 (15)	0.0306 (7)	
C10	0.5151 (2)	0.41244 (19)	0.77676 (16)	0.0371 (8)	
H10	0.5735	0.4200	0.7897	0.045*	
C11	0.4652 (2)	0.46755 (19)	0.74890 (17)	0.0403 (8)	
H11	0.4897	0.5119	0.7442	0.048*	
C12	0.3783 (2)	0.45710 (18)	0.72779 (17)	0.0388 (8)	
H12	0.3445	0.4941	0.7085	0.047*	
C13	0.3423 (2)	0.39116 (17)	0.73569 (16)	0.0334 (7)	
H13	0.2841	0.3841	0.7212	0.040*	
C14	0.3922 (2)	0.33465 (16)	0.76530 (15)	0.0294 (7)	
C15	0.3654 (2)	0.24577 (17)	0.93609 (14)	0.0292 (7)	
C16	0.4342 (2)	0.19229 (17)	0.94884 (14)	0.0283 (6)	
C17	0.5203 (2)	0.21590 (17)	0.98016 (14)	0.0293 (7)	
C18	0.5819(2)	0.16511 (18)	1.00548 (14)	0.0327 (7)	
H18	0.5684	0.1165	1.0024	0.039*	
C19	0.6622 (2)	0.1864 (2)	1.03489 (15)	0.0364 (8)	
H19	0.7032	0.1524	1.0507	0.044*	
C20	0.6815 (2)	0.2591 (2)	1.04065 (17)	0.0410 (8)	
H20	0.7354	0.2738	1.0606	0.049*	
C21	0.6211 (2)	0.3092 (2)	1.01689 (16)	0.0396 (8)	
H21	0.6345	0.3577	1.0218	0.048*	
C22	0.5392 (2)	0.28906 (18)	0.98522 (14)	0.0317 (7)	
C23	0.4766 (2)	0.34208 (17)	0.95560 (15)	0.0317 (7)	
C24	0.4995 (2)	0.41366 (19)	0.94928 (17)	0.0386 (8)	
H24	0.5548	0.4282	0.9663	0.046*	
C25	0.4432 (3)	0.4633 (2)	0.91884 (18)	0.0428 (9)	
H25	0.4599	0.5108	0.9166	0.051*	

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\hat{A}^2)

C26	0.3618 (2)	0.44223 (19)	0.89172 (17)	0.0408 (8)
H26	0.3242	0.4751	0.8696	0.049*
C27	0.3360 (2)	0.37214 (18)	0.89741 (16)	0.0363 (8)
H27	0.2809	0.3585	0.8790	0.044*
C28	0.3911 (2)	0.32129 (17)	0.93032 (15)	0.0312 (7)
C29	0.0764 (2)	0.10264 (17)	0.76379 (14)	0.0291 (7)
C30	0.1064 (2)	0.03055 (17)	0.75287 (15)	0.0309 (7)
C31	0.0435 (2)	-0.02258(18)	0.72335 (15)	0.0346 (7)
C32	0.0685 (2)	-0.0803(2)	0.69200 (18)	0.0429 (9)
H32	0.1254	-0.0851	0.6889	0.051*
C33	0.0088(3)	-0.1305(2)	0.6655(2)	0.051
Н33	0.0255	-0.1689	0.6445	0.064*
C34	-0.0749(3)	-0.1236(2)	0.6701(2)	0.0539(11)
H34	-0.1146	-0.1579	0.6526	0.065*
C35	-0.1000(2)	-0.0668(2)	0.0520	0.005
U25	0.1009 (2)	-0.0628	0.70010 (19)	0.0472(9)
1155 C26	0.1301	0.0028 0.01422(10)	0.7020	0.037°
C30	-0.0422(2)	-0.01423(19)	0.72739(10)	0.0300(8)
C37	-0.0676(2)	0.0483(2)	0.75885(10)	0.0360(8)
C38	-0.14/9 (2)	0.0511 (2)	0.77467 (18)	0.0434 (9)
H38	-0.1845	0.0120	0.7661	0.052*
039	-0.1/41 (2)	0.1100 (2)	0.80252 (18)	0.0462 (9)
H39	-0.2275	0.1102	0.8127	0.055*
C40	-0.1214 (2)	0.1683 (2)	0.81519 (18)	0.0450 (9)
H40	-0.1397	0.2086	0.8331	0.054*
C41	-0.0410 (2)	0.1676 (2)	0.80142 (17)	0.0393 (8)
H41	-0.0054	0.2072	0.8109	0.047*
C42	-0.0122 (2)	0.10822 (18)	0.77336 (15)	0.0323 (7)
C43	0.0836 (2)	0.08681 (19)	0.92655 (15)	0.0345 (7)
C44	0.1243 (2)	0.02254 (19)	0.90908 (15)	0.0346 (7)
C45	0.0739 (2)	-0.04203 (19)	0.89031 (16)	0.0378 (8)
C46	0.1065 (3)	-0.1008 (2)	0.86405 (17)	0.0433 (9)
H46	0.1639	-0.1016	0.8616	0.052*
C47	0.0542 (3)	-0.1577 (2)	0.8417 (2)	0.0555 (11)
H47	0.0760	-0.1960	0.8233	0.067*
C48	-0.0306(3)	-0.1576 (3)	0.8468 (2)	0.0633 (13)
H48	-0.0655	-0.1963	0.8323	0.076*
C49	-0.0636(3)	-0.1003 (2)	0.8731 (2)	0.0535 (11)
H49	-0.1209	-0.1011	0.8758	0.064*
C50	-0.0134(2)	-0.0412(2)	0.89578 (16)	0.0424 (9)
C51	-0.0494(2)	0.0194 (2)	0.92335 (16)	0.0408 (9)
C52	-0.1321(3)	0.0157(2)	0.93653(19)	0.0505 (10)
H52	-0.1643	-0.0258	0.9266	0.061*
C53	-0.1662(2)	0 0706 (3)	0.96312(19)	0.0524(11)
H53	-0.2208	0.0661	0.9712	0.063*
C54	-0.1207(2)	0 1331 (3)	0.97825(18)	0.0507 (10)
С <i>3</i> -т Н54	-0.1443	0 1706	0.9964	0.061*
C 5 5	-0.0392(2)	0 1395 (2)	0.96614(17)	0.0430 (9)
UJJ H55	-0.0084	0.1375 (2)	0.0762	0.052*
1155	0.000-	0.1010	0.9702	0.052

C56	-0.0028(2)	0.0833 (2)	0.93893 (15)	0.0379 (8)	
C57	0.4778 (2)	-0.09599 (17)	0.87975 (15)	0.0311 (7)	
C58	0.4046 (2)	-0.13101 (17)	0.83829 (15)	0.0305 (7)	
C59	0.4055 (2)	-0.20914 (17)	0.83084 (14)	0.0320 (7)	
C60	0.3314 (2)	-0.24949 (19)	0.80879 (17)	0.0412 (8)	
H60	0.2797	-0.2260	0.7955	0.049*	
C61	0.3342 (3)	-0.3230(2)	0.80656 (19)	0.0508 (10)	
H61	0.2842	-0.3488	0.7931	0.061*	
C62	0.4101 (3)	-0.3585(2)	0.82407 (19)	0.0534 (11)	
H62	0.4120	-0.4082	0.8214	0.064*	
C63	0.4839(3)	-0.3203(2)	0.84575 (19)	0.0487 (10)	
H63	0.5352	-0.3449	0.8576	0.058*	
C64	0.4835(2)	-0.24541(18)	0.85028 (16)	0.0367 (8)	
C65	0.5628(2)	-0.20611(19)	0.87678 (16)	0.0367(8)	
C66	0.6421(2)	-0.2409(2)	0.88592 (18)	0.0453(9)	
H66	0.6448	-0.2878	0.8724	0.054*	
C67	0.7154(3)	-0.2078(2)	0.91426(19)	0.021	
H67	0.7673	-0.2321	0.9199	0.058*	
C68	0.7073 0.7123(2)	-0.1384(2)	0.93455(18)	0.038 0.0441 (9)	
H68	0.7622	-0.1167	0.9550	0.053*	
C69	0.7022 0.6367 (2)	-0.10049(19)	0.92512 (16)	0.035	
H69	0.6360	-0.0534	0.92912 (10)	0.0372(0)	
C70	0.5598(2)	-0.13368(18)	0.89503 (15)	0.043	
C71	0.3370(2) 0.1134(3)	0.1368 (4)	0.5502(3)	0.0321(7)	
U71	0.0755	0.1011	0.5302 (5)	0.150*	
H71R	0.1188	0.1755	0.5235	0.150*	
H71C	0.0906	0.1547	0.5233	0.150*	
C72	0.0900	0.1047 0.1037 (3)	0.5052 0.5736 (2)	0.150 0.0596 (12)	
C73	0.2598(3)	0.1037(3) 0.0972(2)	0.5750(2) 0.5371(2)	0.0590(12) 0.0513(10)	
U73	0.2358 (5)	0.1145	0.3371 (2)	0.062*	
C74	0.2400 0.3375(3)	0.1145	0.4980 0.55702 (10)	0.002	
U74	0.3375 (3)	0.0618	0.53792 (19)	0.0501 (10)	
11/4 C75	0.3707 0.3582 (3)	0.0018	0.5550 0.61574(10)	0.000°	
U75	0.3382 (3)	0.0391(2)	0.6204	0.0502 (10)	
1175 C76	0.4100 0.2012 (2)	0.0107	0.0294	0.000°	
U76	0.3013 (3)	0.0400 (3)	0.0327(2)	0.0034 (13)	
П/0 С77	0.3130 0.2225 (2)	0.0287 0.0787 (3)	0.0917 0.6320 (2)	0.070°	
U77	0.2255 (5)	0.0787 (3)	0.0320 (2)	0.0722(13)	
П// С79	-0.1857	0.0842	0.0378 0.4357(4)	0.087°	0.66
C70	-0.1800(3)	0.0902 (4)	0.4337 (4)	0.087 (2)	0.00
П/8А 1179D	-0.1443	0.0383	0.4248	0.131*	0.00
П/0D	-0.1928	0.1314	0.4099	0.131*	0.00
H/8C	-0.2397	0.0050	0.4315	0.131^{*}	0.00
C/9	-0.15/2(3)	0.1133(3)	0.4982 (3)	0.0/2/(19)	0.66
	-0.1/92 (5)	0.0748 (4)	0.5449 (5)	0.080 (2)	0.66
H8U C91	-0.2124	0.0064 (5)	0.5364	0.090*	0.66
	-0.1524 (6)	0.0964 (5)	0.0039 (4)	0.080(3)	0.66
H81	-0.1081	0.0703	0.0348	0.103*	0.66
C82	-0.1021 (6)	0.1574(5)	0.616/(3)	0.082 (3)	0.66

H82	-0.0833	0.1721	0.6562	0.098*	0.66
C83	-0.0800 (5)	0.1963 (4)	0.5703 (3)	0.083 (2)	0.66
H83	-0.0466	0.2373	0.5789	0.100*	0.66
C84	-0.1068 (4)	0.1750 (3)	0.5113 (3)	0.0769 (19)	0.66
H84	-0.0916	0.2015	0.4805	0.092*	0.66
C78A	-0.1035 (9)	0.1587 (8)	0.4495 (6)	0.078 (4)	0.34
H78D	-0.0889	0.2087	0.4523	0.118*	0.34
H78E	-0.1505	0.1513	0.4163	0.118*	0.34
H78F	-0.0551	0.1314	0.4433	0.118*	0.34
C79A	-0.1290 (7)	0.1344 (6)	0.5073 (7)	0.080 (3)	0.34
C81A	-0.2097 (10)	0.0579 (9)	0.5603 (8)	0.082 (3)	0.34
H81A	-0.2488	0.0208	0.5596	0.098*	0.34
C80A	-0.1873 (8)	0.0791 (7)	0.5038 (8)	0.079 (3)	0.34
H80A	-0.2108	0.0567	0.4677	0.094*	0.34
C82A	-0.1790 (13)	0.0872 (12)	0.6100 (11)	0.082 (3)	0.34
H82A	-0.1962	0.0714	0.6444	0.098*	0.34
C83A	-0.1209 (15)	0.1416 (12)	0.6139 (7)	0.082 (3)	0.34
H83A	-0.0979	0.1629	0.6505	0.099*	0.34
C84A	-0.0965 (12)	0.1648 (9)	0.5594 (6)	0.081 (2)	0.34
H84A	-0.0572	0.2019	0.5609	0.097*	0.34
N1	0.27646 (17)	0.24607 (14)	0.76671 (13)	0.0330 (6)	
N2	0.37143 (17)	0.14219 (14)	0.79170 (13)	0.0302 (6)	
N3	0.28561 (17)	0.22388 (14)	0.93075 (12)	0.0312 (6)	
N4	0.42751 (17)	0.12499 (14)	0.93553 (12)	0.0307 (6)	
N5	0.13044 (18)	0.15645 (14)	0.76514 (14)	0.0337 (6)	
N6	0.18439 (17)	0.00730 (14)	0.76635 (13)	0.0308 (6)	
N7	0.13358 (18)	0.14323 (15)	0.93093 (13)	0.0345 (6)	
N8	0.20644 (18)	0.03278 (15)	0.90921 (13)	0.0330 (6)	
N9	0.45806 (17)	-0.03173 (14)	0.89580 (12)	0.0301 (6)	
N10	0.34564 (17)	-0.08566 (14)	0.81453 (13)	0.0329 (6)	
01	0.21502 (15)	0.29635 (12)	0.75199 (13)	0.0424 (6)	
O2	0.41168 (14)	0.07989 (11)	0.80519 (11)	0.0333 (5)	
O3	0.22535 (15)	0.27495 (13)	0.92507 (12)	0.0399 (6)	
O4	0.34924 (14)	0.09770 (12)	0.91047 (11)	0.0342 (5)	
05	0.09813 (16)	0.22254 (13)	0.76779 (14)	0.0454 (7)	
O6	0.24766 (14)	0.05433 (12)	0.79020 (11)	0.0331 (5)	
O7	0.10615 (16)	0.20776 (14)	0.94544 (13)	0.0445 (6)	
08	0.25669 (15)	-0.02059 (12)	0.89833 (11)	0.0361 (5)	
09	0.52088 (17)	0.00669 (14)	0.93202 (12)	0.0404 (6)	
O10	0.27559 (16)	-0.11284 (13)	0.77446 (12)	0.0395 (6)	
Ni1	0.25135 (3)	0.14989 (2)	0.77620 (2)	0.03005 (11)	
Ni2	0.24612 (3)	0.12803 (2)	0.92204 (2)	0.03014 (11)	
Ni3	0.34675 (3)	0.01380 (2)	0.85238 (2)	0.02978 (11)	
H1A	0.149 (4)	0.262 (3)	0.761 (3)	0.10 (2)*	
H2A	0.166 (3)	0.244 (3)	0.935 (2)	0.081 (16)*	
H3A	0.488 (4)	0.043 (3)	0.933 (3)	0.09 (2)*	
H4A	0.233 (3)	-0.070 (3)	0.770 (2)	0.095 (18)*	

Atomic displacement parameters $(Å^2)$

	U^{11}	U ²²	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0263 (16)	0.0267 (15)	0.0376 (17)	-0.0007 (13)	0.0076 (13)	0.0027 (13)
C2	0.0271 (16)	0.0272 (15)	0.0329 (16)	0.0010 (13)	0.0088 (13)	0.0019 (13)
C3	0.0274 (16)	0.0301 (16)	0.0318 (16)	-0.0026 (13)	0.0120 (13)	-0.0014 (13)
C4	0.0296 (17)	0.0359 (18)	0.0379 (18)	0.0044 (14)	0.0095 (14)	0.0015 (14)
C5	0.0292 (18)	0.047 (2)	0.0414 (19)	0.0061 (16)	0.0076 (15)	0.0005 (16)
C6	0.0243 (17)	0.055 (2)	0.046 (2)	-0.0013 (16)	0.0035 (15)	-0.0013 (18)
C7	0.0321 (19)	0.045 (2)	0.046 (2)	-0.0098 (16)	0.0084 (15)	-0.0027 (17)
C8	0.0286 (17)	0.0349 (17)	0.0343 (17)	-0.0010 (14)	0.0122 (13)	-0.0019 (14)
C9	0.0319 (17)	0.0291 (16)	0.0334 (16)	-0.0043 (13)	0.0125 (13)	-0.0049 (13)
C10	0.0361 (19)	0.0343 (18)	0.0448 (19)	-0.0093 (15)	0.0173 (15)	-0.0043 (15)
C11	0.049 (2)	0.0278 (17)	0.048 (2)	-0.0082 (16)	0.0199 (17)	0.0011 (15)
C12	0.048 (2)	0.0265 (16)	0.046 (2)	0.0013 (15)	0.0182 (17)	0.0047 (15)
C13	0.0344 (18)	0.0274 (16)	0.0396 (18)	0.0021 (14)	0.0104 (14)	0.0029 (14)
C14	0.0313 (17)	0.0237 (15)	0.0354 (16)	-0.0018 (13)	0.0118 (13)	0.0005 (13)
C15	0.0300 (17)	0.0285 (16)	0.0298 (15)	-0.0013 (13)	0.0080 (13)	-0.0011 (13)
C16	0.0288 (16)	0.0261 (15)	0.0309 (15)	-0.0018 (13)	0.0081 (13)	-0.0004 (12)
C17	0.0282 (16)	0.0318 (16)	0.0285 (15)	-0.0034 (13)	0.0072 (12)	-0.0013 (13)
C18	0.0335 (18)	0.0324 (17)	0.0321 (16)	-0.0039 (14)	0.0062 (14)	-0.0003 (14)
C19	0.0315 (18)	0.043 (2)	0.0338 (17)	0.0031 (15)	0.0052 (14)	0.0020 (15)
C20	0.0312 (18)	0.049 (2)	0.0400 (19)	-0.0079 (16)	-0.0001 (15)	-0.0026 (16)
C21	0.039 (2)	0.0372 (19)	0.0425 (19)	-0.0116 (16)	0.0077 (16)	-0.0027 (16)
C22	0.0322 (17)	0.0329 (17)	0.0312 (16)	-0.0055 (14)	0.0095 (13)	-0.0022 (13)
C23	0.0345 (18)	0.0299 (16)	0.0328 (16)	-0.0010 (14)	0.0116 (14)	-0.0016 (13)
C24	0.040 (2)	0.0316 (17)	0.047 (2)	-0.0064 (15)	0.0132 (16)	-0.0014 (15)
C25	0.053 (2)	0.0287 (17)	0.050 (2)	-0.0040 (16)	0.0183 (18)	-0.0005 (16)
C26	0.048 (2)	0.0298 (17)	0.046 (2)	0.0042 (16)	0.0131 (17)	0.0056 (15)
C27	0.039 (2)	0.0315 (17)	0.0391 (18)	0.0016 (15)	0.0109 (15)	0.0002 (14)
C28	0.0335 (17)	0.0283 (16)	0.0341 (17)	-0.0011 (14)	0.0123 (14)	-0.0011 (13)
C29	0.0243 (16)	0.0294 (16)	0.0328 (16)	0.0010 (13)	0.0043 (12)	0.0051 (13)
C30	0.0267 (16)	0.0299 (16)	0.0355 (17)	-0.0019 (13)	0.0053 (13)	0.0030 (13)
C31	0.0344 (18)	0.0314 (17)	0.0359 (17)	-0.0036 (14)	0.0026 (14)	0.0023 (14)
C32	0.037 (2)	0.041 (2)	0.048 (2)	-0.0034 (16)	0.0033 (16)	-0.0064 (17)
C33	0.050 (3)	0.040 (2)	0.064 (3)	-0.0043 (19)	0.001 (2)	-0.0108 (19)
C34	0.047 (2)	0.042 (2)	0.065 (3)	-0.0139 (19)	-0.006 (2)	-0.003 (2)
C35	0.034 (2)	0.043 (2)	0.060 (2)	-0.0075 (17)	-0.0004 (17)	0.0005 (19)
C36	0.0312 (18)	0.0355 (18)	0.0403 (19)	-0.0055 (14)	0.0007 (14)	0.0057 (15)
C37	0.0289 (17)	0.0405 (19)	0.0374 (18)	-0.0029 (15)	0.0040 (14)	0.0083 (15)
C38	0.0283 (18)	0.052 (2)	0.049 (2)	-0.0044 (16)	0.0051 (16)	0.0110 (18)
C39	0.0297 (19)	0.063 (3)	0.048 (2)	0.0074 (18)	0.0115 (16)	0.0120 (19)
C40	0.033 (2)	0.057 (2)	0.047 (2)	0.0098 (18)	0.0127 (16)	0.0050 (18)
C41	0.0339 (19)	0.0388 (19)	0.046 (2)	0.0010 (15)	0.0088 (15)	-0.0008 (16)
C42	0.0255 (16)	0.0349 (17)	0.0357 (17)	0.0006 (13)	0.0043 (13)	0.0063 (14)
C43	0.0300 (17)	0.0390 (18)	0.0344 (17)	-0.0011 (14)	0.0059 (13)	0.0044 (14)
C44	0.0328 (18)	0.0368 (18)	0.0328 (17)	-0.0041 (14)	0.0035 (14)	0.0078 (14)
C45	0.038 (2)	0.0362 (18)	0.0372 (18)	-0.0071 (15)	0.0031 (15)	0.0051 (15)

C47 0.063 (3) 0.044 (2) 0.058 (3) -0.017 (2) 0.008 (2) C48 0.056 (3) 0.058 (3) 0.071 (3) -0.030 (2) 0.001 (2) C49 0.043 (2) 0.054 (3) 0.060 (3) -0.017 (2) 0.0017 (19) C50 0.038 (2) 0.047 (2) 0.0387 (19) -0.0151 (17) 0.0003 (15) C51 0.0311 (19) 0.053 (2) 0.049 (2) -0.0115 (19) 0.0057 (17) C52 0.037 (2) 0.064 (3) 0.049 (2) -0.0115 (19) 0.0057 (17) C53 0.030 (2) 0.078 (3) 0.051 (2) 0.002 (2) 0.0115 (17) C54 0.035 (2) 0.074 (3) 0.046 (2) 0.005 (2) 0.0135 (17)	-0.005(2) -0.004(2) 0.002(2) 0.0102(17) 0.0130(16) 0.016(2)
C48 0.056 (3) 0.058 (3) 0.071 (3) -0.030 (2) 0.001 (2) C49 0.043 (2) 0.054 (3) 0.060 (3) -0.017 (2) 0.0017 (19) C50 0.038 (2) 0.047 (2) 0.0387 (19) -0.0151 (17) 0.0003 (15) C51 0.0311 (19) 0.053 (2) 0.0361 (18) -0.0055 (16) 0.0019 (14) C52 0.037 (2) 0.064 (3) 0.049 (2) -0.0115 (19) 0.0057 (17) C53 0.030 (2) 0.078 (3) 0.051 (2) 0.002 (2) 0.0115 (17) C54 0.035 (2) 0.074 (3) 0.046 (2) 0.0055 (2) 0.0135 (17)	-0.004 (2) 0.002 (2) 0.0102 (17) 0.0130 (16) 0.016 (2)
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C530.030 (2)0.078 (3)0.051 (2)0.002 (2)0.0115 (17)C540.035 (2)0.074 (3)0.046 (2)0.005 (2)0.0135 (17)	
C54 0.035 (2) 0.074 (3) 0.046 (2) 0.005 (2) 0.0135 (17)	0.019 (2)
	0.011 (2)
C55 0.0318 (19) 0.054 (2) 0.045 (2) 0.0010 (17) 0.0108 (16)	0.0073 (18)
C56 $0.0281(17)$ $0.050(2)$ $0.0341(17)$ $-0.0037(16)$ $0.0025(14)$	0.0097 (16)
C57 0.0320 (17) 0.0263 (15) 0.0364 (17) 0.0007 (13) 0.0102 (14)	0.0033 (13)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.0031 (13)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.0001 (13)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.0007(16)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	-0.0026(17)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	-0.0054(17)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.0004 (17)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.0008 (14)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.0062 (15)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.0073(17)
C67 0.038 (2) 0.047 (2) 0.060 (2) 0.0133 (18) 0.0112 (18)	0.0134 (19)
C68 0.0332 (19) 0.047 (2) 0.050 (2) 0.0003 (16) 0.0055 (16)	0.0147 (18)
C69 0.0358 (19) 0.0345 (18) 0.0411 (19) 0.0052 (15) 0.0074 (15)	0.0093 (15)
C70 0.0312 (17) 0.0303 (16) 0.0356 (17) 0.0021 (14) 0.0085 (14)	0.0062 (14)
C71 0.055 (3) 0.108 (5) 0.138 (6) 0.015 (3) 0.025 (4)	0.052 (4)
C72 0.047 (3) 0.054 (3) 0.080 (3) 0.000 (2) 0.019 (2)	0.015 (2)
C73 0.055 (3) 0.052 (2) 0.050 (2) -0.001 (2) 0.0157 (19)	0.0116 (19)
C74 0.059 (3) 0.045 (2) 0.053 (2) 0.000 (2) 0.026 (2)	-0.0001 (19)
C75 0.054 (3) 0.045 (2) 0.053 (2) 0.0034 (19) 0.014 (2)	0.0046 (19)
C76 0.070 (3) 0.074 (3) 0.052 (3) 0.018 (3) 0.025 (2)	0.016 (2)
C77 0.075 (3) 0.083 (4) 0.071 (3) 0.017 (3) 0.045 (3)	0.020 (3)
C78 0.110 (6) 0.077 (5) 0.074 (5) -0.002 (5) 0.018 (5)	-0.001 (4)
C79 0.081 (4) 0.068 (4) 0.068 (3) 0.006 (4) 0.014 (3)	-0.012 (3)
C80 0.100 (5) 0.069 (4) 0.067 (4) 0.005 (4) 0.006 (4)	-0.005 (3)
C81 0.105 (5) 0.069 (4) 0.075 (4) 0.007 (4) -0.001 (4)	0.004 (4)
C82 0.097 (5) 0.070 (5) 0.069 (4) 0.009 (4) -0.005 (4)	-0.009 (3)
C83 0.088 (4) 0.078 (5) 0.080 (4) 0.007 (4) 0.007 (4)	-0.015 (4)
C84 0.080 (4) 0.075 (4) 0.076 (4) 0.010 (4) 0.018 (4)	-0.010 (4)
C78A 0.087 (8) 0.089 (8) 0.067 (7) 0.002 (7) 0.035 (6)	0.000(7)
C79A 0.090 (5) 0.074 (5) 0.074 (4) 0.010 (4) 0.015 (4)	-0.008 (4)
C81A 0.098 (6) 0.074 (6) 0.070 (5) 0.002 (5) 0.010 (5)	-0.003 (5)
C80A 0.093 (6) 0.070 (5) 0.072 (5) 0.001 (5) 0.014 (5)	-0.005 (5)
C82A 0.097 (6) 0.072 (5) 0.071 (5) 0.007 (5) 0.004 (5)	0.000 (4)
C83A 0.095 (6) 0.074 (5) 0.071 (4) 0.006 (5) -0.003 (5)	-0.005 (5)
C84A 0.090 (5) 0.074 (5) 0.075 (4) 0.008 (4) 0.007 (4)	-0.009 (4)
N1 0.0260 (14) 0.0272 (14) 0.0453 (16) 0.0056 (11) 0.0061 (12)	0.0058 (12)
N2 0.0292 (14) 0.0235 (13) 0.0386 (15) 0.0031 (11) 0.0087 (12)	0.0020 (11)

N3	0.0283 (14)	0.0289 (14)	0.0366 (15)	0.0044 (11)	0.0072 (11)	0.0010 (11)
N4	0.0267 (14)	0.0288 (13)	0.0366 (14)	-0.0035 (11)	0.0064 (11)	-0.0041 (11)
N5	0.0284 (15)	0.0259 (13)	0.0478 (17)	0.0024 (11)	0.0104 (12)	0.0030 (12)
N6	0.0275 (14)	0.0251 (13)	0.0394 (15)	-0.0040 (11)	0.0060 (12)	0.0006 (11)
N7	0.0304 (15)	0.0345 (15)	0.0379 (15)	-0.0016 (12)	0.0058 (12)	0.0002 (12)
N8	0.0307 (15)	0.0301 (14)	0.0389 (15)	-0.0007 (12)	0.0084 (12)	0.0027 (12)
N9	0.0279 (14)	0.0247 (13)	0.0365 (14)	-0.0019 (11)	0.0034 (11)	0.0023 (11)
N10	0.0294 (15)	0.0264 (13)	0.0415 (16)	-0.0013 (11)	0.0039 (12)	-0.0017 (12)
01	0.0271 (13)	0.0255 (12)	0.0740 (19)	0.0064 (10)	0.0091 (12)	0.0141 (12)
O2	0.0295 (12)	0.0231 (11)	0.0480 (14)	0.0047 (9)	0.0098 (10)	0.0017 (10)
03	0.0292 (13)	0.0282 (12)	0.0627 (17)	0.0058 (10)	0.0104 (11)	0.0020 (11)
O4	0.0255 (12)	0.0296 (12)	0.0474 (14)	-0.0045 (9)	0.0072 (10)	-0.0069 (10)
05	0.0288 (13)	0.0255 (12)	0.084 (2)	0.0048 (10)	0.0163 (13)	0.0077 (13)
06	0.0249 (12)	0.0255 (11)	0.0468 (14)	-0.0039(9)	0.0021 (10)	0.0009 (10)
07	0.0333 (14)	0.0368 (14)	0.0663 (18)	0.0036 (11)	0.0169 (12)	-0.0043 (13)
08	0.0344 (13)	0.0292 (12)	0.0449 (14)	-0.0010 (10)	0.0087 (11)	0.0005 (10)
09	0.0337 (14)	0.0317 (13)	0.0506 (15)	-0.0002 (11)	-0.0038 (11)	-0.0058 (11)
O10	0.0349 (14)	0.0309 (12)	0.0471 (15)	0.0019 (11)	-0.0049 (11)	-0.0071 (11)
Ni1	0.0237 (2)	0.0232 (2)	0.0434 (2)	0.00088 (16)	0.00702 (17)	0.00455 (17)
Ni2	0.0251 (2)	0.0275 (2)	0.0382 (2)	-0.00170 (16)	0.00718 (17)	-0.00033 (17)
Ni3	0.0274 (2)	0.0223 (2)	0.0386 (2)	0.00040 (16)	0.00435 (17)	0.00011 (17)

Geometric parameters (Å, °)

C1—N1	1.310 (4)	С55—Н55	0.9300
C1—C14	1.461 (4)	C57—N9	1.310 (4)
C1—C2	1.464 (4)	C57—C70	1.465 (5)
C2—N2	1.322 (4)	C57—C58	1.495 (5)
C2—C3	1.462 (4)	C58—N10	1.299 (4)
C3—C4	1.410 (5)	C58—C59	1.467 (4)
C3—C8	1.418 (4)	C59—C60	1.406 (5)
C4—C5	1.373 (5)	C59—C64	1.408 (5)
C4—H4	0.9300	C60—C61	1.372 (5)
C5—C6	1.383 (5)	C60—H60	0.9300
С5—Н5	0.9300	C61—C62	1.368 (6)
C6—C7	1.374 (5)	C61—H61	0.9300
С6—Н6	0.9300	C62—C63	1.380 (6)
С7—С8	1.400 (5)	C62—H62	0.9300
С7—Н7	0.9300	C63—C64	1.399 (5)
С8—С9	1.479 (5)	С63—Н63	0.9300
C9—C10	1.406 (4)	C64—C65	1.481 (5)
C9—C14	1.412 (5)	C65—C66	1.400 (5)
C10-C11	1.373 (5)	C65—C70	1.416 (5)
C10—H10	0.9300	C66—C67	1.364 (6)
C11—C12	1.387 (5)	C66—H66	0.9300
C11—H11	0.9300	C67—C68	1.379 (6)
C12—C13	1.384 (5)	C67—H67	0.9300
C12—H12	0.9300	C68—C69	1.378 (5)

C13—H13 0.9300 C69—C70 C15—N3 1.321 (4) C69—H69 C15—C16 1.468 (4) C71—C72 C15—C28 1.479 (4) C71—H71A C16—N4 1.290 (4) C71—H71B C16—C17 1.481 (4) C71—H71C C17—C22 1.396 (4) C72—C73 C17—C18 1.401 (5) C72—C77 C18—C19 1.380 (5) C73—C74 C18—H18 0.9300 C73—H73	1.420 (5) 0.9300 1.517 (7) 0.9600 0.9600 1.387 (6)
C15N31.321 (4)C69H69C15C161.468 (4)C71C72C15C281.479 (4)C71H71AC16N41.290 (4)C71H71BC16C171.481 (4)C71H71CC17C221.396 (4)C72C73C17C181.401 (5)C72C77C18C191.380 (5)C73C74C18H180.9300C73H73	0.9300 1.517 (7) 0.9600 0.9600 0.9600 1.387 (6)
C15—C16 1.468 (4) C71—C72 C15—C28 1.479 (4) C71—H71A C16—N4 1.290 (4) C71—H71B C16—C17 1.481 (4) C71—H71C C17—C22 1.396 (4) C72—C73 C17—C18 1.401 (5) C72—C77 C18—C19 1.380 (5) C73—C74 C18—H18 0.9300 C73—H73	1.517 (7) 0.9600 0.9600 0.9600 1.387 (6)
C15—C28 1.479 (4) C71—H71A C16—N4 1.290 (4) C71—H71B C16—C17 1.481 (4) C71—H71C C17—C22 1.396 (4) C72—C73 C17—C18 1.401 (5) C72—C77 C18—C19 1.380 (5) C73—C74 C18—H18 0.9300 C73—H73	0.9600 0.9600 0.9600 1.387 (6)
C16—N41.290 (4)C71—H71BC16—C171.481 (4)C71—H71CC17—C221.396 (4)C72—C73C17—C181.401 (5)C72—C77C18—C191.380 (5)C73—C74C18—H180.9300C73—H73	0.9600 0.9600 1.387 (6)
C16—C17 1.481 (4) C71—H71C C17—C22 1.396 (4) C72—C73 C17—C18 1.401 (5) C72—C77 C18—C19 1.380 (5) C73—C74 C18—H18 0.9300 C73—H73	0.9600 1.387 (6)
C17—C22 1.396 (4) C72—C73 C17—C18 1.401 (5) C72—C77 C18—C19 1.380 (5) C73—C74 C18—H18 0.9300 C73—H73	1.387 (6)
C17—C18 1.401 (5) C72—C77 C18—C19 1.380 (5) C73—C74 C18—H18 0.9300 C73—H73	
C18—C19 1.380 (5) C73—C74 C18—H18 0.9300 C73—H73	1.388 (7)
C18—H18 0.9300 C73—H73	1.369 (6)
	0.9300
C19—C20 1.390 (5) C74—C75	1.382 (6)
C19—H19 0.9300 C74—H74	0.9300
C20—C21 1.372 (5) C75—C76	1.365 (6)
С20—Н20 0.9300 С75—Н75	0.9300
C21—C22 1.410 (5) C76—C77	1.377 (7)
C21—H21 0.9300 C76—H76	0.9300
C22—C23 1.469 (5) C77—H77	0.9300
C23—C24 1.399 (5) C78—C79	1.470 (10)
C23—C28 1.424 (5) C78—H78A	0.9600
C24—C25 1.376 (5) C78—H78B	0.9600
C24—H24 0.9300 C78—H78C	0.9600
C_{25} — C_{26} $1.378(5)$ C_{79} — C_{80}	1.386 (6)
C25—H25 0.9300 C79—C84	1.401 (6)
$C_{26} - C_{27}$ 1.384 (5) $C_{80} - C_{81}$	1.386 (7)
C26—H26 0.9300 C80—H80	0.9300
C27-C28 1.402 (5) $C81-C82$	1.390 (7)
C27—H27 0.9300 C81—H81	0.9300
C29-N5 1.319 (4) $C82-C83$	1.386 (7)
C29-C30 1.465 (4) $C82-H82$	0.9300
C29-C42 1.479 (4) $C83-C84$	1.384 (6)
C_{30} —N6 1.295 (4) C_{83} —H83	0.9300
C30—C31 1.472 (5) C84—H84	0.9300
C31—C32 1.394 (5) C78A—C79A	1.523 (18)
C31—C36 1.399 (5) C78A—H78D	0.9600
C32—C33 1.384 (5) C78A—H78E	0.9600
C32—H32 0.9300 C78A—H78F	0.9600
C33—C34 1.367 (6) C79A—C84A	1.32 (2)
C33—H33 0.9300 C79A—C80A	1.382 (8)
C34—C35 1.372 (6) C81A—C82A	1.26 (3)
C34—H34 0.9300 C81A—C80A	1.46 (2)
	0.9300
C35—C36 1.410 (5) C81A—H81A	0.0300
C35—C36 1.410 (5) C81A—H81A C35—H35 0.9300 C80A—H80A	0.7300
C35—C361.410 (5)C81A—H81AC35—H350.9300C80A—H80AC36—C371.469 (5)C82A—C83A	1.36 (3)
C35—C361.410 (5)C81A—H81AC35—H350.9300C80A—H80AC36—C371.469 (5)C82A—C83AC37—C381.403 (5)C82A—H82A	1.36 (3) 0.9300
C35—C36 1.410 (5) C81A—H81A C35—H35 0.9300 C80A—H80A C36—C37 1.469 (5) C82A—C83A C37—C38 1.403 (5) C82A—H82A C37—C42 1.420 (5) C83A—C84A	1.36 (3) 0.9300 1.44 (2)
C35—C36 1.410 (5) C81A—H81A C35—H35 0.9300 C80A—H80A C36—C37 1.469 (5) C82A—C83A C37—C38 1.403 (5) C82A—H82A C37—C42 1.420 (5) C83A—C84A C38—C39 1.375 (6) C83A—H83A	1.36 (3) 0.9300 1.44 (2) 0.9300

C39—C40	1.370 (6)	N1-01	1.349 (3)
С39—Н39	0.9300	N1—Ni1	1.859 (3)
C40—C41	1.382 (5)	N2—O2	1.333 (3)
C40—H40	0.9300	N2—Ni1	1.884 (3)
C41—C42	1.401 (5)	N3—O3	1.341 (3)
C41—H41	0.9300	N3—Ni2	1.892 (3)
C43—N7	1.311 (4)	N4—O4	1.363 (3)
C43—C44	1.457 (5)	N5—O5	1.342 (3)
C43—C56	1.465 (5)	N5—Nil	1.900 (3)
C44—N8	1.325 (4)	N6—O6	1.364 (3)
C44—C45	1.463 (5)	N7—O7	1.344 (4)
C45—C46	1.399 (5)	N7—Ni2	1.871 (3)
C45—C50	1.425 (5)	N8-08	1.333 (4)
C46—C47	1.381(5)	N8—Ni2	1.888(3)
C_{46} H46	0.9300	N9-09	1.000(3) 1.365(4)
C47 - C48	1 383 (6)	N9—Ni3	2.035(3)
C47 H47	0.0300	N10 010	1.388(4)
C_{4}^{2} C_{4}^{2} C_{4}^{2}	1 291 (7)	N10 N32	1.300(+)
C_{40} U_{49}	1.301(7)		2.045(5)
C40 C50	0.9300		1.29(0)
C49 - C30	1.399 (5)	02-N13	2.050(2)
C49—H49	0.9300	O3—HZA	1.17(5)
C50—C51	1.465 (6)	04—N12	1.811 (2)
C51—C56	1.411 (5)	04—N13	2.044 (2)
C51—C52	1.416 (5)	O5—HIA	1.13 (6)
C52—C53	1.359 (6)	O6—Ni1	1.813 (2)
С52—Н52	0.9300	06—Ni3	2.042 (2)
C53—C54	1.378 (6)	O7—H2A	1.24 (6)
С53—Н53	0.9300	O8—Ni3	2.047 (2)
C54—C55	1.389 (5)	O9—H3A	0.85 (6)
C54—H54	0.9300	O10—H4A	1.04 (6)
C55—C56	1.402 (5)		
N1—C1—C14	127.3 (3)	C60—C59—C58	123.0 (3)
N1—C1—C2	112.3 (3)	C64—C59—C58	118.0 (3)
C14—C1—C2	120.4 (3)	C61—C60—C59	121.1 (4)
N2—C2—C3	127.7 (3)	C61—C60—H60	119.5
N2—C2—C1	111.7 (3)	С59—С60—Н60	119.5
C3—C2—C1	120.3 (3)	C62—C61—C60	120.4 (4)
C4—C3—C8	119.3 (3)	С62—С61—Н61	119.8
C4—C3—C2	122.9 (3)	С60—С61—Н61	119.8
C8-C3-C2	1175(3)	C61 - C62 - C63	1198(4)
$C_{5} - C_{4} - C_{3}$	120 8 (3)	C61 - C62 - H62	120.1
C5-C4-H4	119.6	C63 - C62 - H62	120.1
C3—C4—H4	119.6	C62 - C63 - C64	121.7(4)
C4-C5-C6	120 1 (3)	C62 - C63 - H63	110.2
C4—C5—H5	120.1 (3)	C64—C63—H63	119.2
C6_C5_H5	120.0	C63 - C64 - C59	118.2 (3)
C_{0}	120.0 (3)	$C_{03} - C_{04} - C_{05}$	120.2(3)
$\cup - \cup - \cup - \cup $	120.0 (3)	-03 - 004 - 003	120.3 (3)

С7—С6—Н6	120.0	C59—C64—C65	121.3 (3)
С5—С6—Н6	120.0	C66—C65—C70	118.8 (3)
C6—C7—C8	122.1 (4)	C66—C65—C64	120.2 (3)
С6—С7—Н7	119.0	C70—C65—C64	120.9 (3)
С8—С7—Н7	119.0	C67—C66—C65	121.6 (4)
C7—C8—C3	117.6 (3)	С67—С66—Н66	119.2
C7—C8—C9	121.2 (3)	С65—С66—Н66	119.2
C3—C8—C9	121.2 (3)	C66—C67—C68	119.8 (4)
C10—C9—C14	117.7 (3)	С66—С67—Н67	120.1
C10—C9—C8	121.0 (3)	С68—С67—Н67	120.1
C14—C9—C8	121.3 (3)	C69—C68—C67	121.3 (4)
C11—C10—C9	122.2 (3)	С69—С68—Н68	119.3
C11—C10—H10	118.9	С67—С68—Н68	119.3
C9-C10-H10	118.9	C68 - C69 - C70	119.7 (3)
C10-C11-C12	120.0 (3)	C68—C69—H69	120.1
C10-C11-H11	120.0	C70—C69—H69	120.1
C12—C11—H11	120.0	$C_{65} - C_{70} - C_{69}$	120.1 1187(3)
C_{13} C_{12} C_{11}	119 5 (3)	C65 - C70 - C57	118.0(3)
C13 - C12 - H12	120.2	C69 - C70 - C57	1233(3)
C11 - C12 - H12	120.2	C72-C71-H71A	109 5
C12 - C13 - C14	120.2	C72 - C71 - H71B	109.5
C12—C13—H13	119.4	H71A - C71 - H71B	109.5
C14—C13—H13	119.4	C72-C71-H71C	109.5
C13 - C14 - C9	1194(3)	H71A - C71 - H71C	109.5
C_{13} C_{14} C_{1}	122.9 (3)	H71B-C71-H71C	109.5
C9-C14-C1	117.6 (3)	C73 - C72 - C77	117.9 (4)
N3-C15-C16	118.6 (3)	C73 - C72 - C71	121.0(5)
N3-C15-C28	124.4 (3)	C77—C72—C71	121.2 (5)
$C_{16} - C_{15} - C_{28}$	117.0 (3)	C74 - C73 - C72	120.7(4)
N4—C16—C15	126.2 (3)	С74—С73—Н73	119.7
N4—C16—C17	115.3 (3)	С72—С73—Н73	119.7
C15—C16—C17	118.5 (3)	C73—C74—C75	120.6 (4)
C22—C17—C18	120.3 (3)	С73—С74—Н74	119.7
C22—C17—C16	119.6 (3)	С75—С74—Н74	119.7
C18—C17—C16	120.1 (3)	C76—C75—C74	119.6 (4)
C19—C18—C17	120.7 (3)	С76—С75—Н75	120.2
C19—C18—H18	119.6	С74—С75—Н75	120.2
C17—C18—H18	119.6	C75—C76—C77	119.9 (4)
C18—C19—C20	119.5 (3)	С75—С76—Н76	120.0
С18—С19—Н19	120.3	С77—С76—Н76	120.0
С20—С19—Н19	120.3	C76—C77—C72	121.3 (4)
C21—C20—C19	120.2 (3)	С76—С77—Н77	119.3
С21—С20—Н20	119.9	С72—С77—Н77	119.3
С19—С20—Н20	119.9	С79—С78—Н78А	109.5
C20—C21—C22	121.7 (3)	C79—C78—H78B	109.5
C20—C21—H21	119.2	H78A—C78—H78B	109.5
C22—C21—H21	119.2	С79—С78—Н78С	109.5
C17—C22—C21	117.7 (3)	H78A—C78—H78C	109.5

C17 - C22 - C23	120 3 (3)	H78B—C78—H78C	109 5
$C_{21} - C_{22} - C_{23}$	122.0(3)	C80-C79-C84	119.1 (5)
C_{24} C_{23} C_{28}	117.8(3)	C_{80} C_{79} C_{78}	121.0(5)
C_{24} C_{23} C_{20}	1214(3)	C84 - C79 - C78	121.0(5)
$C_{24} = C_{23} = C_{22}$	121.4(3) 120.7(3)	C81 - C80 - C79	117.9(5)
$C_{25} = C_{23} = C_{22}$	120.7(3)	$C_{81} = C_{80} = C_{79}$	121.2(3)
$C_{25} = C_{24} = C_{25}$	122.4 (3)	C70 $C80$ $H80$	119.4
$C_{23} = C_{24} = H_{24}$	110.0	C_{79}^{20} C_{80}^{21} C_{82}^{22}	119.4
$C_{23} = C_{24} = H_{24}$	110.0	$C_{80} = C_{81} = C_{82}$	119.3 (3)
$C_{24} = C_{25} = C_{26}$	119.0 (5)		120.2
C24—C25—H25	120.2	C82—C81—H81	120.2
C26—C25—H25	120.2	C83 - C82 - C81	119.6 (5)
C25—C26—C27	119.9 (4)	C83—C82—H82	120.2
С25—С26—Н26	120.0	С81—С82—Н82	120.2
C27—C26—H26	120.0	C84—C83—C82	121.0 (5)
C26—C27—C28	121.5 (3)	С84—С83—Н83	119.5
С26—С27—Н27	119.3	С82—С83—Н83	119.5
С28—С27—Н27	119.3	C83—C84—C79	119.5 (5)
C27—C28—C23	118.6 (3)	C83—C84—H84	120.2
C27—C28—C15	122.4 (3)	С79—С84—Н84	120.2
C23—C28—C15	118.9 (3)	C79A—C78A—H78D	109.5
N5-C29-C30	117.8 (3)	С79А—С78А—Н78Е	109.5
N5-C29-C42	125.7 (3)	H78D—C78A—H78E	109.5
C30—C29—C42	116.4 (3)	C79A—C78A—H78F	109.5
N6-C30-C29	126.8 (3)	H78D—C78A—H78F	109.5
N6-C30-C31	114.6 (3)	H78E—C78A—H78F	109.5
C29—C30—C31	118.6 (3)	C84A—C79A—C80A	120.4 (17)
C32—C31—C36	120.4 (3)	C84A—C79A—C78A	121.6 (11)
C32—C31—C30	120.9 (3)	C80A—C79A—C78A	117.9 (12)
C36—C31—C30	118.7 (3)	C82A—C81A—C80A	123.7 (16)
$C_{33} = C_{32} = C_{31}$	120.1 (4)	C82A - C81A - H81A	118.2
C33—C32—H32	120.0	C80A - C81A - H81A	118.2
C_{31} C_{32} H_{32}	120.0	C79A - C80A - C81A	115.9(14)
C_{34} C_{33} C_{32}	120.0 120.1(4)	C79A - C80A - H80A	122.1
C_{34} C_{33} H_{33}	110.0	C81A - C80A - H80A	122.1
C32_C33_H33	110.0	C81A - C82A - C83A	122.1 121(2)
$C_{32} = C_{33} = H_{33}$	120.7(4)	C81A $C82A$ $H82A$	110 5
$C_{33} = C_{34} = C_{35}$	120.7 (4)	$C_{81A} = C_{82A} = H_{82A}$	119.5
$C_{35} = C_{34} = H_{34}$	119.7	$C_{02A} = C_{02A} = C_{02A} = C_{02A}$	117.3 117.8(17)
$C_{33} = C_{34} = C_{34}$	119.7	$C_{02A} = C_{03A} = C_{04A}$	117.0(17)
$C_{34} = C_{35} = C_{30}$	121.0 (4)	$C_{02}A = C_{02}A = H_{02}A$	121.1
C34—C35—H35	119.5	C84A—C83A—H83A	121.1
С36—С35—Н35	119.5	C/9A—C84A—C83A	121.3 (18)
$C_{31} - C_{36} - C_{35}$	11/.8 (3)	C/9A—C84A—H84A	119.3
C31—C36—C37	119.5 (3)	C83A—C84A—H84A	119.3
C35—C36—C37	122.8 (3)	CI—NI—Ol	120.2 (3)
C38—C37—C42	118.0 (3)	C1—N1—Ni1	117.4 (2)
C38—C37—C36	120.8 (3)	O1—N1—Ni1	122.4 (2)
C42—C37—C36	121.1 (3)	C2—N2—O2	121.3 (3)
C39—C38—C37	121.9 (4)	C2—N2—Ni1	116.3 (2)

G20 G20 H20	110.1		100.0 (0)
C39—C38—H38	119.1	O2—N2—N11	122.3 (2)
С37—С38—Н38	119.1	C15—N3—O3	116.8 (3)
C40—C39—C38	120.0 (4)	C15—N3—Ni2	126.6 (2)
С40—С39—Н39	120.0	O3—N3—Ni2	116.3 (2)
С38—С39—Н39	120.0	C16—N4—O4	119.1 (3)
C39—C40—C41	120 2 (4)	C29—N5—O5	116 3 (3)
C_{39} C_{40} H_{40}	119.9	C_{29} N5 Ni1	126.6(2)
C_{41} C_{40} H_{40}	110.0	O_{2} N5 N1	120.0(2)
$C_{41} = C_{40} = 1140$	119.9	$C_{20} N_{6} O_{6}$	110.3(2)
C40 - C41 - C42	121.2 (4)	C_{30} No -00	118.0 (3)
C40—C41—H41	119.4	C43—N/—O/	120.8 (3)
C42—C41—H41	119.4	C43—N7—Ni2	117.0 (2)
C41—C42—C37	118.7 (3)	07—N7—Ni2	122.0 (2)
C41—C42—C29	122.4 (3)	C44—N8—O8	121.7 (3)
C37—C42—C29	118.6 (3)	C44—N8—Ni2	115.9 (2)
N7—C43—C44	112.4 (3)	08—N8—Ni2	122.3 (2)
N7—C43—C56	127.1 (3)	C57—N9—O9	117.4 (3)
C44—C43—C56	120.4 (3)	C57—N9—Ni3	118.8 (2)
N8-C44-C43	1122(3)	09 - N9 - Ni3	1225(2)
N8 C44 C45	112.2(3) 127.3(3)	C58 N10 O10	122.3(2) 117.1(3)
C_{42} C_{44} C_{45}	127.3(3)	$C_{58} = N_{10} = 0.00$	117.1(3)
C43 - C44 - C43	120.4(3)	C_{30} N10 N13	118.1(2)
C46—C45—C50	120.2 (3)	010—N10—N13	123.1 (2)
C46—C45—C44	122.5 (3)	NI—OI—HIA	101 (3)
C50—C45—C44	117.1 (3)	N2—O2—Ni3	111.88 (18)
C47—C46—C45	120.6 (4)	N3—O3—H2A	103 (2)
C47—C46—H46	119.7	N4—O4—Ni2	127.34 (19)
C45—C46—H46	119.7	N4	116.88 (18)
C46—C47—C48	119.8 (4)	Ni2—O4—Ni3	115.78 (11)
C46—C47—H47	120.1	N5—O5—H1A	107 (3)
C48—C47—H47	120.1	N6-06-Ni1	127 55 (19)
C_{49} C_{48} C_{47}	120.1 120.4(4)	N606Ni3	127.53(19) 116.43(18)
C_{49} C_{48} H_{48}	110.8	Ni1 O6 Ni3	110.43(10) 115.03(11)
C49 - C40 - H48	119.0	NII-00-NI3	113.93(11)
C47 - C48 - H48	119.8	N/-O/-HZA	98 (2)
C48—C49—C50	121.9 (4)	N8-08-N13	111.97 (18)
C48—C49—H49	119.0	N9—O9—H3A	93 (4)
С50—С49—Н49	119.0	N10—O10—H4A	102 (3)
C49—C50—C45	117.1 (4)	O6—Ni1—N1	169.58 (11)
C49—C50—C51	121.4 (4)	O6—Ni1—N2	87.73 (10)
C45—C50—C51	121.5 (3)	N1—Ni1—N2	81.94 (11)
C56—C51—C52	117.1 (4)	O6—Ni1—N5	91.03 (11)
C56—C51—C50	121.7 (3)	N1—Ni1—N5	99.21 (12)
C52—C51—C50	121.2 (4)	N2—Ni1—N5	176.81 (12)
C53—C52—C51	122.3 (4)	04—Ni2—N7	170.22 (11)
C53_C52_H52	118.9	04 Ni2 N8	88 31 (11)
$C_{51} = C_{52} = H_{52}$	118.0	N7 Nj2 N8	81.03(17)
$C_{51} = C_{52} = C_{54}$	110.7	$\frac{1}{1} \frac{1}{1} \frac{1}$	01.75(12)
$C_{22} = C_{23} = C_{24}$	120.0 (4)	U4—IN12—IN3	90.94 (11)
С52—С53—Н53	119./	N/-N12-N3	98.79 (12)
С54—С53—Н53	119.7	N8—N12—N3	177.16 (12)
C53—C54—C55	119.4 (4)	N9—Ni3—O6	165.53 (11)

С53—С54—Н54	120.3	N9—Ni3—N10	76.02 (11)
С55—С54—Н54	120.3	O6—Ni3—N10	95.97 (10)
C54—C55—C56	120.9 (4)	N9—Ni3—O4	96.00 (10)
С54—С55—Н55	119.5	O6—Ni3—O4	94.37 (10)
С56—С55—Н55	119.5	N10—Ni3—O4	164.78 (11)
C55—C56—C51	119.8 (3)	N9—Ni3—O8	104.85 (10)
C55—C56—C43	123.0 (3)	O6—Ni3—O8	86.82 (10)
C51—C56—C43	117.3 (3)	N10—Ni3—O8	89.16 (10)
N9—C57—C70	128.8 (3)	04—Ni3—08	80.29 (9)
N9—C57—C58	112.0 (3)	N9—Ni3—O2	91.16 (10)
C70-C57-C58	119.2 (3)	06-Ni3-02	79 47 (9)
N10-C58-C59	119.2(3) 128.4(3)	N10 - Ni3 - O2	106 60 (10)
N10 C58 C57	120.7(3)	04 Ni ₃ 02	86 25 (9)
C_{59} C_{58} C_{57}	112.7(3) 118.8(3)	O_{4} N_{13} O_{2}	150.00(9)
$C_{5}^{6} = C_{5}^{6} = C_{5}^{6}$	110.0(3) 118.9(2)	08-1115-02	139.99 (9)
00-039-04	110.0 (3)		
N1 C1 C2 N2	6 9 (1)	C50 C51 C56 C43	-0.1.(5)
NI = CI = C2 = N2	0.0(4)	$C_{30} - C_{31} - C_{30} - C_{43}$	-0.1(3)
C14 - C1 - C2 - N2	-1/4.0(3)	N = C43 = C30 = C33	-9.7(0)
NI = CI = C2 = C3	-100.9(3)	C44 - C43 - C56 - C55	108.1(3)
C14 - C1 - C2 - C3	12.4 (5)	N/-C43-C56-C51	1/1.5 (3)
$N_2 = C_2 = C_3 = C_4$	-0.7(5)	C44 - C43 - C56 - C51	-10.7(5)
C1—C2—C3—C4	171.9 (3)	N9—C57—C58—N10	16.8 (4)
N2-C2-C3-C8	-174.4 (3)	C70—C57—C58—N10	-160.8(3)
C1—C2—C3—C8	-1.8 (4)	N9—C57—C58—C59	-161.1 (3)
C8—C3—C4—C5	3.3 (5)	C70—C57—C58—C59	21.3 (4)
C2—C3—C4—C5	-170.3 (3)	N10-C58-C59-C60	-17.7 (6)
C3—C4—C5—C6	-0.6(5)	C57—C58—C59—C60	159.9 (3)
C4—C5—C6—C7	-1.8 (6)	N10-C58-C59-C64	166.4 (3)
C5—C6—C7—C8	1.4 (6)	C57—C58—C59—C64	-16.0 (5)
C6—C7—C8—C3	1.3 (5)	C64—C59—C60—C61	0.6 (6)
C6—C7—C8—C9	-179.2 (3)	C58—C59—C60—C61	-175.3 (4)
C4—C3—C8—C7	-3.6 (5)	C59—C60—C61—C62	-2.1 (6)
C2—C3—C8—C7	170.4 (3)	C60—C61—C62—C63	1.9 (7)
C4—C3—C8—C9	176.9 (3)	C61—C62—C63—C64	-0.2 (7)
C2—C3—C8—C9	-9.1 (4)	C62—C63—C64—C59	-1.3 (6)
C7—C8—C9—C10	13.2 (5)	C62—C63—C64—C65	176.6 (4)
C3—C8—C9—C10	-167.3(3)	C60—C59—C64—C63	1.1 (5)
C7—C8—C9—C14	-169.4(3)	C58—C59—C64—C63	177.2 (3)
C3-C8-C9-C14	10.1 (5)	C60—C59—C64—C65	-176.9(3)
C14 - C9 - C10 - C11	10(5)	C58—C59—C64—C65	-0.8(5)
C8-C9-C10-C11	178 6 (3)	C63 - C64 - C65 - C66	135(5)
C9-C10-C11-C12	-16(6)	C_{59} C_{64} C_{65} C_{66}	-1686(3)
C10-C11-C12-C13	0.8 (5)	C63 - C64 - C65 - C70	-1645(3)
C_{11} C_{12} C_{13} C_{14}	0.6 (5)	C_{59} C_{64} C_{65} C_{70}	134(5)
$C_{12} = C_{13} = C_{14} = C_{14}$	-12(5)	C70 C65 C66 C67	30(6)
$C_{12} = C_{13} = C_{14} = C_{9}$	-1.2(3) -1787(2)	$C_{10} = C_{00} = C_{00} = C_{01}$	-175 1 (4)
$C_{12} - C_{13} - C_{14} - C_{12}$	1/0.7(3)	C_{0}	-0.1(6)
$C_{10} = C_{14} = C_{13}$	0.4(3)		-0.1(0)
0-09-014-013	-1/1.2(3)	UUU-UU/-UU3-U09	-2.1 (0)

C10 C0 C14 C1	170.0 (2)		1 2 (0)
010-09-014-01	1/8.0 (3)	C6/	1.3 (6)
C8—C9—C14—C1	0.5 (5)	C66—C65—C70—C69	-3.7(5)
N1—C1—C14—C13	-14.7 (5)	C64—C65—C70—C69	174.3 (3)
C2-C1-C14-C13	166.2 (3)	C66—C65—C70—C57	174.0 (3)
N1-C1-C14-C9	167.7 (3)	C64—C65—C70—C57	-7.9 (5)
C2-C1-C14-C9	-11.4(5)	C68—C69—C70—C65	1.6 (5)
N3—C15—C16—N4	26.0 (5)	C68—C69—C70—C57	-176.0(3)
C_{28} C15 C16 N4	-1542(3)	N9-C57-C70-C65	173 9 (3)
N_{3} C_{15} C_{16} C_{17}	-1537(3)	$C_{58} - C_{57} - C_{70} - C_{65}$	-90(5)
$C_{10}^{20} = C_{10}^{10} = C_{10}^{10} = C_{17}^{10}$	260(4)	N0 C57 C70 C60	-85(6)
$C_{20} = C_{10} = C_{10} = C_{17}$	20.0(4)	$N_{2} = C_{3}^{2} = C_{10}^{2} = C_{0}^{2}$	8.5(0)
N4-C16-C17-C22	100.5 (3)	$C_{3} = C_{3} = C_{10} = C_{00}$	108.0 (3)
C15—C16—C17—C22	-13.8 (4)	C//C/2C/3C/4	1.6 (7)
N4—C16—C17—C18	-14.7 (4)	C71—C72—C73—C74	-178.3(5)
C15—C16—C17—C18	165.1 (3)	C72—C73—C74—C75	0.3 (7)
C22—C17—C18—C19	-1.0 (5)	C73—C74—C75—C76	-1.4 (7)
C16—C17—C18—C19	-179.9 (3)	C74—C75—C76—C77	0.5 (8)
C17—C18—C19—C20	1.6 (5)	C75—C76—C77—C72	1.5 (9)
C18—C19—C20—C21	-0.4(5)	C73—C72—C77—C76	-2.5(8)
C19—C20—C21—C22	-1.3 (6)	C71—C72—C77—C76	177.3 (6)
C18—C17—C22—C21	-0.6(5)	C84—C79—C80—C81	0.2 (4)
$C_{16} - C_{17} - C_{22} - C_{21}$	178 2 (3)	C78—C79—C80—C81	-179.8(3)
C_{18} C_{17} C_{22} C_{23}	176.2(3)	C79 - C80 - C81 - C82	-0.6(6)
C_{16} C_{17} C_{22} C_{23}	-4.9(5)	$C_{80} = C_{81} = C_{82} = C_{83}$	0.0(0)
$C_{10} = C_{17} = C_{22} = C_{23}$	(3)	$C_{80} - C_{81} - C_{82} - C_{83}$	0.7(7)
$C_{20} = C_{21} = C_{22} = C_{17}$	1.8 (3)	$C_{01} = C_{02} = C_{03} = C_{04} = C_{04}$	-0.4(8)
C20—C21—C22—C23	-175.1(3)	C82—C83—C84—C79	0.1 (7)
C17—C22—C23—C24	-166.6 (3)	C80—C79—C84—C83	0.0 (5)
C21—C22—C23—C24	10.2 (5)	C78—C79—C84—C83	-180.0(3)
C17—C22—C23—C28	11.0 (5)	C84A—C79A—C80A—C81A	0.3 (6)
C21—C22—C23—C28	-172.2 (3)	C78A—C79A—C80A—C81A	-179.9 (2)
C28—C23—C24—C25	-1.3 (5)	C82A—C81A—C80A—C79A	0.0 (6)
C22—C23—C24—C25	176.4 (3)	C80A—C81A—C82A—C83A	-0.5 (7)
C23—C24—C25—C26	-2.0(6)	C81A—C82A—C83A—C84A	0.5 (8)
C24—C25—C26—C27	2.8 (6)	C80A—C79A—C84A—C83A	-0.2(7)
C25—C26—C27—C28	-0.1 (6)	C78A—C79A—C84A—C83A	180.0 (4)
$C_{26} - C_{27} - C_{28} - C_{23}$	-33(5)	C82A—C83A—C84A—C79A	-0.2(8)
C_{26} C_{27} C_{28} C_{15}	-1789(3)	C14-C1-N1-O1	-1.9(5)
$C_{20} = C_{21} = C_{20} = C_{13}$	39(5)	C_{2} C_{1} N_{1} O_{1}	1.7(3)
$C_{24} = C_{23} = C_{20} = C_{27} = C$	-173.8(3)	C_{14} C_{1} N_{1} N_{11}	177.5(3)
$C_{22} = C_{23} = C_{26} = C_{27}$	173.8(3)	C14-C1-N1-N11	170.0(3)
$C_{24} = C_{23} = C_{26} = C_{15}$	1/9.0(5)	$C_2 = C_1 = N_1 = N_1$	-4.2(4)
022-023-028-015	1.9 (5)	C3-C2-N2-O2	-9.3 (5)
N3—C15—C28—C27	-24.8 (5)	C1—C2—N2—O2	177.6 (3)
C16—C15—C28—C27	155.5 (3)	C3—C2—N2—Nil	166.7 (3)
N3—C15—C28—C23	159.6 (3)	C1—C2—N2—Ni1	-6.4 (4)
C16—C15—C28—C23	-20.1 (4)	C16—C15—N3—O3	173.6 (3)
N5-C29-C30-N6	28.4 (5)	C28—C15—N3—O3	-6.2 (5)
C42—C29—C30—N6	-150.6 (3)	C16—C15—N3—Ni2	-13.4 (4)
N5-C29-C30-C31	-151.3 (3)	C28—C15—N3—Ni2	166.8 (2)
C42—C29—C30—C31	29.7 (4)	C15—C16—N4—O4	-4.3 (5)

N6-C30-C31-C32	-24.0 (5)	C17—C16—N4—O4	175.4 (3)
C29—C30—C31—C32	155.8 (3)	C30-C29-N5-O5	172.0 (3)
N6-C30-C31-C36	155.9 (3)	C42—C29—N5—O5	-9.0 (5)
C29—C30—C31—C36	-24.4 (5)	C30-C29-N5-Ni1	-18.5 (4)
C36—C31—C32—C33	-1.1 (6)	C42—C29—N5—Ni1	160.5 (3)
C30—C31—C32—C33	178.8 (4)	C29—C30—N6—O6	-4.0 (5)
C31—C32—C33—C34	-0.1 (7)	C31—C30—N6—O6	175.7 (3)
C32—C33—C34—C35	0.8 (7)	C44—C43—N7—O7	178.7 (3)
C33—C34—C35—C36	-0.3 (7)	C56—C43—N7—O7	-3.4 (5)
C32—C31—C36—C35	1.5 (5)	C44—C43—N7—Ni2	-6.0(4)
C30—C31—C36—C35	-178.3 (3)	C56—C43—N7—Ni2	172.0 (3)
C32—C31—C36—C37	-177.9(3)	C43—C44—N8—O8	175.6 (3)
C_{30} C_{31} C_{36} C_{37}	2.3 (5)	C45—C44—N8—O8	-8.5(5)
$C_{34} - C_{35} - C_{36} - C_{31}$	-0.8(6)	C43—C44—N8—Ni2	-7.2(4)
$C_{34} - C_{35} - C_{36} - C_{37}$	178 5 (4)	C45-C44-N8-Ni2	168.7(3)
C_{31} $-C_{36}$ $-C_{37}$ $-C_{38}$	-166.6(3)	C70-C57-N9-O9	-0.4(5)
$C_{35} - C_{36} - C_{37} - C_{38}$	14 1 (5)	C58-C57-N9-O9	-1777(3)
C_{31} $-C_{36}$ $-C_{37}$ $-C_{42}$	143(5)	C70-C57-N9-Ni3	1665(3)
C_{35} C_{36} C_{37} C_{42}	-1651(3)	C58 - C57 - N9 - Ni3	-10.8(4)
C42-C37-C38-C39	11(5)	C59 - C58 - N10 - O10	-35(5)
$C_{36} - C_{37} - C_{38} - C_{39}$	-1780(3)	C57 - C58 - N10 - O10	1789(3)
C_{37} C_{38} C_{39} C_{40}	04(6)	C59 - C58 - N10 - Ni3	1621(3)
C_{38} C_{39} C_{40} C_{41}	-16(6)	C57 - C58 - N10 - Ni3	-156(4)
C_{39} C_{40} C_{41} C_{42}	12(6)	$C^2 = N^2 = O^2 = N^3$	143.0(3)
C40-C41-C42-C37	0.3(5)	Ni1— $N2$ — $O2$ — $Ni3$	-327(3)
C40-C41-C42-C29	-1735(3)	C16-N4-O4-Ni2	-30.5(4)
$C_{38} - C_{37} - C_{42} - C_{41}$	-15(5)	C16-N4-O4-Ni3	148.7(2)
$C_{36} - C_{37} - C_{42} - C_{41}$	1777(3)	C30—N6—O6—Nil	-30.9(4)
$C_{38} - C_{37} - C_{42} - C_{29}$	177.6(3)	C30 - N6 - O6 - Ni3	1454(2)
$C_{36} - C_{37} - C_{42} - C_{29}$	-83(5)	C44 - N8 - O8 - Ni3	145.0(3)
N_{5} C_{29} C_{42} C_{41}	-185(5)	Ni2-N8-08-Ni3	-321(3)
C_{30} C_{29} C_{42} C_{41}	160.4(3)	N6-06-Ni1-N1	-158.8(6)
N_{5} C_{29} C_{42} C_{37}	167.6(3)	Ni3-06-Ni1-N1	249(7)
C_{30} C_{29} C_{42} C_{37}	-134(4)	N6-06-Ni1-N2	-1510(3)
N7-C43-C44-N8	8.4 (4)	Ni3-06-Ni1-N2	32.64 (14)
C56-C43-C44-N8	-169.7(3)	N6-06-Ni1-N5	31.9 (3)
N7-C43-C44-C45	-167.8(3)	Ni3-06-Ni1-N5	-144.41(14)
C56-C43-C44-C45	14.1 (5)	C1-N1-Ni1-O6	8.5 (8)
N8-C44-C45-C46	-6.1(6)	O1-N1-Ni1-O6	-173.0(6)
C43—C44—C45—C46	169.4 (3)	C1-N1-Ni1-N2	0.7 (3)
N8-C44-C45-C50	178.3 (3)	O1-N1-Ni1-N2	179.1 (3)
C43—C44—C45—C50	-6.2(5)	C1—N1—Ni1—N5	177.7 (3)
C50-C45-C46-C47	1.8 (6)	01—N1—Ni1—N5	-3.9(3)
C44—C45—C46—C47	-173.7 (4)	C2—N2—Ni1—O6	-175.1 (2)
C45—C46—C47—C48	-1.8 (6)	O2—N2—Ni1—O6	0.8 (2)
C46—C47—C48—C49	1.1 (7)	C2—N2—Ni1—N1	3.5 (2)
C47—C48—C49—C50	-0.4 (7)	O2—N2—Ni1—N1	179.4 (3)
C48—C49—C50—C45	0.4 (6)	C29—N5—Ni1—O6	-5.9 (3)
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C48—C49—C50—C51	179.7 (4)	O5—N5—Ni1—O6	163.6 (3)
C46—C45—C50—C49	-1.0 (5)	C29—N5—Ni1—N1	176.0 (3)
C44—C45—C50—C49	174.6 (3)	O5—N5—Ni1—N1	-14.5 (3)
C46—C45—C50—C51	179.7 (3)	N4—O4—Ni2—N8	-149.4 (3)
C44—C45—C50—C51	-4.7 (5)	Ni3—O4—Ni2—N8	31.36 (14)
C49—C50—C51—C56	-171.2 (4)	N4—O4—Ni2—N3	33.3 (3)
C45—C50—C51—C56	8.0 (5)	Ni3—O4—Ni2—N3	-145.90 (14)
C49—C50—C51—C52	9.9 (6)	C43—N7—Ni2—N8	1.8 (3)
C45—C50—C51—C52	-170.8 (3)	O7—N7—Ni2—N8	177.1 (3)
C56—C51—C52—C53	-0.1 (6)	C43—N7—Ni2—N3	179.0 (3)
C50—C51—C52—C53	178.7 (4)	O7—N7—Ni2—N3	-5.7 (3)
C51—C52—C53—C54	0.2 (6)	C44—N8—Ni2—O4	-176.1 (3)
C52—C53—C54—C55	-0.1 (6)	O8—N8—Ni2—O4	1.1 (2)
C53—C54—C55—C56	-0.1 (6)	C44—N8—Ni2—N7	3.3 (2)
C54—C55—C56—C51	0.2 (5)	08—N8—Ni2—N7	-179.5 (3)
C54—C55—C56—C43	-178.5 (3)	C15—N3—Ni2—O4	-10.2 (3)
C52—C51—C56—C55	-0.1 (5)	O3—N3—Ni2—O4	162.8 (2)
C50-C51-C56-C55	-179.0 (3)	C15—N3—Ni2—N7	170.8 (3)
C52—C51—C56—C43	178.7 (3)	O3—N3—Ni2—N7	-16.1 (2)

Hydrogen-bond geometry (Å, °)

D—H···A	D—H	H···A	D···A	D—H···A
01—H1A····O5	1.29 (6)	1.13 (6)	2.405 (4)	169 (6)
O3—H2 <i>A</i> …O7	1.17 (5)	1.23 (5)	2.402 (4)	176 (5)
O9—H3 <i>A</i> ···N4	0.86 (6)	1.82 (6)	2.672 (4)	175 (7)
O10—H4A…N6	1.04 (5)	1.63 (5)	2.657 (4)	168 (4)

(2) $Bis(\mu_2-bis\{[10-(oxidoimino)-9,10-dihydrophenanthren-9-ylidene]amino\}difluoroborato)(phenanthrene-9,10-dione dioxime)trinickel(II) dichloromethane trisolvate$

Crystal data

$[Ni_{3}(C_{28}H_{16}BF_{2}N_{4}O_{4})_{2}(C_{14}H_{10}N_{2}O_{2})]\cdot 3CH_{2}Cl_{2}$	F(000) = 2968
$M_{r} = 1456.88$	$D_x = 1.473 \text{ Mg m}^{-3}$
Monoclinic, $P2_{1}/c$	Cu K α radiation, $\lambda = 1.54184 \text{ Å}$
a = 15.6414 (8) Å	Cell parameters from 25728 reflections
b = 30.8358 (11) A	$\theta = 3.4 - 76.5^{\circ}$
c = 14.7380 (8) Å	$\mu = 1.67 \text{ mm}^{-1}$
$\beta = 112.411 (6)^{\circ}$	T = 100 K
$V = 6571.5 (6) Å^{3}$	Block, red
Z = 4	$0.29 \times 0.07 \times 0.04 \text{ mm}$
Data collection	
Agilent SuperNova Dual Source diffractometer with an Atlas detector Radiation source: sealed X-ray tube Detector resolution: 5.2940 pixels mm ⁻¹ ω scans	Absorption correction: multi-scan (CrysAlis PRO; Agilent, 2014) $T_{min} = 0.303$, $T_{max} = 1.000$ 25337 measured reflections 13041 independent reflections 8782 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.060$ $\theta_{\text{max}} = 75.0^{\circ}, \ \theta_{\text{min}} = 3.4^{\circ}$ $h = -19 \rightarrow 19$	$k = -26 \rightarrow 38$ $l = -18 \rightarrow 18$
Refinement	

Refinement on F^2	Hydrogen site location: inferred from
Least-squares matrix: full	neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.077$	H atoms treated by a mixture of independent
$wR(F^2) = 0.226$	and constrained refinement
<i>S</i> = 1.03	$w = 1/[\sigma^2(F_o^2) + (0.108P)^2 + 8.2009P]$
13041 reflections	where $P = (F_o^2 + 2F_c^2)/3$
896 parameters	$(\Delta/\sigma)_{\rm max} = 0.006$
0 restraints	$\Delta \rho_{\rm max} = 1.16 \text{ e} \text{ Å}^{-3}$
	$\Delta \rho_{\rm min} = -0.83 \ {\rm e} \ {\rm \AA}^{-3}$

Special details

Experimental. CrysAlisPro. (Agilent, 2014). Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm.

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > 2$ sigma(F^2) is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger. Three molecules of what appeared to be dichloromethane were found to be badly disordered. Attempts to model the disorder were unsatisfactory. The contributions to the scattering factors due to these solvent molecules were removed by use of the utility SQUEEZE (Sluis and Spek, 1990) in PLATON98 (Spek, 1998). PLATON98 was used as incorporated in WinGX (Farrugia, 1999).

	x	у	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	
C1	0.3411 (3)	0.33690 (15)	0.7321 (4)	0.0323 (10)	
C2	0.4326 (3)	0.33998 (16)	0.7254 (4)	0.0336 (11)	
C3	0.4765 (4)	0.29952 (16)	0.7127 (4)	0.0347 (11)	
C4	0.5720 (4)	0.29722 (18)	0.7393 (4)	0.0423 (12)	
H4	0.6093	0.3220	0.7661	0.051*	
C5	0.6130 (4)	0.25876 (18)	0.7267 (5)	0.0499 (14)	
H5	0.6781	0.2571	0.7446	0.060*	
C6	0.5574 (4)	0.22266 (18)	0.6875 (5)	0.0498 (15)	
H6	0.5847	0.1963	0.6785	0.060*	
C7	0.4637 (4)	0.22483 (17)	0.6619 (4)	0.0460 (14)	
H7	0.4274	0.1997	0.6361	0.055*	
C8	0.4198 (4)	0.26291 (15)	0.6727 (4)	0.0362 (11)	
C9	0.3201 (4)	0.26641 (15)	0.6448 (4)	0.0357 (11)	
C10	0.2594 (4)	0.23476 (16)	0.5841 (4)	0.0394 (12)	
H10	0.2844	0.2106	0.5627	0.047*	
C11	0.1655 (4)	0.23823 (18)	0.5555 (4)	0.0438 (13)	
H11	0.1267	0.2162	0.5158	0.053*	

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters $(Å^2)$

C12	0.1264 (4)	0.27305 (18)	0.5831 (4)	0.0422 (12)
H12	0.0611	0.2754	0.5620	0.051*
C13	0.1837 (4)	0.30485 (16)	0.6426 (4)	0.0361 (11)
H13	0.1571	0.3291	0.6619	0.043*
C14	0.2797 (4)	0.30177 (15)	0.6744 (4)	0.0336(11)
C15	0.2126 (3)	0.36498 (16)	0.4672 (3)	0.0310(10)
C16	0.3085(3)	0.37760 (15)	0.4870(3)	0.0295 (10)
C17	0.3700(3)	0 34790 (16)	0.4634(3)	0.0304(10)
C18	0.5700(3) 0.4630(3)	0.35767 (17)	0.1037(3) 0.4827(4)	0.0357(10)
H18	0.4869	0.3853	0.5083	0.042*
C19	0.4009 0.5208 (4)	0.3033 0.32743(18)	0.3609	0.042
H19	0.5832	0.32/45 (10)	0.4049(4)	0.053*
C20	0.3852 0.4864 (4)	0.3340	0.4770	0.033
U20	0.4804 (4)	0.26085 (19)	0.4294 (4)	0.0472(14) 0.057*
H20	0.3201	0.2037	0.4190	0.037°
C21	0.3930 (4)	0.27098 (18)	0.4081 (4)	0.0423 (12)
H21	0.3725	0.2491	0.3827	0.051*
C22	0.3341(3)	0.30/00 (16)	0.4230 (4)	0.0330 (10)
C23	0.2352 (4)	0.29719 (16)	0.3911 (4)	0.0335 (10)
C24	0.1973 (4)	0.25910 (18)	0.3381 (4)	0.0423 (12)
H24	0.2378	0.2386	0.3275	0.051*
C25	0.1044 (4)	0.25044 (19)	0.3013 (4)	0.0464 (13)
H25	0.0816	0.2242	0.2667	0.056*
C26	0.0436 (4)	0.28018 (19)	0.3149 (4)	0.0455 (13)
H26	-0.0211	0.2750	0.2874	0.055*
C27	0.0781 (4)	0.31762 (18)	0.3690 (4)	0.0380 (11)
H27	0.0365	0.3376	0.3794	0.046*
C28	0.1730 (3)	0.32636 (16)	0.4084 (4)	0.0341 (11)
C29	0.3296 (3)	0.47670 (15)	0.9399 (4)	0.0325 (10)
C30	0.3915 (3)	0.50074 (15)	0.9040 (3)	0.0316 (10)
C31	0.4137 (3)	0.54622 (15)	0.9300 (3)	0.0322 (10)
C32	0.4728 (4)	0.57048 (16)	0.8987 (4)	0.0386 (12)
H32	0.4988	0.5573	0.8569	0.046*
C33	0.4942 (4)	0.61279 (17)	0.9267 (4)	0.0448 (13)
H33	0.5354	0.6284	0.9053	0.054*
C34	0.4558 (5)	0.63252 (18)	0.9862 (5)	0.0557 (17)
H34	0.4704	0.6618	1.0061	0.067*
C35	0.3969 (4)	0.60974 (17)	1.0163 (4)	0.0454 (14)
H35	0.3712	0.6239	1.0573	0.055*
C36	0.3721 (4)	0.56675 (16)	0.9903 (3)	0.0340 (11)
C37	0.3088(4)	0 54261 (16)	1 0242 (4)	0.0360(11)
C38	0 2666 (4)	0 56390 (18)	1.02.12(1) 1.0798(4)	0.0418(13)
H38	0.2772	0 5941	1.0922	0.050*
C39	0.2105(4)	0.5425(2)	1 1168 (4)	0.020
H39	0.1828	0.5579	1 1545	0.060*
C40	0.1020	0.3377 0.4987 (2)	1.1040	0.000
U40	0.1552	0.4838	1 1261	0.053*
C/1	0.1332 0.2320 (A)	0.7050	1.1201 1.0430(4)	0.035
U41	0.2339 (4)	0.47002 (17)	1.0437 (4)	0.0372(11)
1141	0.2230	0.4404	1.0320	0.045

C42	0.2897 (4)	0.49811 (16)	1.0043 (3)	0.0355 (11)
C43	0.1661 (3)	0.49600 (19)	0.6807 (4)	0.0400 (12)
C44	0.2317 (3)	0.52960 (18)	0.6755 (4)	0.0392 (12)
C45	0.2153 (4)	0.5756 (2)	0.6941 (4)	0.0511 (16)
C46	0.2581 (4)	0.6091 (2)	0.6633 (5)	0.0581 (18)
H46	0.2979	0.6023	0.6301	0.070*
C47	0.2434(5)	0.6522 (2)	0.6804(7)	0.077(3)
H47	0 2730	0 6749	0.6597	0.093*
C48	0.1844(5)	0.6615(3)	0.7286(7)	0.099
H48	0.1729	0.6908	0.7403	0.112*
C49	0.1727 (5)	0.6285(3)	0.7595(6)	0.083(3)
H49	0.1032	0.6356	0.7928	0.005 (5)
C50	0.1032 0.1571 (4)	0.5330	0.7/20	0.100
C51	0.1371(4) 0.1151(4)	0.5045(2) 0.5488(3)	0.7429(3) 0.7768(4)	0.001(2)
C52	0.1131(4) 0.0734(5)	0.5488(3)	0.7708(4) 0.8439(5)	0.002(2)
U52	0.0734(3)	0.5378 (5)	0.8439 (3)	0.073 (2)
П32 С52	0.0721	0.5800	0.8038	0.088 (2)
C35	0.0348 (3)	0.5250 (4)	0.8770(3)	0.088 (5)
H33	0.004/	0.5317	0.9209	0.105*
C54	0.0385 (4)	0.4820 (4)	0.8502 (5)	0.077(3)
H54	0.012/	0.4596	0.8/61	0.093*
C55	0.0804 (4)	0.4722 (3)	0.7841 (4)	0.0604 (19)
H55	0.0838	0.4430	0.7650	0.072*
C56	0.1175 (4)	0.5059 (2)	0.7462 (4)	0.0500 (16)
C57	0.6326 (3)	0.48466 (16)	0.6652 (4)	0.0315 (10)
C58	0.5935 (3)	0.52870 (16)	0.6698 (3)	0.0315 (10)
C59	0.6591 (3)	0.56487 (15)	0.7051 (3)	0.0304 (10)
C60	0.6390 (4)	0.60134 (17)	0.7479 (4)	0.0395 (12)
H60	0.5788	0.6050	0.7481	0.047*
C61	0.7056 (4)	0.63235 (18)	0.7902 (5)	0.0463 (13)
H61	0.6915	0.6570	0.8207	0.056*
C62	0.7926 (4)	0.62788 (17)	0.7885 (4)	0.0436 (13)
H62	0.8380	0.6496	0.8171	0.052*
C63	0.8141 (4)	0.59185 (17)	0.7452 (4)	0.0363 (11)
H63	0.8743	0.5889	0.7446	0.044*
C64	0.7473 (3)	0.55957 (15)	0.7021 (3)	0.0298 (10)
C65	0.7705 (3)	0.52067 (16)	0.6569 (3)	0.0320 (10)
C66	0.8480 (3)	0.51969 (17)	0.6315 (4)	0.0346 (11)
H66	0.8867	0.5445	0.6430	0.041*
C67	0.8690 (3)	0.48300 (19)	0.5900 (4)	0.0390 (12)
H67	0.9216	0.4828	0.5728	0.047*
C68	0.8134 (4)	0.44667 (18)	0.5735 (4)	0.0409 (12)
H68	0.8281	0.4215	0.5450	0.049*
C69	0.7366 (3)	0.44671 (18)	0.5982 (4)	0.0389 (12)
H69	0.6988	0.4216	0.5867	0.047*
C70	0.7146 (3)	0.48316 (16)	0.6398 (4)	0.0332 (11)
N1	0.3219 (3)	0.36493 (12)	0.7882 (3)	0.0321 (9)
N2	0.4789 (3)	0.37504 (13)	0.7289 (3)	0.0339 (9)
N3	0.1712 (3)	0.39220 (13)	0.5053 (3)	0.0316 (9)

N4	0.3296 (3)	0.41530 (13)	0.5315 (3)	0.0290 (8)
N5	0.3158 (3)	0.43671 (13)	0.9063 (3)	0.0337 (9)
N6	0.4215 (3)	0.47684 (11)	0.8493 (3)	0.0288 (8)
N7	0.1558 (3)	0.46191 (15)	0.6258 (3)	0.0364 (10)
N8	0.3047 (3)	0.52333 (14)	0.6565 (3)	0.0359 (9)
N9	0.5880 (3)	0.45401 (13)	0.6849 (3)	0.0346 (9)
N10	0.5047 (3)	0.52610 (13)	0.6498 (3)	0.0378 (10)
01	0.2406 (3)	0.35856 (11)	0.8007 (3)	0.0397 (8)
O2	0.4437 (2)	0.41270 (10)	0.7481 (3)	0.0329 (7)
03	0.0833 (2)	0.38339 (12)	0.4931 (3)	0.0436 (9)
O4	0.4102 (2)	0.43359 (11)	0.5495 (2)	0.0319 (7)
05	0.2580 (3)	0.41112 (11)	0.9310(3)	0.0422 (9)
06	0.4795 (2)	0.49203 (10)	0.8119 (3)	0.0324 (7)
07	0.0898 (3)	0.43277 (13)	0.6260 (3)	0.0448 (9)
08	0.3220 (2)	0.48291 (11)	0.6323 (3)	0.0347 (8)
09	0.6191 (2)	0.41229 (11)	0.6839 (3)	0.0373 (8)
H1A	0.582 (4)	0.3945 (14)	0.694 (5)	0.056*
O10	0.4608 (3)	0.56409 (12)	0.6526 (3)	0.0463 (9)
H2A	0.395 (5)	0.5572 (5)	0.654 (5)	0.069*
Ni1	0.37718 (6)	0.41976 (2)	0.82564 (6)	0.0306 (2)
Ni2	0.23963 (5)	0.44031 (3)	0.57163 (6)	0.0301 (2)
Ni3	0.45817 (5)	0.46682 (2)	0.67739 (6)	0.0277 (2)
F1	0.3252 (3)	0.34124 (11)	0.9649 (3)	0.0556 (9)
F2	0.1678 (3)	0.35125 (12)	0.9084 (3)	0.0601 (10)
F3	-0.0369 (2)	0.39118 (12)	0.5394 (3)	0.0613 (10)
F4	0.0009 (2)	0.44772 (11)	0.4612 (3)	0.0506 (8)
B1	0.2493 (5)	0.3650 (2)	0.9034 (6)	0.0486 (16)
B2	0.0335 (4)	0.4143 (2)	0.5275 (6)	0.0437 (15)

Atomic displacement parameters $(Å^2)$

	U^{11}	U^{22}	U ³³	U^{12}	U^{13}	U^{23}
C1	0.044 (3)	0.021 (2)	0.029 (2)	0.0022 (19)	0.011 (2)	0.0065 (19)
C2	0.042 (3)	0.026 (2)	0.030 (3)	0.002 (2)	0.011 (2)	0.004 (2)
C3	0.052 (3)	0.023 (2)	0.028 (2)	0.003 (2)	0.014 (2)	0.000(2)
C4	0.051 (3)	0.030 (3)	0.042 (3)	0.005 (2)	0.013 (2)	-0.002 (2)
C5	0.058 (3)	0.035 (3)	0.052 (4)	0.010 (3)	0.015 (3)	-0.004 (3)
C6	0.064 (4)	0.029 (3)	0.055 (4)	0.017 (3)	0.022 (3)	-0.001 (3)
C7	0.068 (4)	0.024 (3)	0.038 (3)	0.001 (2)	0.012 (3)	-0.002 (2)
C8	0.061 (3)	0.016 (2)	0.029 (3)	0.001 (2)	0.015 (2)	0.0017 (19)
C9	0.058 (3)	0.020 (2)	0.028 (2)	-0.002 (2)	0.014 (2)	0.006 (2)
C10	0.065 (3)	0.019 (2)	0.032 (3)	-0.006(2)	0.016 (2)	0.002 (2)
C11	0.059 (3)	0.033 (3)	0.035 (3)	-0.016 (2)	0.014 (2)	0.001 (2)
C12	0.054 (3)	0.036 (3)	0.036 (3)	-0.010 (2)	0.015 (2)	0.002 (2)
C13	0.049 (3)	0.028 (3)	0.031 (3)	-0.005 (2)	0.015 (2)	0.003 (2)
C14	0.049 (3)	0.023 (2)	0.024 (2)	-0.006 (2)	0.010(2)	0.0060 (19)
C15	0.036 (2)	0.032 (3)	0.023 (2)	0.0051 (19)	0.0079 (18)	0.006 (2)
C16	0.041 (2)	0.026 (2)	0.025 (2)	0.0059 (19)	0.0159 (19)	0.0028 (19)

C17	0.041 (2)	0.029(2)	0.022 (2)	0.0046 (19)	0.0133 (19)	0.0010 (19)
C18	0.039 (2)	0.034 (3)	0.032 (3)	0.001 (2)	0.013 (2)	-0.002 (2)
C19	0.041 (3)	0.041 (3)	0.049 (3)	0.006 (2)	0.016 (2)	-0.007(3)
C20	0.050 (3)	0.041 (3)	0.050 (3)	0.011 (2)	0.018 (3)	-0.010 (3)
C21	0.048 (3)	0.033 (3)	0.044 (3)	0.001 (2)	0.015 (2)	-0.010 (2)
C22	0.045 (3)	0.030 (3)	0.025 (2)	0.002 (2)	0.015 (2)	-0.002(2)
C23	0.048 (3)	0.029 (3)	0.026 (2)	0.000 (2)	0.016 (2)	0.003 (2)
C24	0.054 (3)	0.038 (3)	0.040 (3)	-0.005 (2)	0.023 (3)	-0.009(2)
C25	0.061 (3)	0.042 (3)	0.038 (3)	-0.013 (3)	0.021 (3)	-0.014 (3)
C26	0.050 (3)	0.049 (3)	0.033 (3)	-0.011 (3)	0.012 (2)	-0.005 (3)
C27	0.043 (3)	0.037 (3)	0.033 (3)	0.002 (2)	0.013 (2)	0.004 (2)
C28	0.042 (3)	0.029 (2)	0.028 (2)	0.002 (2)	0.010 (2)	0.008 (2)
C29	0.045 (3)	0.023 (2)	0.026 (2)	0.0069 (19)	0.011 (2)	0.0031 (19)
C30	0.042 (2)	0.024 (2)	0.025 (2)	0.0043 (19)	0.0094 (19)	0.0014 (19)
C31	0.048 (3)	0.021 (2)	0.021 (2)	0.0047 (19)	0.0066 (19)	0.0026 (19)
C32	0.058 (3)	0.025 (2)	0.030 (3)	0.006 (2)	0.014 (2)	-0.001 (2)
C33	0.072 (4)	0.023 (3)	0.039 (3)	-0.004 (2)	0.021 (3)	-0.001 (2)
C34	0.095 (5)	0.020 (3)	0.051 (4)	-0.003 (3)	0.027 (3)	-0.009 (3)
C35	0.075 (4)	0.028 (3)	0.033 (3)	0.006 (3)	0.020 (3)	-0.006 (2)
C36	0.050(3)	0.024 (2)	0.023 (2)	0.003 (2)	0.009 (2)	-0.0012 (19)
C37	0.045 (3)	0.032 (3)	0.021 (2)	0.009 (2)	0.002 (2)	0.001 (2)
C38	0.053 (3)	0.038 (3)	0.028 (3)	0.009 (2)	0.008 (2)	-0.009(2)
C39	0.065 (4)	0.054 (4)	0.030 (3)	0.014 (3)	0.018 (3)	-0.008 (3)
C40	0.054 (3)	0.054 (4)	0.023 (2)	0.011 (3)	0.011 (2)	0.004 (2)
C41	0.046 (3)	0.031 (3)	0.033 (3)	0.004 (2)	0.013 (2)	0.004 (2)
C42	0.052 (3)	0.028 (3)	0.019 (2)	0.010(2)	0.006 (2)	0.0008 (19)
C43	0.032 (2)	0.056 (3)	0.026 (2)	0.014 (2)	0.0044 (19)	-0.007(2)
C44	0.032 (2)	0.046 (3)	0.031 (3)	0.011 (2)	0.003 (2)	-0.010 (2)
C45	0.036 (3)	0.057 (4)	0.045 (3)	0.017 (3)	-0.002 (2)	-0.021 (3)
C46	0.043 (3)	0.039 (3)	0.074 (5)	0.008 (2)	0.002 (3)	-0.024 (3)
C47	0.052 (4)	0.056 (4)	0.103 (6)	0.009 (3)	0.006 (4)	-0.036 (4)
C48	0.054 (4)	0.070 (5)	0.123 (8)	0.019 (4)	-0.005 (4)	-0.062 (5)
C49	0.053 (4)	0.093 (6)	0.082 (6)	0.024 (4)	0.001 (4)	-0.048 (5)
C50	0.037 (3)	0.080 (5)	0.048 (4)	0.018 (3)	-0.003 (3)	-0.034 (4)
C51	0.043 (3)	0.104 (6)	0.028 (3)	0.023 (3)	0.002 (2)	-0.024 (3)
C52	0.045 (3)	0.129 (7)	0.036 (3)	0.022 (4)	0.005 (3)	-0.028 (4)
C53	0.041 (3)	0.186 (11)	0.031 (3)	0.028 (5)	0.007 (3)	-0.022 (5)
C54	0.045 (3)	0.155 (9)	0.028 (3)	0.007 (4)	0.009 (3)	-0.001 (4)
C55	0.037 (3)	0.115 (6)	0.025 (3)	0.006 (3)	0.007 (2)	-0.010 (3)
C56	0.031 (2)	0.087 (5)	0.027 (3)	0.013 (3)	0.006 (2)	-0.012 (3)
C57	0.034 (2)	0.031 (3)	0.027 (2)	0.0035 (19)	0.0096 (19)	-0.003 (2)
C58	0.037 (2)	0.034 (3)	0.026 (2)	0.000(2)	0.0141 (19)	-0.002 (2)
C59	0.041 (2)	0.027 (2)	0.025 (2)	0.0019 (19)	0.0137 (19)	0.0026 (19)
C60	0.054 (3)	0.032 (3)	0.037 (3)	-0.002 (2)	0.023 (2)	0.001 (2)
C61	0.065 (4)	0.031 (3)	0.048 (3)	0.000 (3)	0.028 (3)	-0.007 (3)
C62	0.061 (3)	0.027 (3)	0.042 (3)	-0.012 (2)	0.018 (3)	-0.004 (2)
C63	0.044 (3)	0.034 (3)	0.030 (3)	-0.004 (2)	0.012 (2)	0.005 (2)
C64	0.041 (2)	0.028 (2)	0.018 (2)	0.0001 (19)	0.0091 (18)	0.0051 (19)

C65	0.041 (2)	0.031 (3)	0.022 (2)	0.001 (2)	0.0104 (19)	0.0010 (19)
C66	0.037 (2)	0.037 (3)	0.028 (2)	0.000 (2)	0.0108 (19)	0.006 (2)
C67	0.033 (2)	0.052 (3)	0.031 (3)	0.009 (2)	0.011 (2)	0.005 (2)
C68	0.044 (3)	0.042 (3)	0.038 (3)	0.002 (2)	0.017 (2)	-0.011 (2)
C69	0.039 (3)	0.041 (3)	0.036 (3)	-0.004 (2)	0.013 (2)	-0.009 (2)
C70	0.032 (2)	0.034 (3)	0.030 (2)	-0.0023 (19)	0.0083 (19)	-0.003 (2)
N1	0.042 (2)	0.0176 (18)	0.039 (2)	-0.0029 (16)	0.0183 (18)	0.0038 (17)
N2	0.047 (2)	0.020 (2)	0.034 (2)	0.0032 (17)	0.0156 (18)	-0.0009 (17)
N3	0.0329 (19)	0.029 (2)	0.032 (2)	-0.0009 (16)	0.0119 (17)	0.0026 (17)
N4	0.0354 (19)	0.027 (2)	0.026 (2)	0.0023 (16)	0.0130 (16)	0.0018 (16)
N5	0.053 (2)	0.0196 (19)	0.032 (2)	0.0056 (17)	0.0206 (19)	0.0067 (17)
N6	0.038 (2)	0.0146 (17)	0.032 (2)	-0.0032 (15)	0.0109 (16)	0.0013 (16)
N7	0.034 (2)	0.046 (3)	0.029 (2)	0.0027 (18)	0.0125 (17)	0.0015 (19)
N8	0.038 (2)	0.032 (2)	0.033 (2)	0.0069 (17)	0.0086 (17)	-0.0043 (18)
N9	0.042 (2)	0.027 (2)	0.036 (2)	-0.0039 (17)	0.0156 (18)	-0.0092 (18)
N10	0.052 (2)	0.025 (2)	0.040 (2)	0.0038 (18)	0.021 (2)	-0.0010 (19)
01	0.049 (2)	0.0295 (18)	0.046 (2)	-0.0043 (15)	0.0243 (17)	-0.0047 (16)
O2	0.055 (2)	0.0144 (15)	0.0378 (19)	0.0011 (13)	0.0274 (16)	0.0017 (14)
O3	0.0344 (18)	0.039 (2)	0.058 (2)	-0.0016 (15)	0.0185 (17)	-0.0068 (19)
04	0.0332 (16)	0.0310 (17)	0.0326 (18)	-0.0019 (13)	0.0137 (14)	-0.0042 (15)
05	0.064 (2)	0.0268 (18)	0.047 (2)	-0.0021 (16)	0.034 (2)	0.0006 (17)
O6	0.0429 (18)	0.0201 (15)	0.0347 (18)	-0.0032 (13)	0.0155 (15)	-0.0033 (14)
O7	0.043 (2)	0.054 (2)	0.043 (2)	-0.0077 (17)	0.0235 (17)	-0.0037 (19)
08	0.0345 (16)	0.0273 (17)	0.041 (2)	0.0049 (13)	0.0123 (15)	-0.0044 (15)
09	0.0384 (18)	0.0263 (17)	0.048 (2)	0.0018 (14)	0.0167 (16)	-0.0074 (16)
O10	0.057 (2)	0.0255 (18)	0.059 (3)	0.0003 (16)	0.025 (2)	0.0035 (18)
Ni1	0.0451 (5)	0.0169 (4)	0.0317 (4)	0.0006 (3)	0.0168 (4)	0.0009 (3)
Ni2	0.0342 (4)	0.0274 (4)	0.0293 (4)	0.0031 (3)	0.0127 (3)	-0.0014 (3)
Ni3	0.0348 (4)	0.0182 (4)	0.0311 (4)	0.0013 (3)	0.0137 (3)	0.0008 (3)
F1	0.071 (2)	0.0393 (18)	0.058 (2)	0.0011 (16)	0.0272 (18)	0.0146 (16)
F2	0.072 (2)	0.048 (2)	0.072 (3)	-0.0111 (17)	0.041 (2)	-0.0011 (19)
F3	0.053 (2)	0.056 (2)	0.083 (3)	-0.0035 (16)	0.0350 (19)	-0.004 (2)
F4	0.0507 (18)	0.0436 (19)	0.049 (2)	0.0051 (15)	0.0097 (15)	-0.0025 (16)
B1	0.061 (4)	0.037 (4)	0.051 (4)	-0.002 (3)	0.025 (3)	-0.001 (3)
B2	0.045 (3)	0.035 (3)	0.059 (4)	-0.005 (3)	0.028 (3)	-0.003 (3)

Geometric parameters (Å, °)

C1—N1	1.307 (6)	C44—C45	1.485 (8)
C1—C2	1.475 (7)	C45—C50	1.385 (9)
C1-C14	1.482 (7)	C45—C46	1.399 (10)
C2—N2	1.291 (6)	C46—C47	1.388 (9)
C2—C3	1.470 (7)	C46—H46	0.9500
C3—C4	1.394 (7)	C47—C48	1.392 (12)
C3—C8	1.418 (7)	C47—H47	0.9500
C4—C5	1.394 (7)	C48—C49	1.375 (14)
C4—H4	0.9500	C48—H48	0.9500
C5—C6	1.396 (9)	C49—C50	1.413 (10)

С5—Н5	0.9500	C49—H49	0.9500
C6—C7	1.367 (8)	C50—C51	1.464 (11)
С6—Н6	0.9500	C51—C52	1.403 (9)
С7—С8	1.400 (7)	C51—C56	1.405 (10)
С7—Н7	0.9500	C52—C53	1.364 (13)
C8—C9	1.458 (8)	С52—Н52	0.9500
C9—C14	1.410 (7)	C53—C54	1.395 (13)
C9—C10	1.417 (7)	С53—Н53	0.9500
C10—C11	1.369 (8)	C54—C55	1.399 (9)
С10—Н10	0.9500	C54—H54	0.9500
C11—C12	1.372 (8)	C55—C56	1.404 (10)
С11—Н11	0.9500	С55—Н55	0.9500
C12—C13	1.391 (7)	C57—N9	1.273 (6)
С12—Н12	0.9500	C57—C70	1.466 (7)
C13—C14	1.395 (7)	C57—C58	1.502 (7)
С13—Н13	0.9500	C58—N10	1.308 (6)
C15—N3	1,309 (6)	C58—C59	1.470 (7)
C15—C28	1.464 (7)	C59—C60	1.383 (7)
C15—C16	1.467 (7)	C59—C64	1.407(7)
C16—N4	1 314 (6)	C60—C61	1 376 (8)
C16—C17	1.464 (6)	C60—H60	0.9500
C17—C18	1.404 (7)	C61—C62	1.377 (8)
C17—C22	1.416 (7)	C61—H61	0.9500
C18 - C19	1 391 (7)	C62 - C63	1 385 (8)
C18—H18	0.9500	C62 - H62	0.9500
C19-C20	1 384 (8)	C63 - C64	1407(7)
C19—H19	0.9500	C63—H63	0.9500
C20—C21	1.376 (8)	C64—C65	1.483 (7)
C20—H20	0.9500	C65—C66	1.105(7) 1 398(7)
$C_{21} - C_{22}$	1 404 (7)	C65-C70	1.396(7) 1 414(7)
C_{21} H21	0.9500	C66—C67	1 384(7)
C^{22} C^{23}	1 465 (7)	C66—H66	0.9500
C_{23} C_{24}	1410(7)	C67—C68	1 381 (8)
C_{23} C_{28}	1.418(7)	C67—H67	0.9500
C24—C25	1.370 (8)	C68—C69	1.382 (7)
C24—H24	0.9500	C68—H68	0.9500
C_{25} C_{26}	1.389 (8)	C69—C70	1.385 (7)
C25—H25	0.9500	C69—H69	0.9500
C26—C27	1,390 (8)	N1-01	1.367 (5)
C26—H26	0.9500	N1—Ni1	1.883 (4)
$C_{27} - C_{28}$	1 399 (7)	N2-02	1 360 (5)
C27—H27	0.9500	N3-03	1.346(5)
$C_{29} N_{5}$	1 315 (6)	N3—Ni2	1.872 (4)
C29—C30	1.469 (7)	N4-04	1.312 (5)
C29—C42	1.476 (7)	N4—Ni2	1.887 (4)
C30—N6	1.304 (6)	N5-05	1.350 (5)
C30-C31	1 460 (7)	N5—Ni1	1 865 (4)
$C_{31} - C_{32}$	1 398 (8)	N6	1 314 (5)
0.51 0.52	1.570 (0)	1.0 00	

C31—C36	1.432 (7)	N6—Ni1	1.875 (4)
C32—C33	1.371 (7)	N7—O7	1.370 (6)
С32—Н32	0.9500	N7—Ni2	1.897 (4)
C33—C34	1.379 (8)	N8—O8	1.352 (5)
С33—Н33	0.9500	N9—O9	1.378 (5)
C34—C35	1.361 (9)	N9—Ni3	2.029 (4)
C34—H34	0.9500	N10-010	1.367(5)
C_{35} C_{36}	1 393 (7)	N10—Ni3	2.064(4)
C35—H35	0.9500	01B1	1.480(8)
C36 C37	1 470 (8)	$O_2 $ Ni1	1.400 (0)
$C_{30} = C_{37}$	1.470(0) 1.207(7)	$O_2 = N_1^2$	1.029(3)
$C_{37} = C_{38}$	1.397(7)	$O_2 = NI_3$	2.025(3)
$C_{3}^{-} - C_{42}^{-}$	1.411(7)	$O_3 - B_2$	1.440(7)
C_{28} U_{28}	1.308 (9)	04—N13	2.022(3)
C38—H38	0.9500	05—B1	1.4/1(8)
C39—C40	1.379 (9)	06—N13	2.036 (3)
С39—Н39	0.9500	0/—B2	1.493 (8)
C40—C41	1.390 (8)	08—Ni2	1.819 (3)
C40—H40	0.9500	08—Ni3	2.036 (3)
C41—C42	1.389 (8)	O9—H1A	0.85 (7)
C41—H41	0.9500	O10—H2A	1.06 (7)
C43—N7	1.298 (7)	F1—B1	1.395 (8)
C43—C56	1.471 (7)	F2—B1	1.372 (8)
C43—C44	1.481 (8)	F3—B2	1.378 (7)
C44—N8	1.290 (7)	F4—B2	1.377 (8)
N1—C1—C2	117.6 (4)	C48—C49—H49	119.1
N1—C1—C14	126.3 (5)	С50—С49—Н49	119.1
C2—C1—C14	116.2 (4)	C45—C50—C49	117.4 (8)
N2—C2—C3	115.7 (5)	C45—C50—C51	119.8 (6)
N2—C2—C1	126.5 (4)	C49—C50—C51	122.8 (7)
C3—C2—C1	117.8 (4)	C52—C51—C56	119.2 (8)
C4—C3—C8	120 7 (5)	C52 - C51 - C50	1189(7)
C4-C3-C2	120.7(5)	$C_{56} - C_{51} - C_{50}$	121.9 (6)
$C_{8} - C_{3} - C_{2}$	120.7(5)	C_{53} C_{52} C_{51}	121.9(0) 120.0(8)
C_{3} C_{4} C_{5}	120.3(5)	C_{53} C_{52} C_{51}	120.0 (0)
$C_3 = C_4 = C_3$	110.0	C51 C52 H52	120.0
$C_5 = C_4 = H_4$	119.9	$C_{51} = C_{52} = C_{54}$	120.0 121.7(7)
C_{3}	119.9	$C_{52} = C_{53} = C_{54}$	121.7(7)
C4 = C5 = U5	119.2 (0)	$C_{52} = C_{53} = H_{53}$	119.1
	120.4	С54—С55—П55	119.1
C6C5H5	120.4	C53—C54—C55	119.2 (9)
C/C6C5	120.6 (5)	С53—С54—Н54	120.4
С/—С6—Н6	119.7	С55—С54—Н54	120.4
С5—С6—Н6	119.7	C54—C55—C56	119.5 (8)
C6—C7—C8	122.0 (5)	С54—С55—Н55	120.2
С6—С7—Н7	119.0	С56—С55—Н55	120.2
С8—С7—Н7	119.0	C55—C56—C51	120.2 (6)
C7—C8—C3	117.3 (5)	C55—C56—C43	120.3 (6)
C7 - C8 - C9	123.3 (5)	C51—C56—C43	119.3 (6)

C3—C8—C9	119.4 (4)	N9—C57—C70	130.1 (5)
C14—C9—C10	117.2 (5)	N9—C57—C58	113.1 (4)
C14—C9—C8	121.8 (4)	C70—C57—C58	116.8 (4)
C10—C9—C8	121.0 (5)	N10-C58-C59	131.6 (4)
C11—C10—C9	121.5 (5)	N10-C58-C57	110.5 (4)
$C_{11} - C_{10} - H_{10}$	119.3	C_{59} C_{58} C_{57}	1174(4)
C9-C10-H10	119.3	C60-C59-C64	1204(5)
C10-C11-C12	121.1 (5)	C60-C59-C58	120.1(3) 121.9(4)
C10-C11-H11	119.4	C64 - C59 - C58	121.5(4)
C_{12} C_{11} H_{11}	119.4	C_{61} C_{60} C_{59}	117.5(4)
$C_{12} = C_{11} = C_{13}$	119.4	C_{61} C_{60} H_{60}	110.8
C11 - C12 - C13	120.5	C_{59} C_{60} H_{60}	119.8
$C_{12} = C_{12} = H_{12}$	120.5	$C_{5}^{60} = C_{50}^{61} = C_{52}^{62}$	119.0 120.3(5)
$C_{13} = C_{12} = C_{14}$	120.3	C60 - C61 - C62	120.3 (3)
$C_{12} = C_{13} = C_{14}$	121.1 (5)	C62 C61 H61	119.8
C12 - C13 - H13	119.5	C_{02} C_{01} C_{01} C_{01} C_{02} C	119.0
C12 - C14 - C0	119.5	$C_{01} = C_{02} = C_{03}$	120.3 (3)
$C_{13} = C_{14} = C_{9}$	120.0(3)	C62 - C62 - H62	119.9
$C_{13} - C_{14} - C_{1}$	121.3(3) 118.2(5)	C63 - C62 - C64	119.9
C_{9} C_{14} C_{15} C_{29}	110.5(5)	$C_{02} = C_{03} = C_{04}$	120.4 (3)
N_{3} $-C_{15}$ $-C_{28}$	127.5(5)	$C_{02} = C_{03} = H_{03}$	119.8
$N_{3} = C_{13} = C_{16}$	112.3 (4)	C64—C63—H63	119.8
$C_{28} = C_{15} = C_{16}$	120.1 (4)	C59 - C64 - C63	118.1 (5)
N4	127.5 (4)	C59—C64—C65	121.3 (4)
N4—C16—C15	112.2 (4)	063-064-065	120.6 (4)
C17 - C16 - C15	120.2 (4)	C66—C65—C70	118.4 (5)
C18 - C17 - C22	119.4 (4)	C66—C65—C64	121.5 (4)
C18—C17—C16	122.8 (5)	C/0—C65—C64	120.0 (4)
C22—C17—C16	117.7 (4)	C67—C66—C65	120.8 (5)
C19—C18—C17	121.0 (5)	С67—С66—Н66	119.6
С19—С18—Н18	119.5	С65—С66—Н66	119.6
С17—С18—Н18	119.5	C68—C67—C66	120.1 (5)
C20—C19—C18	119.4 (5)	С68—С67—Н67	119.9
С20—С19—Н19	120.3	С66—С67—Н67	119.9
С18—С19—Н19	120.3	C67—C68—C69	120.3 (5)
C21—C20—C19	120.4 (5)	С67—С68—Н68	119.9
С21—С20—Н20	119.8	С69—С68—Н68	119.9
С19—С20—Н20	119.8	C68—C69—C70	120.4 (5)
C20—C21—C22	121.8 (5)	С68—С69—Н69	119.8
C20—C21—H21	119.1	С70—С69—Н69	119.8
C22—C21—H21	119.1	C69—C70—C65	120.0 (5)
C21—C22—C17	117.9 (5)	C69—C70—C57	121.9 (4)
C21—C22—C23	120.6 (5)	C65—C70—C57	118.1 (4)
C17—C22—C23	121.3 (4)	C1—N1—O1	116.1 (4)
C24—C23—C28	117.1 (5)	C1—N1—Ni1	126.1 (3)
C24—C23—C22	121.3 (5)	O1—N1—Ni1	115.8 (3)
C28—C23—C22	121.6 (5)	C2—N2—O2	117.3 (4)
C25—C24—C23	122.7 (5)	C15—N3—O3	117.4 (4)
С25—С24—Н24	118.6	C15—N3—Ni2	117.0 (3)

	110.6		
C23—C24—H24	118.6	O3—N3—N12	125.6 (3)
C24—C25—C26	119.7 (5)	O4—N4—C16	121.4 (4)
C24—C25—H25	120.2	O4—N4—Ni2	122.3 (3)
C26—C25—H25	120.2	C16—N4—Ni2	116.2 (3)
C25—C26—C27	119.6 (5)	C29—N5—O5	118.5 (4)
С25—С26—Н26	120.2	C29—N5—Ni1	116.9 (4)
C27—C26—H26	120.2	O5—N5—Ni1	124.6 (3)
C26—C27—C28	121.1 (5)	C30—N6—O6	121.7 (4)
С26—С27—Н27	119.5	C30—N6—Ni1	116.8 (3)
C28—C27—H27	119.5	06—N6—Ni1	121.5(3)
C_{27} C_{28} C_{23}	119.5	$C43$ _N7_O7	121.3(3)
$C_{27} C_{28} C_{25}$	117.7(5) 123.0(5)	C_{43} N7 Nj2	110.+(+) 127.2(4)
$C_{27} = C_{28} = C_{15}$	123.0(3)	C43 - N7 - N2	127.2(4)
$C_{23} = C_{20} = C_{13}$	117.5 (4)	O/-N/-NI2	114.4(3)
N5-C29-C30	112.0 (4)	C44—N8—O8	118.6 (4)
N5—C29—C42	127.8 (5)	C57—N9—O9	117.6 (4)
C30—C29—C42	120.3 (4)	C57—N9—Ni3	117.9 (3)
N6-C30-C31	126.9 (5)	O9—N9—Ni3	122.1 (3)
N6—C30—C29	112.3 (4)	C58—N10—O10	116.3 (4)
C31—C30—C29	120.9 (4)	C58—N10—Ni3	115.7 (3)
C32—C31—C36	118.8 (4)	O10—N10—Ni3	122.3 (3)
C32—C31—C30	123.2 (5)	N1	113.0 (4)
C36—C31—C30	118.0 (5)	N2—O2—Ni1	127.1 (3)
C33—C32—C31	121.8 (5)	N2—O2—Ni3	117.6 (3)
С33—С32—Н32	119.1	Ni1—O2—Ni3	115.17 (16)
С31—С32—Н32	119.1	N3—O3—B2	118.8 (4)
C32—C33—C34	119.7 (6)	N4-04-Ni3	113.1 (3)
C32—C33—H33	120.1	N5-05-B1	119.5 (4)
C34_C33_H33	120.1	N6-06-Ni3	112.9(3)
$C_{35} = C_{34} = C_{33}$	110 5 (5)	N7 07 B2	112.9(3)
$C_{35} = C_{34} = C_{35}$	119.5 (5)	$N_{1} = 07 = B_{2}$	114.4(4) 128.3(3)
$C_{33} = C_{34} = H_{34}$	120.2	$\frac{100-00}{1012}$	120.3(3)
C33—C34—H34	120.2	No-00-NIS	115.4 (5)
$C_{34} - C_{35} - C_{36}$	123.6 (5)	N12-08-N13	116.21 (17)
C34—C35—H35	118.2	N9-09-HIA	109.5
C36—C35—H35	118.2	N10—O10—H2A	109.5
C35—C36—C31	116.6 (5)	O2—N11—N5	170.30 (16)
C35—C36—C37	122.7 (5)	O2—Ni1—N6	88.27 (16)
C31—C36—C37	120.7 (4)	N5—Ni1—N6	82.07 (18)
C38—C37—C42	118.0 (5)	O2—Ni1—N1	90.96 (16)
C38—C37—C36	119.4 (5)	N5—Ni1—N1	98.53 (18)
C42—C37—C36	122.6 (5)	N6—Ni1—N1	173.25 (18)
C39—C38—C37	121.8 (5)	O8—Ni2—N3	170.42 (16)
С39—С38—Н38	119.1	08—Ni2—N4	88.60 (16)
С37—С38—Н38	119.1	N3—Ni2—N4	81.88 (17)
C38—C39—C40	120.3 (5)	O8—Ni2—N7	90.78 (17)
С38—С39—Н39	119.8	N3—Ni2—N7	98.56 (19)
C40—C39—H39	119.8	N4—Ni2—N7	173 36 (18)
C39-C40-C41	119.4 (6)	04—Ni3— 02	89 28 (14)
C39 C40 H40	120.3	04 Ni3 02	87 74 (15)
	120.5		07.77(13)

C41—C40—H40	120.3	O2—Ni3—N9	96.72 (16)
C42—C41—C40	120.9 (5)	O4—Ni3—O6	165.12 (14)
C42—C41—H41	119.6	O2—Ni3—O6	79.77 (13)
C40—C41—H41	119.6	N9—Ni3—O6	103.43 (16)
C41—C42—C37	119.6 (5)	O4—Ni3—O8	81.29 (14)
C41—C42—C29	123.0 (5)	O2—Ni3—O8	93.48 (15)
C37—C42—C29	117.4 (5)	N9—Ni3—O8	164.91 (17)
N7—C43—C56	127.1 (6)	O6—Ni3—O8	89.28 (14)
N7—C43—C44	117.3 (5)	O4—Ni3—N10	107.36 (16)
C56—C43—C44	115.5 (5)	02—Ni3—N10	161.43 (17)
N8-C44-C43	126.5 (5)	N9—Ni3—N10	76 28 (17)
N8-C44-C45	114.4 (5)	06-Ni3-N10	85.12 (16)
C_{43} C_{44} C_{45}	119.1 (5)	08—Ni3—N10	97.08 (16)
C_{50} C_{45} C_{46}	120.8 (6)	$F^2 - B^1 - F^1$	112.4 (5)
C_{50} C_{45} C_{44}	1187(7)	$F_2 = B_1 = O_5$	105.9(5)
C46-C45-C44	120.5(5)	$F_1 = B_1 = O_5$	103.3(5)
C47 - C46 - C45	120.5(3)	$F_2 = B_1 = O_1$	106.8(5)
$C_{47} = C_{40} = C_{45}$	110.5	$F_1 = B_1 = O_1$	100.8(5)
$C_{47} = C_{40} = 1140$	119.5	05 B1 01	108.5(5)
$C_{45} = C_{40} = 1140$	119.5	F_4 F_2 F_3	112.0(5)
$C_{40} = C_{47} = C_{48}$	118.3 (9)	$F_4 = B_2 = F_3$	112.3(5)
$C_{40} = C_{47} = H_{47}$	120.7	$F_2 = P_2 = O_3$	110.4(3)
$C_{40} = C_{47} = H_{47}$	120.7	$F_3 = B_2 = O_3$	103.4(3)
$C_{49} = C_{48} = C_{47}$	120.3 (7)	F4 = B2 = O7	109.1(3)
C49—C48—H48	119.7	$F_3 = B_2 = 07$	106.1 (5)
C47 - C48 - H48	119.7	03—B2—07	113.4 (5)
C48—C49—C30	121.7 (8)		
N1—C1—C2—N2	-33.6(7)	C54—C55—C56—C43	176.8 (5)
C14-C1-C2-N2	146.8 (5)	C52—C51—C56—C55	-1.5(8)
N1-C1-C2-C3	146.9 (5)	C50—C51—C56—C55	176.8 (5)
$C_{14} - C_{1} - C_{2} - C_{3}$	-32.8(6)	C52—C51—C56—C43	-176.3(5)
N2-C2-C3-C4	22.9(7)	C_{50} C_{51} C_{56} C_{43}	2.0.(8)
C1-C2-C3-C4	-157.5(5)	N7-C43-C56-C55	24.5 (8)
N_{2} C_{2} C_{3} C_{8}	-1571(5)	C44-C43-C56-C55	-1574(5)
C1-C2-C3-C8	22.6 (7)	N7-C43-C56-C51	-160.7(5)
C8-C3-C4-C5	0.1 (8)	C44—C43—C56—C51	17.4 (7)
$C_2 - C_3 - C_4 - C_5$	-179.8(5)	N9—C57—C58—N10	-28.2(6)
$C_{3}-C_{4}-C_{5}-C_{6}$	-0.2(9)	C70-C57-C58-N10	152.2(4)
C4—C5—C6—C7	-0.2(10)	N9—C57—C58—C59	145.3 (4)
C5-C6-C7-C8	0.7 (10)	C70—C57—C58—C59	-34.4(6)
C6-C7-C8-C3	-0.7(8)	N10-C58-C59-C60	17 3 (8)
C6-C7-C8-C9	179.0 (5)	C57 - C58 - C59 - C60	-1545(5)
C4-C3-C8-C7	03(8)	N10-C58-C59-C64	-1683(5)
$C_{2} = C_{3} = C_{8} = C_{7}$	-1797(5)	C_{57} C_{58} C_{59} C_{64}	199(6)
$-\frac{1}{2}$	-179 5 (5)	C64 - C59 - C60 - C61	-1.7(8)
$C_{2} = C_{3} = C_{8} = C_{9}$	05(7)	C_{58} C_{59} C_{60} C_{61}	172.6 (5)
C_{7} C_{8} C_{9} C_{14}	166 8 (5)	C_{59} C_{60} C_{61} C_{62}	14(9)
C_{3} C_{8} C_{9} C_{14}	-134(7)	C60-C61-C62-C63	-0.8(9)
05 00 07 017	1 J T (/ J	-00 -002 -003	0.0 (7)

C7—C8—C9—C10	-15.0 (8)	C61—C62—C63—C64	0.3 (8)
C3—C8—C9—C10	164.8 (5)	C60—C59—C64—C63	1.2 (7)
C14—C9—C10—C11	-0.2 (7)	C58—C59—C64—C63	-173.3 (4)
C8—C9—C10—C11	-178.5 (5)	C60—C59—C64—C65	179.8 (4)
C9-C10-C11-C12	1.3 (8)	C58—C59—C64—C65	5.3 (6)
C10-C11-C12-C13	-1.1 (8)	C62—C63—C64—C59	-0.6 (7)
C11—C12—C13—C14	-0.1 (8)	C62—C63—C64—C65	-179.2 (5)
C12—C13—C14—C9	1.2 (7)	C59—C64—C65—C66	164.0 (5)
C12—C13—C14—C1	175.8 (5)	C63—C64—C65—C66	-17.5 (7)
C10-C9-C14-C13	-1.0 (7)	C59—C64—C65—C70	-17.0(7)
C8—C9—C14—C13	177.2 (5)	C63—C64—C65—C70	161.6 (5)
C10-C9-C14-C1	-175.8 (4)	C70—C65—C66—C67	0.5 (7)
C8—C9—C14—C1	2.5 (7)	C64—C65—C66—C67	179.6 (4)
N1—C1—C14—C13	26.0 (7)	C65—C66—C67—C68	-0.3 (8)
C2-C1-C14-C13	-154.4 (5)	C66—C67—C68—C69	0.0 (8)
N1—C1—C14—C9	-159.3 (5)	C67—C68—C69—C70	0.0 (8)
C2-C1-C14-C9	20.3 (6)	C68—C69—C70—C65	0.2 (8)
N3-C15-C16-N4	-5.2 (6)	C68—C69—C70—C57	178.7 (5)
C_{28} C15 C16 N4	173.2 (4)	C66—C65—C70—C69	-0.5(7)
N3-C15-C16-C17	170.7 (4)	C64—C65—C70—C69	-179.5(4)
C_{28} C_{15} C_{16} C_{17}	-10.9(7)	C66-C65-C70-C57	-179.1(4)
N4-C16-C17-C18	-31(8)	C64 - C65 - C70 - C57	19(7)
C_{15} C_{16} C_{17} C_{18}	-1784(4)	N9-C57-C70-C69	24 8 (8)
N4-C16-C17-C22	174 2 (5)	$C_{58} - C_{57} - C_{70} - C_{69}$	-1556(5)
C_{15} C_{16} C_{17} C_{22}	-10(7)	N9-C57-C70-C65	-156.6(5)
C_{22} C_{17} C_{18} C_{19}	-1.5(8)	$C_{58} = C_{57} = C_{70} = C_{65}$	23.0(6)
C_{16} C_{17} C_{18} C_{19}	175.9 (5)	C_{2} C_{1} N_{1} O_{1}	-1737(4)
$C_{17} - C_{18} - C_{19} - C_{20}$	-13(9)	C_{14} C_{1} N_{1} O_{1}	60(7)
$C_{17} = C_{10} = C_{17} = C_{20}$	1.5(0)	$C_2 = C_1 = N_1 = N_1$	0.0(7)
C19 - C20 - C21	2.3(9)	$C_2 = C_1 = N_1 = N_1$	-157.2(0)
$C_{13} = C_{20} = C_{21} = C_{22}$	-1.7(9)	$C_1 = C_1 = N_1 = N_1$	-175 1 (4)
$C_{20} = C_{21} = C_{22} = C_{17}$	1.7(6) 174.2(5)	$C_3 = C_2 = N_2 = O_2$	173.1(4) 5 3 (7)
$C_{20} = C_{21} = C_{22} = C_{23}$	1/4.2(3)	$C_1 = C_2 = N_2 = O_2$	3.3(7)
$C_{16} = C_{17} = C_{22} = C_{21}$	2.9(7) -174.6(5)	$C_{20} = C_{13} = N_3 = O_3$	2.9(7)
$C_{10} = C_{17} = C_{22} = C_{21}$	-172.0(3)	$C_{10} = C_{13} = N_3 = 0_3$	178.9(4)
$C_{16} = C_{17} = C_{22} = C_{23}$	1/2.9(4)	$C_{20} = C_{13} = N_3 = N_{12}$	170.9(4)
$C_{10} = C_{17} = C_{22} = C_{23}$	9.0(7)	C17 C16 N4 O4	1.4(3)
$C_{21} = C_{22} = C_{23} = C_{24}$	-3.0(8)	C17 - C10 - N4 - O4	9.2(7)
C17 - C22 - C23 - C24	1/0.7(3)	C13 - C16 - N4 - 04	-1/3.3(4)
$C_{21} = C_{22} = C_{23} = C_{28}$	1/7.0(3)	C17 - C10 - N4 - N2	-108.8(4)
C17 - C22 - C23 - C28	-0.0(7)	C13 - C10 - N4 - N12	0.8(3)
$C_{28} = C_{23} = C_{24} = C_{25}$	2.0(8)	$C_{30} = C_{29} = N_5 = O_5$	-1/8.1(4)
$C_{22} = C_{23} = C_{24} = C_{25}$	-1/3.3(3)	$C_{42} = C_{29} = IN_{3} = O_{3}$	U.0 (0)
123 - 124 - 125 - 126	0.8(9)	$C_{20} = C_{29} = N_{20} = N_{11}$	1.0 (5)
124 - 125 - 126 - 127	-2.5 (9)	C42 - C29 - IN3 - IN11	1/9.9 (4)
123 - 126 - 127 - 128	1.4 (ð)	C_{20} C_{20} N_{0} N_{0} C_{0}	1./(/)
$C_{20} - C_{27} - C_{28} - C_{23}$	1.5 (8)	C29-C30-N6-O6	-1/8.3(4)
C26—C27—C28—C15	-1/8.9 (5)	C31—C30—N6—N11	-17/8.0(4)
C24—C23—C28—C27	-3.1 (7)	C29—C30—N6—Ni1	2.0 (5)

C^{22} C^{22} C^{29} C^{27}	174.2(5)	C56 C42 N7 O7	20(0)
$C_{22} = C_{23} = C_{26} = C_{27}$	174.2(3)	$C_{30} - C_{43} - N_{7} - O_{7}$	2.0(0) -175.3(4)
$C_{24} = C_{23} = C_{26} = C_{15}$	-5.4(7)	$C_{44} = C_{43} = N7 = 07$	-160.1(4)
$N_{22} = C_{23} = C_{26} = C_{13}$	3.4(7)	$C_{44} C_{43} N_7 N_2^{-1}$	100.1(4)
13 - 13 - 23 - 27	-165.7(5)	$C_{44} = C_{43} = N_7 = N_1^2$	21.8(7)
C10 - C13 - C28 - C27	-103.7(3)	C45 - C44 - N8 - O8	4.0(0)
$N_{3} = C_{13} = C_{20} = C_{23}$	-108.0(3)	$C_{43} = C_{44} = N_8 = 0_8$	-1/3.8(4)
C10-C13-C20-C23	13.9(7)	$C_{10} = C_{31} = N_{9} = 0_{9}$	0.4(8)
$N_{3} = C_{29} = C_{30} = N_{6}$	-1.9(6)	$C_{38} = C_{37} = N_{9} = 0_{9}$	-1/9.2(4)
C42 - C29 - C30 - N6	1/9.1 (4)	C/0 - C5/ - N9 - N13	-162.4 (4)
N5-C29-C30-C31	1/8.1 (4)	C58 - C57 - N9 - N13	18.0 (5)
C42 - C29 - C30 - C31	-0.9(7)	C59—C58—N10—O10	7.1 (8)
N6—C30—C31—C32	-0.6 (8)	C57—C58—N10—O10	179.3 (4)
C29—C30—C31—C32	179.4 (5)	C59—C58—N10—Ni3	-147.0 (4)
N6—C30—C31—C36	179.1 (5)	C57—C58—N10—Ni3	25.2 (5)
C29—C30—C31—C36	-0.9 (7)	C1—N1—O1—B1	137.4 (5)
C36—C31—C32—C33	2.2 (8)	Ni1—N1—O1—B1	-57.6 (5)
C30—C31—C32—C33	-178.1 (5)	C2—N2—O2—Ni1	32.8 (6)
C31—C32—C33—C34	-1.0 (9)	C2—N2—O2—Ni3	-142.9 (4)
C32—C33—C34—C35	-0.1 (10)	C15—N3—O3—B2	-175.3 (5)
C33—C34—C35—C36	0.0 (10)	Ni2—N3—O3—B2	4.4 (7)
C34—C35—C36—C31	1.2 (8)	C16—N4—O4—Ni3	-145.0 (4)
C34—C35—C36—C37	179.5 (6)	Ni2—N4—O4—Ni3	32.9 (4)
C32—C31—C36—C35	-2.2 (7)	C29—N5—O5—B1	-173.4 (5)
C30—C31—C36—C35	178.1 (5)	Ni1—N5—O5—B1	7.6 (7)
C32—C31—C36—C37	179.4 (4)	C30—N6—O6—Ni3	-146.0 (4)
C30—C31—C36—C37	-0.3 (7)	Ni1—N6—O6—Ni3	33.6 (4)
C35—C36—C37—C38	4.1 (7)	C43—N7—O7—B2	139.1 (5)
C31—C36—C37—C38	-177.6 (4)	Ni2—N7—O7—B2	-55.7 (5)
C35—C36—C37—C42	-174.7(5)	C44—N8—O8—Ni2	29.2 (6)
C_{31} C_{36} C_{37} C_{42}	3.6(7)	C44—N8—O8—Ni3	-146.7(4)
C42 - C37 - C38 - C39	19(8)	$N^2 = O^2 = Ni1 = N6$	1527(4)
$C_{12} = C_{37} = C_{38} = C_{39}$	-1770(5)	Ni3 = 02 = Ni1 = N6	-31.5(2)
C_{37} C_{38} C_{39} C_{40}	-0.1(9)	N2 = O2 = Ni1 = N1	-340(4)
C_{38} C_{39} C_{40} C_{41}	-0.6(8)	$Ni3_02_Ni1_N1$	141.8(2)
$C_{39} - C_{40} - C_{41} - C_{42}$	-0.6(8)	C_{29} N5 Ni1 N6	141.0(2)
C_{40} C_{41} C_{42} C_{37}	24(8)	05-N5-Ni1-N6	1790(4)
$C_{40} = C_{41} = C_{42} = C_{37}$	-176 A (5)	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	-173.2(4)
$C_{40} = C_{41} = C_{42} = C_{23}$	1/0.4(3)	C_{29} NJ NJ NJ	173.2 (4) 5 8 (4)
$C_{36} = C_{37} = C_{42} = C_{41}$	3.0(7)	C_{20} NG NG C_{20}	3.8(4)
$C_{30} = C_{37} = C_{42} = C_{41}$	175.0(4)	C_{30} NG NG C_{30}	177.9(4)
$C_{38} = C_{37} = C_{42} = C_{29}$	1/3.9 (4) 5 2 (7)	00 NG NI N5	-1.7(3)
$C_{30} = C_{37} = C_{42} = C_{29}$	-5.2(7)	C_{30} NG Ni1 N5	-1.2(4)
N5-C29-C42-C41	4.0 (8)	06-N6-N11-N5	1/9.1 (4)
$C_{30} - C_{29} - C_{42} - C_{41}$	-1//.2 (4)	$U_1 - N_1 - N_1 - O_2$	4.2 (4)
N5-C29-C42-C37	-1/4.9(5)	O1— $N1$ — $N1$ — $O2$	-159.1 (3)
C30—C29—C42—C37	3.9 (7)	C1 - N1 - N1 - N5	-177.8 (4)
N/C43C44N8	-29.8 (8)	01—N1—N1—N5	19.0 (4)
C56—C43—C44—N8	151.8 (5)	N8—O8—Ni2—N4	157.2 (4)
N7—C43—C44—C45	150.5 (5)	Ni3—O8—Ni2—N4	-27.0(2)

C56—C43—C44—C45	-27.8 (7)	N8—O8—Ni2—N7	-29.5 (4)
N8—C44—C45—C50	-161.0 (5)	Ni3—O8—Ni2—N7	146.4 (2)
C43—C44—C45—C50	18.7 (7)	C15—N3—Ni2—N4	1.7 (4)
N8—C44—C45—C46	18.7 (8)	O3—N3—Ni2—N4	-178.0 (4)
C43—C44—C45—C46	-161.6 (5)	C15—N3—Ni2—N7	-171.6 (4)
C50—C45—C46—C47	-0.2 (9)	O3—N3—Ni2—N7	8.7 (4)
C44—C45—C46—C47	-179.9 (6)	O4—N4—Ni2—O8	-3.9(3)
C45—C46—C47—C48	-0.4 (11)	C16—N4—Ni2—O8	174.1 (4)
C46—C47—C48—C49	0.8 (12)	O4—N4—Ni2—N3	177.1 (4)
C47—C48—C49—C50	-0.6 (12)	C16—N4—Ni2—N3	-4.9 (3)
C46—C45—C50—C49	0.4 (9)	C43—N7—Ni2—O8	2.3 (5)
C44—C45—C50—C49	-179.9 (5)	O7—N7—Ni2—O8	-161.0 (3)
C46—C45—C50—C51	-178.3 (5)	C43—N7—Ni2—N3	-179.8 (5)
C44—C45—C50—C51	1.4 (8)	O7—N7—Ni2—N3	16.9 (4)
C48—C49—C50—C45	0.0 (10)	N5—O5—B1—F2	-161.0 (5)
C48—C49—C50—C51	178.7 (7)	N5—O5—B1—F1	76.6 (7)
C45—C50—C51—C52	166.2 (5)	N5	-45.0 (7)
C49—C50—C51—C52	-12.4 (9)	N1—O1—B1—F2	-170.9 (4)
C45—C50—C51—C56	-12.1 (8)	N1-O1-B1-F1	-49.6 (6)
C49—C50—C51—C56	169.3 (6)	N1-01-B1-05	73.6 (6)
C56—C51—C52—C53	-0.8 (9)	N3—O3—B2—F4	80.1 (6)
C50—C51—C52—C53	-179.1 (6)	N3—O3—B2—F3	-158.4 (5)
C51—C52—C53—C54	2.5 (10)	N3—O3—B2—O7	-42.7 (7)
C52—C53—C54—C55	-1.9 (10)	N7—O7—B2—F4	-50.1 (6)
C53—C54—C55—C56	-0.4 (9)	N7—O7—B2—F3	-171.4 (4)
C54—C55—C56—C51	2.1 (8)	N7—O7—B2—O3	73.4 (6)

Hydrogen-bond geometry (Å, °)

D—H···A	D—H	H···A	D····A	<i>D</i> —H··· <i>A</i>
O9—H1A…N2	0.85	1.96	2.771 (6)	158
O10—H2A…N8	1.06	1.77	2.765 (6)	155