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Trimethyl 3,3',3''-(benzene-1,3,5-triyl)tripropyn- oate

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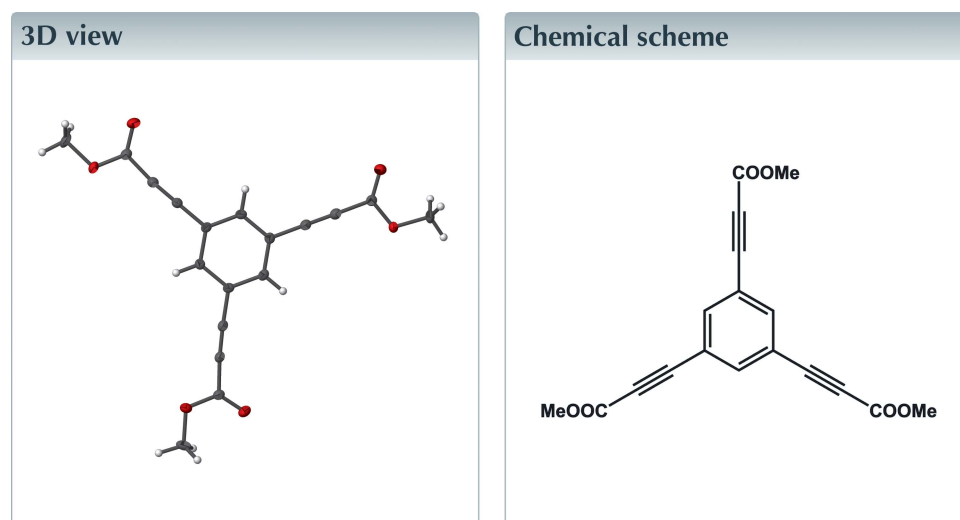
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Keywords: crystal structure; trimesic acid homologue; C—H···O hydrogen bond; C—H··· π contact.

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Structural data: full structural data are available from iucrdata.iucr.org

In the title compound, C₁₈H₁₂O₆, the alkyne bonds are distorted, featuring bond angles around the C—C≡C—C group of 173.6 (1)/179.0 (1), 178.1 (1)/178.4 (1) and 174.9 (1)/175.9 (1)°, and the ester groups make angles of 3.5 (1), 13.8 (1) and 14.5 (1)° with the central benzene ring. In the crystal, molecules are connected in layers parallel to (131) by weak C—H···O hydrogen bonds, giving rise to a system of hydrogen-bonded ring motifs with graph sets R₂²(14) and R₄⁴(22). The layers are linked by C—H···O and C—H··· π contacts.



Structure description

The title compound, C₁₈H₁₂O₆, is an interesting synthetic intermediate for the preparation of application-oriented solid materials involving both porous coordination polymers (MacGillivray, 2010) or metal-organic frameworks (Noro & Kitagawa, 2010) and crystalline inclusion hosts (Weber, 1996; Katzsch *et al.*, 2016).

In the structure of the molecule (Fig. 1), the alkyne bonds are distorted which is shown by the corresponding bond angles [C2—C3—C4 = 173.6 (1), C3—C4—C13 = 179.0 (1), C6—C7—C8 = 178.1 (1), C7—C8—C15 = 178.4 (1), C10—C11—C12 = 174.9 (1), C11—C12—C17 = 175.9 (1)°] and the ester functions are not arranged in the benzene plane [interplanar angles: 3.5 (1) (C2, O1, O2), 13.8 (1) (C6, O3, O4) and 14.5 (1)° (C10, O5, O10)].

In the crystal, molecules are connected in layers parallel to (131) by weak C—H···O hydrogen bonds (Desiraju & Steiner, 1999) (Table 1) giving rise to hydrogen-bonded ring motifs with graph sets R₂²(14) and R₄⁴(22) (Fig. 2). The layers are linked by weak C—H···O contacts and additionally by C—H··· π interactions.

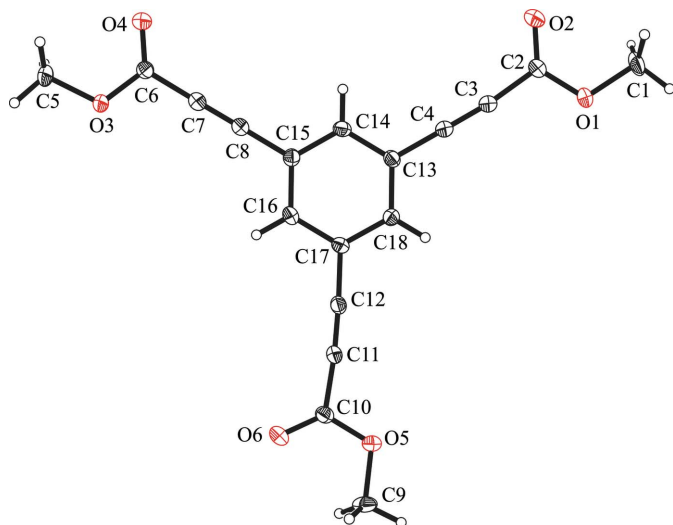


Figure 1
A view of the molecular structure of the title compound, with atom labelling. Displacement ellipsoids are drawn at the 50% probability level.

Synthesis and crystallization

The title compound was prepared from 1,3,5-triethynylbenzene (Münch *et al.*, 2013) and methyl chloroformate as described in the literature (Katzsch *et al.*, 2016). Colorless single crystals of prismatic shape suitable for X-ray diffraction were obtained by slow crystallization from a solvent mixture of acetone, ethyl acetate and *n*-hexane. For the synthesis of a related compound, see: Welti & Diederich (2003).

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

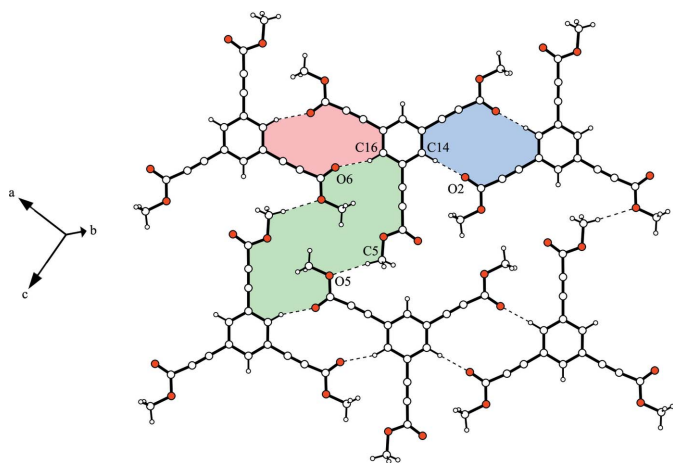


Figure 2
A partial view of the crystal packing of the title compound, showing the formation of the C—H···O bonded layer structure, enclosing the system of $R_2^2(14)$ and $R_4^4(22)$ ring motifs. Hydrogen bonds are indicated by dashed lines.

Table 1
Hydrogen-bond geometry (Å, °).

Cg1 is the centroid of the C13–C18 ring and Cg3 is the mid-point of atoms C7 and C8.

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
C1—H1C···O6 ⁱ	0.98	2.70	3.4378 (15)	133
C5—H5B···O5 ⁱⁱ	0.98	2.66	3.4517 (15)	138
C14—H14···O2 ⁱⁱⁱ	0.95	2.31	3.2278 (13)	162
C16—H16···O6 ^{iv}	0.95	2.35	3.2390 (13)	155
C1—H1A···Cg3 ^v	0.98	2.85	3.5506 (13)	130
C9—H9A···Cg1 ^{vi}	0.98	2.76	3.6302 (13)	148

Symmetry codes: (i) $x - 1, y + 1, z - 1$; (ii) $x - 1, y, z + 1$; (iii) $-x + 1, -y + 1, -z + 1$; (iv) $-x + 3, -y, -z + 2$; (v) $x, y, z - 1$; (vi) $x + 1, y, z$.

Table 2
Experimental details.

Crystal data	
Chemical formula	$C_{18}H_{12}O_6$
M_r	324.28
Crystal system, space group	Triclinic, $P\bar{1}$
Temperature (K)	100
<i>a</i> , <i>b</i> , <i>c</i> (Å)	8.5765 (2), 9.8469 (2), 10.2677 (2)
α , β , γ (°)	78.903 (1), 79.552 (1), 68.655 (1)
<i>V</i> (Å ³)	786.65 (3)
<i>Z</i>	2
Radiation type	Mo $K\alpha$
μ (mm ⁻¹)	0.10
Crystal size (mm)	0.54 × 0.43 × 0.37
Data collection	
Diffractometer	Bruker APEXII CCD
Absorption correction	Multi-scan (<i>SADABS</i> ; Sheldrick, 2004)
T_{min} , T_{max}	0.946, 0.963
No. of measured, independent and observed [$I > 2\sigma(I)$] reflections	16379, 2769, 2558
R_{int}	0.022
($\sin \theta/\lambda$) _{max} (Å ⁻¹)	0.595
Refinement	
$R[F^2 > 2\sigma(F^2)]$, $wR(F^2)$, <i>S</i>	0.030, 0.082, 1.08
No. of reflections	2769
No. of parameters	220
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{max}$, $\Delta\rho_{min}$ (e Å ⁻³)	0.15, -0.24

Computer programs: *APEX2* and *SAINT* (Bruker, 2008), *SHELXS97*, *SHELXL97* and *SHELXTL* (Sheldrick, 2008).

Acknowledgements

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full crystallographic data

IUCrData (2016). **1**, x160629 [doi:10.1107/S2414314616006295]

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Crystal data

$C_{18}H_{12}O_6$	$Z = 2$
$M_r = 324.28$	$F(000) = 336$
Triclinic, $P\bar{1}$	$D_x = 1.369 \text{ Mg m}^{-3}$
Hall symbol: $-P 1$	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
$a = 8.5765 (2) \text{ \AA}$	Cell parameters from 9912 reflections
$b = 9.8469 (2) \text{ \AA}$	$\theta = 2.8\text{--}35.9^\circ$
$c = 10.2677 (2) \text{ \AA}$	$\mu = 0.10 \text{ mm}^{-1}$
$\alpha = 78.903 (1)^\circ$	$T = 100 \text{ K}$
$\beta = 79.552 (1)^\circ$	Prism, colourless
$\gamma = 68.655 (1)^\circ$	$0.54 \times 0.43 \times 0.37 \text{ mm}$
$V = 786.65 (3) \text{ \AA}^3$	

Data collection

Bruker APEXII CCD diffractometer	16379 measured reflections
Radiation source: fine-focus sealed tube	2769 independent reflections
Graphite monochromator	2558 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{\text{int}} = 0.022$
Absorption correction: multi-scan (<i>SADABS</i> ; Sheldrick, 2004)	$\theta_{\text{max}} = 25.0^\circ$, $\theta_{\text{min}} = 2.0^\circ$
$T_{\text{min}} = 0.946$, $T_{\text{max}} = 0.963$	$h = -10 \rightarrow 10$
	$k = -11 \rightarrow 11$
	$l = -12 \rightarrow 12$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.030$	H-atom parameters constrained
$wR(F^2) = 0.082$	$w = 1/[\sigma^2(F_o^2) + (0.0423P)^2 + 0.2206P]$
$S = 1.08$	where $P = (F_o^2 + 2F_c^2)/3$
2769 reflections	$(\Delta/\sigma)_{\text{max}} = 0.001$
220 parameters	$\Delta\rho_{\text{max}} = 0.15 \text{ e \AA}^{-3}$
0 restraints	$\Delta\rho_{\text{min}} = -0.24 \text{ e \AA}^{-3}$
Primary atom site location: structure-invariant direct methods	

Special details

Geometry. All s.u.'s (except the s.u. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R-factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.83566 (10)	0.52293 (9)	0.16775 (8)	0.0236 (2)
O2	0.57685 (10)	0.57849 (10)	0.28339 (8)	0.0293 (2)
O3	0.65384 (9)	0.22546 (9)	1.31248 (7)	0.02084 (19)
O4	0.41386 (10)	0.31372 (9)	1.21571 (8)	0.0268 (2)
O5	1.77215 (9)	-0.03869 (9)	0.67379 (8)	0.0236 (2)
O6	1.70937 (10)	-0.04808 (10)	0.89660 (8)	0.0299 (2)
C1	0.75769 (16)	0.59765 (14)	0.04628 (11)	0.0278 (3)
H1A	0.6774	0.5528	0.0338	0.042*
H1B	0.8452	0.5882	-0.0308	0.042*
H1C	0.6982	0.7021	0.0544	0.042*
C2	0.72764 (14)	0.52404 (12)	0.27834 (11)	0.0190 (2)
C3	0.81227 (13)	0.45041 (12)	0.39566 (11)	0.0194 (2)
C4	0.86741 (13)	0.38829 (12)	0.49835 (11)	0.0185 (2)
C5	0.55244 (15)	0.23382 (15)	1.44175 (11)	0.0278 (3)
H5A	0.4759	0.3352	1.4466	0.042*
H5B	0.6263	0.2036	1.5123	0.042*
H5C	0.4865	0.1683	1.4539	0.042*
C6	0.56511 (13)	0.26911 (12)	1.20822 (11)	0.0188 (2)
C7	0.67830 (14)	0.25513 (12)	1.08442 (11)	0.0203 (2)
C8	0.77122 (13)	0.24785 (11)	0.98194 (11)	0.0188 (2)
C9	1.94875 (14)	-0.09886 (14)	0.69328 (13)	0.0277 (3)
H9A	1.9750	-0.0336	0.7404	0.042*
H9B	2.0182	-0.1071	0.6062	0.042*
H9C	1.9726	-0.1965	0.7465	0.042*
C10	1.66647 (14)	-0.01980 (12)	0.78696 (11)	0.0190 (2)
C11	1.49194 (13)	0.04128 (12)	0.76135 (10)	0.0193 (2)
C12	1.34504 (14)	0.09501 (12)	0.74997 (10)	0.0183 (2)
C13	0.93204 (13)	0.31191 (12)	0.62162 (11)	0.0175 (2)
C14	0.82157 (13)	0.31487 (12)	0.73950 (11)	0.0183 (2)
H14	0.7041	0.3661	0.7379	0.022*
C15	0.88415 (13)	0.24233 (12)	0.85995 (11)	0.0178 (2)
C16	1.05675 (14)	0.16784 (12)	0.86244 (11)	0.0180 (2)
H16	1.0992	0.1189	0.9445	0.022*
C17	1.16685 (13)	0.16532 (11)	0.74421 (11)	0.0171 (2)
C18	1.10490 (13)	0.23615 (12)	0.62325 (11)	0.0178 (2)
H18	1.1797	0.2329	0.5425	0.021*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0233 (4)	0.0293 (4)	0.0145 (4)	-0.0059 (3)	-0.0046 (3)	0.0018 (3)
O2	0.0194 (5)	0.0351 (5)	0.0271 (5)	-0.0035 (4)	-0.0074 (3)	0.0034 (4)
O3	0.0171 (4)	0.0290 (4)	0.0140 (4)	-0.0066 (3)	-0.0007 (3)	-0.0010 (3)
O4	0.0179 (4)	0.0345 (5)	0.0225 (4)	-0.0033 (4)	-0.0029 (3)	-0.0018 (3)
O5	0.0146 (4)	0.0310 (4)	0.0222 (4)	-0.0039 (3)	-0.0028 (3)	-0.0036 (3)
O6	0.0230 (4)	0.0434 (5)	0.0209 (5)	-0.0074 (4)	-0.0089 (3)	-0.0005 (4)
C1	0.0363 (7)	0.0292 (6)	0.0153 (6)	-0.0082 (5)	-0.0099 (5)	0.0038 (5)
C2	0.0204 (6)	0.0181 (5)	0.0186 (6)	-0.0061 (4)	-0.0053 (4)	-0.0007 (4)
C3	0.0162 (5)	0.0209 (5)	0.0196 (6)	-0.0046 (4)	-0.0020 (4)	-0.0025 (4)
C4	0.0153 (5)	0.0205 (5)	0.0185 (6)	-0.0050 (4)	-0.0015 (4)	-0.0026 (4)
C5	0.0248 (6)	0.0411 (7)	0.0133 (5)	-0.0097 (5)	0.0017 (4)	-0.0005 (5)
C6	0.0191 (6)	0.0179 (5)	0.0177 (5)	-0.0050 (4)	-0.0023 (4)	-0.0009 (4)
C7	0.0203 (6)	0.0211 (6)	0.0187 (6)	-0.0059 (4)	-0.0048 (5)	-0.0008 (4)
C8	0.0186 (5)	0.0191 (5)	0.0178 (6)	-0.0052 (4)	-0.0042 (4)	-0.0012 (4)
C9	0.0139 (5)	0.0309 (6)	0.0358 (7)	-0.0024 (5)	-0.0033 (5)	-0.0090 (5)
C10	0.0176 (5)	0.0185 (5)	0.0203 (6)	-0.0055 (4)	-0.0049 (4)	-0.0002 (4)
C11	0.0194 (6)	0.0225 (6)	0.0149 (5)	-0.0070 (5)	-0.0035 (4)	0.0008 (4)
C12	0.0200 (6)	0.0201 (5)	0.0151 (5)	-0.0076 (4)	-0.0034 (4)	-0.0003 (4)
C13	0.0175 (5)	0.0185 (5)	0.0163 (5)	-0.0057 (4)	-0.0044 (4)	-0.0009 (4)
C14	0.0147 (5)	0.0193 (5)	0.0199 (6)	-0.0039 (4)	-0.0037 (4)	-0.0026 (4)
C15	0.0184 (5)	0.0190 (5)	0.0166 (5)	-0.0071 (4)	-0.0020 (4)	-0.0024 (4)
C16	0.0204 (6)	0.0188 (5)	0.0157 (5)	-0.0073 (4)	-0.0055 (4)	-0.0005 (4)
C17	0.0156 (5)	0.0174 (5)	0.0188 (5)	-0.0053 (4)	-0.0046 (4)	-0.0015 (4)
C18	0.0166 (5)	0.0207 (5)	0.0162 (5)	-0.0070 (4)	-0.0013 (4)	-0.0021 (4)

Geometric parameters (Å, °)

O1—C2	1.3280 (14)	C6—C7	1.4490 (15)
O1—C1	1.4612 (13)	C7—C8	1.1961 (16)
O2—C2	1.2018 (14)	C8—C15	1.4345 (15)
O3—C6	1.3379 (13)	C9—H9A	0.9800
O3—C5	1.4478 (13)	C9—H9B	0.9800
O4—C6	1.2015 (13)	C9—H9C	0.9800
O5—C10	1.3359 (13)	C10—C11	1.4485 (15)
O5—C9	1.4478 (13)	C11—C12	1.1938 (16)
O6—C10	1.1980 (14)	C12—C17	1.4371 (15)
C1—H1A	0.9800	C13—C14	1.3931 (15)
C1—H1B	0.9800	C13—C18	1.3978 (15)
C1—H1C	0.9800	C14—C15	1.3963 (15)
C2—C3	1.4496 (15)	C14—H14	0.9500
C3—C4	1.1965 (16)	C15—C16	1.3950 (15)
C4—C13	1.4351 (15)	C16—C17	1.3945 (15)
C5—H5A	0.9800	C16—H16	0.9500
C5—H5B	0.9800	C17—C18	1.3955 (15)
C5—H5C	0.9800	C18—H18	0.9500

C2—O1—C1	114.65 (9)	H9A—C9—H9B	109.5
C6—O3—C5	114.53 (8)	O5—C9—H9C	109.5
C10—O5—C9	114.15 (9)	H9A—C9—H9C	109.5
O1—C1—H1A	109.5	H9B—C9—H9C	109.5
O1—C1—H1B	109.5	O6—C10—O5	124.66 (10)
H1A—C1—H1B	109.5	O6—C10—C11	123.62 (10)
O1—C1—H1C	109.5	O5—C10—C11	111.72 (9)
H1A—C1—H1C	109.5	C12—C11—C10	174.86 (11)
H1B—C1—H1C	109.5	C11—C12—C17	175.86 (11)
O2—C2—O1	125.16 (10)	C14—C13—C18	120.36 (10)
O2—C2—C3	122.68 (10)	C14—C13—C4	119.56 (9)
O1—C2—C3	112.16 (9)	C18—C13—C4	120.08 (10)
C4—C3—C2	173.59 (11)	C13—C14—C15	119.76 (10)
C3—C4—C13	178.97 (12)	C13—C14—H14	120.1
O3—C5—H5A	109.5	C15—C14—H14	120.1
O3—C5—H5B	109.5	C16—C15—C14	120.15 (10)
H5A—C5—H5B	109.5	C16—C15—C8	119.93 (10)
O3—C5—H5C	109.5	C14—C15—C8	119.90 (10)
H5A—C5—H5C	109.5	C17—C16—C15	119.89 (10)
H5B—C5—H5C	109.5	C17—C16—H16	120.1
O4—C6—O3	125.17 (10)	C15—C16—H16	120.1
O4—C6—C7	124.83 (10)	C16—C17—C18	120.25 (9)
O3—C6—C7	110.00 (9)	C16—C17—C12	119.01 (9)
C8—C7—C6	178.08 (11)	C18—C17—C12	120.67 (10)
C7—C8—C15	178.35 (11)	C17—C18—C13	119.58 (10)
O5—C9—H9A	109.5	C17—C18—H18	120.2
O5—C9—H9B	109.5	C13—C18—H18	120.2
C1—O1—C2—O2	1.09 (16)	C13—C14—C15—C8	178.64 (10)
C1—O1—C2—C3	-179.11 (9)	C14—C15—C16—C17	-0.29 (16)
C5—O3—C6—O4	0.66 (16)	C8—C15—C16—C17	-178.57 (9)
C5—O3—C6—C7	-179.49 (9)	C15—C16—C17—C18	-0.46 (16)
C9—O5—C10—O6	0.41 (16)	C15—C16—C17—C12	176.57 (9)
C9—O5—C10—C11	179.68 (9)	C16—C17—C18—C13	1.13 (15)
C18—C13—C14—C15	0.32 (16)	C12—C17—C18—C13	-175.85 (10)
C4—C13—C14—C15	-179.16 (10)	C14—C13—C18—C17	-1.06 (16)
C13—C14—C15—C16	0.35 (16)	C4—C13—C18—C17	178.42 (10)

Hydrogen-bond geometry (\AA , $^\circ$)

Cg1 is the centroid of the C13–C18 ring and Cg3 is the mid-point of atoms C7 and C8.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C1—H1C \cdots O6 ⁱ	0.98	2.70	3.4378 (15)	133
C5—H5B \cdots O5 ⁱⁱ	0.98	2.66	3.4517 (15)	138
C14—H14 \cdots O2 ⁱⁱⁱ	0.95	2.31	3.2278 (13)	162
C16—H16 \cdots O6 ^{iv}	0.95	2.35	3.2390 (13)	155

C1—H1A...Cg3 ^v	0.98	2.85	3.5506 (13)	130
C9—H9A...Cg1 ^{vi}	0.98	2.76	3.6302 (13)	148

Symmetry codes: (i) $x-1, y+1, z-1$; (ii) $x-1, y, z+1$; (iii) $-x+1, -y+1, -z+1$; (iv) $-x+3, -y, -z+2$; (v) $x, y, z-1$; (vi) $x+1, y, z$.