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Crystal structure of octakis(N,N-dimethylformamide- κO)europium(III) tetracosa- μ_2 -oxidododecaoxido- μ_{12} -phosphato-dodecamolybdate(VI)

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In the title salt, $[Eu(C_3H_7NO)_8][PMo_{12}O_{40}]$, the asymmetric unit comprises one α -Keggin-type $[PMo_{12}O_{40}]^{3-}$ polyoxidometalate anion and one distorted dodecahedral $[Eu(C_3H_7NO)_8]^{3+}$ complex cation. In the crystal, the isolated polyoxidometalate anions are packed into hexagonally arranged rows extending parallel to [001]. The complex cations are situated between the rows and are linked to the neighbouring anions through weak C-H···O hydrogen-bonding interactions, leading to the formation of a three-dimensional network structure.

1. Chemical context

Polyoxidometalates (POMs) are versatile metal-oxygen complexes which have attracted interest due to their topological properties and their potential applications in catalysis, photoluminescence, electrochromism and magnetism (Long *et al.*, 2010; Pope & Müller, 2010; Coronado & Gómez-García, 1998). Up to date, a variety of strategies have been developed and used to assemble POM-based hybrid materials by controlling reaction factors such as metal ions, organic ligands, POM species, pH, molar ratio of raw materials or reaction environments (Wang *et al.*, 2013; Liu *et al.*, 2013). Even with these approaches, the design and synthesis of new stable polyoxidomolybdate complexes are still challenging.





Herein, we report on the synthesis, UV–vis and IR spectra along with the crystal structure of the hybrid europium(III) POM title compound, $[Eu(C_3H_7NO)_8][PMo_{12}O_{40}]$, (I).



Figure 1

The molecular structures of the cation and anion in compound (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 40% probability level. H atoms have been omitted for clarity.

2. Structural commentary

The structures of the molecular components of compound (I) are illustrated in Fig. 1. The $[PMo_{12}O_{40}]^{3-}$ polyoxidoanion of (I) exhibits a classical α -Keggin-type structure. The central P atom is tetrahedrally surrounded with all four oxygen atoms (O_a) linked to four Mo₃O₁₃ moieties. The latter species are fused together by sharing corner atoms (O_b) and consist of three MoO₆ octahedra condensed in a triangular arrangement by sharing edges (O_c). There is also a terminal oxygen atom (O_d) in every MoO₆ octahedron. The P–O bond lengths range from 1.521 (5) Å to 1.536 (4) Å and the Mo–O bond lengths from 1.690 (5) Å to 2.438 (4) Å. The O–P–O angles [109.1 (2)–109.8 (3)°] indicate only a slight distortion of the central PO₄ tetrahedron. The Eu^{III} cation is coordinated by eight dimethylformamide ligands through their oxygen atoms



Figure 2 The contents of the unit cell of complex (I). H atoms have been omitted for clarity.

with Eu–O distances from 2.369 (5) to 2.416 (6) Å. These values are comparable to those of related oxido-europium(III) species, *e.g* for the $[Eu(thd)_3(DMF)_2]$ complex (thd is the ion of 2,2,6,6-tetramethyl-3,5-heptanedione) with Eu–O = 2.494 (5)–2.442 (5) Å (Cunningham & Siever, 1980). Calculations with the *SHAPE* software (Alvarez *et al.*, 2005) indicate that the coordination polyhedron of Eu^{III} is a slightly distorted dodecahedron approaching molecular D_{2d} symmetry (Casanova *et al.*, 2005).





C-H···O hydrogen bonds (dashed lines) link one $[Eu(dmf)_8]^{3+}$ cation to four neighbouring α -Keggin-type $[PMo_{12}O_{40}]^{3-}$ anions. Symmetry codes refer to Table 1.

research communications

Table 1	
Hydrogen-bond	geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$C3-H4\cdots O31^{i}$	0.96	2.72	3.289 (11)	118
C4-H7···O36 ⁱⁱ	0.96	2.61	3.476 (12)	150
C7−H13···O7	0.96	2.65	3.473 (12)	145
$C8-H15\cdots O28^{ii}$	0.96	2.51	3.316 (13)	141
$C8-H16\cdots O5$	0.96	2.58	3.424 (11)	147
C10−H18···O28 ⁱⁱ	0.93	2.62	3.516 (10)	162
$C15-H29\cdots O16^{i}$	0.96	2.55	3.299 (12)	135
$C15-H27\cdots O32^{iii}$	0.96	2.52	3.440 (10)	160
$C15-H28\cdots O52^{iv}$	0.96	2.29	3.174 (10)	153
C16-H31···O6 ⁱⁱⁱ	0.96	2.63	3.541 (11)	159
C22-H41···O55	0.96	2.53	3.280 (12)	135
$C22-H42\cdots O11^{iv}$	0.96	2.45	3.305 (11)	149
C22-H43···O45	0.96	2.61	3.567 (13)	172
$C23-H44\cdots O17^{v}$	0.96	2.52	3.443 (11)	161
C25-H55···O24	0.93	2.53	3.337 (11)	145
C25-H55···O44	0.93	2.58	3.091 (10)	115
$C28-H56\cdots O36^{v}$	0.93	2.62	3.476 (11)	153
$C30-H60\cdots O36^{v}$	0.96	2.58	3.458 (14)	152

Symmetry codes: (i) $x + \frac{1}{2}, -y + \frac{1}{2}, z - 1$; (ii) $-x + 1, -y + 1, z - \frac{1}{2}$; (iii) x, y, z - 1; (iv) $-x + 1, -y, z - \frac{1}{2}$; (v) $x + \frac{1}{2}, -y + \frac{1}{2}, z$.

3. Supramolecular features

The unit cell content of the title compound is illustrated in Fig. 2. In the crystal structure of (I), each $[Eu(DMF)_8]^{3+}$ cation is linked to four neighbouring α -Keggin-type $[PMo_{12}O_{40}]^{3-}$ anions through C-H···O hydrogen-bonding interactions between the methyl groups of the DMF ligands and the terminal-oxygen (O_d) and the bridging-oxygen atoms $(O_{b,c})$ of the $[PMo_{12}O_{40}]^{3-}$ anions (Fig. 3, Table 1). The C(donor)···O_d(acceptor) distances are between 3.174 (10) and 3.541 (11) Å while the C···O(_{b,c}) distances are between 3.289 (11) and 3.473 (12) Å. In the crystal packing, the POM anions are packed into hexagonally arranged rows extending parallel to [001] with the $[Eu(DMF)_8]^{3+}$ cations located between the rows (Fig. 4).

4. Synthesis and crystallization

The starting material $[(C_4H_9)_4N)_4H_3][PMo_{11}O_{39}]$ was prepared using a literature method (Combs-Walker & Hill, 1998). EuCl₃·6H₂O (361.41 mg, 1 mmol) and isonicotinic acid



Figure 4 The crystal packing of (I) with the $[PMo_{12}O_{40}]^{3-}$ anions in polyhedral representation.



The FT-IR spectrum of (I).

 $(C_6H_5NO_2)$ (123.11 mg, 1 mmol) were dissolved in 10 ml of dimethylformamide. This solution was added dropwise to a yellow dimethylformamide solution of $[(C_4H_9)_4N)_4H_3]$ - $[PMo_{11}O_{39}]$ (0.33 mmol in 10 ml). The mixture was heated under stirring for 1 h at 333 K. Single crystals of the title compounds were obtained by slow diffusion of 2-propanol through the dimethylformamide solution. UV-vis spectrum in dimethylformamide: λ_{max} (nm) 315 and 205.

5. FT-IR spectroscopy

The FT–IR spectrum was recorded in the range 4000– 400 cm^{-1} on a Nicolet 470 FT–IR spectrophotometer with pressed KBr pellets.

The FT–IR spectrum of (I) (Fig. 5) exhibits characteristic bands attributed to the stretching and deformation modes of the Mo–O bond vibration of the $[PMo_{12}O_{40}]^{3-}$ anion in the region 1100–400 cm⁻¹. Thus, the asymmetric vibration $v_{as}(P-O_a)$, $v_{as}(Mo=O_d)$, $v_{as}(Mo-O_b-Mo)$ and $v_{as}(Mo-O_c-Mo)$ appear at 1065, 951, 885 and 974 cm⁻¹, respectively (Masteri-Farahani & Shahbazi, 2012). The absorption bands at 1265 and 1657 cm⁻¹ are characteristic of the asymmetric vibration of the C–N and the C=O bonds, respectively. The vibration bands at 1115, 1440, 1385 and 2964 cm⁻¹ are attributed to the vibration ρ (CH₃) (rocking vibration), δ_a (CH₃), δ_s (CH₃) and ν (C-H) of the dimethylformamide ligand (Durgaprasad *et al.*, 1971).

6. Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2. Hydrogen atoms were placed in calculated positions and refined as riding atoms: C-H =0.93 Å and $U_{iso}(H) = 1.2U_{eq}(C)$ for methine groups and C-H= 0.96 Å and $U_{iso}(H) = 1.5U_{eq}(C)$ for methyl groups. The refined Flack parameter (Parsons *et al.*, 2013) of -0.015 (7) indicates the correct determination of the absolute structure. Table 2Experimental details.

Crystal data Chemical formula [Eu(C₃H₇NO)₈][PMo₁₂O₄₀] 2558.97 М., Crystal system, space group Orthorhombic, Pna21 Temperature (K) 296 26.9108 (10), 18.3506 (6), *a*, *b*, *c* (Å) 13.4494 (4) $V(Å^3)$ 6641.7 (4) Z4 Radiation type Μο Κα μ (mm⁻¹) 3 24 Crystal size (mm) $0.20 \times 0.18 \times 0.17$ Data collection Bruker APEXII CCD Diffractometer Absorption correction Multi-scan (SADABS; Bruker, 2006) 0.667. 0.747 T_{\min}, T_{\max} No. of measured, independent and 88354, 33285, 19583 observed $[I > 2\sigma(I)]$ reflections 0.075 R_{int} $(\sin \theta / \lambda)_{\text{max}} (\text{\AA}^{-1})$ 0.849 Refinement $R[F^2 > 2\sigma(F^2)], wR(F^2), S$ 0.047, 0.095, 0.98 No. of reflections 33285 No. of parameters 863 No. of restraints 1 H-atom treatment H-atom parameters constrained $\Delta \rho_{\rm max}, \Delta \rho_{\rm min}$ (e Å 2.08, -1.80Flack x determined using 6924 Absolute structure quotients $[(I^+)-(I^-)]/[(I^+)+(I^-)]$ Absolute structure parameter -0.015(7)

Computer programs: *APEX2* and *SAINT* (Bruker, 2006), *SIR2004* (Burla *et al.*, 2005), *SHELXL2014* (Sheldrick, 2015), *ORTEP-3 for Windows* (Farrugia, 2012), *DIAMOND* (Putz & Brandenburg, 2014) and *WinGX* (Farrugia, 2012).

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Crystal structure of octakis (N, N-dimethylformamide- κO) europium (III) tetracosa- μ_2 -oxido-dodecaoxido- μ_{12} -phosphato-dodecamolybdate(VI)

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Computing details

Data collection: APEX2 (Bruker, 2006); cell refinement: SAINT (Bruker, 2006); data reduction: SAINT (Bruker, 2006); program(s) used to solve structure: SIR2004 (Burla et al., 2005); program(s) used to refine structure: SHELXL2014 (Sheldrick, 2015); molecular graphics: ORTEP-3 for Windows (Farrugia, 2012) and DIAMOND (Putz & Brandenburg, 2014); software used to prepare material for publication: WinGX (Farrugia, 2012).

Octakis(N,N-dimethylformamide- κO)europium(III) tetracosa- μ_2 -oxido-dodecaoxido- μ_{12} -phosphatododecamolybdate(VI)

Crystal data $D_{\rm x} = 2.559 {\rm Mg} {\rm m}^{-3}$ $[Eu(C_3H_7NO)_8][PMo_{12}O_{40}]$ $M_r = 2558.97$ Orthorhombic, $Pna2_1$ a = 26.9108 (10) Å $\theta = 3.0 - 32.3^{\circ}$ b = 18.3506 (6) Å $\mu = 3.24 \text{ mm}^{-1}$ T = 296 Kc = 13.4494 (4) Å V = 6641.7 (4) Å³ Prism, yellow $0.20 \times 0.18 \times 0.17 \text{ mm}$ Z = 4F(000) = 4888Data collection Bruker APEXII CCD diffractometer Radiation source: fine-focus sealed tube $R_{\rm int} = 0.075$ $\theta_{\rm max} = 37.1^\circ, \, \theta_{\rm min} = 2.2^\circ$ φ and ω scans $h = -41 \rightarrow 45$ Absorption correction: multi-scan $k = -31 \rightarrow 31$ (SADABS; Bruker, 2006) $T_{\rm min} = 0.667, T_{\rm max} = 0.747$ $l = -21 \rightarrow 22$ 88354 measured reflections Refinement Refinement on F^2 1 restraint Least-squares matrix: full $R[F^2 > 2\sigma(F^2)] = 0.047$ direct methods $wR(F^2) = 0.095$ S = 0.98map 33285 reflections 863 parameters

Mo *K* α radiation, $\lambda = 0.71073$ Å Cell parameters from 9528 reflections

33285 independent reflections 19583 reflections with $I > 2\sigma(I)$

Primary atom site location: structure-invariant Secondary atom site location: difference Fourier Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.035P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} = 0.001$ $\Delta\rho_{max} = 2.08 \text{ e} \text{ Å}^{-3}$

Special details

 $\Delta \rho_{\min} = -1.80 \text{ e } \text{Å}^{-3}$ Absolute structure: Flack *x* determined using 6924 quotients [(*I*⁺)-(*I*)]/[(*I*⁺)+(*I*)]
Absolute structure parameter: -0.015 (7)

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F² against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F², conventional R-factors R are based on F, with F set to zero for negative F². The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F² are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	
Eu1	0.62226 (2)	0.24667 (2)	0.27301 (3)	0.03117 (7)	
Mo1	0.26680 (2)	0.18597 (3)	0.63650 (5)	0.03332 (13)	
Mo2	0.30169 (2)	0.35458 (3)	0.55755 (5)	0.03366 (13)	
Mo3	0.41829 (3)	0.43690 (3)	0.64996 (5)	0.03387 (13)	
Mo4	0.47559 (2)	0.36195 (3)	0.85122 (5)	0.03004 (12)	
Mo5	0.49335 (2)	0.20234 (3)	0.68682 (5)	0.03324 (13)	
Mo6	0.39245 (3)	0.10613 (3)	0.60967 (5)	0.03505 (14)	
Mo7	0.32991 (3)	0.10420 (3)	0.85410 (5)	0.03838 (15)	
Mo8	0.43839 (3)	0.17833 (4)	0.93808 (5)	0.03661 (14)	
Mo9	0.32843 (3)	0.26018 (4)	0.99282 (5)	0.04260 (17)	
Mo10	0.36044 (2)	0.43617 (3)	0.87490 (5)	0.03517 (14)	
Mo11	0.25072 (2)	0.33286 (4)	0.78760 (5)	0.03541 (14)	
Mo12	0.42166 (3)	0.26738 (4)	0.49994 (5)	0.03432 (13)	
P1	0.37299 (6)	0.26883 (8)	0.74508 (12)	0.0207 (3)	
01	0.6764 (2)	0.1551 (4)	0.2093 (5)	0.0526 (16)	
O2	0.2228 (2)	0.1295 (3)	0.5946 (5)	0.0453 (14)	
03	0.26515 (19)	0.2607 (3)	0.5440 (4)	0.0350 (11)	
O4	0.34964 (18)	0.4169 (3)	0.6033 (4)	0.0320 (10)	
05	0.47195 (19)	0.4273 (3)	0.7347 (4)	0.0325 (11)	
O6	0.5310(2)	0.3873 (3)	0.8969 (4)	0.0420 (13)	
O7	0.49746 (19)	0.2908 (3)	0.7483 (4)	0.0330 (11)	
08	0.45951 (19)	0.1127 (2)	0.6394 (4)	0.0355 (11)	
09	0.32412 (19)	0.1439 (3)	0.5897 (4)	0.0333 (11)	
O10	0.3722 (2)	0.0854 (3)	0.7370 (4)	0.0372 (12)	
011	0.3952 (2)	0.1009 (3)	0.9240 (4)	0.0407 (13)	
012	0.45956 (19)	0.2814 (3)	0.9252 (4)	0.0349 (11)	
013	0.3888 (2)	0.2209 (3)	1.0310 (4)	0.0420 (13)	
014	0.3515 (2)	0.3523 (3)	0.9652 (4)	0.0374 (12)	
015	0.30057 (19)	0.4081 (3)	0.8207 (4)	0.0337 (11)	
016	0.1992 (2)	0.3743 (4)	0.8289 (5)	0.0527 (16)	

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\hat{A}^2)

O17	0.25496 (19)	0.3787 (3)	0.6547 (4)	0.0363 (11)
O18	0.22650 (19)	0.2506 (3)	0.7233 (4)	0.0377 (12)
O19	0.32137 (16)	0.2804 (2)	0.7018 (3)	0.0231 (9)
O20	0.36948 (17)	0.2233 (2)	0.8397 (3)	0.0257 (9)
O21	0.39628 (16)	0.3431 (2)	0.7695 (3)	0.0241 (8)
022	0.40539 (17)	0.2288 (2)	0.6696 (3)	0.0248 (9)
023	0.35674 (19)	0.2989(3)	0.5004 (4)	0.0344(11)
024	0 4401 (3)	0.2850(4)	0.3828(4)	0.0512 (15)
025	0.43539(19)	0.3610(3)	0.5671 (4)	0.0342(11)
026	0 48598 (18)	0.2386(3)	0.5597(4)	0.0340(11)
027	0.4063(2)	0.1693(3)	0.3937(1) 0.4917(4)	0.0349(11)
028	0.3447(2)	0.1095(3)	0.1517(1) 0.9514(5)	0.0319(11) 0.0499(15)
029	0.38277(19)	0.5055(5) 0.4838(2)	0.7624(4)	0.0361(11)
030	0.30277(19) 0.43123(19)	0.4030(2) 0.4244(3)	0.7024(4) 0.9129(4)	0.0344(11)
031	0.3075(2)	0.1211(3) 0.1592(3)	0.9607(4)	0.0511(11) 0.0440(14)
032	0.3073(2) 0.4708(2)	0.1552(3)	0.9007(4) 0.8178(4)	0.0341(11)
033	0.4701(2)	0.1059(5)	1.0216(5)	0.0547(11)
034	0.4791(3) 0.2843(2)	0.1402(4) 0.1380(3)	0.7637(4)	0.0347(10)
035	0.2343(2) 0.4338(2)	0.1389(3)	0.7037(4) 0.5822(5)	0.0307(11)
035	0.4550(2) 0.2790(2)	0.3074(3)	0.5622(5)	0.0466(13)
N1	0.2790(2) 0.6761(3)	0.3974(3)	0.4300 (4)	0.0440(14) 0.0457(17)
N2	0.0701(3)	0.4077(4)	0.0480(0) 0.5746(5)	0.0438(16)
N3	0.3903(3) 0.7647(3)	0.3509(4)	0.3740(5) 0.3484(6)	0.0437(16)
N/	0.7047(3)	0.3014(4) 0.1673(3)	-0.0459(5)	0.0437(10)
N5	0.3971(3) 0.4821(3)	0.1075(3)	0.0459(5)	0.0390(13)
NJ 047	0.4821(3) 0.6329(3)	0.3901(4)	0.2330(0) 0.4103(5)	0.0470(18) 0.0617(19)
N7	0.0329(3)	0.1003(4)	0.4103(5) 0.5252(5)	0.0017(19)
N8	0.0224(3) 0.4832(3)	0.0301(4) 0.1029(4)	0.3232(5) 0.2582(6)	0.0400(13) 0.0474(17)
NO	0.7415(3)	0.1029(4) 0.0838(4)	0.2382 (0)	0.0471(17)
037	0.7413(3) 0.2743(2)	0.0000(4) 0.2842(3)	0.1778(0) 0.8971(4)	0.0471(17) 0.0382(12)
038	0.2743(2) 0.2080(3)	0.2042(3)	1 1010 (4)	0.0532(12) 0.0632(19)
030	0.2989(3)	0.2710(4)	1.1019(4)	0.0052(19)
055	0.5080(3)	0.0201(3) 0.1732(3)	0.8810(5)	0.0339(17) 0.0459(14)
033	0.5525(2)	0.1732(3) 0.2481(3)	0.0850(5)	0.0439(14)
C2	0.0330(2) 0.6452(3)	0.3481(3) 0.3714(4)	0.1039(3) 0.1012(7)	0.0490(14)
U2	0.6142	0.3714 (4)	0.1012 (7)	0.040(2)
118 C3	0.0142 0.7237 (4)	0.3009	0.0743	0.055°
С5 U2	0.7237 (4)	0.4299 (3)	0.0907 (9)	0.001 (3)
H3	0.7105	0.4550	0.1320	0.092
П.5 ЦЛ	0.7412	0.4000	0.1032	0.092
114 C4	0.7429	0.4320 (6)	-0.0517(7)	0.092
U4 H6	0.0038 (3)	0.4329 (0)	-0.0980	0.007 (3)
H7	0.6670	0.4849	-0.0564	0.101*
H5	0.6303	0.4101	-0.0668	0.101*
041	0.6001 (3)	0.3175(A)	0.4158 (5)	0.101
C6	0.6001(3)	0.3173(4) 0.3158(5)	0.7130(3) 0.5037(7)	0.0500(17) 0.051(2)
H10	0.6400	0.2836	0.5057 (7)	0.051(2)
C7	0.0400	0.2030	0.5200	0.002°
\cup	0.0137(4)	0.3322(7)	0.0/47(/)	0.007(3)

H12	0.6340	0.3092	0.6823	0.100*
H13	0.5861	0.3496	0.7195	0.100*
H11	0.6335	0.3945	0.6902	0.100*
C8	0.5581 (4)	0.4096 (6)	0.5489 (8)	0.068 (3)
H15	0.5724	0.4481	0.5099	0.102*
H16	0.5441	0.4297	0.6086	0.102*
H14	0.5324	0.3860	0.5112	0.102*
O42	0.7020 (2)	0.2798 (3)	0.3417 (5)	0.0472 (14)
C10	0.7180 (3)	0.3417 (4)	0.3558 (6)	0.0430 (18)
H18	0.6951	0.3774	0.3731	0.052*
C11	0.8025 (4)	0.3079 (6)	0.3206 (8)	0.062 (3)
H19	0.7881	0.2600	0.3203	0.093*
H20	0.8293	0.3093	0.3678	0.093*
H21	0.8151	0.3190	0.2555	0.093*
C12	0.7807 (4)	0.4351 (6)	0.3660 (9)	0.067 (3)
H22	0.7522	0.4657	0.3761	0.100*
H23	0.7992	0.4523	0.3096	0.100*
H24	0.8014	0.4365	0.4241	0.100*
O43	0.6047 (3)	0.2262 (3)	0.0990 (4)	0.0514 (15)
C14	0.5960 (3)	0.1715 (4)	0.0515 (6)	0.0422 (18)
H26	0.5881	0.1294	0.0866	0.051*
C15	0.5833 (4)	0.1013 (4)	-0.0986 (6)	0.046 (2)
H28	0.5798	0.0620	-0.0519	0.069*
H29	0.6087	0.0894	-0.1461	0.069*
H27	0.5523	0.1088	-0.1325	0.069*
C16	0.6085 (4)	0.2314 (5)	-0.1036(7)	0.055 (2)
H31	0.5807	0.2641	-0.1021	0.082*
H30	0.6153	0.2175	-0.1710	0.082*
H32	0.6372	0.2551	-0.0759	0.082*
O44	0.5498 (2)	0.3180 (4)	0.2319 (5)	0.0528 (16)
C18	0.5248 (3)	0.3674 (4)	0.2680 (8)	0.0468 (19)
H34	0.5377	0.3907	0.3237	0.056*
C19	0.4532 (4)	0.4425 (5)	0.2893 (9)	0.065 (3)
H35	0.4220	0.4211	0.3081	0.098*
H36	0.4472	0.4842	0.2479	0.098*
H37	0.4709	0.4573	0.3478	0.098*
C20	0.4613 (6)	0.3597 (9)	0.1481 (9)	0.114 (6)
H39	0.4498	0.3982	0.1055	0.171*
H40	0.4338	0.3289	0.1656	0.171*
H38	0.4861	0.3316	0.1140	0.171*
C21	0.6486 (4)	0.1287 (5)	0.4775 (6)	0.0453 (19)
H47	0.6816	0.1349	0.4962	0.054*
C22	0.5722 (3)	0.0645 (5)	0.4960 (9)	0.058 (2)
H41	0.5499	0.0791	0.5481	0.087*
H42	0.5687	0.0132	0.4841	0.087*
H43	0.5645	0.0909	0.4364	0.087*
C23	0.6432 (4)	0.0379 (6)	0.6040 (8)	0.064 (3)
H44	0.6776	0.0503	0.6120	0.096*

H46	0.6403	-0.0130	0.5884	0.096*
H45	0.6257	0.0480	0.6646	0.096*
O45	0.5562 (2)	0.1600 (3)	0.2673 (5)	0.0503 (14)
C25	0.5106 (3)	0.1617 (4)	0.2671 (7)	0.0449 (19)
H55	0.4948	0.2066	0.2735	0.054*
C26	0.5044 (5)	0.0336 (5)	0.2361 (11)	0.088 (4)
H49	0.5385	0.0398	0.2165	0.133*
H51	0.4862	0.0111	0.1830	0.133*
H50	0.5029	0.0031	0.2941	0.133*
C27	0.4286 (4)	0.1112 (7)	0.2546 (9)	0.080 (4)
H52	0.4201	0.1618	0.2609	0.120*
H53	0.4139	0.0842	0.3081	0.120*
H54	0.4164	0.0930	0.1923	0.120*
C28	0.7155 (4)	0.1265 (5)	0.2333 (7)	0.050(2)
H56	0.7275	0.1363	0.2968	0.060*
C29	0.7248 (4)	0.0661 (8)	0.0795 (9)	0.084 (4)
H58	0.7093	0.1081	0.0501	0.126*
H59	0.7012	0.0270	0.0829	0.126*
H57	0.7526	0.0514	0.0396	0.126*
C30	0.7896 (5)	0.0544 (8)	0.2084 (9)	0.088 (4)
H61	0.8144	0.0676	0.1603	0.132*
H62	0.7875	0.0023	0.2129	0.132*
H60	0.7985	0.0741	0.2721	0.132*
O52	0.3886 (3)	0.0253 (3)	0.5521 (5)	0.0571 (17)

Atomic displacement parameters $(Å^2)$

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Eu1	0.03342 (16)	0.02974 (14)	0.03036 (15)	-0.00241 (13)	-0.00034 (15)	-0.00084 (15)
Mo1	0.0306 (3)	0.0302 (3)	0.0392 (3)	-0.0058 (2)	-0.0060 (3)	-0.0007 (3)
Mo2	0.0306 (3)	0.0352 (3)	0.0352 (3)	0.0008 (2)	-0.0079 (3)	0.0104 (3)
Mo3	0.0381 (3)	0.0250 (3)	0.0385 (3)	-0.0063 (2)	-0.0028 (3)	0.0059 (3)
Mo4	0.0258 (3)	0.0310 (3)	0.0333 (3)	0.0000 (2)	-0.0034 (2)	-0.0058 (3)
Mo5	0.0281 (3)	0.0369 (3)	0.0347 (3)	0.0098 (2)	-0.0026 (3)	-0.0085 (3)
Mo6	0.0397 (3)	0.0252 (3)	0.0403 (3)	0.0005 (2)	0.0023 (3)	-0.0095 (3)
Mo7	0.0455 (4)	0.0297 (3)	0.0399 (4)	-0.0055 (3)	0.0003 (3)	0.0121 (3)
Mo8	0.0409 (4)	0.0384 (3)	0.0306 (3)	0.0043 (3)	-0.0081 (3)	0.0091 (3)
Mo9	0.0560 (4)	0.0450 (4)	0.0268 (3)	0.0115 (3)	0.0128 (3)	0.0052 (3)
Mo10	0.0317 (3)	0.0291 (3)	0.0447 (4)	0.0035 (2)	0.0021 (3)	-0.0131 (3)
Mo11	0.0238 (3)	0.0406 (3)	0.0419 (4)	0.0031 (2)	0.0032 (3)	-0.0048 (3)
Mo12	0.0406 (3)	0.0386 (3)	0.0238 (3)	0.0044 (3)	0.0064 (3)	0.0030 (3)
P1	0.0222 (7)	0.0199 (6)	0.0198 (7)	0.0017 (5)	-0.0004 (5)	0.0009 (5)
01	0.050 (4)	0.057 (4)	0.051 (4)	0.014 (3)	-0.006 (3)	-0.014 (3)
O2	0.040 (3)	0.040 (3)	0.056 (4)	-0.009 (2)	-0.007 (3)	-0.006 (3)
O3	0.034 (3)	0.036 (3)	0.035 (3)	0.000 (2)	-0.009 (2)	0.001 (2)
O4	0.034 (3)	0.027 (2)	0.035 (3)	0.0005 (19)	-0.003 (2)	0.005 (2)
05	0.032 (3)	0.029 (2)	0.037 (3)	-0.0049 (19)	0.000 (2)	-0.002 (2)
O6	0.031 (3)	0.043 (3)	0.052 (3)	-0.004 (2)	-0.013 (2)	-0.009 (3)

O7	0.030(3)	0.032 (2)	0.037 (3)	0.0019 (19)	0.002 (2)	-0.004(2)
08	0.038 (3)	0.027 (2)	0.042 (3)	0.008 (2)	0.002 (2)	-0.007(2)
O9	0.035 (3)	0.031 (2)	0.033 (3)	0.001 (2)	-0.005(2)	-0.006(2)
O10	0.044 (3)	0.029 (2)	0.039 (3)	0.004 (2)	0.000 (2)	0.001 (2)
O11	0.051 (3)	0.032 (3)	0.040 (3)	0.004 (2)	-0.005 (3)	0.009 (2)
O12	0.037 (3)	0.035 (3)	0.032 (3)	0.004 (2)	0.001 (2)	0.000 (2)
O13	0.054 (3)	0.051 (3)	0.021 (2)	0.007 (3)	-0.003 (2)	0.004 (2)
O14	0.043 (3)	0.040 (3)	0.030 (3)	0.006 (2)	0.002 (2)	-0.003(2)
O15	0.029 (3)	0.032 (2)	0.040 (3)	0.003 (2)	0.002 (2)	-0.003 (2)
O16	0.032 (3)	0.061 (4)	0.065 (4)	0.007 (3)	0.011 (3)	-0.009 (3)
O17	0.031 (3)	0.034 (2)	0.044 (3)	0.007 (2)	-0.002 (2)	0.005 (2)
O18	0.026 (3)	0.040 (3)	0.047 (3)	-0.004 (2)	0.003 (2)	0.000 (2)
O19	0.025 (2)	0.0205 (19)	0.024 (2)	0.0017 (16)	-0.0012 (17)	0.0037 (17)
O20	0.031 (2)	0.023 (2)	0.023 (2)	0.0016 (17)	0.0007 (18)	0.0042 (18)
O21	0.027 (2)	0.0232 (18)	0.022 (2)	0.0028 (16)	0.0001 (19)	-0.0043 (18)
O22	0.030(2)	0.0222 (19)	0.022 (2)	0.0018 (17)	-0.0004 (18)	-0.0021 (17)
O23	0.035 (3)	0.038 (3)	0.031 (3)	0.002 (2)	-0.003 (2)	-0.001 (2)
O24	0.064 (4)	0.061 (4)	0.028 (3)	0.005 (3)	0.016 (3)	0.006 (3)
O25	0.032 (3)	0.035 (2)	0.036 (3)	-0.001 (2)	0.000 (2)	-0.003 (2)
O26	0.029 (2)	0.039 (3)	0.033 (3)	0.006 (2)	0.007 (2)	-0.004(2)
O27	0.042 (3)	0.038 (3)	0.025 (2)	0.006 (2)	-0.002(2)	-0.008(2)
O28	0.050 (4)	0.038 (3)	0.062 (4)	0.006 (3)	0.004 (3)	-0.024(3)
O29	0.038 (3)	0.0220 (19)	0.048 (3)	0.0043 (19)	-0.001 (2)	-0.002(2)
O30	0.032 (3)	0.035 (2)	0.036 (3)	0.005 (2)	-0.003 (2)	-0.013 (2)
O31	0.051 (4)	0.039 (3)	0.042 (3)	-0.002 (2)	0.014 (3)	0.013 (2)
O32	0.042 (3)	0.030 (2)	0.030 (2)	0.011 (2)	-0.002(2)	0.001 (2)
O33	0.060 (4)	0.061 (4)	0.043 (3)	0.013 (3)	-0.018 (3)	0.010 (3)
O34	0.041 (3)	0.029 (2)	0.039 (3)	-0.005 (2)	0.005 (2)	0.006 (2)
O35	0.054 (4)	0.038 (3)	0.055 (4)	-0.012 (3)	-0.002(3)	0.014 (3)
O36	0.047 (3)	0.047 (3)	0.040 (3)	0.001 (3)	-0.016 (3)	0.015 (3)
N1	0.060 (5)	0.033 (3)	0.044 (4)	-0.002(3)	0.008 (4)	0.000 (3)
N2	0.040 (4)	0.056 (4)	0.035 (3)	0.000 (3)	0.004 (3)	-0.007(3)
N3	0.040 (4)	0.041 (3)	0.050 (4)	-0.002(3)	-0.010 (3)	-0.007(3)
N4	0.041 (4)	0.034 (3)	0.043 (4)	0.005 (3)	-0.008(3)	-0.001(3)
N5	0.049 (4)	0.039 (3)	0.054 (4)	0.007 (3)	0.000 (3)	0.012 (3)
O47	0.072 (5)	0.053 (4)	0.060 (4)	-0.010 (3)	-0.008 (4)	0.023 (3)
N7	0.043 (4)	0.037 (3)	0.040 (4)	-0.005(3)	0.004 (3)	-0.002(3)
N8	0.045 (4)	0.047 (4)	0.050 (4)	-0.016 (3)	0.006 (3)	-0.013 (3)
N9	0.052 (4)	0.043 (4)	0.047 (4)	0.005 (3)	0.002 (3)	0.001 (3)
O37	0.040 (3)	0.038 (3)	0.036 (3)	0.003 (2)	0.011 (2)	0.003 (2)
O38	0.075 (5)	0.085 (5)	0.030 (3)	0.017 (4)	0.026 (3)	0.003 (3)
O39	0.072 (4)	0.034 (3)	0.061 (4)	-0.017(3)	-0.003 (3)	0.017 (3)
O55	0.036 (3)	0.054 (3)	0.048 (3)	0.017 (3)	-0.003 (3)	-0.015 (3)
O40	0.059 (4)	0.041 (3)	0.047 (3)	-0.016 (3)	-0.004 (3)	0.011 (3)
C2	0.049 (5)	0.029 (3)	0.060 (6)	-0.006 (3)	-0.002 (4)	-0.003 (4)
C3	0.051 (6)	0.053 (5)	0.079 (7)	-0.004 (4)	0.017 (5)	0.007 (5)
C4	0.105 (9)	0.059 (6)	0.037 (5)	-0.012 (6)	0.002 (5)	0.008 (5)
O41	0.057 (4)	0.077 (4)	0.040 (3)	0.019 (4)	-0.004 (3)	-0.016 (3)
	× /	· /	× /	× /	× /	× /

C6	0.050 (5)	0.058 (5)	0.046 (5)	0.014 (4)	-0.006 (4)	-0.009 (4)
C7	0.062 (7)	0.103 (9)	0.036 (5)	-0.012 (6)	0.012 (5)	-0.015 (5)
C8	0.067 (7)	0.080 (7)	0.056 (6)	0.019 (6)	0.018 (5)	-0.007 (6)
O42	0.041 (3)	0.045 (3)	0.056 (4)	-0.006 (3)	-0.012 (3)	-0.001 (3)
C10	0.040 (4)	0.042 (4)	0.046 (5)	0.000 (3)	-0.003 (4)	-0.005 (4)
C11	0.047 (6)	0.074 (7)	0.066 (6)	0.006 (5)	-0.002 (5)	-0.018 (5)
C12	0.057 (6)	0.065 (6)	0.078 (7)	-0.023 (5)	-0.004 (6)	-0.009 (6)
O43	0.073 (4)	0.047 (3)	0.033 (3)	0.003 (3)	-0.010 (3)	-0.007(3)
C14	0.053 (5)	0.039 (4)	0.034 (4)	0.003 (4)	-0.010 (4)	0.000 (4)
C15	0.061 (6)	0.040 (4)	0.038 (4)	0.011 (4)	-0.013 (4)	-0.003 (3)
C16	0.078 (7)	0.042 (4)	0.045 (5)	0.005 (5)	-0.003 (5)	0.004 (4)
O44	0.052 (4)	0.058 (4)	0.049 (4)	0.019 (3)	-0.002 (3)	0.000 (3)
C18	0.045 (5)	0.042 (4)	0.053 (5)	-0.003 (3)	-0.008 (4)	0.000 (4)
C19	0.058 (6)	0.044 (5)	0.093 (8)	0.007 (4)	0.007 (6)	0.016 (5)
C20	0.128 (13)	0.149 (14)	0.064 (8)	0.055 (11)	-0.058 (9)	-0.021 (9)
C21	0.049 (5)	0.045 (4)	0.042 (5)	-0.010 (4)	0.001 (4)	0.007 (4)
C22	0.050 (5)	0.044 (5)	0.081 (7)	-0.011 (4)	0.005 (5)	-0.006(5)
C23	0.075 (7)	0.065 (6)	0.051 (6)	-0.018 (5)	-0.011 (5)	0.023 (5)
O45	0.041 (3)	0.046 (3)	0.064 (4)	-0.013 (3)	-0.001 (3)	0.008 (3)
C25	0.050 (5)	0.039 (4)	0.046 (5)	-0.014 (3)	0.003 (4)	-0.007 (4)
C26	0.084 (9)	0.041 (5)	0.140 (13)	-0.009 (6)	0.007 (8)	-0.008 (6)
C27	0.052 (6)	0.106 (9)	0.083 (9)	-0.023 (6)	0.014 (6)	-0.027 (7)
C28	0.065 (6)	0.044 (4)	0.042 (5)	0.009 (4)	0.001 (4)	-0.013 (4)
C29	0.070 (8)	0.119 (10)	0.063 (7)	0.003 (7)	0.001 (6)	-0.047 (7)
C30	0.095 (10)	0.100 (9)	0.069 (8)	0.057 (8)	-0.008 (7)	-0.006 (7)
O52	0.069 (4)	0.035 (3)	0.067 (4)	-0.004 (3)	0.007 (4)	-0.025 (3)

Geometric parameters (Å, °)

Eu1—O40	2.369 (5)	N1—C3	1.450 (12)
Eu1—O47	2.378 (6)	N1—C4	1.452 (12)
Eu1—O1	2.383 (6)	N2—C6	1.311 (11)
Eu1—O45	2.387 (5)	N2—C7	1.433 (12)
Eu1—O41	2.395 (6)	N2—C8	1.455 (12)
Eu1—044	2.413 (6)	N3—C10	1.312 (11)
Eu1—O42	2.415 (6)	N3—C12	1.438 (11)
Eu1—O43	2.416 (6)	N3—C11	1.463 (12)
Mo1—O2	1.672 (5)	N4—C14	1.313 (10)
Mo1-09	1.836 (5)	N4—C16	1.441 (11)
Mo1-03	1.852 (5)	N4—C15	1.452 (10)
Mo1-034	1.973 (5)	N5—C18	1.297 (11)
Mo1-018	1.986 (5)	N5—C20	1.417 (14)
Mo1-019	2.436 (4)	N5—C19	1.433 (12)
Mo2—O36	1.690 (5)	O47—C21	1.214 (11)
Mo2—O4	1.831 (5)	N7—C21	1.307 (10)
Mo2—O17	1.866 (5)	N7—C23	1.427 (12)
Mo2—O23	1.957 (5)	N7—C22	1.435 (11)
Mo2—O3	1.993 (5)	N8—C25	1.313 (10)

Mo2—O19	2.429 (4)	N8—C26	1.427 (13)
Mo3—O35	1.672 (5)	N8—C27	1.477 (13)
Mo3—O25	1.842 (5)	N9—C28	1.290 (11)
Mo3—O5	1.848 (5)	N9—C29	1.435 (13)
Mo3	1.985 (5)	N9—C30	1.461 (14)
Mo3-029	1 985 (5)	040	1 245 (11)
Mo3-021	2428(4)	C2—H8	0.9300
Mo3 021 Mo4—06	1 677 (5)	C3_H2	0.9600
Mo4012	1.077(5) 1.834(5)	C3_H3	0.9600
$M_{04} = 0.12$	1.851 (5)		0.9600
Mo4_05	1.031(3)		0.9000
Mo4_07	1.970(3)		0.9600
M04	1.992(5)	C4—H7	0.9600
Mo4—O21	2.426 (4)	C4—H5	0.9600
M05	1.6/5 (5)	041	1.245 (11)
Mo5—O7	1.825 (5)	C6—H10	0.9300
Mo5—O26	1.845 (5)	С7—Н12	0.9600
Mo5—O32	1.979 (5)	С7—Н13	0.9600
Mo5—O8	1.985 (5)	C7—H11	0.9600
Mo5—O22	2.428 (5)	C8—H15	0.9600
Mo6—O52	1.676 (5)	C8—H16	0.9600
Mo6—O10	1.837 (5)	C8—H14	0.9600
Mo6—O8	1.852 (5)	O42—C10	1.228 (9)
Mo6—O9	1.983 (5)	C10—H18	0.9300
Mo6—O27	2.000 (5)	C11—H19	0.9600
Mo6—O22	2.416 (4)	С11—Н20	0.9600
Mo7—O39	1.686 (5)	С11—Н21	0.9600
Mo7-034	1.842 (6)	С12—Н22	0.9600
Mo7-031	1 854 (6)	C12—H23	0.9600
Mo7-010	1 973 (6)	C12—H24	0.9600
Mo7-011	1 994 (6)	043 - 014	1.214(10)
Mo7 020	2.438(4)	C_{14} H26	0.0300
$M_0^{\circ} = 020$	2.438 (4)	C15 H28	0.9500
Mo8_011	1.075(0) 1.844(6)	C15 H20	0.9000
Mo8_022	1.044(0) 1.952(5)	C15_U27	0.9000
M08-012	1.033(3) 1.082(5)	C15-H27	0.9600
M08-012	1.985 (5)		0.9600
M08-013	1.987 (6)	C16—H30	0.9600
M08—020	2.423 (4)	C10—H32	0.9600
M09—038	1.680 (6)	044	1.229 (10)
Mo9—O14	1.838 (6)	С18—Н34	0.9300
Mo9—O13	1.851 (6)	C19—H35	0.9600
Mo9—O31	1.984 (6)	С19—Н36	0.9600
Mo9—O37	1.993 (6)	С19—Н37	0.9600
Mo9—O20	2.433 (5)	С20—Н39	0.9600
Mo10	1.690 (5)	С20—Н40	0.9600
Mo10-015	1.842 (5)	С20—Н38	0.9600
Mo10	1.847 (6)	C21—H47	0.9300
Mo10-014	1.976 (5)	C22—H41	0.9600
Mo10-O30	1.984 (5)	C22—H42	0.9600

Mo10-021	2.420 (4)	С22—Н43	0.9600
Mo11—O16	1.676 (6)	С23—Н44	0.9600
Mo11—O37	1.836 (5)	С23—Н46	0.9600
Mo11	1.857 (5)	С23—Н45	0.9600
Mo11-015	1.976 (5)	O45—C25	1.227 (10)
Mo11—O17	1.979 (5)	С25—Н55	0.9300
Mo11—O19	2.423 (4)	C26—H49	0.9600
Mo12—O24	1.684 (5)	C26—H51	0.9600
Mo12—O23	1.840 (5)	C26—H50	0.9600
Mo12-027	1.850 (5)	С27—Н52	0.9600
Mo12-025	1 976 (5)	C27—H53	0.9600
Mo12-026	1.970(5)	C27—H54	0.9600
Mo12 020 Mo12-022	2428(4)	C_{28} —H56	0.9300
P1019	1.521(5)	C29_H58	0.9500
P1 020	1.521(5) 1.526(5)	C20 H50	0.9600
$P_1 = O_2^2$	1.520(5) 1.527(5)	C29 H57	0.9000
P1 021	1.527(5)	C_{29} H(1	0.9000
P1-021	1.530 (4)	C30—H01	0.9000
01-028	1.219 (11)	C30—H62	0.9600
NI—C2	1.297 (11)	C30—H60	0.9600
040—Fu1—047	145 6 (2)	026—Mo12—022	72 45 (18)
040—Fu1—01	98.6 (2)	019 - P1 - 020	109.8(3)
$047 - F_{11} - 01$	76.7(2)	019 - P1 - 022	109.6(3)
040 = Eu1 = 045	141.7(2)	$O_{10} = 11 = O_{22}$	109.5(3) 100.1(2)
040 - Eu1 - 045	141.7(2)	020 - 11 - 022	109.1(2) 100.2(2)
04/-Eu1-045	72.7(2)	019 - 1 - 021	109.5(2)
01 - Eu1 - 043	88.3(2)	020 - P1 - 021	109.3(3)
040 - Eu1 - 041	93.6 (2) 75.1 (2)	022 - P1 - 021	109.6(3)
04/—Eu1—041	/5.1 (3)		137.4 (6)
OI—EuI—O4I	145.4 (2)	Mo1-03-Mo2	124.5 (3)
045—Eu1—041	101.7 (2)	Mo2—O4—Mo3	151.4 (3)
O40—Eu1—O44	76.2 (2)	Mo3—O5—Mo4	125.8 (3)
O47—Eu1—O44	127.6 (2)	Mo5—O7—Mo4	151.7 (3)
O1—Eu1—O44	142.6 (2)	Mo6—O8—Mo5	124.8 (2)
O45—Eu1—O44	75.6 (2)	Mo1	151.6 (3)
O41—Eu1—O44	71.8 (2)	Mo6—O10—Mo7	151.5 (3)
O40—Eu1—O42	70.2 (2)	Mo8—O11—Mo7	125.5 (3)
O47—Eu1—O42	75.6 (2)	Mo4—O12—Mo8	152.1 (3)
O1—Eu1—O42	76.8 (2)	Mo9-013-Mo8	124.6 (3)
O45—Eu1—O42	147.4 (2)	Mo9-014-Mo10	151.8 (3)
O41—Eu1—O42	77.2 (2)	Mo10-015-Mo11	151.4 (3)
O44—Eu1—O42	132.0 (2)	Mo2-017-Mo11	124.8 (3)
O40—Eu1—O43	73.5 (2)	Mo11-O18-Mo1	124.6 (3)
O47—Eu1—O43	132.8 (2)	P1-019-Mo11	126.2 (2)
O1—Eu1—O43	70.2 (2)	P1	125.7 (2)
O45—Eu1—O43	73.7 (2)	Mo11—O19—Mo2	89.23 (14)
O41—Eu1—O43	144.3 (2)	P1-019-Mo1	126.1 (2)
O44—Eu1—O43	72.8 (2)	Mo11—019—Mo1	88.93 (15)
O42—Eu1—O43	125.7 (2)	Mo2-019-Mo1	88.80 (15)
			20.00 (12)

O2—Mo1—O9	102.7 (3)	P1	126.5 (3)
O2—Mo1—O3	102.4 (3)	P1	125.6 (2)
O9—Mo1—O3	95.8 (2)	Mo8-020-Mo9	88.87 (15)
O2—Mo1—O34	101.0 (3)	P1	125.7 (3)
O9—Mo1—O34	85.0 (2)	Mo8—O20—Mo7	89.20 (14)
O3—Mo1—O34	155.8 (2)	Mo9—O20—Mo7	89.09 (15)
O2—Mo1—O18	100.5 (3)	P1	126.1 (3)
O9—Mo1—O18	155.7 (2)	P1	125.6 (2)
O3—Mo1—O18	86.5 (2)	Mo10-021-Mo4	89.12 (14)
O34—Mo1—O18	83.2 (2)	P1-021-Mo3	126.0 (3)
$02 - M_01 - 019$	171.9 (2)	Mo10-021-Mo3	89.13 (13)
09—Mo1—019	85.20 (18)	Mo4—O21—Mo3	89.11 (14)
03 - Mo1 - 019	74.36 (18)	P1-022-Mo6	126.0(3)
034—Mo1—O19	81.65 (18)	P1-022-Mo5	126.1(3)
018 - Mo1 - 019	72.08 (18)	Mo6-022-Mo5	89 19 (14)
$0.36 - M_0 - 0.4$	103.7(2)	P1 = O22 = Mo12	1260(2)
$0.36 - M_0^2 - 0.17$	102.7(2)	Mo6-022-Mo12	89 10 (14)
$04 - M_0^2 - 017$	953(2)	Mo5-022-Mo12	88 42 (15)
036 - Mo2 - 023	101.5(3)	$M_{012} = 0.022 - M_{012}$	1521(3)
0.000 MO2 0.025	85 7 (2)	Mo12 025 Mo2	152.1(3) 152.5(3)
017 - Mo2 - 023	155.4(2)	Mo5-026-Mo12	132.5(3) 124 5(3)
$0.36 - M_0 2 - 0.3$	98.6(3)	Mo12Mo6	124.9(3) 1239(3)
$04 - M_0 2 - 03$	156.8 (2)	Mo12-027-M00 Mo10-029-Mo3	125.9(3) 125.1(2)
017 - Mo2 - 03	86.4 (2)	Mo4Mo10	123.1(2) 124.8(3)
017 - M02 - 03 023 - M02 - 03	83.5 (2)	Mo7-031-Mo9	124.8(3) 125.8(3)
025 - M02 - 019	1701(2)	Mo8-032-Mo5	123.8(3) 151.7(3)
0.000 - 0.00	170.1(2) 85.87(18)	Mo7 O34 Mo1	151.7(3) 152.0(3)
017 - M02 - 019	73 79 (19)	$C_2 N_1 C_3$	132.0(3) 119.4(8)
$O_{17} = MO_{2} = O_{17}$	81 74 (18)	$C_2 = N_1 = C_3$	117.4(0) 121.4(0)
025 - M02 - 019	72 30 (18)	C_2 N1 C_4	121.4(9) 119.2(8)
03 - Mo2 = 012	102.30(10)	C6-N2-C7	119.2(0) 121.7(8)
035 - Mo3 - 05	102.5(3) 102.5(3)	C6 - N2 - C8	121.7(0) 118.4(8)
025—Mo3—05	96.1.(2)	C_{1}^{-1} C_{2}^{-1} C_{3}^{-1} $C_{$	110.4 (0)
0.025 Mo 3 - 0.04	1020(3)	$C_{10} = N_{3} = C_{12}$	119.9(8) 122.2(8)
025 M 03 04	84 4 (2)	C10 - N3 - C12	122.2(0) 120.0(7)
$05-M_03-04$	154.8(2)	C12 - N3 - C11	120.0(7) 117.8(8)
$035 - M_03 - 029$	100.9(3)	C12 = N3 = C11 C14 = N4 = C16	117.8(0) 119.6(7)
035 - M03 - 029	155.3(2)	C14 - N4 - C10	119.0(7) 122.0(7)
023 - M03 - 029	133.3(2) 87.0(2)	C14 - N4 - C15	122.0(7) 118 2(7)
$04 - M_0 = 029$	87.0(2)	C18 - N5 - C20	110.2(7) 120.2(9)
$O_{1}^{3} = MO_{2}^{3} = O_{2}^{3}$	171.5(2)	$C_{10} = N_5 = C_{20}$	120.2(9) 121.8(9)
035 - M03 - 021	85 71 (10)	$C_{10} = N_{5} = C_{19}$	121.0(9) 117.9(10)
$025 - M_03 - 021$	73 44 (10)	$C_{20} - C_{10} - C_{10}$	1665(7)
$04 - M_0 3 - 021$	81 45 (17)	$C_{21} - O_{7} - E_{01}$	100.3(7) 1216(8)
$0.29 - M_0 3 - 0.21$	71 66 (17)	C_{21} N_{7} C_{23}	121.0(0) 120.6(8)
$02^{-1}M03 = 021$ 06 $- M04 = 012$	1035(3)	C_{21} C_{22} C_{23} N_{7} C_{22}	120.0(0) 1177(8)
$06 - M_0 4 - 030$	103.3(3) 103.7(2)	C_{25} N8 C_{25}	117.7(0) 1217(0)
$012 - M_0 4 - 030$	960(2)	C_{25} N8 C_{27}	121.7(9) 118 5 (0)
012 1007 000	JU.U (4)	-10 - 021	110.3 (2)

O6—Mo4—O5	99.6 (2)	C26—N8—C27	118.9 (9)
O12—Mo4—O5	155.3 (2)	C28—N9—C29	120.0 (9)
O30—Mo4—O5	87.0 (2)	C28—N9—C30	122.9 (9)
O6—Mo4—O7	100.0 (2)	C29—N9—C30	117.0 (9)
O12—Mo4—O7	85.3 (2)	Mo11-O37-Mo9	151.5 (3)
O30—Mo4—O7	155.2 (2)	C2—O40—Eu1	130.2 (5)
O5—Mo4—O7	82.0 (2)	O40—C2—N1	123.4 (9)
O6—Mo4—O21	170.8 (2)	O40—C2—H8	118.3
$012 - M_04 - 021$	85.6 (2)	N1—C2—H8	118.3
$0.30 - M_0 4 - 0.21$	73 97 (19)	N1-C3-H2	109.5
$05-M_04-021$	71 54 (18)	N1_C3_H3	109.5
0.5 Mo4 0.21 0.7 Mo4 0.21	81 47 (18)	H2_C3_H3	109.5
$0.55 M_{0}5 0.7$	103 A (3)	$\frac{112}{C^2} = \frac{113}{H^4}$	109.5
055 Mo5 026	103.4(3) 102.0(2)	$H^2 = C^2 = H^4$	109.5
033 - M05 - 020	102.0(3)	$H_2 = C_3 = H_4$	109.5
0/M05020	90.1 (2)	H3—C3—H4	109.5
055—M05—032	101.0 (3)	NI—C4—H6	109.5
07—Mo5—032	85.1 (2)	NI-C4-H7	109.5
O26—Mo5—O32	156.0 (2)	H6—C4—H7	109.5
O55—Mo5—O8	99.6 (3)	N1—C4—H5	109.5
O7—Mo5—O8	155.5 (2)	H6—C4—H5	109.5
O26—Mo5—O8	87.2 (2)	H7—C4—H5	109.5
O32—Mo5—O8	82.3 (2)	C6—O41—Eu1	132.0 (6)
O55—Mo5—O22	170.7 (2)	O41—C6—N2	123.9 (9)
O7—Mo5—O22	85.67 (19)	O41—C6—H10	118.1
O26—Mo5—O22	74.59 (18)	N2—C6—H10	118.1
O32—Mo5—O22	81.61 (18)	N2—C7—H12	109.5
O8—Mo5—O22	71.80 (17)	N2—C7—H13	109.5
O52—Mo6—O10	103.3 (3)	H12—C7—H13	109.5
O52—Mo6—O8	102.5 (3)	N2—C7—H11	109.5
Q10—Mo6—Q8	95.8 (2)	H12—C7—H11	109.5
052—Mo6—09	100.9(3)	H13—C7—H11	109.5
$010 - M_06 - 09$	856(2)	N2-C8-H15	109.5
$08 - M_0 6 - 09$	155.5(2)	N2-C8-H16	109.5
052 Mo6 027	90.1(3)	$H_{12} \subset G H_{16}$	109.5
0.000 - 0.000 - 0.00000 - 0.000000 - 0.00000 - 0.00000 - 0.00000 - 0.0000 - 0.0000	156.2(2)	$N_2 C_8 H_1 4$	109.5
0^{8} Mag 0^{27}	150.2(2)	112 - 00 - 1114	109.5
$00 - M_0 = 027$	87.2(2)	113 - 0.0 - 0.014	109.5
09 - M00 - 027	62.1(2)	10-0.000	109.5
0.000 - 0.000 - 0.00000 - 0.000000 - 0.00000 - 0.00000 - 0.00000 - 0.0000 - 0.0000	1/0.8(3)	C10-042-E01	12/.1(5)
010—M06—022	85.69 (19)	042 - 010 - N3	125.4 (8)
08—Mo6—022	74.14 (18)	042—C10—H18	117.3
O9—Mo6—O22	81.56 (18)	N3—C10—H18	117.3
O27—Mo6—O22	72.40 (17)	N3—C11—H19	109.5
O39—Mo7—O34	103.4 (3)	N3—C11—H20	109.5
O39—Mo7—O31	102.8 (3)	H19—C11—H20	109.5
O34—Mo7—O31	96.1 (3)	N3—C11—H21	109.5
O39—Mo7—O10	102.0 (3)	H19—C11—H21	109.5
O34—Mo7—O10	85.3 (2)	H20—C11—H21	109.5
O31—Mo7—O10	154.1 (2)	N3—C12—H22	109.5

O39—Mo7—O11	99.8 (3)	N3—C12—H23	109.5
O34—Mo7—O11	155.4 (2)	H22—C12—H23	109.5
O31—Mo7—O11	86.4 (3)	N3—C12—H24	109.5
O10—Mo7—O11	82.1 (2)	H22—C12—H24	109.5
O39—Mo7—O20	170.4 (3)	H23—C12—H24	109.5
O34—Mo7—O20	85.93 (18)	C14—O43—Eu1	132.6 (5)
O31—Mo7—O20	73.5 (2)	O43—C14—N4	124.7 (8)
O10—Mo7—O20	80.88 (19)	O43—C14—H26	117.6
O11—Mo7—O20	71.30 (18)	N4—C14—H26	117.6
Q33—Mo8—Q11	102.1 (3)	N4—C15—H28	109.5
$033 - M_0 8 - 032$	103.5 (3)	N4—C15—H29	109.5
$011 - M_0 8 - 032$	96.5 (2)	H28—C15—H29	109.5
$0.33 - M_0 8 - 0.12$	1019(3)	N4—C15—H27	109.5
$011 - M_0 8 - 012$	155.0 (2)	H28—C15—H27	109.5
$032 - M_0 8 - 012$	84 6 (2)	H29—C15—H27	109.5
$033 - M_0 8 - 013$	989(3)	N4—C16—H31	109.5
$011 - M_0 8 - 013$	86 8 (3)	N4—C16—H30	109.5
$032 - M_0 8 - 013$	1560(2)	H_{31} $-C_{16}$ H_{30}	109.5
$012 - M_0 8 - 013$	827(2)	N4-C16-H32	109.5
$0.33 - M_0 8 - 0.20$	170.4(3)	H_{31} (16 H_{32}	109.5
$011 - M_0 8 - 020$	740(2)	H30-C16-H32	109.5
$032 - M_0 8 - 020$	85 74 (19)	C18 - O44 - Fu1	138.8 (6)
0.00000000000000000000000000000000000	81 23 (18)	044-018-044	136.0(0) 126.1(0)
012 - M00 - 020	72 31 (10)	044 C18 H34	117.0
$O_{13}^{-1008} O_{20}^{-020}$	103.1(3)	N5 C18 H34	117.0
$038 M_0 013$	103.1(3) 102.6(3)	N5 C10 H35	100.5
038 - M09 - 013	102.0(3)	N5 C10 H36	109.5
014 - 1009 - 013 028 - 1009 - 021	90.7(3)	$H_{25} = C_{10} = H_{26}$	109.5
0.14 Ma0 0.021	39.0 (3) 155.6 (2)	N5 C10 H27	109.5
014 - 1009 - 031 012 - Ma0 - 031	155.0(2)	$N_{3} = C_{19} = H_{37}$	109.5
013 - 1009 - 031	30.9(3)	$H_{33} - C_{19} - H_{37}$	109.5
0.14 Mag 0.27	101.1(3)	N5 C20 U20	109.5
014 - M09 - 037	83.0(2)	N5 C20 1140	109.5
013 - 1009 - 037	133.1(2)	$N_{3} = C_{20} = H_{40}$	109.5
031 - 1009 - 037	81.9(2)	$H_{39} = C_{20} = H_{40}$	109.5
038 - M09 - 020	1/0.0(3)	$N_{3} = C_{20} = H_{38}$	109.5
014 - M09 - 020	80.07 (19) 74.10 (10)	H39 - C20 - H38	109.5
013 - M09 - 020	74.19 (19)	H40 - C20 - H38	109.5
031 - M09 - 020	/1.00 (19)	047 - 021 - N7	124.0 (9)
037 - M09 - 020	81.18 (18)	04/-021-H4/	11/./
028—M010—015	103.4 (3)	N = C21 = H47	11/./
028—M010—029	103.0(3)	N = C22 = H41	109.5
015—M010—029	95.3 (2)	$N = C_{22} = H_{42}$	109.5
028 - M010 - 014	100.5(3)	H41 - C22 - H42	109.5
013 - M010 - 014	03.4 (<i>2</i>)	IN / - U22 - H43	109.5
029 - M010 - 014	155.7(2)	H41 - C22 - H43	109.5
028 - M010 - 030	99.0 (3) 155 7 (2)	H42 - C22 - H43	109.5
015—M010—030	155.7 (2)	N/	109.5
O29—Mo10—O30	87.1 (2)	N/C23H46	109.5

Q14—Mo10—Q30	82.7 (2)	H44—C23—H46	109.5
$0.28 - M_0 10 - 0.21$	171.0(2)	N7—C23—H45	109.5
$015 - M_0 10 - 021$	85 38 (19)	H44—C23—H45	109.5
029 - Mo10 - 021	74 00 (18)	$H46-C^{23}-H45$	109.5
014 Mo10 021	81 87 (18)	$C_{25} = 0.45 = E_{11}$	136.7 (6)
O_{14}^{-1} Mo10 O_{21}^{-1}	72.04(17)	045 025 N8	130.7(0) 122.8(8)
0.16 Mo11 027	12.04(17)	045 - 025 - 1155	122.8 (8)
010 - M011 - 037	104.0(3)	045-C25-H55	110.0
016-M011-018	103.5(3)	N8-C25-H35	118.0
03/M011018	95.7 (2)	N8-C26-H49	109.5
016—Mo11—015	99.8 (3)	N8—C26—H51	109.5
O37—Mo11—O15	85.6 (2)	H49—C26—H51	109.5
O18—Mo11—O15	155.6 (2)	N8—C26—H50	109.5
O16—Mo11—O17	98.8 (3)	H49—C26—H50	109.5
O37—Mo11—O17	155.7 (2)	H51—C26—H50	109.5
O18—Mo11—O17	86.8 (2)	N8—C27—H52	109.5
O15—Mo11—O17	82.4 (2)	N8—C27—H53	109.5
O16—Mo11—O19	170.7 (3)	Н52—С27—Н53	109.5
O37—Mo11—O19	85.24 (19)	N8—C27—H54	109.5
O18—Mo11—O19	74.38 (19)	Н52—С27—Н54	109.5
O15—Mo11—O19	81.50 (18)	Н53—С27—Н54	109.5
O17—Mo11—O19	72.16 (18)	O1—C28—N9	125.3 (9)
O24—Mo12—O23	102.9 (3)	O1—C28—H56	117.4
O24—Mo12—O27	101.4 (3)	N9—C28—H56	117.4
O23—Mo12—O27	95.4 (2)	N9—C29—H58	109.5
0.24 Mo12 0.27	101.9(3)	N9-C29-H59	109.5
023 - Mo12 - 025	84 4 (2)	H58-C29-H59	109.5
025 Mol2 025	1562(2)	N9_C29_H57	109.5
O_2^{-1} Mo12 O_2^{-1}	100.2(2)	$H_{58} = C_{29} = H_{57}$	109.5
O_{24}^{23} Mo12 O_{20}^{26}	100.0(3) 155.8(2)	H50 C20 H57	109.5
025 - M012 - 020	133.8(2)	N0 C20 H61	109.5
027 - M012 - 020	87.7(2)	N9 - C30 - HC2	109.5
023 - M012 - 020	85.2 (2)	N9-C30-H62	109.5
024—Mo12—022	1/1.3 (3)	H61—C30—H62	109.5
023—Mo12—022	85.27 (19)	N9—C30—H60	109.5
027—Mo12—022	74.48 (18)	H61—C30—H60	109.5
O25—Mo12—O22	81.80 (18)	H62—C30—H60	109.5
02 Mal 02 Mal	171.0.(2)	022 B1 021 Ma10	175 = (2)
02 - M01 - 03 - M02	1/1.0(3)	022 - P1 - 021 - M010	175.3(3)
09—M01—03—M02	-84.6(3)	019—P1—021—M04	1/5.3(3)
034—Mo1—03—Mo2	6.0 (8)	020—P1—021—M04	55.0 (4)
018—Mo1—03—Mo2	71.1 (3)	022—P1—021—Mo4	-64.6 (4)
O19—Mo1—O3—Mo2	-1.3 (3)	O19—P1—O21—Mo3	-64.9 (3)
O36—Mo2—O4—Mo3	-129.9 (6)	O20—P1—O21—Mo3	174.7 (3)
O17—Mo2—O4—Mo3	126.1 (6)	O22—P1—O21—Mo3	55.1 (3)
O23—Mo2—O4—Mo3	-29.1 (6)	O19—P1—O22—Mo6	-64.7 (3)
O3—Mo2—O4—Mo3	33.1 (10)	O20—P1—O22—Mo6	55.5 (4)
O19—Mo2—O4—Mo3	52.9 (6)	O21—P1—O22—Mo6	175.3 (3)
O35—Mo3—O5—Mo4	169.8 (3)	O19—P1—O22—Mo5	174.7 (3)
O25—Mo3—O5—Mo4	-86.2 (3)	O20—P1—O22—Mo5	-65.1 (3)

O4—Mo3—O5—Mo4	3.4 (7)	O21—P1—O22—Mo5	54.8 (4)
O29—Mo3—O5—Mo4	69.3 (3)	O19—P1—O22—Mo12	55.5 (4)
O21—Mo3—O5—Mo4	-2.5 (3)	O20-P1-O22-Mo12	175.7 (3)
O55—Mo5—O7—Mo4	-129.4 (6)	O21—P1—O22—Mo12	-64.4 (4)
O26—Mo5—O7—Mo4	126.7 (6)	O24—Mo12—O23—Mo2	-130.2 (6)
O32—Mo5—O7—Mo4	-29.2 (6)	O27—Mo12—O23—Mo2	126.8 (6)
O8—Mo5—O7—Mo4	30.1 (10)	O25—Mo12—O23—Mo2	-29.3 (6)
O22—Mo5—O7—Mo4	52.8 (6)	O26—Mo12—O23—Mo2	30.3 (10)
O52—Mo6—O8—Mo5	168.5 (4)	O22—Mo12—O23—Mo2	52.9 (6)
O10—Mo6—O8—Mo5	-86.5 (3)	O35—Mo3—O25—Mo12	-132.4 (7)
O9—Mo6—O8—Mo5	5.7 (8)	O5-Mo3-O25-Mo12	123.3 (7)
O27—Mo6—O8—Mo5	69.8 (3)	O4—Mo3—O25—Mo12	-31.3 (6)
O22—Mo6—O8—Mo5	-2.7 (3)	O29—Mo3—O25—Mo12	27.3 (10)
O2—Mo1—O9—Mo6	-129.2 (6)	O21—Mo3—O25—Mo12	50.5 (6)
O3—Mo1—O9—Mo6	126.6 (6)	O55—Mo5—O26—Mo12	169.1 (3)
O34—Mo1—O9—Mo6	-29.1 (6)	O7—Mo5—O26—Mo12	-85.8 (3)
O18—Mo1—O9—Mo6	32.2 (10)	O32—Mo5—O26—Mo12	5.8 (7)
O19—Mo1—O9—Mo6	52.9 (6)	O8—Mo5—O26—Mo12	69.8 (3)
O52—Mo6—O10—Mo7	-128.1 (7)	O22—Mo5—O26—Mo12	-2.0(3)
O8—Mo6—O10—Mo7	127.5 (7)	O24—Mo12—O27—Mo6	169.1 (3)
O9—Mo6—O10—Mo7	-27.9 (7)	O23—Mo12—O27—Mo6	-86.6(3)
O27—Mo6—O10—Mo7	31.2 (11)	O25—Mo12—O27—Mo6	1.8 (8)
O22—Mo6—O10—Mo7	54.0 (7)	O26—Mo12—O27—Mo6	69.4 (3)
O33—Mo8—O11—Mo7	168.8 (4)	O22—Mo12—O27—Mo6	-3.0(3)
O32—Mo8—O11—Mo7	-85.8 (4)	O28—Mo10—O29—Mo3	168.8 (3)
O12—Mo8—O11—Mo7	5.2 (8)	O15—Mo10—O29—Mo3	-86.1(3)
O13—Mo8—O11—Mo7	70.3 (4)	O14—Mo10—O29—Mo3	4.4 (8)
O20-Mo8-O11-Mo7	-2.2 (3)	O30—Mo10—O29—Mo3	69.6 (3)
O6—Mo4—O12—Mo8	-128.2 (6)	O21—Mo10—O29—Mo3	-2.5(3)
O30—Mo4—O12—Mo8	126.1 (6)	O6—Mo4—O30—Mo10	168.6 (3)
O5—Mo4—O12—Mo8	30.3 (10)	O12-Mo4-O30-Mo10	-85.9 (4)
O7—Mo4—O12—Mo8	-29.0 (6)	O5—Mo4—O30—Mo10	69.4 (3)
O21—Mo4—O12—Mo8	52.8 (6)	O7—Mo4—O30—Mo10	5.8 (8)
O38—Mo9—O13—Mo8	169.0 (4)	O21—Mo4—O30—Mo10	-2.2(3)
O14—Mo9—O13—Mo8	-85.8 (4)	O39—Mo7—O31—Mo9	168.4 (4)
O31—Mo9—O13—Mo8	69.9 (4)	O34—Mo7—O31—Mo9	-86.3 (4)
O37—Mo9—O13—Mo8	6.8 (9)	O10—Mo7—O31—Mo9	5.5 (8)
O20—Mo9—O13—Mo8	-1.9 (3)	O11—Mo7—O31—Mo9	69.2 (4)
O38—Mo9—O14—Mo10	-131.0 (7)	O20—Mo7—O31—Mo9	-2.4(3)
O13—Mo9—O14—Mo10	124.3 (7)	O33—Mo8—O32—Mo5	-132.1 (7)
O31—Mo9—O14—Mo10	27.0 (11)	O11—Mo8—O32—Mo5	123.8 (7)
O37—Mo9—O14—Mo10	-30.7 (6)	O12—Mo8—O32—Mo5	-31.1 (7)
O20—Mo9—O14—Mo10	50.8 (6)	O13—Mo8—O32—Mo5	27.1 (11)
O28—Mo10—O15—Mo11	-128.4 (6)	O20—Mo8—O32—Mo5	50.5 (6)
O29—Mo10—O15—Mo11	126.9 (6)	O39—Mo7—O34—Mo1	-131.2 (6)
O14—Mo10—O15—Mo11	-28.7 (6)	O31—Mo7—O34—Mo1	124.0 (6)
O30-Mo10-O15-Mo11	32.1 (10)	O10—Mo7—O34—Mo1	-30.0 (6)
O21—Mo10—O15—Mo11	53.5 (6)	O11—Mo7—O34—Mo1	29.3 (10)

O36—Mo2—O17—Mo11	168.2 (3)	O20-Mo7-O34-Mo1	51.1 (6)
O4—Mo2—O17—Mo11	-86.5 (3)	O16—Mo11—O37—Mo9	-127.0 (6)
O23—Mo2—O17—Mo11	4.5 (7)	O18—Mo11—O37—Mo9	127.6 (6)
O3—Mo2—O17—Mo11	70.2 (3)	O15—Mo11—O37—Mo9	-28.0 (6)
O19—Mo2—O17—Mo11	-2.4 (3)	O17—Mo11—O37—Mo9	32.5 (10)
O16-Mo11-O18-Mo1	168.6 (4)	O19—Mo11—O37—Mo9	53.8 (6)
O37—Mo11—O18—Mo1	-85.5 (4)	Eu1—O40—C2—N1	-154.9 (6)
O15-Mo11-O18-Mo1	6.6 (8)	C3—N1—C2—O40	-3.1 (13)
O17—Mo11—O18—Mo1	70.3 (3)	C4—N1—C2—O40	177.2 (8)
O19-Mo11-O18-Mo1	-2.1 (3)	Eu1—O41—C6—N2	-176.3 (7)
O20-P1-O19-Mo11	55.7 (3)	C7—N2—C6—O41	179.1 (10)
O22-P1-O19-Mo11	175.5 (3)	C8—N2—C6—O41	-2.5 (15)
O21—P1—O19—Mo11	-64.4 (3)	Eu1—O42—C10—N3	-146.6 (7)
O20—P1—O19—Mo2	175.9 (3)	C12—N3—C10—O42	180.0 (9)
O22—P1—O19—Mo2	-64.3 (3)	C11—N3—C10—O42	0.5 (14)
O21—P1—O19—Mo2	55.8 (4)	Eu1—O43—C14—N4	-165.4 (6)
O20-P1-O19-Mo1	-64.6 (3)	C16—N4—C14—O43	-0.9 (14)
O22—P1—O19—Mo1	55.2 (3)	C15—N4—C14—O43	-176.1 (8)
O21—P1—O19—Mo1	175.3 (3)	Eu1—O44—C18—N5	-169.7 (7)
O19—P1—O20—Mo8	175.4 (3)	C20—N5—C18—O44	-3.2 (16)
O22—P1—O20—Mo8	55.4 (4)	C19—N5—C18—O44	172.7 (9)
O21—P1—O20—Mo8	-64.6 (4)	Eu1—O47—C21—N7	-167 (2)
O19—P1—O20—Mo9	-64.7 (4)	C23—N7—C21—O47	-179.5 (9)
O22—P1—O20—Mo9	175.3 (3)	C22—N7—C21—O47	4.0 (14)
O21—P1—O20—Mo9	55.3 (4)	Eu1—O45—C25—N8	176.5 (6)
O19—P1—O20—Mo7	54.7 (4)	C26—N8—C25—O45	-7.7 (16)
O22—P1—O20—Mo7	-65.4 (4)	C27—N8—C25—O45	-177.4 (9)
O21—P1—O20—Mo7	174.7 (3)	Eu1—O1—C28—N9	-169.7 (7)
O19—P1—O21—Mo10	55.4 (4)	C29—N9—C28—O1	-0.5 (16)
O20-P1-O21-Mo10	-64.9 (3)	C30—N9—C28—O1	176.3 (11)

Hydrogen-bond geometry (Å, °)

D—H···A	<i>D</i> —Н	H···A	$D \cdots A$	D—H···A
C3—H4…O31 ⁱ	0.96	2.72	3.289 (11)	118
C4—H7…O36 ⁱⁱ	0.96	2.61	3.476 (12)	150
С7—Н13…О7	0.96	2.65	3.473 (12)	145
C8—H15…O28 ⁱⁱ	0.96	2.51	3.316 (13)	141
C8—H16…O5	0.96	2.58	3.424 (11)	147
C10—H18…O28 ⁱⁱ	0.93	2.62	3.516 (10)	162
C15—H29…O16 ⁱ	0.96	2.55	3.299 (12)	135
C15—H27···O32 ⁱⁱⁱ	0.96	2.52	3.440 (10)	160
C15—H28····O52 ^{iv}	0.96	2.29	3.174 (10)	153
С16—Н31…Обііі	0.96	2.63	3.541 (11)	159
C22—H41…O55	0.96	2.53	3.280 (12)	135
C22—H42…O11 ^{iv}	0.96	2.45	3.305 (11)	149
C22—H43…O45	0.96	2.61	3.567 (13)	172
C23—H44…O17 ^v	0.96	2.52	3.443 (11)	161

C25—H55…O24	0.93	2.53	3.337 (11)	145
C25—H55…O44	0.93	2.58	3.091 (10)	115
C28—H56···O36 ^v	0.93	2.62	3.476 (11)	153
C30—H60····O36 ^v	0.96	2.58	3.458 (14)	152

Symmetry codes: (i) x+1/2, -y+1/2, z-1; (ii) -x+1, -y+1, z-1/2; (iii) x, y, z-1; (iv) -x+1, -y, z-1/2; (v) x+1/2, -y+1/2, z.