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Kai-Xiao Li<sup>a</sup> and Xiao-Juan Wang<sup>b</sup>\*

<sup>a</sup>College of Physical Education and Health Sciences, Zhejiang Normal University, Jinhua, Zhejiang 321004, People's Republic of China, and <sup>b</sup>College of Chemistry and Life Science, Zhejiang Normal University, Jinhua, Zhejiang 321004, People's Republic of China. \*Correspondence e-mail: wangxj@zjnu.cn

The title compound,  $C_{19}H_{29}BO_3$ , was prepared by the reaction of 1-(hex-1-yn-1-yloxy)-3-methylbenzene in tetrahydrofuran with 4,4,5,5-tetramethyl-1,3,2-dioxaborolane under a nitrogen atmosphere. In the molecule, the butyl group adopts an extended conformation, with a torsion angle of 179.52 (19)°. The dioxaborolane ring has a twisted conformation on the C–C bond, and its mean plane is inclined to the 3-methylphenyl ring by 44.79 (19)°. In the crystal, there are no significant intermolecular interactions present.



## **Structure description**

The title compound was previously prepared by Cui *et al.* (2013), and its derivatives were synthesized by Yang & Cheng (2001) and Whelligan *et al.* (2010). We present here the crystal structure of the compound (Fig. 1). In the molecule, the conformation of the dioxaborolane ring is similar to half chair, and the butyl group adopts an extended conformation with a C10-C11-C12-C13 torsion angle =  $179.52 (19)^{\circ}$ . No hydrogen bonding is observed in the crystal.

Synthesis and crystallization

To a solution of 1-(hex-1-yn-1-yloxy)-3-methylbenzene (94 mg, 0.5 mmol) in 2.0 ml of tetrahydrofuran (THF) was added neat 4,4,5,5-tetramethyl-1,3,2-dioxaborolane (HBpin) (71 mg, 0.55 mmol) under nitrogen atmosphere. After stirring at room temperature for 1.5 h, the reaction mixture was concentrated and purified by on silica gel (petroleum ether/EtOAc = 60/1).



Edited by D.-J. Xu, Zhejiang University (Yuquan Campus), China

**IUCrData** 

**Keywords:** crystal structure; dioxaborolane; extended conformation.

CCDC reference: 1473545

Structural data: full structural data are available from iucrdata.iucr.org





#### Figure 1

Perspective view of the structure of the title compound, showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

### Refinement

Crystal data, data collection and structure refinement details are summarized in Table 1.

### **Acknowledgements**

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Table 1		
Experime	ental	details.

Crystal data C19H29BO3 Chemical formula 316.23  $M_{\rm r}$ Crystal system, space group Monoclinic,  $P2_1/c$ Temperature (K) 296 10.0293 (9), 18.1230 (17), a, b, c (Å) 10.7973 (10) 97.475 (6)  $\beta$  (°  $V(Å^3)$ 1945.9 (3) Ζ 4 Radiation type Μο Κα  $\mu \,({\rm mm}^{-1})$ 0.07 Crystal size (mm)  $0.3 \times 0.2 \times 0.1$ Data collection Bruker APEXII Diffractometer Absorption correction Multi-scan (SADABS; Sheldrick, 1996) 0.983, 0.993  $T_{\min}, T_{\max}$ No. of measured, independent and 30133, 4375, 2665 observed  $[I > 2\sigma(I)]$  reflections Rint 0.044  $(\sin \theta / \lambda)_{max} (\text{\AA}^{-1})$ 0.648 Refinement  $R[F^2 > 2\sigma(F^2)], wR(F^2), S$ 0.052, 0.168, 1.09 No. of reflections 4375 No. of parameters 208 H-atom treatment H-atom parameters constrained  $\Delta \rho_{\rm max}, \, \Delta \rho_{\rm min} \ ({\rm e} \ {\rm \AA}^{-3})$ 0.18, -0.18

Computer programs: APEX2 and SAINT (Bruker, 2006), SHELXS97, SHELXL97 and SHELXTL (Sheldrick, 2008).

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# full crystallographic data

IUCrData (2016). 1, x160611 [doi:10.1107/S2414314616006118]

4,4,5,5-Tetramethyl-2-[(*Z*)-1-(3-methylphenoxy)hex-1-en-2-yl]-1,3,2dioxaborolane

F(000) = 688

 $\theta = 2.1 - 27.4^{\circ}$  $\mu = 0.07 \text{ mm}^{-1}$ 

Block, colourless

 $0.3 \times 0.2 \times 0.1 \text{ mm}$ 

T = 296 K

 $D_{\rm x} = 1.079 {\rm Mg m^{-3}}$ 

Mo *K* $\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 4375 reflections

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4,4,5,5-Tetramethyl-2-[(Z)-1-(3-methylphenoxy)hex-1-en-2-yl]-1,3,2-dioxaborolane

## Crystal data

C<sub>19</sub>H<sub>29</sub>BO<sub>3</sub>  $M_r = 316.23$ Monoclinic,  $P2_1/c$  a = 10.0293 (9) Å b = 18.1230 (17) Å c = 10.7973 (10) Å  $\beta = 97.475$  (6)° V = 1945.9 (3) Å<sup>3</sup> Z = 4

## Data collection

Bruker APEXII 30133 measured reflections 4375 independent reflections diffractometer Radiation source: fine-focus sealed tube 2665 reflections with  $I > 2\sigma(I)$ Graphite monochromator  $R_{\rm int} = 0.044$  $\omega$  scans  $\theta_{\rm max} = 27.4^{\circ}, \ \theta_{\rm min} = 2.1^{\circ}$  $h = -13 \rightarrow 12$ Absorption correction: multi-scan  $k = -23 \rightarrow 21$ (SADABS: Sheldrick, 1996)  $l = -13 \rightarrow 13$  $T_{\rm min} = 0.983, T_{\rm max} = 0.993$ 

# Refinement

Refinement on  $F^2$ Secondary atom site location: difference Fourier Least-squares matrix: full map  $R[F^2 > 2\sigma(F^2)] = 0.052$ Hydrogen site location: inferred from  $wR(F^2) = 0.168$ neighbouring sites S = 1.09H-atom parameters constrained 4375 reflections  $w = 1/[\sigma^2(F_0^2) + (0.0826P)^2 + 0.1148P]$ 208 parameters where  $P = (F_0^2 + 2F_c^2)/3$ 0 restraints  $(\Delta/\sigma)_{\rm max} < 0.001$ Primary atom site location: structure-invariant  $\Delta \rho_{\rm max} = 0.18 \ {\rm e} \ {\rm \AA}^{-3}$ direct methods  $\Delta \rho_{\rm min} = -0.18 \ {\rm e} \ {\rm \AA}^{-3}$ 

# Special details

**Geometry**. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement**. Refinement of  $F^2$  against ALL reflections. The weighted R-factor wR and goodness of fit S are based on  $F^2$ , conventional R-factors R are based on F, with F set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2$ sigma( $F^2$ ) is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

	x	v	Z	$U_{\rm iso}^*/U_{\rm eq}$	
 B1	0.66602 (17)	0 11935 (10)	0 19395 (16)	0.0479 (4)	
01	0.49825(11)	0 27876 (7)	0.01441(12)	0.0662(4)	
02	0.19022(11) 0.56737(11)	0.06699 (6)	0.18524(11)	0.0622(1) 0.0620(4)	
03	0.78374(10)	0.09596 (6)	0.25964(11)	0.0573(3)	
C1	0.36387(17)	0.29473 (9)	-0.02574(16)	0.0558(4)	
C2	0.33922(18)	0.33782 (9)	-0.13101(16)	0.0587(4)	
H2A	0.4104	0.3531	-0.1721	0.070*	
C3	0.2087(2)	0.35875(10)	-0.17633(18)	0.0664(5)	
C4	0.1053(2)	0.33435 (12)	-0.1136(2)	0.0820(7)	
H4A	0.0170	0.3466	-0.1438	0.098*	
C5	0.1308(2)	0.29256 (12)	-0.0081(3)	0.0882(7)	
H5A	0.0597	0.2772	0.0331	0.106*	
C6	0.26130 (19)	0.27263 (11)	0.0386 (2)	0.0732(5)	
H6A	0.2789	0.2450	0.1115	0.088*	
C7	0.1814(3)	0.40687 (14)	-0.2895(2)	0.1007 (8)	
H7C	0.0864	0.4152	-0.3082	0.121*	
H7B	0.2137	0.3830	-0.3592	0.121*	
H7A	0.2266	0.4533	-0.2738	0.121*	
C8	0.52791 (17)	0.21201 (9)	0.07200 (15)	0.0548 (4)	
H8A	0.4622	0.1755	0.0649	0.066*	
C9	0.64616 (15)	0.19761 (9)	0.13719 (14)	0.0493 (4)	
C10	0.75572 (16)	0.25522 (10)	0.15952 (15)	0.0563 (4)	
H10B	0.8418	0.2317	0.1547	0.068*	
H10A	0.7419	0.2917	0.0933	0.068*	
C11	0.76153 (18)	0.29400 (10)	0.28296 (16)	0.0632 (5)	
H11B	0.7786	0.2579	0.3494	0.076*	
H11A	0.6747	0.3163	0.2891	0.076*	
C12	0.8690 (2)	0.35332 (11)	0.30207 (18)	0.0763 (6)	
H12B	0.9557	0.3308	0.2964	0.092*	
H12A	0.8523	0.3890	0.2349	0.092*	
C13	0.8758 (3)	0.39323 (13)	0.4244 (2)	0.1006 (8)	
H13C	0.9454	0.4299	0.4295	0.121*	
H13B	0.8954	0.3587	0.4917	0.121*	
H13A	0.7911	0.4167	0.4304	0.121*	
C14	0.62832 (19)	-0.00170 (9)	0.23442 (18)	0.0635 (5)	
C15	0.75651 (16)	0.02563 (10)	0.31787 (16)	0.0565 (4)	
C16	0.6612 (3)	-0.04561 (13)	0.1223 (2)	0.1110 (9)	
H16C	0.5792	-0.0618	0.0741	0.133*	
H16B	0.7147	-0.0877	0.1505	0.133*	
H16A	0.7103	-0.0150	0.0715	0.133*	

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(\mathring{A}^2)$ 

C17	0.5277 (2)	-0.04152 (15)	0.3024 (3)	0.1157 (10)	
H17C	0.4526	-0.0567	0.2436	0.139*	
H17B	0.4971	-0.0091	0.3630	0.139*	
H17A	0.5694	-0.0841	0.3438	0.139*	
C18	0.8788 (2)	-0.02308 (12)	0.3206 (2)	0.0831 (6)	
H18C	0.8927	-0.0352	0.2367	0.100*	
H18B	0.8653	-0.0675	0.3656	0.100*	
H18A	0.9562	0.0025	0.3612	0.100*	
C19	0.7334 (3)	0.04403 (14)	0.45048 (18)	0.0917 (7)	
H19C	0.6552	0.0747	0.4487	0.110*	
H19B	0.8103	0.0697	0.4918	0.110*	
H19A	0.7201	-0.0007	0.4948	0.110*	

Atomic displacement parameters  $(Å^2)$ 

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
B1	0.0444 (10)	0.0490 (10)	0.0490 (9)	-0.0016 (8)	0.0007 (8)	-0.0027 (8)
O1	0.0523 (7)	0.0530 (7)	0.0881 (9)	-0.0010 (5)	-0.0104 (6)	0.0185 (6)
O2	0.0509 (7)	0.0507 (7)	0.0791 (8)	-0.0044 (5)	-0.0114 (6)	0.0111 (6)
O3	0.0468 (6)	0.0527 (7)	0.0689 (7)	-0.0018 (5)	-0.0065 (5)	0.0086 (6)
C1	0.0504 (10)	0.0473 (9)	0.0661 (10)	0.0005 (7)	-0.0056 (8)	-0.0005 (8)
C2	0.0585 (11)	0.0536 (10)	0.0615 (10)	0.0058 (8)	-0.0012 (8)	0.0016 (8)
C3	0.0666 (12)	0.0567 (11)	0.0705 (11)	0.0123 (9)	-0.0117 (9)	-0.0080 (9)
C4	0.0536 (12)	0.0669 (13)	0.1195 (18)	0.0088 (10)	-0.0116 (12)	-0.0070 (13)
C5	0.0605 (13)	0.0714 (14)	0.136 (2)	0.0021 (11)	0.0232 (12)	0.0120 (14)
C6	0.0678 (12)	0.0633 (12)	0.0886 (14)	0.0002 (10)	0.0104 (10)	0.0164 (10)
C7	0.1099 (18)	0.0981 (18)	0.0862 (15)	0.0364 (15)	-0.0170 (13)	0.0121 (13)
C8	0.0571 (10)	0.0424 (9)	0.0620 (10)	-0.0049 (7)	-0.0037 (8)	0.0053 (7)
C9	0.0471 (9)	0.0484 (9)	0.0506 (9)	-0.0010 (7)	-0.0004 (7)	0.0011 (7)
C10	0.0530 (10)	0.0553 (10)	0.0593 (10)	-0.0056 (8)	0.0020(7)	0.0041 (8)
C11	0.0690 (12)	0.0538 (10)	0.0662 (11)	-0.0074 (8)	0.0065 (9)	-0.0014 (8)
C12	0.0868 (14)	0.0681 (13)	0.0716 (12)	-0.0258 (11)	0.0013 (10)	-0.0003 (10)
C13	0.136 (2)	0.0864 (16)	0.0765 (14)	-0.0390 (15)	0.0029 (14)	-0.0076 (12)
C14	0.0657 (11)	0.0445 (9)	0.0749 (12)	-0.0028 (8)	-0.0108 (9)	0.0092 (8)
C15	0.0546 (10)	0.0541 (10)	0.0586 (9)	0.0042 (8)	-0.0011 (7)	0.0093 (8)
C16	0.160 (2)	0.0626 (14)	0.0960 (17)	0.0129 (15)	-0.0365 (16)	-0.0184 (12)
C17	0.0818 (16)	0.0952 (18)	0.164 (3)	-0.0247 (14)	-0.0082 (16)	0.0607 (18)
C18	0.0690 (13)	0.0736 (14)	0.1030 (16)	0.0160 (11)	-0.0028 (11)	0.0132 (12)
C19	0.1162 (18)	0.0980 (17)	0.0586 (12)	0.0114 (14)	0.0023 (11)	0.0079 (11)

# Geometric parameters (Å, °)

B1—O3	1.363 (2)	C11—C12	1.517 (2)	
B1—O2	1.365 (2)	C11—H11B	0.9700	
B1—C9	1.548 (2)	C11—H11A	0.9700	
O1—C8	1.3749 (19)	C12—C13	1.500 (3)	
01—C1	1.3914 (19)	C12—H12B	0.9700	
O2—C14	1.456 (2)	C12—H12A	0.9700	

O3—C15	1.463 (2)	C13—H13C	0.9600
C1—C6	1.374 (3)	C13—H13B	0.9600
C1—C2	1.375 (2)	C13—H13A	0.9600
C2—C3	1.390 (2)	C14—C17	1.506 (3)
C2—H2A	0.9300	C14—C16	1.520 (3)
C3—C4	1.383 (3)	C14—C15	1.552 (2)
C3—C7	1.498 (3)	C15—C18	1.508(2)
C4—C5	1 364 (3)	$C_{15} - C_{19}$	1517(3)
C4—H4A	0.9300	C16—H16C	0.9600
C5-C6	1 388 (3)	C16—H16B	0.9600
C5—H5A	0.9300	C16—H16A	0.9600
C6—H6A	0.9300	C17—H17C	0.9600
C7_H7C	0.9600	C17H17B	0.9600
C7H7B	0.9600	C17H17A	0.9600
	0.9600	$C_{18}$ $H_{18}C$	0.9600
$C_{1}^{2}$	1,324(2)	C18 H18B	0.9000
	1.324(2)		0.9000
$C_0 = C_{10}$	1.512(2)		0.9000
$C_{10}$	1.512(2) 1.501(2)	C10 H10P	0.9000
C10_U10P	1.301(2)	C10 H10A	0.9000
	0.9700	С19—п19А	0.9000
CI0—HI0A	0.9700		
03 B1 02	112 00 (15)	C13 C12 C11	11/ 32 (18)
$O_3 = B_1 = O_2$	112.99(15) 123.30(15)	$C_{13} = C_{12} = C_{11}$	108 7
$O_3 = B_1 = C_9$	123.30(13) 122.70(14)	C13 - C12 - H12B	108.7
02-B1-C9	125.70(14) 117.80(12)	C12 - C12 - H12B	108.7
	117.69(13)	C13 - C12 - D12A	108.7
B1 = 02 = C14	107.04(12) 107.28(12)	CII - CI2 - HI2A	108.7
	107.28 (12)	H12B - C12 - H12A	107.6
$C_{6}$	121.19 (17)	C12 - C13 - H13C	109.5
	122.72 (16)		109.5
C2-C1-01	116.00 (15)	HI3C—CI3—HI3B	109.5
C1—C2—C3	120.40 (18)	C12—C13—H13A	109.5
C1—C2—H2A	119.8	H13C—C13—H13A	109.5
C3—C2—H2A	119.8	H13B—C13—H13A	109.5
C4—C3—C2	118.19 (18)	O2—C14—C17	108.13 (16)
C4—C3—C7	121.14 (19)	O2—C14—C16	106.33 (15)
C2—C3—C7	120.7 (2)	C17—C14—C16	111.4 (2)
C5—C4—C3	121.04 (19)	O2—C14—C15	102.45 (13)
С5—С4—Н4А	119.5	C17—C14—C15	115.27 (17)
C3—C4—H4A	119.5	C16—C14—C15	112.36 (18)
C4—C5—C6	120.9 (2)	O3—C15—C18	108.75 (14)
C4—C5—H5A	119.6	O3—C15—C19	105.91 (15)
С6—С5—Н5А	119.6	C18—C15—C19	109.50 (16)
C1—C6—C5	118.26 (19)	O3—C15—C14	102.45 (12)
С1—С6—Н6А	120.9	C18—C15—C14	115.77 (16)
С5—С6—Н6А	120.9	C19—C15—C14	113.67 (16)
С3—С7—Н7С	109.5	C14—C16—H16C	109.5
С3—С7—Н7В	109.5	C14—C16—H16B	109.5

	100 5		100 -
H'/C - C' - H'/B	109.5	H16C-C16-H16B	109.5
С3—С7—Н7А	109.5	C14—C16—H16A	109.5
Н7С—С7—Н7А	109.5	H16C—C16—H16A	109.5
H7B—C7—H7A	109.5	H16B—C16—H16A	109.5
C9—C8—O1	122.69 (15)	C14—C17—H17C	109.5
С9—С8—Н8А	118.7	C14—C17—H17B	109.5
O1—C8—H8A	118.7	H17C—C17—H17B	109.5
C8—C9—C10	122.06 (15)	C14—C17—H17A	109.5
C8—C9—B1	116.84 (14)	H17C—C17—H17A	109.5
C10—C9—B1	121.07 (13)	H17B—C17—H17A	109.5
C11—C10—C9	114.10 (14)	C15—C18—H18C	109.5
C11—C10—H10B	108.7	C15—C18—H18B	109.5
C9—C10—H10B	108.7	H18C—C18—H18B	109.5
C11—C10—H10A	108.7	C15—C18—H18A	109.5
C9—C10—H10A	108.7	H18C—C18—H18A	109.5
H10B-C10-H10A	107.6	H18B—C18—H18A	109.5
C10—C11—C12	113.41 (15)	С15—С19—Н19С	109.5
C10—C11—H11B	108.9	C15—C19—H19B	109.5
C12—C11—H11B	108.9	H19C—C19—H19B	109.5
C10—C11—H11A	108.9	С15—С19—Н19А	109.5
C12—C11—H11A	108.9	H19C—C19—H19A	109.5
H11B—C11—H11A	107.7	H19B—C19—H19A	109.5
03 - B1 - 02 - C14	-847(19)	O3 - B1 - C9 - C10	-43(2)
C9-B1-O2-C14	172.46 (15)	$\Omega^2 = B1 = C9 = C10$	1.3(2) 174 71 (15)
02-B1-O3-C15	-9.78(18)	C8 - C9 - C10 - C11	96 8 (2)
C9-B1-O3-C15	169 30 (15)	B1-C9-C10-C11	-81.07(19)
$C_{8} = 01 = C_{1} = C_{6}$	35 4 (2)	C9-C10-C11-C12	-178 12 (16)
$C_{8} = 01 = C_{1} = C_{2}$	-148.01.(16)	$C_{10}$ $C_{11}$ $C_{12}$ $C_{13}$	179 52 (19)
C6-C1-C2-C3	-14(3)	B1 - O2 - C14 - C17	143.32(19)
01 - C1 - C2 - C3	-178.00(15)	B1 = O2 = C14 = C16	-9649(19)
C1 - C2 - C3 - C4	-0.7(3)	B1 = O2 = C14 = C15	21 59 (17)
C1 - C2 - C3 - C7	178 83 (18)	B1 = 02 = C14 = C15 B1 = 03 = C15 = C18	145 33 (15)
$C_1 = C_2 = C_3 = C_7$	18(3)	B1 = 03 = C15 = C18	-97.07(16)
$C_2 - C_3 - C_4 - C_5$	-177.8(2)	B1 = 03 = C15 = C14	22.30(17)
$C_{1}^{2} = C_{2}^{2} = C_{1}^{2} = C_{2}^{2}$	-0.7(3)	$D_1 = 05 = 015 = 014$	-26.26(17)
$C_{2} = C_{1} = C_{2} = C_{0}$	24(3)	$C_{17} = C_{14} = C_{15} = 0.03$	-1/3/3 (17)
$C_2 - C_1 - C_0 - C_3$	2.7(3)	$C_{1}^{-} = C_{1}^{-} = C_{1$	1+3.+3(17) 97 46 (19)
$C_{1} = C_{1} = C_{0} = C_{3}$	1/0.01(10) -1 4 (2)	$C_{10} - C_{14} - C_{15} - C_{18}$	67.40(10) -144.42(16)
$C_{4} = C_{3} = C_{0} = C_{1}$	-1.4(3)	02 - 014 - 015 - 018	-144.43(10)
C1 = C1 = C3 = C9	-104.93(10)	C1/-C14-C15-C18	98.4 (2)
01 - 08 - 09 - 010	3.3(3)	10 - 14 - 15 - 18	-30.7(2)
$O_1 - O_2 - O_3 - B_1$	-1/8.55(15)	02 - 014 - 015 - 019	87.33 (18)
03 - B1 - C9 - C8	1//./3(13)	C1/-C14-C15-C19	-29.6 (2)
02—B1—C9—C8	-5.3 (2)	C16—C14—C15—C19	-158.76 (18)