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## 2,4-Dichloro-*N*-{[1-(2,4-dichlorobenzoyl)piperidin-4-yl]methyl}benzamide

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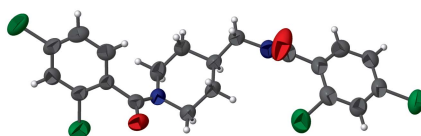
Keywords: crystal structure; piperidine; piperidone; benzamide; N—H···O hydrogen bonding.

CCDC reference: 1484074

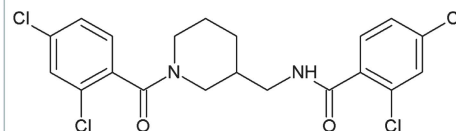
Structural data: full structural data are available from [iucrdata.iucr.org](http://iucrdata.iucr.org)

In the title compound, C<sub>20</sub>H<sub>18</sub>Cl<sub>4</sub>N<sub>2</sub>O<sub>2</sub>, the piperidine ring adopts a chair conformation. The dihedral angles between its mean plane and the two benzene rings are 45.5 (3) and 28.0 (3)°, while the benzene rings are inclined to one another by 54.5 (3)°. In the crystal, molecules are linked by N—H···O hydrogen bonds, forming chains along the *a*-axis direction. Neighbouring chains are linked by C—H···π interactions, forming double-stranded chains along [100].

### 3D view



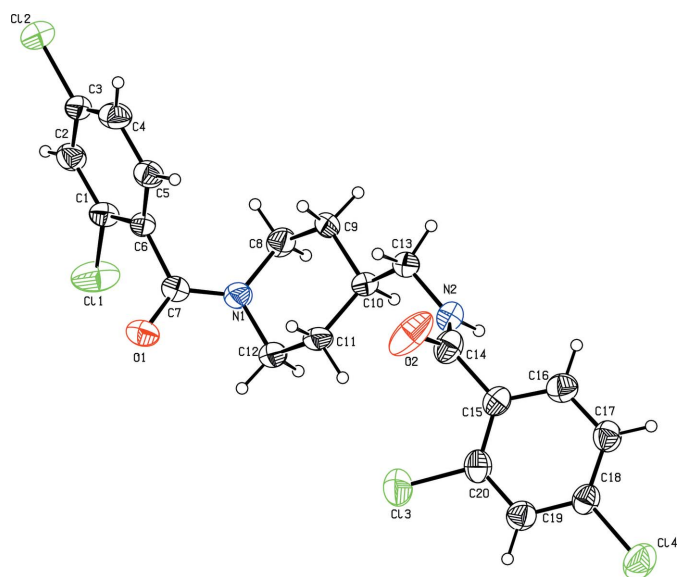
### Chemical scheme



### Structure description

Piperidones are an important group of heterocyclic compounds in the field of medicinal chemistry due to their biological activities which include cytotoxic and anticancer properties (Dimmock *et al.*, 2001). They are also reported to possess analgesic, anti-inflammatory, central nervous system (CNS), local anaesthetic, anticancer and antimicrobial activities (Perumal *et al.*, 2001). Piperidine derivatives have been observed to exhibit antimicrobial, anti-inflammatory, antiviral, antimalarial and general anesthetic activities (Aridoss *et al.*, 2009). Functionalized piperidines are familiar substructures found in biologically active natural products and synthetic pharmaceuticals (Michael, 2001; Pinder, 1992; Rubiralta *et al.*, 1991). Piperidines have also been found to exhibit blood-cholesterol-lowering activities (Nalanishi *et al.*, 1974). Herein, we report on the synthesis and crystal structure of the title piperidone derivative.

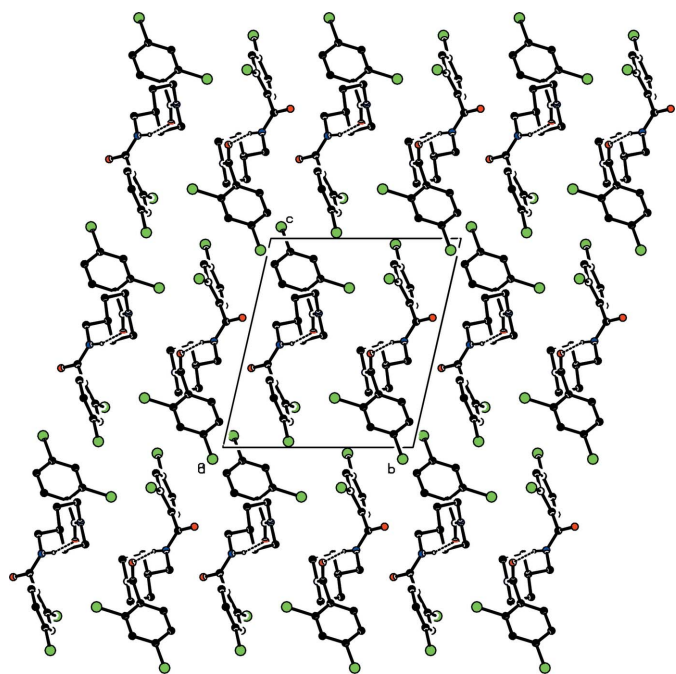
The molecular structure of the title compound is illustrated in Fig. 1. The bond distances and bond angles are close to those observed for similar structures (Revathi *et al.*, 2015; Prathebha *et al.*, 2015). The piperidine ring (N1/C8–C12) adopts a chair conformation. The dihedral angles between its mean plane and the two benzene rings



**Figure 1**  
The molecular structure of the title compound, with atom labelling. Displacement ellipsoids are drawn at the 30% probability level.

(C1–C6 and C15–C20) are 44.5 (3) and 28.0 (3)°, respectively, while the benzene rings are inclined to one another by 54.5 (3)°.

In the crystal, adjacent molecules are linked through N–H···O hydrogen bonds, forming chains along the *a*-axis direction (Table 1 and Fig. 2). Neighbouring chains are linked via C–H··· $\pi$  interactions, forming double-stranded chains along [100]; see Table 1 and Fig. 3.



**Figure 2**  
A view along the *a* axis of the crystal packing of the title compound, with hydrogen bonds shown as dashed lines (see Table 1). For clarity, the C-bound H atoms have been omitted.

**Table 1**  
Hydrogen-bond geometry (Å, °).

Cg1 is the centroid of the C15–C20 ring.

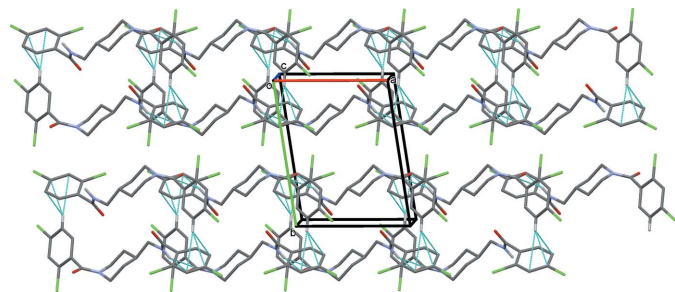
<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
N2–H2A···O1 <sup>i</sup>	0.90 (6)	1.94 (6)	2.809 (8)	163 (5)
C4–H4···Cg1	0.93	2.93	3.628 (8)	132

Symmetry code: (i) *x* + 1, *y*, *z*.

**Table 2**  
Experimental details.

<b>Crystal data</b>	
Chemical formula	C <sub>20</sub> H <sub>18</sub> Cl <sub>4</sub> N <sub>2</sub> O <sub>2</sub>
<i>M<sub>r</sub></i>	460.16
Crystal system, space group	Triclinic, <i>P</i> $\bar{1}$
Temperature (K)	293
<i>a</i> , <i>b</i> , <i>c</i> (Å)	8.2338 (2), 10.9617 (2), 12.2646 (3)
$\alpha$ , $\beta$ , $\gamma$ (°)	76.239 (1), 83.571 (1), 80.407 (1)
<i>V</i> (Å <sup>3</sup> )	1057.12 (4)
<i>Z</i>	2
Radiation type	Mo <i>K</i> $\alpha$
$\mu$ (mm <sup>−1</sup> )	0.58
Crystal size (mm)	0.30 × 0.20 × 0.20
<b>Data collection</b>	
Diffractometer	Bruker Kappa APEXII CCD diffractometer
Absorption correction	Multi-scan ( <i>SADABS</i> ; Bruker, 2004)
<i>T<sub>min</sub></i> , <i>T<sub>max</sub></i>	0.870, 0.891
No. of measured, independent and observed [ <i>I</i> > 2 $\sigma$ ( <i>I</i> )] reflections	3729, 3729, 2242
<i>R<sub>int</sub></i>	0.000
( <i>sin</i> $\theta$ / $\lambda$ ) <sub>max</sub> (Å <sup>−1</sup> )	0.595
<b>Refinement</b>	
<i>R</i> [ <i>F</i> <sup>2</sup> > 2 $\sigma$ ( <i>F</i> <sup>2</sup> )], <i>wR</i> ( <i>F</i> <sup>2</sup> ), <i>S</i>	0.086, 0.186, 1.07
No. of reflections	3726
No. of parameters	258
H-atom treatment	H atoms treated by a mixture of independent and constrained refinement
$\Delta\rho_{\max}$ , $\Delta\rho_{\min}$ (e Å <sup>−3</sup> )	0.37, −0.34

Computer programs: *APEX2*, *SAINT* and *XPREP* (Bruker, 2004), *SHELXS97* and *SHELXL97* (Sheldrick, 2008), *ORTEP-3 for Windows* (Farrugia, 2012), *Mercury* (Macrae *et al.*, 2008) and *PLATON* (Spek, 2009).



**Figure 3**  
A view along the *c* axis of the crystal packing of the title compound, with hydrogen bonds and C–H··· $\pi$  interactions shown as dashed lines (see Table 1). For clarity, H atoms not involved in these interactions have been omitted.

## Synthesis and crystallization

The title compound was synthesized following a published procedure (Revathi *et al.*, 2015). In a 250 ml round-bottomed flask, 120 ml of ethylmethylketone was added to 4-aminomethylpiperidine (0.02 mol) and stirred at room temperature. After 5 min, triethylamine (0.04 mol) was added and the mixture was stirred for 15 min. Then 2,4-dichloro benzoyl chloride (0.04 mol) was added and the reaction mixture was stirred at room temperature for 2 h. A white precipitate of triethylammonium chloride was formed, which was filtered and the filtrate was evaporated to give the crude title product. It was recrystallized twice from ethyl methyl ketone to give colourless block-like crystals of the title compound (yield: 82%).

## Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

## Acknowledgements

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## full crystallographic data

*IUCrData* (2016). **1**, x160919 [doi:10.1107/S2414314616009196]

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### 2,4-Dichloro-*N*-{[1-(2,4-dichlorobenzoyl)piperidin-4-yl]methyl}benzamide

#### Crystal data

$C_{20}H_{18}Cl_4N_2O_2$	$Z = 2$
$M_r = 460.16$	$F(000) = 472$
Triclinic, $P\bar{1}$	$D_x = 1.446 \text{ Mg m}^{-3}$
$a = 8.2338 (2) \text{ \AA}$	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
$b = 10.9617 (2) \text{ \AA}$	Cell parameters from 2242 reflections
$c = 12.2646 (3) \text{ \AA}$	$\theta = 1.7\text{--}25.0^\circ$
$\alpha = 76.239 (1)^\circ$	$\mu = 0.58 \text{ mm}^{-1}$
$\beta = 83.571 (1)^\circ$	$T = 293 \text{ K}$
$\gamma = 80.407 (1)^\circ$	Block, colourless
$V = 1057.12 (4) \text{ \AA}^3$	$0.30 \times 0.20 \times 0.20 \text{ mm}$

#### Data collection

Bruker Kappa APEXII CCD diffractometer	3729 measured reflections
Radiation source: fine-focus sealed tube	3729 independent reflections
Graphite monochromator	2242 reflections with $I > 2\sigma(I)$
$\omega$ and $\chi$ scan	$R_{\text{int}} = 0.000$
Absorption correction: multi-scan (SADABS; Bruker, 2004)	$\theta_{\text{max}} = 25.0^\circ$ , $\theta_{\text{min}} = 1.7^\circ$
$T_{\text{min}} = 0.870$ , $T_{\text{max}} = 0.891$	$h = -9 \rightarrow 9$
	$k = -12 \rightarrow 13$
	$l = 0 \rightarrow 14$

#### Refinement

Refinement on $F^2$	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.086$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.186$	$w = 1/[\sigma^2(F_o^2) + (0.P)^2 + 4.198P]$
$S = 1.07$	where $P = (F_o^2 + 2F_c^2)/3$
3726 reflections	$(\Delta/\sigma)_{\text{max}} < 0.001$
258 parameters	$\Delta\rho_{\text{max}} = 0.37 \text{ e \AA}^{-3}$
0 restraints	$\Delta\rho_{\text{min}} = -0.34 \text{ e \AA}^{-3}$
Primary atom site location: structure-invariant direct methods	

#### Special details

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted R-factor wR and goodness of fit S are based on  $F^2$ , conventional R-factors R are based on F, with F set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on F, and R-factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	−0.1778 (8)	0.3093 (6)	0.8072 (5)	0.0539 (16)
C2	−0.2618 (8)	0.2486 (7)	0.9025 (5)	0.0619 (18)
H2	−0.3430	0.2935	0.9428	0.074*
C3	−0.2213 (8)	0.1181 (7)	0.9365 (5)	0.0622 (19)
C4	−0.1047 (9)	0.0510 (7)	0.8774 (6)	0.0659 (19)
H4	−0.0819	−0.0369	0.9006	0.079*
C5	−0.0200 (8)	0.1148 (6)	0.7821 (5)	0.0578 (17)
H5	0.0618	0.0697	0.7425	0.069*
C6	−0.0572 (7)	0.2457 (6)	0.7458 (5)	0.0454 (14)
C7	0.0273 (8)	0.3122 (6)	0.6384 (5)	0.0493 (15)
C8	0.2759 (8)	0.2995 (7)	0.7375 (5)	0.0634 (18)
H8A	0.2080	0.2654	0.8040	0.076*
H8B	0.3193	0.3711	0.7510	0.076*
C9	0.4171 (7)	0.1987 (6)	0.7159 (5)	0.0564 (17)
H9A	0.3731	0.1228	0.7133	0.068*
H9B	0.4879	0.1773	0.7778	0.068*
C10	0.5195 (8)	0.2414 (6)	0.6063 (5)	0.0517 (16)
H10	0.5754	0.3109	0.6138	0.062*
C11	0.4077 (8)	0.2907 (7)	0.5107 (5)	0.0644 (19)
H11A	0.3605	0.2208	0.4971	0.077*
H11B	0.4726	0.3254	0.4428	0.077*
C12	0.2685 (8)	0.3931 (7)	0.5361 (6)	0.068 (2)
H12A	0.3141	0.4666	0.5434	0.082*
H12B	0.1960	0.4193	0.4751	0.082*
C13	0.6490 (8)	0.1333 (6)	0.5858 (5)	0.0601 (18)
H13A	0.7081	0.0977	0.6530	0.072*
H13B	0.5960	0.0672	0.5703	0.072*
C14	0.7855 (9)	0.1351 (7)	0.3983 (7)	0.068 (2)
C15	0.9081 (8)	0.1928 (6)	0.3085 (6)	0.0558 (16)
C16	1.0746 (9)	0.1642 (6)	0.3215 (6)	0.0663 (19)
H16	1.1103	0.1139	0.3893	0.080*
C17	1.1912 (9)	0.2076 (7)	0.2368 (6)	0.0652 (19)
H17	1.3034	0.1860	0.2471	0.078*
C18	1.1380 (8)	0.2830 (6)	0.1379 (6)	0.0588 (17)
C19	0.9714 (9)	0.3206 (7)	0.1231 (6)	0.0642 (18)
H19	0.9362	0.3746	0.0566	0.077*
C20	0.8589 (8)	0.2759 (7)	0.2094 (6)	0.0631 (18)
N1	0.1756 (7)	0.3417 (5)	0.6411 (4)	0.0586 (14)
N2	0.7647 (7)	0.1751 (6)	0.4921 (5)	0.0603 (15)
O1	−0.0451 (6)	0.3363 (4)	0.5513 (4)	0.0681 (13)

O2	0.7133 (9)	0.0541 (6)	0.3794 (5)	0.120 (3)
Cl1	-0.2283 (3)	0.47199 (19)	0.76344 (18)	0.1046 (9)
Cl2	-0.3249 (3)	0.0404 (3)	1.05567 (19)	0.1082 (9)
Cl3	0.6511 (3)	0.3285 (3)	0.1925 (2)	0.1024 (8)
Cl4	1.2777 (3)	0.3376 (2)	0.02837 (18)	0.0933 (7)
H2A	0.822 (7)	0.237 (5)	0.497 (5)	0.042 (17)*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.064 (4)	0.051 (4)	0.043 (4)	-0.003 (3)	-0.005 (3)	-0.007 (3)
C2	0.061 (4)	0.073 (5)	0.047 (4)	0.000 (4)	-0.004 (3)	-0.012 (3)
C3	0.057 (4)	0.074 (5)	0.050 (4)	-0.022 (4)	-0.013 (3)	0.008 (4)
C4	0.086 (6)	0.050 (4)	0.061 (5)	-0.016 (4)	-0.016 (4)	-0.003 (4)
C5	0.066 (4)	0.054 (4)	0.055 (4)	-0.008 (3)	-0.012 (3)	-0.012 (3)
C6	0.047 (4)	0.050 (4)	0.040 (3)	-0.008 (3)	-0.004 (3)	-0.008 (3)
C7	0.056 (4)	0.046 (4)	0.045 (4)	-0.001 (3)	-0.005 (3)	-0.013 (3)
C8	0.065 (4)	0.087 (5)	0.044 (4)	-0.023 (4)	0.000 (3)	-0.018 (4)
C9	0.048 (4)	0.075 (5)	0.043 (4)	-0.011 (3)	-0.012 (3)	-0.002 (3)
C10	0.053 (4)	0.055 (4)	0.044 (4)	-0.012 (3)	-0.003 (3)	-0.001 (3)
C11	0.063 (4)	0.077 (5)	0.040 (4)	-0.006 (4)	0.002 (3)	0.006 (3)
C12	0.054 (4)	0.077 (5)	0.059 (4)	-0.005 (4)	0.001 (3)	0.010 (4)
C13	0.051 (4)	0.064 (4)	0.056 (4)	-0.006 (3)	-0.004 (3)	0.003 (3)
C14	0.060 (5)	0.064 (5)	0.082 (5)	-0.010 (4)	0.010 (4)	-0.025 (4)
C15	0.059 (4)	0.050 (4)	0.062 (4)	-0.015 (3)	0.004 (3)	-0.019 (3)
C16	0.074 (5)	0.060 (4)	0.064 (5)	-0.013 (4)	-0.005 (4)	-0.011 (4)
C17	0.055 (4)	0.066 (5)	0.072 (5)	-0.004 (4)	-0.001 (4)	-0.018 (4)
C18	0.056 (4)	0.067 (4)	0.057 (4)	-0.013 (3)	0.005 (3)	-0.022 (4)
C19	0.066 (5)	0.073 (5)	0.056 (4)	-0.014 (4)	-0.002 (3)	-0.016 (4)
C20	0.055 (4)	0.068 (5)	0.072 (5)	-0.008 (4)	-0.003 (4)	-0.027 (4)
N1	0.055 (3)	0.066 (4)	0.049 (3)	-0.003 (3)	0.002 (3)	-0.005 (3)
N2	0.053 (4)	0.064 (4)	0.064 (4)	-0.015 (3)	0.002 (3)	-0.013 (3)
O1	0.082 (3)	0.069 (3)	0.051 (3)	-0.008 (3)	-0.016 (2)	-0.004 (2)
O2	0.154 (6)	0.125 (5)	0.110 (5)	-0.092 (5)	0.054 (4)	-0.061 (4)
Cl1	0.157 (2)	0.0591 (12)	0.0783 (14)	0.0125 (13)	0.0236 (14)	-0.0114 (10)
Cl2	0.0897 (16)	0.135 (2)	0.0803 (15)	-0.0405 (14)	-0.0055 (12)	0.0306 (14)
Cl3	0.0602 (12)	0.143 (2)	0.0965 (16)	-0.0136 (13)	-0.0045 (11)	-0.0129 (15)
Cl4	0.0793 (14)	0.130 (2)	0.0720 (13)	-0.0303 (13)	0.0173 (11)	-0.0250 (13)

*Geometric parameters (Å, °)*

C1—C2	1.375 (9)	C11—C12	1.526 (9)
C1—C6	1.376 (8)	C11—H11A	0.9700
C1—Cl1	1.729 (6)	C11—H11B	0.9700
C2—C3	1.385 (9)	C12—N1	1.462 (8)
C2—H2	0.9300	C12—H12A	0.9700
C3—C4	1.360 (9)	C12—H12B	0.9700
C3—Cl2	1.720 (7)	C13—N2	1.445 (8)

C4—C5	1.391 (9)	C13—H13A	0.9700
C4—H4	0.9300	C13—H13B	0.9700
C5—C6	1.389 (8)	C14—O2	1.224 (8)
C5—H5	0.9300	C14—N2	1.310 (9)
C6—C7	1.503 (8)	C14—C15	1.504 (9)
C7—O1	1.234 (7)	C15—C16	1.373 (9)
C7—N1	1.320 (8)	C15—C20	1.390 (9)
C8—N1	1.458 (8)	C16—C17	1.386 (9)
C8—C9	1.511 (9)	C16—H16	0.9300
C8—H8A	0.9700	C17—C18	1.367 (9)
C8—H8B	0.9700	C17—H17	0.9300
C9—C10	1.521 (8)	C18—C19	1.384 (9)
C9—H9A	0.9700	C18—C14	1.728 (7)
C9—H9B	0.9700	C19—C20	1.380 (9)
C10—C13	1.505 (8)	C19—H19	0.9300
C10—C11	1.512 (8)	C20—C13	1.732 (7)
C10—H10	0.9800	N2—H2A	0.90 (6)
C2—C1—C6	122.6 (6)	C10—C11—H11B	109.2
C2—C1—C11	118.1 (5)	C12—C11—H11B	109.2
C6—C1—C11	119.3 (5)	H11A—C11—H11B	107.9
C1—C2—C3	117.7 (6)	N1—C12—C11	108.9 (5)
C1—C2—H2	121.2	N1—C12—H12A	109.9
C3—C2—H2	121.2	C11—C12—H12A	109.9
C4—C3—C2	121.8 (6)	N1—C12—H12B	109.9
C4—C3—C12	119.8 (6)	C11—C12—H12B	109.9
C2—C3—C12	118.4 (6)	H12A—C12—H12B	108.3
C3—C4—C5	119.4 (6)	N2—C13—C10	111.2 (5)
C3—C4—H4	120.3	N2—C13—H13A	109.4
C5—C4—H4	120.3	C10—C13—H13A	109.4
C6—C5—C4	120.3 (6)	N2—C13—H13B	109.4
C6—C5—H5	119.9	C10—C13—H13B	109.4
C4—C5—H5	119.9	H13A—C13—H13B	108.0
C1—C6—C5	118.2 (6)	O2—C14—N2	124.6 (7)
C1—C6—C7	122.0 (5)	O2—C14—C15	118.3 (7)
C5—C6—C7	119.7 (5)	N2—C14—C15	117.1 (7)
O1—C7—N1	122.9 (6)	C16—C15—C20	117.2 (6)
O1—C7—C6	118.3 (6)	C16—C15—C14	120.8 (6)
N1—C7—C6	118.8 (5)	C20—C15—C14	122.0 (6)
N1—C8—C9	110.3 (5)	C15—C16—C17	122.2 (7)
N1—C8—H8A	109.6	C15—C16—H16	118.9
C9—C8—H8A	109.6	C17—C16—H16	118.9
N1—C8—H8B	109.6	C18—C17—C16	118.7 (7)
C9—C8—H8B	109.6	C18—C17—H17	120.7
H8A—C8—H8B	108.1	C16—C17—H17	120.7
C8—C9—C10	112.5 (5)	C17—C18—C19	121.3 (6)
C8—C9—H9A	109.1	C17—C18—C14	120.7 (5)
C10—C9—H9A	109.1	C19—C18—C14	117.9 (5)

C8—C9—H9B	109.1	C20—C19—C18	118.3 (7)
C10—C9—H9B	109.1	C20—C19—H19	120.9
H9A—C9—H9B	107.8	C18—C19—H19	120.9
C13—C10—C11	112.4 (5)	C19—C20—C15	122.0 (6)
C13—C10—C9	109.5 (5)	C19—C20—C13	118.0 (6)
C11—C10—C9	109.8 (5)	C15—C20—C13	120.0 (5)
C13—C10—H10	108.3	C7—N1—C8	124.9 (6)
C11—C10—H10	108.3	C7—N1—C12	119.9 (6)
C9—C10—H10	108.3	C8—N1—C12	113.5 (5)
C10—C11—C12	112.0 (6)	C14—N2—C13	124.7 (6)
C10—C11—H11A	109.2	C14—N2—H2A	117 (4)
C12—C11—H11A	109.2	C13—N2—H2A	118 (4)
C6—C1—C2—C3	0.0 (10)	O2—C14—C15—C20	71.3 (10)
C11—C1—C2—C3	179.6 (5)	N2—C14—C15—C20	-109.3 (8)
C1—C2—C3—C4	-1.1 (10)	C20—C15—C16—C17	-4.8 (10)
C1—C2—C3—C12	179.9 (5)	C14—C15—C16—C17	174.8 (7)
C2—C3—C4—C5	1.9 (11)	C15—C16—C17—C18	1.0 (11)
C12—C3—C4—C5	-179.1 (5)	C16—C17—C18—C19	2.9 (10)
C3—C4—C5—C6	-1.7 (10)	C16—C17—C18—C14	-178.8 (5)
C2—C1—C6—C5	0.1 (10)	C17—C18—C19—C20	-2.6 (10)
C11—C1—C6—C5	-179.4 (5)	C14—C18—C19—C20	179.0 (5)
C2—C1—C6—C7	176.7 (6)	C18—C19—C20—C15	-1.5 (11)
C11—C1—C6—C7	-2.8 (9)	C18—C19—C20—C13	177.4 (5)
C4—C5—C6—C1	0.7 (9)	C16—C15—C20—C19	5.0 (10)
C4—C5—C6—C7	-175.9 (6)	C14—C15—C20—C19	-174.5 (7)
C1—C6—C7—O1	-79.8 (8)	C16—C15—C20—C13	-173.9 (5)
C5—C6—C7—O1	96.7 (7)	C14—C15—C20—C13	6.6 (9)
C1—C6—C7—N1	100.6 (7)	O1—C7—N1—C8	-170.7 (6)
C5—C6—C7—N1	-83.0 (8)	C6—C7—N1—C8	8.9 (9)
N1—C8—C9—C10	53.7 (7)	O1—C7—N1—C12	-6.7 (9)
C8—C9—C10—C13	-175.6 (5)	C6—C7—N1—C12	172.9 (5)
C8—C9—C10—C11	-51.7 (7)	C9—C8—N1—C7	106.5 (7)
C13—C10—C11—C12	175.5 (6)	C9—C8—N1—C12	-58.4 (7)
C9—C10—C11—C12	53.3 (8)	C11—C12—N1—C7	-106.3 (7)
C10—C11—C12—N1	-56.6 (8)	C11—C12—N1—C8	59.4 (8)
C11—C10—C13—N2	66.5 (7)	O2—C14—N2—C13	-2.5 (13)
C9—C10—C13—N2	-171.1 (5)	C15—C14—N2—C13	178.1 (6)
O2—C14—C15—C16	-108.2 (9)	C10—C13—N2—C14	-116.8 (7)
N2—C14—C15—C16	71.2 (9)		

*Hydrogen-bond geometry* ( $\text{\AA}$ ,  $^\circ$ )

Cg1 is the centroid of the C15–C20 ring.

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
N2—H2A $\cdots$ O1 <sup>i</sup>	0.90 (6)	1.94 (6)	2.809 (8)	163 (5)



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C4—H4···Cg1	0.93	2.93	3.628 (8)	132
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Symmetry code: (i)  $x+1, y, z$ .