

ISSN 2414-3146

Received 31 March 2016 Accepted 10 April 2016

Edited by H. Stoeckli-Evans, University of Neuchâtel, Switzerland

Keywords: crystal structure; acenaphthylene-1,2-dione; α -diimine catalyst; potential bidentate ligand; C—H···N hydrogen bonding; π - π interactions.

CCDC reference: 1473249

Structural data: full structural data are available from iucrdata.iucr.org

(1*E*,2*E*)-*N*¹,*N*²-Bis(5'-methyl-[1,1':3',1"-terphenyl]-4'-yl)acenaphthylene-1,2-diimine unknown solvent

Liqiong Zhu* and Yan Zhao

Key Laboratory of Bioelectrochemical and Environmental Analysis of Gansu Province, College of Chemistry & Chemical Engineering, Northwest Normal University, Lanzhou 730070, People's Republic of China. *Correspondence e-mail: zhaoyan2532@126.com

The title compound, C50H36N2, synthesized by the condensation reaction of 2-methyl-4,6-diphenylaniline and acenaphthylene-1,2-dione, crystallizes with two independent molecules (A and B) in the asymmetric unit. The two molecules differ essentially in the orientation of the phenyl ring at position 3' of the terphenyl group with respect to the central ring of this unit. In molecule A this dihedral angle is 16.68 (14)°, while in molecule B the corresponding angle is 33.10 (16)°. The three-fused-ring 1,2-dihydroacenaphthylene units are planar in each molecule; r.m.s. deviation of 0.025 Å in molecule A and 0.017 Å in molecule B. The central rings of the terphenyl groups are almost normal to the mean plane of the three-fused-ring units with dihedral angles of 79.43 (12) and $82.66 (13)^{\circ}$ in molecule A and 88.99 (13) and $87.98 (12)^{\circ}$ in molecule B. In the crystal, the two molecules are linked via a $C-H \cdots N$ hydrogen bond. These A-B units are linked by a pair of $C-H \cdots \pi$ interactions, forming a four-molecule unit located about an inversion center. These four-molecule units are linked by weak π - π interactions [most significant intercentroid distance = 3.794 (2) Å]. forming columns along direction [010]. A region of disordered electron density was corrected for using the SQUEEZE routine in PLATON [Spek (2015). Acta Cryst. C71, 9–18]. The formula mass and unit-cell characteristics of this unknown solvent were not be taken into account during the refinement.



Structure description

In the past few decades, there has been a rapid development of a series of α -diimine palladium and nickel complex catalysts $[MX_2(\alpha\text{-diimine})]$ (where M = Ni, Pd; X = halide) for the polymerization of α -olefins since the original discovery of highly active α -diimine nickel catalysts (Johnson *et al.*, 1995). The catalytic activity and properties of the resulting





Figure 1

The condensation reaction of 2-methyl-4,6-diphenylaniline (1) and acenaphthylene-1,2-dione (2), giving the title compound.

polymers are greatly dependent on the reaction conditions (Helldörfer *et al.*, 2003) and ligand structures (Meinhard *et al.*, 2007; Popeney *et al.*, 2011; Yuan *et al.*, 2005; 2013). Nickel and palladium metal complex catalysts have a high catalytic activity for ethylene polymerization which gives high branched polyethylene, and the copolymerization of ethylene and polar monomers have also high catalytic activity. In this study, we describe the synthesis and crystal structure of the title compound, a new potential bidentate ligand prepared by the condensation reaction of 2-methyl-4,6-diphenylaniline (**1**) and acenaphthylene-1,2-dione (**2**); Fig. 1.

The title compound crystallizes with two independent molecules in the asymmetric unit. The molecular structure of molecule A is illustrated in Fig. 2, while that of molecule B is illustrated in Fig. 3. The *AutoMolFit* of molecule B inverted on molecule A gives the best fit (Fig. 4; Spek, 2009). The two molecules having weighted and unit weight r.m.s. fits of 0.317 and 0.278 Å, respectively, for the 52 non-H atoms. The two molecules differ essentially in the orientation of the phenyl ring at position 3' of the terphenyl group with respect to the



Figure 2

The molecular structure of molecule A of the title compound, with atom labelling. Displacement ellipsoids are drawn at the 50% probability level.



Figure 3

The molecular structure of molecule B of the title compound, with atom labelling. Displacement ellipsoids are drawn at the 50% probability level.



Figure 4

The *AutoMolFit* of molecule *B* (red) inverted on molecule *A* (black) gives the best fit (*PLATON*; Spek, 2009).

central ring of this unit. In molecule *A* this dihedral angle (C7–C12/C1-C6) is 16.68 (14)°, while in molecule *B* the corresponding angle (C94–C99/C82–C87) is 33.10 (16)°. The three-fused-ring 1,2-dihydroacenaphthylene units (C20–C31 in molecule *A*, and C70–C81 in molecule *B*) are planar in each molecule with r.m.s. deviation of 0.025 Å in molecule *A* and 0.017 Å in molecule *B*. The central ring of the terphenyl groups (C1–C6 and C32–C37 in molecule *A*, and C82–C88 and C51–C56 in molecule *B*) are almost normal to the mean plane of the three-fused-ring units, with dihedral angles of 79.43 (12) and 82.66 (13)°, respectively, in molecule *A*.

In the crystal, the two molecules are linked *via* a C-H···N hydrogen bond (Table 1 and Fig. 5). These A-B units are linked about a center of inversion by a pair of C-H··· π interactions, forming a four-molecule unit (Table 1 and Fig. 5). These four-molecule units are linked by slipped parallel π - π interactions forming columns along the *b*-axis direction $[Cg3 \cdots Cg18^{i} = 3.794 (2) \text{ Å}$, interplanar distance = 3.590 (1) Å, slippage = 0.787 Å, Cg3 is the centroid of ring C7-C12 in molecule *A* and Cg18 is the centroid of ring C71-C75/ C80 in molecule *B*, symmetry code (i): -x + 1, -y + 2, -z + 1].





A view along the *c* axis of the crystal packing of the title compound. The $C-H\cdots N$ hydrogen bonds and $C-H\cdots \pi$ interactions are represented by dashed lines (see Table 1). H atoms not involved in these interactions have been omitted for clarity.

Table 1 Hydrogen-bond geometry (Å, °).

Cg2 is the centroid of ring C1–C6 in molecule A.

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$C16-H16\cdots N3^i$	0.93	2.60	3.517 (6)	168
$C92 - H92 \cdot \cdot \cdot Cg2^{ii}$	0.93	2.95	3.789 (5)	151

Symmetry codes: (i) -x + 1, -y + 1, -z + 1; (ii) x + 1, y, z - 1.

Synthesis and crystallization

The synthesis of the title compound is illustrated in Fig. 1.

Synthesis of 2-methyl-4,6-diphenylaniline (1): 2-methyl-4,6dibromo-aniline (2 mmol, 0.53 g) was dissolved in PEG-400 (10 ml) containing phenylboronic acid (0.586 g, 4.8 mmol), K_2CO_3 (0,828 g;0.6 mmol) PdCl₂ (50 µg). The solution was placed in a round-bottomed flask and stirred at room temperature for 12 h. After the reaction, the solution was eluted with ethyl acetate/petroleum ether (v/v = 1:15) through a column, giving compound (1) as a colourless liquid (yield: 0.75 g, 79%).

Synthesis of the title compound: Formic acid (0.5 ml) was added to a stirred solution of acenaphthylene-1,2-dione (2) [0.18 g, 1.00 mmol] and compound (1) [0.57 g, 2.2 mmol) in ethanol (10 ml]. The mixture was refluxed for 24 h, and then cooled and the precipitate separated by filtration. The solid was recrystallized from EtOH/CH₂Cl₂ ($\nu/\nu = 10$:1), washed and dried under vacuum (yield: 0.43 g, 72%). Crystals suitable for *X*-ray structure analysis were grown from a cyclohexane/ dichloromethane ($\nu/\nu = 1$:2) solution by slow evaporation.

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2. A region of disordered electron density was corrected for using the SQUEEZE routine in *PLATON* (Spek, 2015): volume *ca* 690 Å³ for two regions of 36 and 65 electron counts. The formula mass and unit-cell characteristics of this unknown solvent were not taken into account during the refinement.

Acknowledgements

We thank the Key Laboratory of Bioelectrochemical and Environmental Analysis of Gansu Province, College of Chemistry and Chemical Engineering, (Northwest Normal University), for financial support.

Table 2	
Experimental details.	
Crystal data	
Chemical formula	$C_{50}H_{36}N_2$
M _r	664.81
Crystal system, space group	Triclinic, P1
Temperature (K)	296
<i>a</i> , <i>b</i> , <i>c</i> (Å)	15.579 (3), 16.799 (4), 17.332 (4)
α, β, γ (°)	72.622 (2), 73.116 (2), 84.324 (2)
$V(\text{\AA}^3)$	4142.0 (15)
Ζ	4
Radiation type	Μο Κα
$\mu (\text{mm}^{-1})$	0.06
Crystal size (mm)	$0.23\times0.21\times0.16$
Data collection	
Diffractometer	Bruker APEXII CCD
Absorption correction	Multi-scan (<i>SADABS</i> ; Bruker, 2008)
T_{\min}, T_{\max}	0.986, 0.990
No. of measured, independent and observed $[I > 2\sigma(I)]$ reflections	29357, 14707, 5868
R _{int}	0.041
$(\sin \theta / \lambda)_{\rm max} ({\rm \AA}^{-1})$	0.599
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.065, 0.144, 1.04
No. of reflections	14707
No. of parameters	930
H-atom treatment	H-atom parameters constrained
$\Delta \rho_{\rm max}, \Delta \rho_{\rm min} ({\rm e} {\rm \AA}^{-3})$	0.24, -0.29

Computer programs: APEX2 and SAINT (Bruker, 2008), SHELXS97 and SHELXL97 (Sheldrick, 2008), PLATON (Spek, 2009) and Mercury (Macrae et al., 2008).

References

- Bruker (2008). SAINT, APEX2 and SADABS. Bruker AXS Inc., Madison, Wisconsin, USA.
- Helldörfer, M., Milius, W. & Alt, H. G. (2003). J. Mol. Catal. A Chem. 197, 1–13.
- Johnson, L. K., Killian, C. M. & Brookhart, M. (1995). J. Am. Chem. Soc. 117, 6414–6415.
- Macrae, C. F., Bruno, I. J., Chisholm, J. A., Edgington, P. R., McCabe, P., Pidcock, E., Rodriguez-Monge, L., Taylor, R., van de Streek, J. & Wood, P. A. (2008). J. Appl. Cryst. 41, 466–470.
- Meinhard, D., Wegner, M., Kipiani, G., Hearley, A., Reuter, P., Fischer, S., Marti, O. & Rieger, B. (2007). J. Am. Chem. Soc. 129, 9182–9191.
- Popeney, C. S., Levins, C. M. & Guan, Z. B. (2011). Organometallics, 30, 2432–2452.
- Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.
- Spek, A. L. (2009). Acta Cryst. D65, 148-155.
- Spek, A. L. (2015). Acta Cryst. C71, 9-18.
- Yuan, J. C., Jia, Z., Li, J., Song, F. Y., Wang, F. Z. & Yuan, B. N. (2013). *Transition Met. Chem.* 38, 341–350.
- Yuan, J. C., Silva, L. C., Gomes, P. T., Valerga, P., Campos, J. M., Ribeiro, M. R., Chien, J. C. W. & Marques, M. M. (2005). *Polymer*, 46, 2122–2132.

full crystallographic data

IUCrData (2016). **1**, x160594 [doi:10.1107/S2414314616005940]

(1*E*,2*E*)-*N*¹,*N*²-Bis(5'-methyl-[1,1':3',1''-terphenyl]-4'-yl)acenaphthylene-1,2-diimine unknown solvent

Liqiong Zhu and Yan Zhao

 $(1E, 2E)-N^1, N^2$ -Bis(5'-methyl- $[1, 1': 3', 1''- \ terphenyl]-4'-yl)$ acenaphthylene-1,2-diimine unknown solvent

Crystal data

C₅₀H₃₆N₂ $M_r = 664.81$ Triclinic, *P*1 Hall symbol: -P1 a = 15.579 (3) Å b = 16.799 (4) Å c = 17.332 (4) Å $\alpha = 72.622$ (2)° $\beta = 73.116$ (2)° $\gamma = 84.324$ (2)° V = 4142.0 (15) Å³

Data collection

Bruker APEXII CCD diffractometer Radiation source: fine-focus sealed tube Graphite monochromator φ and ω scans Absorption correction: multi-scan (*SADABS*; Bruker, 2008) $T_{\min} = 0.986, T_{\max} = 0.990$

Refinement

Refinement on F^2 Least-squares matrix: full $R[F^2 > 2\sigma(F^2)] = 0.065$ $wR(F^2) = 0.144$ S = 1.0414707 reflections 930 parameters 0 restraints Primary atom site location: structure-invariant direct methods Secondary atom site location: difference Fourier map Z = 4 F(000) = 1400 $D_x = 1.066 \text{ Mg m}^{-3}$ Mo K α radiation, $\lambda = 0.71073 \text{ Å}$ Cell parameters from 3879 reflections $\theta = 2.3-21.0^{\circ}$ $\mu = 0.06 \text{ mm}^{-1}$ T = 296 K Block, colorless $0.23 \times 0.21 \times 0.16 \text{ mm}$

29357 measured reflections 14707 independent reflections 5868 reflections with $I > 2\sigma(I)$ $R_{int} = 0.041$ $\theta_{max} = 25.2^{\circ}, \ \theta_{min} = 2.3^{\circ}$ $h = -18 \rightarrow 18$ $k = -20 \rightarrow 20$ $l = -20 \rightarrow 20$

Hydrogen site location: inferred from neighbouring sites H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.040P)^2 + 0.3174P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} = 0.001$ $\Delta\rho_{max} = 0.24$ e Å⁻³ $\Delta\rho_{min} = -0.29$ e Å⁻³ Extinction correction: *SHELXL*, Fc*=kFc[1+0.001xFc^2\lambda^3/sin(2\theta)]^{-1/4} Extinction coefficient: 0.0017 (2)

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > 2 \text{sigma}(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$
C7	0.31195 (15)	0.84290 (16)	0.84423 (15)	0.0510 (8)
C8	0.34685 (16)	0.92224 (15)	0.82139 (13)	0.0807 (11)
H8	0.3571	0.9556	0.7661	0.097*
С9	0.36642 (16)	0.95177 (14)	0.8812 (2)	0.0916 (13)
H9	0.3898	1.0049	0.8659	0.110*
C10	0.35108 (18)	0.9020 (2)	0.96375 (17)	0.0931 (14)
H10	0.3642	0.9217	1.0037	0.112*
C11	0.3162 (2)	0.82262 (19)	0.98659 (12)	0.1235 (18)
H11	0.3059	0.7893	1.0419	0.148*
C12	0.29661 (18)	0.79309 (13)	0.92683 (18)	0.0998 (14)
H12	0.2733	0.7400	0.9421	0.120*
C1	0.26934 (19)	0.7493 (2)	0.65182 (19)	0.0488 (8)
C2	0.2781 (2)	0.8342 (2)	0.6375 (2)	0.0611 (9)
C3	0.2890 (2)	0.8636 (2)	0.7015 (2)	0.0630 (10)
Н3	0.2928	0.9208	0.6922	0.076*
C4	0.29437 (19)	0.8097 (2)	0.77868 (19)	0.0521 (8)
C5	0.28410 (19)	0.7258 (2)	0.79027 (19)	0.0534 (9)
Н5	0.2872	0.6884	0.8413	0.064*
C6	0.26940 (19)	0.6942 (2)	0.72966 (19)	0.0483 (8)
C13	0.2546 (2)	0.6033 (2)	0.74996 (19)	0.0557 (9)
C14	0.3040 (2)	0.5452 (2)	0.7955 (2)	0.0742 (11)
H14	0.3488	0.5632	0.8119	0.089*
C15	0.2878 (3)	0.4594 (3)	0.8174 (2)	0.0891 (13)
H15	0.3204	0.4210	0.8492	0.107*
C16	0.2237 (4)	0.4331 (3)	0.7916 (3)	0.0936 (14)
H16	0.2140	0.3763	0.8043	0.112*
C17	0.1739 (3)	0.4891 (3)	0.7474 (3)	0.1001 (14)
H17	0.1293	0.4707	0.7312	0.120*
C18	0.1896 (3)	0.5727 (3)	0.7268 (2)	0.0771 (11)
H18	0.1552	0.6102	0.6961	0.093*
C19	0.2785 (3)	0.8948 (2)	0.5528 (2)	0.1027 (14)
H19A	0.3331	0.8876	0.5119	0.154*
H19B	0.2746	0.9509	0.5569	0.154*
H19C	0.2282	0.8842	0.5361	0.154*
C20	0.3182 (2)	0.68981 (18)	0.54082 (19)	0.0470 (8)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters $(Å^2)$

C21	0.4137 (2)	0.67354 (19)	0.5368 (2)	0.0521 (8)
C22	0.4692 (2)	0.6909 (2)	0.5773 (2)	0.0704 (10)
H22	0.4471	0.7172	0.6195	0.084*
C23	0.5600(3)	0.6685 (3)	0.5545 (3)	0.0869 (12)
H23	0.5978	0.6810	0.5819	0.104*
C24	0.5952(3)	0.6293(2)	0.4938(3)	0.0850(12)
H24	0.6558	0.6148	0.4811	0.102*
C25	0.5400(3)	0.6105(2)	0.4502(2)	0.0668(10)
C26	0.5655 (3)	0.5709(2)	0.3845(3)	0.0000(10) 0.0845(12)
U20 H26	0.6245	0.5527	0.3673	0.101*
C27	0.0243	0.5527	0.3468 (2)	0.101 0.0863 (12)
U27	0.5042 (5)	0.5337 (2)	0.3408 (2)	0.0803 (12)
C28	0.3220	0.5551	0.3040	0.104°
C28	0.4140(3)	0.3802 (2)	0.3080 (2)	0.0707 (10)
П28 С20	0.3740	0.3783	0.3408	0.083°
C29	0.38/1(2)	0.62395(19)	0.4323(2)	0.0542 (9)
C30	0.4493 (2)	0.63508 (18)	0.4/23(2)	0.0532 (8)
C31	0.3009 (2)	0.65909 (18)	0.47245 (18)	0.0504 (8)
C32	0.2104 (2)	0.6412 (2)	0.3931 (2)	0.0538 (9)
C33	0.2297 (2)	0.6958 (2)	0.3117 (2)	0.0534 (9)
C34	0.2151 (2)	0.6676 (2)	0.2489 (2)	0.0624 (9)
H34	0.2279	0.7033	0.1945	0.075*
C35	0.1830 (2)	0.5901 (2)	0.2629 (2)	0.0639 (10)
C36	0.1594 (2)	0.5397 (2)	0.3458 (2)	0.0689 (10)
H36	0.1341	0.4881	0.3574	0.083*
C37	0.1723 (2)	0.5637 (2)	0.4115 (2)	0.0630 (10)
C38	0.2641 (2)	0.7811 (2)	0.2913 (2)	0.0509 (8)
C39	0.2268 (2)	0.8331 (2)	0.3420 (2)	0.0675 (10)
H39	0.1814	0.8135	0.3913	0.081*
C40	0.2569 (3)	0.9140 (2)	0.3193 (3)	0.0781 (11)
H40	0.2316	0.9480	0.3539	0.094*
C41	0.3232 (3)	0.9444 (2)	0.2471 (3)	0.0763 (11)
H41	0.3427	0.9989	0.2321	0.092*
C42	0.3601 (2)	0.8938 (2)	0.1975 (2)	0.0729 (11)
H42	0.4052	0.9138	0.1481	0.088*
C43	0.3314 (2)	0.8131 (2)	0.2199 (2)	0.0639 (10)
H43	0.3584	0.7792	0.1855	0.077*
C44	0.1732 (3)	0.5603 (2)	0.1918 (2)	0.0604 (10)
C45	0.2328 (3)	0.5863(2)	0.1142 (3)	0.0896 (13)
H45	0.2776	0.6233	0.1055	0.108*
C46	0.2273 (4)	0.5581 (3)	0.0477(3)	0.1095 (16)
H46	0.2688	0.5753	-0.0047	0.131*
C47	0.1602(5)	0.5755 0.5047 (3)	0.0605 (4)	0.1186(19)
H47	0.1555	0.4865	0.0162	0.142*
C48	0.0998(4)	0.4778(3)	0.1380(4)	0.112 0.1112 (17)
H48	0.0549	0.4407	0.1473	0.133*
C49	0.0577	0 5081 (2)	0 2038 (3)	0.0855 (12)
U-17 H40	0.0665	0.4015	0.2564	0.103*
C50	0.0003 0.1477(2)	0.7713	0.2304	0.105
0.50	0.14//(3)	0.3070(2)	U.7777 (4)	0.0777 (13)

H50A	0.2011	0.4901	0.5180	0.142*
H50B	0.1183	0.4587	0.5016	0.142*
H50C	0.1081	0.5361	0.5366	0.142*
C51	0.8308 (2)	0.76630 (18)	0.26840 (19)	0.0506 (8)
C52	0.9136 (2)	0.72957 (19)	0.2736 (2)	0.0539 (9)
C53	0.9371 (2)	0.71489 (19)	0.3476 (2)	0.0591 (9)
Н53	0.9926	0.6904	0.3509	0.071*
C54	0.8800 (2)	0.73574 (19)	0.4176 (2)	0.0538 (9)
C55	0.7973(2)	0.77178 (18)	0.41044(19)	0.0545(9)
H55	0.7581	0.7857	0.4565	0.065*
C56	0.7713(2)	0.78774 (17)	0.33696 (19)	0.0471 (8)
C57	0.9076(2)	0.71895(18)	0.33090(13)	0.0527(9)
C58	0.9978(2)	0.7233(2)	0.1972(2) 0.4962(2)	0.0527(3)
H58	1.0368	0.7340	0.4451	0.092*
C59	1.0182(3)	0.7122(2)	0.5705(3)	0.092
Н59	1.0780	0.7163	0.5682	0.109*
C60	0.9536(3)	0.6954(2)	0.5002	0.0825(12)
H60	0.9693	0.6886	0.6959	0.0025 (12)
C61	0.8648 (3)	0.6886 (2)	0.6488(2)	0.0790(11)
H61	0.8048 (5)	0.6760	0.0400(2)	0.095*
C62	0.8209	0.0700	0.7000	0.095
С02 Н62	0.7819	0.6965	0.5766	0.000+(10)
C63	0.7819	0.8266 (2)	0.33531 (19)	0.0552 (9)
C64	0.6407(3)	0.8200(2)	0.33331(1)) 0.3721(2)	0.0552(9)
U04 H64	0.6874	0.8902(2)	0.3721(2) 0.3940	0.0710(10)
C65	0.0874 0.5653 (3)	0.9114 0.0220 (3)	0.3340 0.3773(2)	0.085°
U05	0.5055 (5)	0.9229 (3)	0.3773(2)	0.0985 (14)
C66	0.5404	0.9039 0.8027 (4)	0.4020	0.118° 0.1140(18)
U66	0.3083 (3)	0.0927(4)	0.3402 (3)	0.1140 (18)
П00 С67	0.4303 0.5375 (2)	0.9142	0.3307 0.3084 (3)	0.137
07	0.3373 (3)	0.8304 (4)	0.3064 (3)	0.1008 (10)
П07 С69	0.4994	0.8100	0.2800	0.128°
	0.6237(3)	0.7904 (2)	0.3032 (2)	0.0784 (11)
H08	0.0425	0.7555	0.2781	0.094°
	0.9770(2)	0.7041(2)	0.1997 (2)	0.0746 (11)
H09A	1.0330	0.0932	0.2084	0.112*
H09B	0.9801	0.7484	0.1489	0.112*
H69C	0.9552	0.6547	0.1952	0.112*
C70	0.81620 (19)	0.8452 (2)	0.13499 (19)	0.0456 (8)
C/I	0.8550 (2)	0.92518 (18)	0.12345 (19)	0.0486 (8)
C/2	0.8870(2)	0.9589 (2)	0.1719(2)	0.0654 (10)
H72	0.8871	0.9283	0.2263	0.078*
C/3	0.9201 (2)	1.0411 (2)	0.1382 (3)	0.0797 (11)
H73	0.9414	1.0641	0.1715	0.096*
C/4	0.9218 (3)	1.0874 (2)	0.0591 (3)	0.0862 (12)
H74	0.9444	1.1411	0.0393	0.103*
C75	0.8898 (2)	1.0551 (2)	0.0065 (2)	0.0670 (10)
C76	0.8858 (3)	1.0956 (2)	-0.0769 (3)	0.0879 (12)
H76	0.9072	1.1496	-0.1029	0.105*

077	0.9512 (2)	1.05(9.(2))	0 1102 (2)	0.0004 (10)
C//	0.8513 (3)	1.0508 (5)	-0.1192 (2)	0.0824 (12)
H//	0.849/	1.0851	-0.1/38	0.099*
C/8	0.81/8 (2)	0.9754 (2)	-0.0835 (2)	0.0629 (9)
H78	0.7947	0.9500	-0.1139	0.075*
C79	0.82002 (19)	0.93409 (19)	-0.0025 (2)	0.0493 (8)
C80	0.8562 (2)	0.9742 (2)	0.0411 (2)	0.0502 (8)
C81	0.79069 (19)	0.8515 (2)	0.05548 (18)	0.0450 (8)
C82	0.7271 (2)	0.80212 (19)	-0.02622 (19)	0.0504 (8)
C83	0.7878 (2)	0.77977 (18)	-0.09352 (19)	0.0480 (8)
C84	0.7617 (2)	0.79215 (18)	-0.16637 (19)	0.0517 (8)
H84	0.8020	0.7784	-0.2120	0.062*
C85	0.6780 (2)	0.82414 (19)	-0.17419 (19)	0.0520 (8)
C86	0.6187 (2)	0.8422 (2)	-0.1044 (2)	0.0616 (9)
H86	0.5614	0.8619	-0.1073	0.074*
C87	0.6419 (2)	0.8317 (2)	-0.0305 (2)	0.0618 (9)
C88	0.8793 (2)	0.7470 (2)	-0.0903 (2)	0.0512 (8)
C89	0.8943 (2)	0.6866 (2)	-0.0210 (2)	0.0653 (10)
H89	0.8462	0.6673	0.0259	0.078*
C90	0.9794 (3)	0.6543 (2)	-0.0199 (3)	0.0821 (12)
H90	0.9881	0.6137	0.0272	0.099*
C91	1.0508 (3)	0.6828 (3)	-0.0891 (3)	0.0894 (13)
H91	1.1082	0.6617	-0.0887	0.107*
C92	1.0378 (3)	0.7421 (3)	-0.1585 (2)	0.0778 (11)
H92	1.0859	0.7609	-0.2056	0.093*
C93	0.9524 (3)	0.7738(2)	-0.1581(2)	0.0657 (10)
H93	0.9441	0.8147	-0.2052	0.079*
C94	0.6529 (2)	0.83902 (18)	-0.2551(2)	0.0498 (8)
C95	0.7146(2)	0.86355 (19)	-0.3299(2)	0.0637(10)
H95	0 7736	0.8703	-0 3313	0.076*
C96	0.6921 (3)	0.8790(2)	-0.4048(2)	0.0782(11)
H96	0.7359	0.8949	-0.4560	0.094*
C97	0.7557 0.6048 (3)	0.8706 (2)	-0.4028(3)	0.094 0.0884 (13)
H07	0.5888	0.8806	-0.4525	0.0004 (15)
C08	0.5420(3)	0.8474(2)	-0.3274(3)	0.100
U08	0.3420 (3)	0.8474 (2)	-0.3274 (3)	0.0878 (12)
C00	0.4023	0.0423	-0.2534(2)	0.103°
U00	0.5058 (5)	0.8313 (2)	-0.2334(2)	0.0717 (10)
П99	0.3224	0.0130	-0.2020	0.080°
	0.5755 (2)	0.8339 (3)	0.0424 (2)	0.0992 (14)
	0.5740	0.8131	0.0938	0.149
HIUB	0.5108	0.8023	0.0338	0.149*
HIOC	0.3932	0.90/6	0.0458	0.149*
IN I	0.2343/(1/)	0.72250(15)	0.58/08 (16)	0.0530(7)
NZ	0.22359 (19)	0.00008 (15)	0.46041 (15)	0.0557(7)
N3	0.80571 (17)	0.77649 (16)	0.19320 (16)	0.0552 (7)
N4	0.74955 (16)	0.79299 (16)	0.04974 (14)	0.0514 (7)

Atomic displacement parameters $(Å^2)$

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C7	0.048 (2)	0.060 (2)	0.054 (2)	0.0065 (18)	-0.0176 (18)	-0.0279 (19)
C8	0.110 (3)	0.089 (3)	0.063 (3)	-0.006(2)	-0.034 (2)	-0.040 (2)
C9	0.116 (3)	0.096 (3)	0.084 (3)	-0.004 (3)	-0.035 (3)	-0.047 (3)
C10	0.103 (3)	0.122 (4)	0.086 (3)	0.032 (3)	-0.050 (3)	-0.063 (3)
C11	0.217 (6)	0.098 (4)	0.073 (3)	-0.002 (4)	-0.063 (3)	-0.027 (3)
C12	0.167 (4)	0.083 (3)	0.074 (3)	-0.003 (3)	-0.055 (3)	-0.036 (3)
C1	0.053 (2)	0.055 (2)	0.042 (2)	0.0005 (17)	-0.0165 (17)	-0.0169 (18)
C2	0.080 (3)	0.064 (3)	0.050(2)	0.006 (2)	-0.034 (2)	-0.019 (2)
C3	0.084 (3)	0.048 (2)	0.062 (2)	-0.0010 (19)	-0.028 (2)	-0.0161 (19)
C4	0.052 (2)	0.058 (2)	0.049 (2)	0.0044 (18)	-0.0151 (17)	-0.0205 (19)
C5	0.057 (2)	0.057 (2)	0.041 (2)	0.0026 (18)	-0.0085 (17)	-0.0130 (17)
C6	0.050 (2)	0.053 (2)	0.046 (2)	-0.0001 (16)	-0.0105 (17)	-0.0227 (18)
C13	0.063 (2)	0.053 (2)	0.047 (2)	-0.001 (2)	-0.0061 (19)	-0.0175 (18)
C14	0.089 (3)	0.063 (3)	0.068 (3)	0.004 (2)	-0.016 (2)	-0.021 (2)
C15	0.116 (4)	0.060 (3)	0.072 (3)	0.007 (3)	-0.006 (3)	-0.013 (2)
C16	0.119 (4)	0.066 (3)	0.082 (3)	-0.030 (3)	0.003 (3)	-0.022 (3)
C17	0.108 (4)	0.084 (4)	0.103 (4)	-0.037 (3)	-0.015 (3)	-0.020 (3)
C18	0.083 (3)	0.071 (3)	0.080 (3)	-0.018 (2)	-0.024 (2)	-0.019 (2)
C19	0.179 (4)	0.068 (3)	0.073 (3)	-0.007 (3)	-0.059 (3)	-0.010 (2)
C20	0.058 (2)	0.042 (2)	0.040 (2)	-0.0046 (17)	-0.0121 (18)	-0.0100 (16)
C21	0.047 (2)	0.058 (2)	0.053 (2)	-0.0069 (17)	-0.0139 (18)	-0.0170 (18)
C22	0.059 (3)	0.085 (3)	0.070 (3)	-0.005 (2)	-0.013 (2)	-0.030 (2)
C23	0.063 (3)	0.106 (3)	0.097 (3)	-0.013 (2)	-0.025 (3)	-0.030 (3)
C24	0.053 (3)	0.095 (3)	0.096 (3)	0.000 (2)	-0.008 (3)	-0.023 (3)
C25	0.060 (3)	0.069 (3)	0.063 (3)	0.001 (2)	-0.008(2)	-0.016 (2)
C26	0.067 (3)	0.086 (3)	0.084 (3)	0.015 (2)	0.002 (2)	-0.027 (2)
C27	0.082 (3)	0.097 (3)	0.078 (3)	0.010 (3)	-0.002 (3)	-0.044 (2)
C28	0.080 (3)	0.074 (3)	0.061 (2)	0.007 (2)	-0.015 (2)	-0.030 (2)
C29	0.061 (2)	0.052 (2)	0.048 (2)	-0.0013 (18)	-0.0097 (19)	-0.0166 (17)
C30	0.055 (2)	0.046 (2)	0.052 (2)	-0.0002 (17)	-0.0056 (19)	-0.0113 (17)
C31	0.065 (3)	0.046 (2)	0.038 (2)	-0.0045 (18)	-0.0092 (19)	-0.0123 (16)
C32	0.064 (2)	0.056 (2)	0.057 (2)	0.0057 (19)	-0.0253 (19)	-0.033 (2)
C33	0.064 (2)	0.054 (2)	0.051 (2)	0.0033 (18)	-0.0215 (19)	-0.0237 (19)
C34	0.083 (3)	0.058 (2)	0.055 (2)	0.001 (2)	-0.029 (2)	-0.0207 (19)
C35	0.087 (3)	0.049 (2)	0.068 (3)	0.001 (2)	-0.034 (2)	-0.021 (2)
C36	0.094 (3)	0.050 (2)	0.077 (3)	-0.004(2)	-0.036 (2)	-0.026 (2)
C37	0.082 (3)	0.056 (2)	0.060 (3)	0.001 (2)	-0.026(2)	-0.023 (2)
C38	0.057 (2)	0.055 (2)	0.049 (2)	0.0020 (19)	-0.0208 (19)	-0.0228 (19)
C39	0.076 (3)	0.062 (3)	0.065 (2)	-0.002 (2)	-0.013 (2)	-0.025 (2)
C40	0.099 (3)	0.064 (3)	0.079 (3)	0.001 (2)	-0.022(3)	-0.034(2)
C41	0.104 (3)	0.056 (3)	0.072 (3)	-0.013 (2)	-0.029 (3)	-0.014 (2)
C42	0.084 (3)	0.073 (3)	0.065 (3)	-0.013 (2)	-0.017 (2)	-0.022 (2)
C43	0.074 (3)	0.067 (3)	0.062 (3)	-0.003 (2)	-0.021 (2)	-0.033 (2)
C44	0.088 (3)	0.048 (2)	0.064 (3)	0.001 (2)	-0.042 (2)	-0.025 (2)
C45	0.138 (4)	0.069 (3)	0.080 (3)	-0.013(3)	-0.043 (3)	-0.032 (3)

C46	0.172 (5)	0.089 (3)	0.085 (3)	-0.016 (3)	-0.050(3)	-0.033 (3)
C47	0.214 (6)	0.078 (4)	0.102 (4)	0.002 (4)	-0.097(4)	-0.030 (3)
C48	0.178 (5)	0.075 (3)	0.116 (4)	-0.013 (3)	-0.085 (4)	-0.030 (3)
C49	0.117 (4)	0.073 (3)	0.084 (3)	0.006 (3)	-0.051(3)	-0.027 (2)
C50	0.142 (4)	0.073 (3)	0.072 (3)	-0.027(3)	-0.028(3)	-0.018(2)
C51	0.067 (2)	0.047 (2)	0.043 (2)	-0.0067 (18)	-0.0208(19)	-0.0126 (16)
C52	0.066 (3)	0.051 (2)	0.047 (2)	-0.0028(18)	-0.016 (2)	-0.0160 (17)
C53	0.060 (2)	0.064 (2)	0.057 (2)	0.0094 (18)	-0.024(2)	-0.0199 (19)
C54	0.060 (2)	0.058 (2)	0.048 (2)	0.0054 (18)	-0.0225(19)	-0.0172(17)
C55	0.060 (2)	0.058(2)	0.047(2)	-0.0003(18)	-0.0162(18)	-0.0149(17)
C56	0.058(2)	0.045(2)	0.040(2)	-0.0047(17)	-0.0185(18)	-0.0086(16)
C57	0.063(3)	0.056(2)	0.046(2)	0.0036 (18)	-0.025(2)	-0.0168(17)
C58	0.065(3)	0.111(3)	0.052(2)	0.002.0(10)	-0.017(2)	-0.023(2)
C59	0.077(3)	0.129 (4)	0.078(3)	0.015(3)	-0.034(3)	-0.039(3)
C60	0.098(3)	0.027(3)	0.069(3)	0.023(3)	-0.044(3)	-0.036(2)
C61	0.097(3)	0.086(3)	0.056(3)	0.022(2)	-0.026(2)	-0.027(2)
C62	0.078(3)	0.072(2)	0.059(3)	0.017(2)	-0.035(2)	-0.023(2)
C63	0.059(2)	0.061(2)	0.023(2)	-0.007(2)	-0.0239(19)	-0.0011(18)
C64	0.073(3)	0.001(2) 0.075(3)	0.062(2)	0.018(2)	-0.025(2)	-0.015(2)
C65	0.098(4)	0.075(3) 0.115(4)	0.002(2) 0.076(3)	0.040(3)	-0.037(3)	-0.018(3)
C66	0.076 (4)	0.116(1)	0.070(3)	0.010(3) 0.022(4)	-0.027(3)	-0.009(3)
C67	0.071 (4)	0.164(5)	0.080(3)	-0.014(3)	-0.039(3)	-0.004(3)
C68	0.074(3)	0.092(3)	0.066(3)	-0.010(2)	-0.030(2)	-0.005(2)
C69	0.080(3)	0.083(3)	0.064(2)	0.005(2)	-0.018(2)	-0.029(2)
C70	0.051(2)	0.045(2)	0.042(2)	-0.0048(17)	-0.0136(17)	-0.0131(17)
C71	0.057(2)	0.043(2)	0.053(2)	-0.0048(17)	-0.0235(18)	-0.0168(18)
C72	0.082(3)	0.063(3)	0.067(2)	-0.005(2)	-0.038(2)	-0.023(2)
C73	0.101(3)	0.060(3)	0.101(3)	-0.015(2)	-0.054(3)	-0.028(2)
C74	0.097(3)	0.054(3)	0.113 (4)	-0.016(2)	-0.042(3)	-0.013(3)
C75	0.071(3)	0.051(2)	0.080(3)	-0.009(2)	-0.029(2)	-0.010(2)
C76	0.101(3)	0.063(3)	0.084(3)	-0.025(2)	-0.030(3)	0.013(2)
C77	0.094(3)	0.078(3)	0.066(3)	-0.008(2)	-0.028(2)	0.002(2)
C78	0.069(3)	0.064(3)	0.051(2)	-0.003(2)	-0.0188(19)	-0.0078(19)
C79	0.051(2)	0.051(2)	0.046(2)	-0.0044(17)	-0.0161(17)	-0.0109(18)
C80	0.050(2)	0.044(2)	0.055(2)	-0.0045(17)	-0.0164(18)	-0.0079(18)
C81	0.044(2)	0.053(2)	0.044(2)	-0.0024(17)	-0.0110(16)	-0.0239(18)
C82	0.051 (2)	0.069(2)	0.040(2)	-0.0087(18)	-0.0155(18)	-0.0228(17)
C83	0.053(2)	0.058(2)	0.043(2)	-0.0071(17)	-0.0157(18)	-0.0230(17)
C84	0.052(2)	0.064(2)	0.043(2)	-0.0032(18)	-0.0102(18)	-0.0237(17)
C85	0.053(2)	0.064(2)	0.048(2)	-0.0019(18)	-0.0176(19)	-0.0242(18)
C86	0.050(2)	0.082(3)	0.059(2)	0.0000 (18)	-0.014(2)	-0.031(2)
C87	0.058(2)	0.093(3)	0.046(2)	-0.002(2)	-0.0128(19)	-0.037(2)
C88	0.058(2)	0.061(2)	0.044(2)	-0.0061(19)	-0.016(2)	-0.0242(18)
C89	0.065(3)	0.072(3)	0.058(2)	-0.002(2)	-0.012(2)	-0.022(2)
C90	0.087 (3)	0.089 (3)	0.077 (3)	0.019 (3)	-0.039(3)	-0.024(2)
C91	0.060 (3)	0.118 (4)	0.100 (4)	0.017 (3)	-0.029 (3)	-0.044(3)
C92	0.059 (3)	0.103 (3)	0.069 (3)	0.000 (2)	-0.015 (2)	-0.024 (2)
C93	0.065 (3)	0.080 (3)	0.054 (2)	-0.001 (2)	-0.016 (2)	-0.023 (2)
	× /	× /	× /	× /	× /	× /

C94	0.049 (2)	0.051 (2)	0.051 (2)	-0.0012 (17)	-0.012 (2)	-0.0198 (17)
C95	0.079 (3)	0.066 (2)	0.055 (2)	0.000 (2)	-0.028 (2)	-0.0213 (19)
C96	0.101 (4)	0.080 (3)	0.051 (3)	0.005 (2)	-0.021 (2)	-0.017 (2)
C97	0.108 (4)	0.107 (3)	0.066 (3)	0.028 (3)	-0.047 (3)	-0.035 (3)
C98	0.083 (3)	0.124 (4)	0.072 (3)	0.015 (3)	-0.037 (3)	-0.041 (3)
C99	0.076 (3)	0.091 (3)	0.061 (3)	0.003 (2)	-0.026 (2)	-0.034 (2)
C100	0.070 (3)	0.177 (4)	0.075 (3)	0.016 (3)	-0.022 (2)	-0.075 (3)
N1	0.0632 (19)	0.0567 (18)	0.0465 (17)	0.0013 (14)	-0.0177 (15)	-0.0238 (14)
N2	0.061 (2)	0.0595 (18)	0.0519 (18)	-0.0010 (15)	-0.0178 (16)	-0.0212 (14)
N3	0.072 (2)	0.0536 (19)	0.0468 (18)	-0.0049 (15)	-0.0221 (15)	-0.0173 (15)
N4	0.0555 (18)	0.0631 (19)	0.0427 (17)	-0.0100 (14)	-0.0117 (14)	-0.0243 (14)

Geometric parameters (Å, °)

С7—С8	1.3900	C51—C52	1.389 (4)
C7—C12	1.3900	C51—C56	1.397 (4)
C7—C4	1.505 (3)	C51—N3	1.425 (3)
C8—C9	1.3900	C52—C53	1.380 (4)
С8—Н8	0.9300	C52—C69	1.518 (4)
C9—C10	1.3900	C53—C54	1.397 (4)
С9—Н9	0.9300	С53—Н53	0.9300
C10—C11	1.3900	C54—C55	1.391 (4)
C10—H10	0.9300	C54—C57	1.503 (4)
C11—C12	1.3900	C55—C56	1.390 (4)
C11—H11	0.9300	С55—Н55	0.9300
C12—H12	0.9300	C56—C63	1.495 (4)
C1—C2	1.386 (4)	C57—C58	1.331 (4)
C1—C6	1.390 (4)	C57—C62	1.406 (4)
C1—N1	1.410 (3)	C58—C59	1.410 (4)
C2—C3	1.397 (4)	C58—H58	0.9300
C2—C19	1.514 (4)	C59—C60	1.374 (5)
C3—C4	1.394 (4)	С59—Н59	0.9300
С3—Н3	0.9300	C60—C61	1.387 (5)
C4—C5	1.383 (4)	C60—H60	0.9300
C5—C6	1.390 (4)	C61—C62	1.383 (4)
С5—Н5	0.9300	C61—H61	0.9300
C6—C13	1.486 (4)	C62—H62	0.9300
C13—C14	1.384 (4)	C63—C64	1.382 (4)
C13—C18	1.384 (4)	C63—C68	1.383 (4)
C14—C15	1.405 (5)	C64—C65	1.363 (5)
C14—H14	0.9300	C64—H64	0.9300
C15—C16	1.363 (5)	C65—C66	1.363 (5)
С15—Н15	0.9300	С65—Н65	0.9300
C16—C17	1.357 (5)	C66—C67	1.368 (6)
C16—H16	0.9300	С66—Н66	0.9300
C17—C18	1.370 (5)	C67—C68	1.395 (5)
С17—Н17	0.9300	С67—Н67	0.9300
C18—H18	0.9300	С68—Н68	0.9300

С19—Н19А	0.9600	С69—Н69А	0.9600
С19—Н19В	0.9600	С69—Н69В	0.9600
C19—H19C	0.9600	С69—Н69С	0.9600
C20—N1	1.273 (3)	C70—N3	1.274 (3)
C20—C21	1.472 (4)	C70—C71	1.463 (4)
C20—C31	1.524 (4)	C70—C81	1.514 (4)
C21—C22	1.359 (4)	C71—C72	1.363 (4)
C21—C30	1.405 (4)	C71—C80	1.414 (4)
C22—C23	1.400 (4)	С72—С73	1.413 (4)
С22—Н22	0.9300	С72—Н72	0.9300
C23—C24	1.359 (5)	C73—C74	1.352 (4)
С23—Н23	0.9300	С73—Н73	0.9300
C24—C25	1.408 (5)	C74—C75	1.410 (4)
C24—H24	0.9300	С74—Н74	0.9300
C25—C30	1.407 (4)	C75—C80	1.398 (4)
C25—C26	1.427 (5)	C75—C76	1.418 (5)
C26—C27	1 359 (4)	C76—C77	1 354 (4)
C26—H26	0.9300	C76—H76	0.9300
C_{27} C_{28}	1 401 (4)	C77 - C78	1 405 (4)
C27—H27	0.9300	С77—Н77	0.9300
C_{28} C_{29}	1 376 (4)	C78—C79	1 376 (4)
C28—H28	0.9300	C78—H78	0.9300
C_{29} C_{30}	1 400 (4)	C79 - C80	1 404 (4)
C_{29} C_{31}	1 478 (4)	C79—C81	1 473 (4)
C_{31} N2	1 269 (3)	C81—N4	1.175 (1)
C32—C37	1.394 (4)	C82—C87	1.385 (4)
C32—C33	1.399 (4)	C82—C83	1.395 (4)
C32—N2	1.423 (3)	C82—N4	1.420 (3)
C33—C34	1.390 (4)	C83—C84	1.388 (4)
C33—C38	1.485 (4)	C83—C88	1.488 (4)
C34—C35	1.370 (4)	C84—C85	1.387 (4)
C34—H34	0.9300	C84—H84	0.9300
C35—C36	1.394 (4)	C85—C86	1.386 (4)
C35—C44	1.511 (4)	C85—C94	1.508 (4)
C36—C37	1.388 (4)	C86—C87	1.387 (4)
С36—Н36	0.9300	С86—Н86	0.9300
C37—C50	1.503 (4)	C87—C100	1.515 (4)
C38—C43	1.371 (4)	C88—C93	1.374 (4)
C38—C39	1.391 (4)	C88—C89	1.383 (4)
C39—C40	1.386 (4)	C89—C90	1.384 (4)
С39—Н39	0.9300	С89—Н89	0.9300
C40—C41	1.367 (4)	С90—С91	1.377 (5)
C40—H40	0.9300	С90—Н90	0.9300
C41—C42	1.359 (4)	C91—C92	1.366 (5)
C41—H41	0.9300	С91—Н91	0.9300
C42—C43	1.377 (4)	С92—С93	1.382 (4)
C42—H42	0.9300	С92—Н92	0.9300
С43—Н43	0.9300	С93—Н93	0.9300

C44—C49	1.341 (4)	C94—C95	1.344 (4)
C44—C45	1.365 (5)	С94—С99	1.365 (4)
C45—C46	1.397 (5)	C95—C96	1.387 (4)
C45—H45	0.9300	С95—Н95	0.9300
C46—C47	1.371 (6)	C96—C97	1.370 (5)
C46—H46	0.9300	С96—Н96	0.9300
C47—C48	1.372 (6)	C97—C98	1.359 (5)
C47—H47	0.9300	С97—Н97	0.9300
C48—C49	1.420 (5)	C98—C99	1.381 (4)
C48—H48	0.9300	C98—H98	0.9300
C49—H49	0.9300	C99—H99	0.9300
C50—H50A	0.9600	C100—H10A	0.9600
C50_H50R	0.9600	C100—H10B	0.9600
C50 H50C	0.9600	C100 H10C	0.9600
C30—1150C	0.9000		0.9000
C8—C7—C12	120.0	C56—C51—N3	120.7(3)
$C_{8} - C_{7} - C_{4}$	110.0(2)	C_{53} C_{52} C_{51} C_{51}	120.7(3) 1101(3)
C_{12} C_{7} C_{4}	119.9(2) 1201(2)	$C_{53} = C_{52} = C_{51}$	117.1(3) 120.2(3)
$C_{12} = C_{7} = C_{7}$	120.1 (2)	$C_{55} - C_{52} - C_{69}$	120.2(3) 120.7(3)
$C_{2} = C_{3} = C_{1}$	120.0	$C_{51} = C_{52} = C_{69}$	120.7(3) 121.0(3)
$C_7 C_8 H_8$	120.0	$C_{32} = C_{33} = C_{34}$	121.9 (5)
$C_{1}^{2} = C_{1}^{2} = C_{1}^{2}$	120.0	$C_{32} - C_{33} - H_{33}$	119.0
	120.0	$C_{54} = C_{53} = 1155$	117.0 117.4(3)
$C_{0} = C_{0} = 119$	120.0	$C_{55} = C_{54} = C_{55}$	117.4(3)
C_{10} C_{10} C_{11}	120.0	$C_{53} = C_{54} = C_{57}$	121.9(3)
	120.0	$C_{55} = C_{54} = C_{54}$	120.0(3)
C9-C10-H10	120.0	$C_{50} = C_{55} = C_{54}$	122.4 (3)
CII—CI0—HI0	120.0	С56—С55—Н55	118.8
	120.0	С54—С55—Н55	118.8
С12—С11—Н11	120.0	C55—C56—C51	118.2 (3)
С10—С11—Н11	120.0	C55—C56—C63	118.5 (3)
C11—C12—C7	120.0	C51—C56—C63	123.3 (3)
С11—С12—Н12	120.0	C58—C57—C62	118.9 (3)
C7—C12—H12	120.0	C58—C57—C54	121.8 (3)
C2—C1—C6	120.2 (3)	C62—C57—C54	119.3 (3)
C2-C1-N1	117.7 (3)	C57—C58—C59	121.9 (4)
C6—C1—N1	122.0 (3)	С57—С58—Н58	119.0
C1—C2—C3	119.4 (3)	С59—С58—Н58	119.0
C1—C2—C19	120.7 (3)	C60—C59—C58	119.3 (4)
C3—C2—C19	119.9 (3)	С60—С59—Н59	120.3
C4—C3—C2	121.9 (3)	С58—С59—Н59	120.3
C4—C3—H3	119.0	C59—C60—C61	119.5 (4)
С2—С3—Н3	119.0	С59—С60—Н60	120.2
C5—C4—C3	116.5 (3)	С61—С60—Н60	120.2
C5—C4—C7	122.9 (3)	C62—C61—C60	120.0 (4)
C3—C4—C7	120.6 (3)	С62—С61—Н61	120.0
C4—C5—C6	123.4 (3)	С60—С61—Н61	120.0
C4—C5—H5	118.3	C61—C62—C57	120.4 (3)
С6—С5—Н5	118.3	С61—С62—Н62	119.8

C1—C6—C5	118.4 (3)	С57—С62—Н62	119.8
C1—C6—C13	122.4 (3)	C64—C63—C68	117.9 (3)
C5—C6—C13	119.2 (3)	C64—C63—C56	120.3 (3)
C14—C13—C18	116.8 (3)	C68—C63—C56	121.6 (3)
C14—C13—C6	121.0 (3)	C65—C64—C63	121.9 (4)
C18—C13—C6	122.2 (3)	С65—С64—Н64	119.1
C13—C14—C15	121.0 (4)	С63—С64—Н64	119.1
C13—C14—H14	119.5	C66—C65—C64	120.3 (4)
C15—C14—H14	119.5	С66—С65—Н65	119.8
C16—C15—C14	119.4 (4)	С64—С65—Н65	119.8
C16—C15—H15	120.3	C65—C66—C67	119.4 (5)
C14—C15—H15	120.3	С65—С66—Н66	120.3
C17—C16—C15	120.6 (4)	С67—С66—Н66	120.3
C17—C16—H16	119.7	C66—C67—C68	120.7 (5)
C15—C16—H16	119.7	С66—С67—Н67	119.7
C16—C17—C18	119.8 (4)	С68—С67—Н67	119.7
С16—С17—Н17	120.1	C63—C68—C67	119.8 (4)
С18—С17—Н17	120.1	С63—С68—Н68	120.1
C17—C18—C13	122.4 (4)	С67—С68—Н68	120.1
С17—С18—Н18	118.8	С52—С69—Н69А	109.5
C13—C18—H18	118.8	С52—С69—Н69В	109.5
C2-C19-H19A	109.5	H69A—C69—H69B	109.5
C2-C19-H19B	109.5	C52—C69—H69C	109.5
H19A—C19—H19B	109.5	H69A—C69—H69C	109.5
C2-C19-H19C	109.5	H69B—C69—H69C	109.5
H19A—C19—H19C	109.5	N3-C70-C71	133.4 (3)
H19B-C19-H19C	109.5	N3-C70-C81	1194(3)
N1-C20-C21	133.3 (3)	C71 - C70 - C81	107.0(3)
N1-C20-C31	120.0 (3)	C72-C71-C80	118.4 (3)
C_{21} C_{20} C_{31}	106.7 (3)	C72 - C71 - C70	135.3 (3)
C_{22} C_{21} C_{30}	119.2 (3)	C80—C71—C70	106.3 (3)
C_{22} C_{21} C_{20}	134.4 (3)	C71—C72—C73	119.0 (3)
C_{30} C_{21} C_{20}	106.3 (3)	С71—С72—Н72	120.5
$C_{21} - C_{22} - C_{23}$	118 9 (3)	C73 - C72 - H72	120.5
C21—C22—H22	120.5	C74 - C73 - C72	122.3 (3)
C23—C22—H22	120.5	С74—С73—Н73	118.8
C_{24} C_{23} C_{22}	122.7 (4)	С72—С73—Н73	118.8
C24—C23—H23	118.6	C73 - C74 - C75	120.8 (4)
$C_{22} = C_{23} = H_{23}$	118.6	C73—C74—H74	119.6
C_{23} C_{24} C_{25}	120 1 (4)	C75—C74—H74	119.6
C_{23} C_{24} H_{24}	120.1 (1)	C80-C75-C74	116.2 (3)
$C_{25} = C_{24} = H_{24}$	120.0	C80 - C75 - C76	116.0(3)
C_{24} C_{25} C_{30} C_{30}	116.6 (3)	C74 - C75 - C76	127.8(4)
C_{24} C_{25} C_{26} C_{26}	127 7 (4)	C77 - C76 - C75	127.0(-7) 121.0(4)
C_{30} C_{25} C_{26} C_{26}	115 7 (4)	С77—С76—Н76	119 5
C_{27} C_{26} C_{25} C	120 3 (4)	C75—C76—H76	119.5
C27—C26—H26	119.8	C76 - C77 - C78	122 3 (4)
$C_{2} = C_{2} = C_{2$	110.8	C76 C77 H77	118.8
023 - 020 - 1120	117.0	$U_{1}U_{1}U_{1}U_{1}U_{1}U_{1}U_{1}U_{1}$	110.0

C26—C27—C28	123.1 (4)	С78—С77—Н77	118.8
С26—С27—Н27	118.4	C79—C78—C77	118.5 (3)
С28—С27—Н27	118.4	С79—С78—Н78	120.7
C29—C28—C27	118.3 (3)	С77—С78—Н78	120.7
C29—C28—H28	120.8	C78—C79—C80	119.1 (3)
C27—C28—H28	120.8	C78—C79—C81	134.4 (3)
C28—C29—C30	119.2 (3)	C80—C79—C81	106.5 (3)
C28—C29—C31	134.2 (3)	C75—C80—C79	123.0 (3)
C30—C29—C31	106.6 (3)	C75—C80—C71	123.3 (3)
C29—C30—C21	114.3 (3)	C79—C80—C71	113.7 (3)
C29—C30—C25	123.3 (3)	N4—C81—C79	132.5 (3)
C21—C30—C25	122.3 (3)	N4—C81—C70	121.1 (3)
N2—C31—C29	133.1 (3)	C79—C81—C70	106.4 (3)
N2-C31-C20	120.8 (3)	C87—C82—C83	121.0 (3)
C29—C31—C20	106.1 (3)	C87—C82—N4	117.8 (3)
C37—C32—C33	121.1 (3)	C83—C82—N4	121.2 (3)
C37—C32—N2	118.2 (3)	C84—C83—C82	117.8 (3)
C_{33} — C_{32} — N_{2}	120.5 (3)	C84—C83—C88	120.0(3)
C_{34} C_{33} C_{32}	117.5 (3)	C82—C83—C88	120.0(3) 122.1(3)
C_{34} C_{33} C_{38}	120.1 (3)	C85 - C84 - C83	122.9(3)
C_{32} C_{33} C_{38}	122.4 (3)	C85—C84—H84	118.5
C_{35} C_{34} C_{33}	123.5(3)	C83—C84—H84	118.5
C35—C34—H34	118.2	C86-C85-C84	117.1(3)
C33—C34—H34	118.2	C86-C85-C94	1214(3)
C_{34} C_{35} C_{36} C_{36}	117.0(3)	C84 - C85 - C94	121.1(3) 121.5(3)
$C_{34} = C_{35} = C_{44}$	121 6 (3)	C85 - C86 - C87	121.3(3) 1221(3)
C_{36} C_{35} C_{44}	121.0(3) 121.4(3)	C85 - C86 - H86	112.1 (5)
C_{37} C_{36} C_{35} C_{44}	121.4(3) 122.5(3)	C87 - C86 - H86	118.9
$C_{37} = C_{30} = C_{35}$	1122.5 (5)	$C_{87} = C_{80} = 1180$	118.0(3)
C_{35} C_{36} H_{36}	118.8	C82 - C87 - C80	110.9(3) 1213(3)
$C_{36} = C_{30} = C_{30}$	118.1 (3)	$C_{82} = C_{87} = C_{100}$	121.3(3) 110.8(3)
$C_{36} = C_{37} = C_{52}$	110.1(3)	$C_{00} = C_{00} = C_{100}$	117.0(3)
$C_{30} = C_{37} = C_{50}$	121.0(3) 120.0(3)	$C_{93} = C_{83} = C_{83}$	117.3(3) 120.7(3)
$C_{32} = C_{37} = C_{30}$	120.9(3) 117.3(3)	$C_{23} = C_{33} = C_{33}$	120.7(3) 122.0(3)
$C_{+3} = C_{38} = C_{33}$	117.3 (3)	$C_{89} = C_{89} = C_{89}$	122.0(3) 121.5(3)
$C_{+5} = C_{55} = C_{55}$	121.3(3) 121.3(3)	$C_{90} = C_{89} = C_{88}$	121.3(3)
$C_{33} = C_{33} = C_{33}$	121.3(3) 120.2(3)		119.2
$C_{40} = C_{39} = C_{38}$	120.2 (3)	$C_{00} = C_{00} = C_{00}$	119.2
$C_{40} = C_{59} = H_{59}$	119.9	$C_{91} = C_{90} = C_{89}$	119.4 (4)
$C_{30} = C_{39} = H_{39}$	119.9	$C_{91} = C_{90} = H_{90}$	120.5
C41 - C40 - C39	121.1 (5)	$C_{89} = C_{90} = H_{90}$	120.5
$C_{41} = C_{40} = H_{40}$	119.4	C92 = C91 = C90	120.2 (4)
$C_{39} = C_{40} = \Pi_{40}$	117.4	C_{92} C_{91} C	119.9
C42 - C41 - C40	118.9 (4)	C90 - C91 - H91	119.9
U42 - U41 - H41	120.0	$C_{91} - C_{92} - C_{93}$	119.4 (4)
C40 - C41 - H41	120.0	$C_{91} - C_{92} - H_{92}$	120.3
C41 - C42 - C43	120.5 (4)	C93—C92—H92	120.3
C41—C42—H42	119.8	C88—C93—C92	122.2 (3)
C43—C42—H42	119.8	С88—С93—Н93	118.9

C38—C43—C42	122.0 (3)	С92—С93—Н93	118.9
C38—C43—H43	119.0	C95—C94—C99	119.0 (3)
C42—C43—H43	119.0	C95—C94—C85	120.8 (3)
C49—C44—C45	119.5 (3)	C99—C94—C85	120.1 (3)
C49—C44—C35	121.1 (4)	C94—C95—C96	121.5 (3)
C45 - C44 - C35	1194(4)	C94—C95—H95	119.3
C_{44} C_{45} C_{46}	120.9(4)	C96—C95—H95	119.3
C_{44} C_{45} H_{45}	110.5	C97 C96 C95	119.3
$C_{44} - C_{45} - 1145$	119.5	C97 - C96 + 196	119.5 (+)
C40-C43-H43	119.3	C97 - C90 - H90	120.3
C4/-C46-C45	119.3 (5)	C95—C96—H96	120.3
C47—C46—H46	120.3	C98—C97—C96	119.4 (4)
C45—C46—H46	120.3	С98—С97—Н97	120.3
C46—C47—C48	120.6 (5)	С96—С97—Н97	120.3
C46—C47—H47	119.7	C97—C98—C99	120.4 (4)
C48—C47—H47	119.7	С97—С98—Н98	119.8
C47—C48—C49	118.2 (5)	С99—С98—Н98	119.8
C47—C48—H48	120.9	C94—C99—C98	120.4 (4)
C49—C48—H48	120.9	С94—С99—Н99	119.8
C44—C49—C48	121.5 (4)	С98—С99—Н99	119.8
C44—C49—H49	119.3	C87—C100—H10A	109.5
C48—C49—H49	119.3	C87—C100—H10B	109.5
C37—C50—H50A	109.5	H10A—C100—H10B	109.5
C37—C50—H50B	109.5	C87—C100—H10C	109.5
H50A_C50_H50B	109.5	H10A - C100 - H10C	109.5
C37 C50 H50C	109.5	HIOR CIOO HIOC	109.5
H50A C50 H50C	109.5	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	109.5 120.5(3)
H50R C50 H50C	109.5	C_{20} N1 C_{1}	120.3(3)
$\begin{array}{c} \text{H30B} \\ \text{C30} \\ \text{H30C} \\ \text{C51} \\ \text{C56} \\ $	109.5	$C_{31} = N_2 = C_{32}$	119.7(3)
$C_{2} = C_{3} = C_{3} = C_{3}$	120.9 (3)	C/0 N3 $-C31$	121.9 (3)
C52—C51—N3	118.2 (3)	C81—N4—C82	118.8 (3)
C12 C7 C8 C0	0.0	C52 C54 C55 C56	05(4)
	0.0	C53—C54—C55—C56	-0.5 (4)
C4—C7—C8—C9	-177.5 (2)	057-054-055-056	179.8 (3)
C/C8C9C10	0.0	C54—C55—C56—C51	0.2 (4)
C8—C9—C10—C11	0.0	C54—C55—C56—C63	179.9 (3)
C9—C10—C11—C12	0.0	C52—C51—C56—C55	0.4 (4)
C10—C11—C12—C7	0.0	N3—C51—C56—C55	176.2 (3)
C8—C7—C12—C11	0.0	C52—C51—C56—C63	-179.4 (3)
C4—C7—C12—C11	177.5 (2)	N3—C51—C56—C63	-3.6 (4)
C6—C1—C2—C3	1.4 (5)	C55—C54—C57—C58	-147.3 (3)
N1-C1-C2-C3	177.3 (3)	C53—C54—C57—C58	33.1 (5)
C6-C1-C2-C19	179.7 (3)	C55—C54—C57—C62	30.7 (4)
N1—C1—C2—C19	-4.4 (5)	C53—C54—C57—C62	-148.9(3)
C1—C2—C3—C4	2.2 (5)	C62—C57—C58—C59	-1.7 (5)
C19—C2—C3—C4	-176.1 (3)	C54—C57—C58—C59	176.3 (3)
C2—C3—C4—C5	-3.0(5)	C57—C58—C59—C60	0.8 (6)
$C_{2}-C_{3}-C_{4}-C_{7}$	176.1 (3)	C58—C59—C60—C61	0.8 (6)
C8-C7-C4-C5	1614(2)	C59 - C60 - C61 - C62	-14(5)
C_{12} C_{7} C_{4} C_{5}	-161(4)	C60 - C61 - C62 - C57	0.6(5)
	10,1 (7)		0.0 (0)

C8—C7—C4—C3	-17.6 (3)	C58—C57—C62—C61	1.0 (5)
C12—C7—C4—C3	164.9 (2)	C54—C57—C62—C61	-177.1 (3)
C3—C4—C5—C6	0.2 (4)	C55—C56—C63—C64	41.2 (4)
C7—C4—C5—C6	-178.8(3)	C51—C56—C63—C64	-139.1(3)
C2-C1-C6-C5	-4.0 (4)	C55—C56—C63—C68	-133.8(3)
N1-C1-C6-C5	-179.7(3)	C51—C56—C63—C68	46.0 (4)
C_{2} C_{1} C_{6} C_{13}	176 1 (3)	C68 - C63 - C64 - C65	01(5)
$N_1 - C_1 - C_6 - C_{13}$	0.4(4)	$C_{56} - C_{63} - C_{64} - C_{65}$	-1751(3)
C4-C5-C6-C1	32(4)	C_{63} C_{64} C_{65} C_{66}	05(6)
C4-C5-C6-C13	-1769(3)	C64 - C65 - C66 - C67	-1.3(7)
C1 - C6 - C13 - C14	1395(3)	C65 - C65 - C67 - C68	1.5(7)
C_{5} C_{6} C_{13} C_{14}	-40.3(4)	C64 - C63 - C68 - C67	1.3(7)
$C_{1} - C_{6} - C_{13} - C_{18}$	-42.4(4)	C_{56} C_{63} C_{68} C_{67}	175 3 (3)
$C_{1} = C_{0} = C_{13} = C_{18}$	+2.+(+)	$C_{50} = C_{65} = C_{68} = C_{67}$	-10(6)
$C_{18} = C_{13} = C_{16} = C_{18}$	-0.5(5)	$N_{2} = C_{0} = C_{0} = C_{0}$	-6.8(6)
$C_{10} - C_{13} - C_{14} - C_{15}$	0.5(3)	$N_{3} - C_{70} - C_{71} - C_{72}$	0.8(0)
$C_{12} = C_{13} = C_{14} = C_{15}$	177.0(5)	$12 \ C70 \ C71 \ C80$	177.2(3)
C13 - C14 - C13 - C10	1.0(3)	$N_{3} = C_{10} = C_{11} = C_{80}$	175.5(3)
C14 - C15 - C16 - C17	-2.1(0)	$C_{81} - C_{70} - C_{71} - C_{80}$	-2.4(3)
C15 - C16 - C17 - C18	1.5 (6)	$C_{80} - C_{1} - C_{2} - C_{3}$	-0.1(5)
C16-C1/-C18-C13	-0.5(6)	C/0 - C/1 - C/2 - C/3	-1/9.7(3)
C14-C13-C18-C17	0.0 (5)	C/1 - C/2 - C/3 - C/4	-0.6(5)
C6—C13—C18—C17	-178.1(3)	C/2—C/3—C/4—C/5	0.3 (6)
N1—C20—C21—C22	-4.0 (6)	C73—C74—C75—C80	0.5 (5)
C31—C20—C21—C22	175.6 (3)	C73—C74—C75—C76	179.1 (4)
N1—C20—C21—C30	179.6 (3)	C80—C75—C76—C77	-0.1(5)
C31—C20—C21—C30	-0.9 (3)	C74—C75—C76—C77	-178.6 (4)
C30—C21—C22—C23	-1.1 (5)	C75—C76—C77—C78	0.0 (6)
C20—C21—C22—C23	-177.2 (3)	C76—C77—C78—C79	0.3 (5)
C21—C22—C23—C24	-0.6 (6)	С77—С78—С79—С80	-0.6 (5)
C22—C23—C24—C25	0.8 (6)	С77—С78—С79—С81	178.6 (3)
C23—C24—C25—C30	0.7 (5)	C74—C75—C80—C79	178.5 (3)
C23—C24—C25—C26	179.1 (4)	C76—C75—C80—C79	-0.3 (5)
C24—C25—C26—C27	-177.8 (4)	C74—C75—C80—C71	-1.2 (5)
C30—C25—C26—C27	0.6 (5)	C76—C75—C80—C71	-179.9 (3)
C25—C26—C27—C28	0.7 (6)	C78—C79—C80—C75	0.6 (5)
C26—C27—C28—C29	-1.3 (6)	C81—C79—C80—C75	-178.8 (3)
C27—C28—C29—C30	0.5 (5)	C78—C79—C80—C71	-179.7 (3)
C27—C28—C29—C31	-179.7 (3)	C81—C79—C80—C71	0.9 (3)
C28—C29—C30—C21	-179.4 (3)	C72—C71—C80—C75	1.0 (5)
C31—C29—C30—C21	0.8 (4)	C70—C71—C80—C75	-179.3 (3)
C28—C29—C30—C25	0.9 (5)	C72—C71—C80—C79	-178.7(3)
C31—C29—C30—C25	-179.0(3)	C70—C71—C80—C79	1.0 (3)
C22—C21—C30—C29	-177.0(3)	C78—C79—C81—N4	-3.9 (6)
C20—C21—C30—C29	0.1 (4)	C80—C79—C81—N4	175.4 (3)
C_{22} C_{21} C_{30} C_{25}	2.7 (5)	C78—C79—C81—C70	178.4 (3)
C_{20} C_{21} C_{30} C_{25}	179.9 (3)	C80—C79—C81—C70	-2.3(3)
C_{24} C_{25} C_{30} C_{29}	177.2 (3)	N3-C70-C81-N4	8.3 (4)
$C_{26} = C_{25} = C_{30} = C_{29}$	-14(5)	C71 - C70 - C81 - N4	-175 1 (3)
		0,1 0,0 001 101	1,2,1 (2)

C24—C25—C30—C21	-2.5 (5)	N3—C70—C81—C79	-173.6(3)
C26—C25—C30—C21	178.9 (3)	C71—C70—C81—C79	3.0 (3)
C28—C29—C31—N2	-0.5 (6)	C87—C82—C83—C84	3.2 (4)
C30-C29-C31-N2	179.3 (3)	N4—C82—C83—C84	-178.7(3)
C28—C29—C31—C20	178.9 (3)	C87—C82—C83—C88	-179.3 (3)
C30—C29—C31—C20	-1.3 (3)	N4—C82—C83—C88	-1.2 (4)
N1—C20—C31—N2	0.4 (4)	C82—C83—C84—C85	-1.1(4)
C21—C20—C31—N2	-179.1 (3)	C88—C83—C84—C85	-178.6(3)
N1—C20—C31—C29	-179.0(3)	C83—C84—C85—C86	-1.5 (4)
C21—C20—C31—C29	1.4 (3)	C83—C84—C85—C94	178.3 (3)
C37—C32—C33—C34	-4.4 (5)	C84—C85—C86—C87	2.1 (5)
N2-C32-C33-C34	-179.9(3)	C94-C85-C86-C87	-177.6(3)
C37—C32—C33—C38	175.4 (3)	C83—C82—C87—C86	-2.7(5)
N2-C32-C33-C38	-0.1(5)	N4—C82—C87—C86	179.2(3)
$C_{32} = C_{33} = C_{34} = C_{35}$	0.2(5)	C83 - C82 - C87 - C100	1799(3)
C_{38} C_{33} C_{34} C_{35}	-179.6(3)	N4-C82-C87-C100	1.7 (5)
C_{33} C_{34} C_{35} C_{36}	37(5)	C85 - C86 - C87 - C82	-0.1(5)
C_{33} C_{34} C_{35} C_{44}	-1765(3)	C85 - C86 - C87 - C100	1774(3)
$C_{34} = C_{35} = C_{36} = C_{37}$	-38(5)	C84 - C83 - C88 - C93	40 2 (4)
$C_{44} = C_{35} = C_{36} = C_{37}$	1764(3)	C82 - C83 - C88 - C93	-1371(3)
C_{35} C_{36} C_{37} C_{37} C_{32}	-0.1(5)	C84 - C83 - C88 - C89	-137.4(3)
$C_{35} = C_{36} = C_{37} = C_{50}$	-1787(3)	C82 - C83 - C88 - C89	45 2 (4)
C_{33} C_{32} C_{37} C_{36}	43(5)	$C_{02} = C_{03} = C$	-0.2(5)
N_{2} C_{32} C_{37} C_{36}	179 9 (3)	C_{83} C_{88} C_{89} C_{90}	177.6(3)
C_{33} C_{32} C_{37} C_{50}	-1771(3)	$C_{88} = C_{89} = C_{90} = C_{91}$	0.1(5)
$N_{2} = C_{32} = C_{37} = C_{50}$	-1.5(5)	$C_{89} = C_{90} = C_{90} = C_{91} = C_{92}$	-0.4(6)
C_{34} C_{33} C_{38} C_{43}	-423(4)	$C_{00} = C_{01} = C_{01} = C_{02}$	0.4(0)
$C_{34} = C_{33} = C_{36} = C_{43}$	42.3(4)	$C_{90} = C_{91} = C_{92} = C_{93}$	0.3(0)
$C_{32} = C_{33} = C_{30} = C_{43}$	137.9(3) 134.9(3)	$C_{83} C_{88} C_{93} C_{92}$	-177.2(3)
C_{32} C_{33} C_{38} C_{39}	-44.9(3)	$C_{33} = C_{33} = C_{33} = C_{32}$	-0.9(5)
$C_{32} = C_{33} = C_{30} = C_{30}$	(+)	$C_{2}^{86} = C_{2}^{85} = C_{2}^{86} = C_{$	145 4 (3)
$C_{45} = C_{58} = C_{59} = C_{40}$	-176.6(3)	$C_{80} = C_{85} = C_{94} = C_{95}$	-344(3)
$C_{35} = C_{36} = C_{39} = C_{40}$	170.0(3)	$C_{84} = C_{85} = C_{94} = C_{95}$	-31.5(4)
$C_{38} = C_{39} = C_{40} = C_{41}$	-0.6(5)	$C_{80} = C_{85} = C_{94} = C_{99}$	$1/8 \ 8 \ (3)$
$C_{40} = C_{40} = C_{41} = C_{42}$	-0.2(5)	$C_{00} = C_{01} = C$	-10(5)
$C_{+0} = C_{+1} = C_{+2} = C_{+3}$	-1.4(5)	$C_{33} = C_{34} = C_{35} = C_{36}$	-1788(3)
$C_{33} = C_{38} = C_{43} = C_{42}$	1.4(3) 175 9 (3)	$C_{85} - C_{94} - C_{95} - C_{90}$	178.8(3) 12(5)
$C_{33} = C_{38} = C_{43} = C_{42}$	173.9(3) 1 2 (5)	$C_{94} C_{95} C_{90} C_{90} C_{97} $	1.2(3)
$C_{41} - C_{42} - C_{43} - C_{38}$	-147.0(3)	$C_{95} - C_{90} - C_{97} - C_{98}$	-0.8(6)
$C_{34} = C_{35} = C_{44} = C_{49}$	147.5(3)	$C_{90} C_{97} C_{98} C_{99} $	0.8(0)
$C_{30} = C_{33} = C_{44} = C_{49}$	32.0(5)	$C_{95} = C_{94} = C_{99} = C_{98}$	1.2(3)
$C_{34} = C_{35} = C_{44} = C_{45}$	52.9(3) -147.2(4)	$C_{00} = C_{00} = C_{00} = C_{00}$	1/0.1(3)
$C_{30} = C_{33} = C_{44} = C_{43}$	-147.3(4)	$C_{97} - C_{98} - C_{99} - C_{94}$	0.2(3)
$C_{43} = C_{44} = C_{43} = C_{40}$	1.2(0) 178 1 (3)	$C_{21} = C_{20} = N_1 = C_1$	2.1(3) 178 5 (3)
$C_{44} = C_{45} = C_{46} = C_{40}$	1/0.1(3)	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	1/0.3(3) 1031(3)
$C_{44} = C_{43} = C_{40} = C_{47} = C_{49}$	-1.2(7)	$C_2 = C_1 = N_1 = C_2 U$	-811(3)
C43 - C40 - C47 - C48	-1.2(7)	$C_0 = C_1 = N_1 = C_2 0$	-81.1(4)
(40 - (4) - (48 - (49))	1.3 (/)	$C_{29} = C_{31} = N_2 = C_{32}$	-4.2(3)
C43-C44-C49-C48	1.3 (6)	C20—C31—N2—C32	1/0.4 (3)

C35—C44—C49—C48	-178.0 (3)	C37—C32—N2—C31	100.5 (4)
C47—C48—C49—C44	-1.4 (6)	C33—C32—N2—C31	-83.9 (4)
C56—C51—C52—C53	-0.5 (4)	C71—C70—N3—C51	2.7 (5)
N3—C51—C52—C53	-176.4 (3)	C81—C70—N3—C51	178.3 (3)
C56—C51—C52—C69	178.2 (3)	C52—C51—N3—C70	-97.4 (3)
N3—C51—C52—C69 C51—C52—C53—C54	2.3 (4) 0.1 (5)	C56—C51—N3—C70 C79—C81—N4—C82	86.7 (4) 2.4 (5) 170 0 (2)
C52—C53—C54—C55 C52—C53—C54—C57	-178.0 (3) 0.4 (5) -180.0 (3)	C70-C81-N4-C82 C87-C82-N4-C81 C83-C82-N4-C81	-95.6 (3) 86.3 (4)

Hydrogen-bond geometry (Å, °)

Cg2 is the centroid of ring C1–C6 in molecule A.

D—H···A	D—H	Н…А	$D \cdots A$	D—H···A
C16—H16…N3 ⁱ	0.93	2.60	3.517 (6)	168
С92—Н92…Сд2 ^{іі}	0.93	2.95	3.789 (5)	151

Symmetry codes: (i) -*x*+1, -*y*+1, -*z*+1; (ii) *x*+1, *y*, *z*-1.