

carboxyl group. A plane can be drawn through the five ring atoms, from which the displacements are

$$\begin{aligned} O_1 &-0.0076, C_2 +0.0077, C_3 -0.0026 \text{ \AA}, \\ C_4 &-0.0035, C_5 +0.0086 \text{ \AA}, \end{aligned}$$

all of which are less than the standard deviations. On the other hand a plane taken through the four carbon atoms of the ring, is very accurately planar; the deviations are

$$C_2 -0.0003, C_3 +0.0005, C_4 -0.0005, C_5 +0.0003 \text{ \AA},$$

and the ring oxygen lies 0.0224 Å below this plane, a displacement which is substantially greater than the standard deviation. In relation to this plane the carboxyl group is twisted by nearly 2° in a direction away from the ring oxygen, the displacements being

$$C_1 -0.0458, O_2 +0.0442, O_3 -0.0261 \text{ \AA}.$$

The geometry of the carboxyl group is somewhat unusual; the longer bond of 1.29 Å is normal, the shorter, 1.18 Å, rather below average, but the relative size of the angles C-C-OH and C-C=O is reversed. Typical values are C-C-OH 116°, C-C=O 122° with O=C-OH 122°. In  $\alpha$ -furoic acid the values are 119°, 117° and 123° respectively. It is also to be noted that the configuration of the carboxyl group is reversed compared with  $\alpha$ -thiophenic acid and  $\alpha$ -selenophenic acid (Nardelli, Fava & Giraldi, 1962), in both of which the hydroxyl group faces the ring hetero-atom. Both sulphur and selenium compounds display normal geometry in their angles, although there is a certain amount of distortion of the angles bringing the carboxyl group nearer to the hetero-atom. It seems therefore that although the configuration is reversed in  $\alpha$ -furoic acid there is still an

attraction between carboxyl group and the ring oxygen atom. The exocyclic bond C<sub>1</sub>-C<sub>2</sub> is remarkably short, 1.41 Å, although it is comparable with the values observed for 1- and 2-naphthoic acids by Trotter (1960, 1961). This bond is usually shortened if the cyclic system is a conjugated one, but one would not expect shortening of quite such magnitude in  $\alpha$ -furoic acid which is not strongly aromatic. The acid forms hydrogen-bonded centrosymmetric dimers, the molecules being nearly co-planar, the distance O-H...O being 2.54 Å.

Because the *c* axis is so very short, the *z* parameters are not very reliable—as the standard deviations indicate. A three-dimensional analysis is needed to make the results really reliable; this would clarify the position of the C<sub>5</sub> atom and make conclusions about the ring more dependable.

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## Notes and News

*Announcements and other items of crystallographic interest will be published under this heading at the discretion of the Editorial Board. The notes (in duplicate) should be sent to the General Secretary of the International Union of Crystallography (D. W. Smits, Mathematisch Instituut, University of Groningen, Reütdiepskade 4, Groningen, The Netherlands).*

### International Union of Crystallography

The Executive Committee of the Union much regrets to have to announce that the Technical Editor of *Acta Crystallographica*, Professor R. W. Asmussen, has expressed a wish to be relieved of his task with the completion of the current volume of the journal. Professor Asmussen has served as Technical Editor since May 1958, when he succeeded Dr R. C. Evans. The Union, and the community of crystallographers, is much indebted to him for

his work on the journal, which during his tenure of office considerably expanded in size, and for the painstaking way in which he has carried out his duties.

No successor to Professor Asmussen has been found. Considering the increasing amount of work involved, the Executive Committee feels that the time has come to appoint a full-time technical editor for *all* Union publications. Attention is drawn to the relevant advertisement in this issue.

### Book Reviews

*Works intended for notice in this column should be sent direct to the Editor (A. J. C. Wilson, Department of Physics, University College, Cathays Park, Cardiff, Great Britain). As far as practicable books will be reviewed in a country different from that of publication.*

**Towards Information Retrieval.** By R. A. FAIRTHORNE. Pp. xxiv+211, 15 figs. London: Butterworths. 1961. Price 40s.

Crystallographers have always been interested in information retrieval, an early example being the list of all struc-

tures known in 1923 in Ewald's book 'Kristalle und Röntgenstrahlen'. Out of this interest has grown the continuing series of *Strukturbericht* and *Structure Reports*; indexes of a conventional but very thorough type have sufficed for retrieval of the information stored in these. Since its inception the International Union of Crystallo-