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(4'-Phenyl-2,2':6',2"-terpyridine- $\kappa^3 N, N', N''$)bis-(thiocyanato- κN)zinc(II) unknown solvate

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The title compound, $[Zn(NCS)_2(C_{21}H_{15}N_3)]$, crystallizes with three independent complex molecules in the asymmetric unit. In each complex molecule, the Zn^{II} atom is coordinated by three N atoms of a 4'-phenyl-2,2':6',2''-terpyridine ligand, and by the N atoms of two NCS⁻ anions. The Zn^{II} atoms are therefore fivecoordinate, ZnN₅, with distorted square-pyramidal geometries. In the crystal, the three independent molecules are linked by a series of offset π - π interactions [intercentroid distances vary between 3.680 (5) and 3.791 (5) Å], forming columns along the *a*-axis direction. The columns are linked *via* C-H···S interactions, forming a fence-like arrangement parallel to the *ab* plane. A small region of disordered electron density was corrected for using the SQUEEZE routine in *PLATON* [Spek (2015). *Acta Cryst.* C71, 9–18], but the formula mass and unit-cell characteristics were not taken into account during the refinement.



Structure description

In recent decades, the design of metal–organic compounds has been an important goal for synthetic chemistry as it provides the opportunity to control the properties of materials at the molecular level (Li *et al.*, 2008). Polydentate ligands containing N-donor heterocyclic rings are versatile building blocks in the construction of coordination compounds because of their ability to form stable complexes with transition metals and their various coordination modes (Hancock, 2013; Li *et al.*, 2011; Bhaumik *et al.*, 2011). One such ligand is 4'-phenyl-2,2':6',2''-terpyridine (pypyt), first synthesized by Constable *et al.* (1990) and used by them for the formation of nickel complexes. The same ligand has been used by many groups to form metal complexes, both mononuclear, binuclear and polymeric (Crystal Structure Database; Groom *et al.*, 2016). We have studied the reaction of



data reports

Table 1 Hydrogen-bond geometry (Å, °).						
$D - H \cdots A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$		
$C5-H5\cdots S6^{i}$	0.93	2.87	3.748 (8)	159		
$C41 - H41 \cdots S6^{ii}$	0.93	2.82	3.612 (9)	144		

Symmetry codes: (i) $x + \frac{1}{2}, -y + \frac{1}{2}, z - \frac{1}{2}$; (ii) $-x + \frac{1}{2}, y - \frac{1}{2}, -z + \frac{3}{2}$.

this ligand with Zn(NCS)₂, and report herein on the crystal structure of the resulting mononuclear complex.

The asymmetric unit of the title compound contains three complex molecules. For simplicity, as the three complexes are very similar, only the structure of one complex, involving atom Zn1, is illustrated in Fig. 1. In each complex molecule, the zinc(II) atom is coordinated by three N atoms of a 4'-phenyl-2,2':6',2''-terpyridine ligand in a tridentate manner, and by the N atoms of two NCS⁻ anions. The zinc atoms are therefore five-coordinate with τ values indicating fivefold coordination (Addison et al., 1984) being 0.33 for the Zn1 complex, 0.30 for Zn2 and 0.13 for Zn3; hence, all have highly distorted squarepyramidal geometries.

In the crystal, the three molecules are linked by offset $\pi - \pi$ interactions [intercentroid distances vary between 3.680 (5) and 3.791 (5) Å], forming columns along the *a*-axis direction. The columns are linked via $C-H \cdot \cdot S$ interactions, forming a fence-like arrangement parallel to the ab plane (Table 1 and Fig. 2).

Synthesis and crystallization

The 4'-phenyl-2,2':6',2"-terpyridine ligand was synthesized according to a literature method (Mutai et al., 2011). For the preparation of the title complex, a methanol solution of Zn(NCS)₂ (0.181 g, 1 mmol) and 4'-phenyl-2,2':6',2"-terpyr-



Figure 1

The molecular structure of one of the three title complex molecules, showing the atom labelling. Displacement ellipsoids are drawn at the 50% probability level.

Table 2 Experimental details.

Crystal data	
Chemical formula	$[Zn(NCS)_2(C_{21}H_{15}N_3)]$
$M_{\rm r}$	490.89
Crystal system, space group	Monoclinic, $P2_1/n$
Temperature (K)	298
<i>a</i> , <i>b</i> , <i>c</i> (Å)	11.518 (5), 25.095 (5), 23.137 (5)
β (°)	98.217 (5)
$V(Å^3)$	6619 (3)
Ζ	12
Radiation type	Μο Κα
$\mu \text{ (mm}^{-1})$	1.32
Crystal size (mm)	$0.30 \times 0.20 \times 0.20$
Data collection	
Diffractometer	Bruker SMART APEX CCD
Absorption correction	Multi-scan (<i>SADABS</i> ; Bruker, 2007)
T_{\min}, T_{\max}	0.692, 0.778
No. of measured, independent and	47455, 11948, 5625
$\frac{1}{2} \sum_{i=1}^{n} \frac{1}{2} \sum_{i=1}^{n} \frac{1}$	0.127
K_{int}	0.127
$(\sin \theta/\lambda)_{\rm max}(A)$	0.000
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.080, 0.213, 0.98
No. of reflections	11948
No. of parameters	838
H-atom treatment	H-atom parameters constrained
$\Delta \rho_{\rm max}, \Delta \rho_{\rm min} ({\rm e} {\rm \AA}^{-3})$	0.78, -0.53

Computer programs: SMART and SAINT (Bruker, 2007), SHELXS97 and SHELXL97 (Sheldrick, 2008), DIAMOND (Brandenburg, 2006) and PLATON (Spek, 2009).

idine (0.64 g, 2 mmol) was refluxed for 2 h. The precipitate that formed was filtered off and recrystallized from methanol solution, giving colourless block-like crystals of the title compound.





A view along the a axis of the crystal packing of the title compound [colour code for the three independent complexes: black (Zn1), red (Zn2), and green (Zn3)]. Hydrogen bonds are shown as dashed lines (see Table 1) and H atoms not involved in these interactions have been omitted for clarity.

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2. Reflections (011), (012), and (020) affected by the beam-stop, were omitted for the final refinement. A small region of disordered electron density was corrected for using the SQUEEZE routine in *PLATON* (Spek, 2015), volume *ca* 60 Å³ for 12 electrons count; the formula mass and unit-cell characteristics were not taken into account during refinement.

Acknowledgements

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full crystallographic data

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(4'-Phenyl-2,2':6',2''-terpyridine- $\kappa^3 N, N', N''$) bis(thiocyanato- κN)zinc(II) unknown solvate

F(000) = 3000

 $\theta = 2.3 - 17.9^{\circ}$

 $\mu = 1.32 \text{ mm}^{-1}$

Block, colourless

 $0.30 \times 0.20 \times 0.20$ mm

 $\theta_{\rm max} = 25.3^\circ, \ \theta_{\rm min} = 1.8^\circ$

47455 measured reflections

11948 independent reflections

5625 reflections with $I > 2\sigma(I)$

T = 298 K

 $R_{\rm int} = 0.127$

 $h = -13 \rightarrow 13$

 $k = -30 \rightarrow 27$

 $l = -27 \rightarrow 27$

 $D_{\rm x} = 1.478 {\rm Mg} {\rm m}^{-3}$

Mo *K* α radiation, $\lambda = 0.71069$ Å

Cell parameters from 2976 reflections

An-ran Wang, Cong Wang and Sheng-Li Li

(4'-Phenyl-2,2':6',2''-terpyridine-κ³N,N',N'')bis(thiocyanato-κN)zinc(II) unknown solvate

Crystal data

 $[Zn(NCS)_2(C_{21}H_{15}N_3)]$ $M_r = 490.89$ Monoclinic, $P2_1/n$ a = 11.518 (5) Å b = 25.095 (5) Å c = 23.137 (5) Å $\beta = 98.217$ (5)° V = 6619 (3) Å³ Z = 12

Data collection

Bruker SMART APEX CCD diffractometer Radiation source: fine-focus sealed tube Graphite monochromator phi and ω scans Absorption correction: multi-scan (SADABS; Bruker, 2007) $T_{\min} = 0.692, T_{\max} = 0.778$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier
Least-squares matrix: full	map
$R[F^2 > 2\sigma(F^2)] = 0.080$	Hydrogen site location: inferred from
$wR(F^2) = 0.213$	neighbouring sites
S = 0.98	H-atom parameters constrained
11948 reflections	$w = 1/[\sigma^2(F_o^2) + (0.093P)^2]$
838 parameters	where $P = (F_o^2 + 2F_c^2)/3$
0 restraints	$(\Delta/\sigma)_{\rm max} = 0.001$
Primary atom site location: structure-invariant	$\Delta ho_{ m max} = 0.78 \ { m e} \ { m \AA}^{-3}$
direct methods	$\Delta \rho_{\rm min} = -0.53 \ {\rm e} \ {\rm \AA}^{-3}$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > 2$ sigma(F^2) is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

	X	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	
Zn1	0.40821 (8)	0.09997 (4)	0.08583 (3)	0.0588 (3)	
S1	0.0197 (3)	0.09060 (15)	0.01774 (13)	0.1266 (13)	
S2	0.7367 (2)	0.15375 (12)	0.00189 (12)	0.1003 (9)	
N1	0.3470 (6)	0.1807 (2)	0.1066 (3)	0.0576 (17)	
N2	0.4488 (5)	0.1072 (2)	0.1756 (2)	0.0443 (14)	
N3	0.4745 (6)	0.0217 (2)	0.1153 (2)	0.0563 (17)	
N10	0.2607 (7)	0.0800 (3)	0.0387 (3)	0.079 (2)	
N11	0.5373 (7)	0.1201 (3)	0.0425 (3)	0.071 (2)	
C1	0.5630 (8)	0.1283 (4)	0.4818 (4)	0.078 (3)	
H1	0.5720	0.1308	0.5223	0.094*	
C2	0.5519 (8)	0.1729 (4)	0.4485 (3)	0.078 (3)	
H2	0.5568	0.2062	0.4663	0.094*	
C3	0.5332 (7)	0.1694 (3)	0.3882 (3)	0.062 (2)	
H3	0.5259	0.2004	0.3660	0.075*	
C4	0.5254 (6)	0.1205 (3)	0.3605 (3)	0.0479 (18)	
C5	0.5406 (7)	0.0750 (3)	0.3955 (3)	0.065 (2)	
Н5	0.5370	0.0416	0.3781	0.078*	
C6	0.5609 (9)	0.0792 (4)	0.4552 (4)	0.084 (3)	
H6	0.5732	0.0486	0.4779	0.101*	
C7	0.5002 (6)	0.1160 (3)	0.2963 (3)	0.0422 (17)	
C8	0.4496 (6)	0.1578 (3)	0.2620 (3)	0.0464 (18)	
H8	0.4319	0.1894	0.2798	0.056*	
C9	0.4256 (6)	0.1530 (3)	0.2026 (3)	0.0439 (17)	
C10	0.4990 (6)	0.0661 (3)	0.2068 (3)	0.0417 (17)	
C11	0.5248 (6)	0.0697 (3)	0.2672 (3)	0.0476 (18)	
H11	0.5589	0.0409	0.2884	0.057*	
C12	0.3702 (6)	0.1956 (3)	0.1630 (3)	0.0487 (19)	
C13	0.3427 (8)	0.2456 (3)	0.1806 (4)	0.070 (2)	
H13	0.3585	0.2549	0.2199	0.084*	
C14	0.2918 (8)	0.2821 (3)	0.1406 (4)	0.073 (3)	
H14	0.2731	0.3162	0.1522	0.088*	
C15	0.2694 (8)	0.2672 (4)	0.0836 (4)	0.083 (3)	
H15	0.2348	0.2911	0.0556	0.100*	
C16	0.2983 (8)	0.2162 (4)	0.0672 (4)	0.077 (3)	
H16	0.2835	0.2066	0.0281	0.092*	
C17	0.5169 (6)	0.0180 (3)	0.1716 (3)	0.0491 (19)	
C18	0.5750 (7)	-0.0266 (3)	0.1945 (3)	0.060 (2)	
H18	0.6055	-0.0280	0.2339	0.072*	
C19	0.5875 (8)	-0.0690 (3)	0.1588 (4)	0.074 (2)	
H19	0.6261	-0.0997	0.1735	0.089*	

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters $(Å^2)$

C20	0.5416 (8)	-0.0652 (4)	0.1007 (4)	0.076 (3)
H20	0.5482	-0.0933	0.0752	0.091*
C21	0.4867 (8)	-0.0196 (4)	0.0815 (3)	0.075 (3)
H21	0.4559	-0.0172	0.0422	0.090*
C69	0.1643 (10)	0.0857 (4)	0.0302 (3)	0.076 (3)
C70	0.6193 (7)	0.1331 (3)	0.0252 (3)	0.056 (2)
Zn2	0.62013 (8)	0.26836 (4)	0.81912 (4)	0.0645 (3)
83	0.3027 (2)	0.19842 (10)	0.90059 (10)	0.0783(7)
S4	0.9905(2)	0.18933 (11)	0.83004 (16)	0.1121 (10)
N4	0 5505 (6)	0 2600 (2)	0 7256 (3)	0.0597(17)
N5	0.6317(5)	0.3445(2)	0.7233(3)	0.0397(17) 0.0493(15)
N6	0.6317(6)	0.3198(3)	0.8944(3)	0.0199(19)
N12	0.0704 (0)	0.3190(3)	0.8506(3)	0.0094(19)
N12	0.4919(0) 0.7654(7)	0.2319(3) 0.2278(3)	0.8300(3)	0.075(2)
C22	0.7054(7)	0.2278(3)	0.6221(5)	0.074(2)
U22	0.7000 (9)	0.5882 (5)	0.0372 (3)	0.083 (3)
П22 С22	0.7642	0.0202	0.0403	0.099
C23	0.7027 (9)	0.5510 (4)	0.6247 (4)	0.082 (3)
H23	0.6764	0.5573	0.5854	0.098*
C24	0.6767 (8)	0.5028 (3)	0.6504 (4)	0.072(3)
H24	0.6345	0.4770	0.6274	0.087*
C25	0.7115 (6)	0.4924 (3)	0.7086 (3)	0.0504 (19)
C26	0.7760 (8)	0.5309 (3)	0.7400 (4)	0.075 (3)
H26	0.8025	0.5251	0.7793	0.089*
C27	0.8031 (10)	0.5786 (4)	0.7144 (5)	0.094 (3)
H27	0.8471	0.6042	0.7369	0.112*
C28	0.6848 (6)	0.4406 (3)	0.7344 (3)	0.0452 (18)
C29	0.6335 (7)	0.3986 (3)	0.7014 (3)	0.055 (2)
H29	0.6158	0.4023	0.6611	0.066*
C30	0.6085 (6)	0.3518 (3)	0.7268 (3)	0.0481 (18)
C31	0.7037 (6)	0.4328 (3)	0.7951 (3)	0.054 (2)
H31	0.7318	0.4606	0.8198	0.065*
C32	0.6803 (6)	0.3834 (3)	0.8181 (3)	0.0485 (18)
C33	0.7007 (7)	0.3705 (3)	0.8812 (3)	0.054 (2)
C34	0.7416 (7)	0.4059 (4)	0.9239 (4)	0.072 (2)
H34	0.7564	0.4409	0.9140	0.087*
C35	0.7615 (9)	0.3903 (5)	0.9823 (4)	0.096 (3)
H35	0.7908	0.4140	1.0117	0.115*
C36	0.7362 (9)	0.3395 (6)	0.9942 (4)	0.101 (4)
H36	0.7445	0.3280	1.0327	0.122*
C37	0.6981 (8)	0.3044 (4)	0.9498(4)	0.084(3)
H37	0.6858	0 2689	0.9588	0.101*
C38	0.5532 (7)	0.3049(3)	0.6939 (4)	0.058(2)
C39	0.5552(7) 0.5141(7)	0.3051(3)	0.6352(4)	0.063(2)
H39	0.5200	0 3361	0.6137	0.075*
C40	0.4666 (8)	0.2605(4)	0.6077(4)	0.079(3)
H40	0 4373	0.2609	0.5681	0.095*
C41	0.4633 (8)	0.2009	0.6404(4)	0.078 (3)
H41	0.4344	0.1833	0.6228	0.078 (3)
11-11	V. T.J. T.T.	0.1033	0.0220	0.075

C43	0.5028 (8)	0.2164 (3)	0.6988 (4)	0.073 (3)
H43	0.4963	0.1858	0.7208	0.088*
C65	0.8564 (9)	0.2108 (3)	0.8268 (4)	0.065(2)
C66	0.4157 (7)	0.2176 (3)	0.8730 (3)	0.055 (2)
Zn3	0.39549 (8)	0.48965 (3)	0.81136 (4)	0.0524 (3)
S5	0.7473 (2)	0.56688 (12)	0.89780 (12)	0.0956 (9)
S6	0.0596(2)	0.57143(10)	0.86627(12)	0.0856 (8)
N7	0.3859(5)	0.5012(2)	0.7189(3)	0.0544 (16)
N8	0.3035(3)	0.3012(2) 0.4158(2)	0.7753(2)	0.0397(13)
N9	0.4070 (6)	0.1130(2) 0.4335(2)	0.8843(2)	0.0597(13)
N14	0.5383 (6)	0.4355(2) 0.5294(3)	0.8402(3)	0.0590(17)
N15	0.3303 (0)	0.5274(3)	0.8320(3)	0.071(2)
C14	0.2009(0)	0.3312(3) 0.1752(3)	0.6320(3)	0.0033(19)
U44	0.1194 (9)	0.1732 (3)	0.0482 (4)	0.083 (3)
П44 С45	0.0901 0.1755 (10)	0.1433	0.0312	0.099°
C45	0.1755 (10)	0.1755 (4)	0.7046 (4)	0.096 (3)
H45	0.1804	0.1439	0.7256	0.116*
C46	0.2160 (8)	0.2231 (3)	0.7302 (3)	0.074 (3)
H46	0.2521	0.2231	0.7688	0.089*
C47	0.2041 (6)	0.2703 (3)	0.6997 (3)	0.0487 (18)
C48	0.1500 (7)	0.2686 (3)	0.6430 (3)	0.055 (2)
H48	0.1422	0.2998	0.6212	0.066*
C49	0.1067 (8)	0.2214 (3)	0.6174 (4)	0.069 (2)
H49	0.0689	0.2214	0.5791	0.082*
C50	0.2509 (6)	0.3211 (3)	0.7260 (3)	0.0415 (17)
C51	0.2798 (6)	0.3269 (3)	0.7869 (3)	0.0464 (18)
H51	0.2671	0.2989	0.8115	0.056*
C52	0.3269 (6)	0.3739 (3)	0.8098 (3)	0.0464 (18)
C53	0.2717 (6)	0.3646 (3)	0.6920 (3)	0.0442 (18)
H53	0.2550	0.3621	0.6516	0.053*
C54	0.3166 (6)	0.4117 (3)	0.7169 (3)	0.0444 (17)
C55	0.3415 (6)	0.4608 (3)	0.6844 (3)	0.0460 (18)
C56	0.3223 (7)	0.4638 (3)	0.6243 (3)	0.066 (2)
H56	0.2926	0.4347	0.6020	0.079*
C57	0.3476 (9)	0.5101 (4)	0.5983 (4)	0.086(3)
H57	0.3346	0.5132	0.5578	0.104*
C58	0.3927 (9)	0.5528 (4)	0.6324 (4)	0.080(3)
H58	0.4110	0.5846	0.6152	0.095*
C59	0.4099(7)	0.5470 (3)	0.6918 (4)	0.062(2)
H59	0.4393	0.5758	0.7146	0.074*
C60	0.3616 (6)	0.3851 (3)	0.8729(3)	0.0449 (18)
C61	0.3562(7)	0.3476(3)	0.9159(3)	0.059(2)
H61	0.3262	0.3137	0.9066	0.071*
C62	0.3959 (8)	0.3611 (4)	0.9726(3)	0.074(3)
U02 Н62	0.3913	0.3364	1.0022	0.088*
C63	0.3713	0.3304	0.9863 (1)	0.086 (3)
Н63	0.4708	0.4200	1 0245	0.000 (3)
C64	0 1113 (8)	0.7200	0 0/03 (/)	0.103
UG4	0 4729	0.4701	0.0400	0.001 (3)
1104	0.4/30	0.4/91	0.9400	0.098.

data reports

C67	0.1803 (8)	0.5478 (3)	0.8479 (3)	0.052 (2)
C68	0.6241 (8)	0.5454 (3)	0.8634 (3)	0.058 (2)

Atomic displacement parameters $(Å^2)$

	U^{11}	U ²²	U^{33}	U^{12}	U^{13}	U^{23}
Zn1	0.0606 (6)	0.0714 (7)	0.0429 (5)	0.0004 (5)	0.0021 (4)	-0.0018 (4)
S1	0.082 (2)	0.190 (4)	0.099 (2)	0.036 (2)	-0.0135 (16)	-0.056 (2)
S2	0.086 (2)	0.109 (2)	0.113 (2)	-0.0146 (17)	0.0391 (16)	-0.0114 (16)
N1	0.068 (5)	0.055 (4)	0.047 (4)	0.007 (4)	0.000 (3)	0.009 (3)
N2	0.040 (4)	0.050 (4)	0.043 (3)	0.004 (3)	0.008 (3)	0.002 (3)
N3	0.069 (5)	0.055 (4)	0.044 (4)	0.004 (3)	0.005 (3)	-0.009 (3)
N10	0.059 (5)	0.098 (6)	0.073 (5)	0.002 (5)	-0.015 (4)	-0.002 (4)
N11	0.070 (5)	0.081 (5)	0.064 (5)	-0.001 (4)	0.010 (4)	-0.005 (4)
C1	0.079 (7)	0.109 (8)	0.043 (5)	0.006 (6)	0.000 (4)	0.001 (5)
C2	0.104 (8)	0.075 (6)	0.050 (5)	0.016 (6)	-0.010 (5)	-0.009 (5)
C3	0.076 (6)	0.054 (5)	0.053 (5)	0.017 (5)	-0.006 (4)	-0.001 (4)
C4	0.053 (5)	0.047 (5)	0.044 (4)	0.011 (4)	0.006 (3)	0.001 (4)
C5	0.092 (7)	0.053 (5)	0.050 (5)	0.014 (5)	0.008 (4)	0.000 (4)
C6	0.114 (8)	0.084 (7)	0.049 (5)	0.026 (6)	-0.007 (5)	0.014 (5)
C7	0.041 (4)	0.040 (4)	0.046 (4)	0.002 (4)	0.006 (3)	-0.002 (3)
C8	0.048 (5)	0.042 (4)	0.048 (4)	0.008 (4)	0.001 (3)	-0.001 (3)
C9	0.038 (4)	0.047 (5)	0.049 (4)	-0.002 (4)	0.013 (3)	0.005 (3)
C10	0.037 (4)	0.042 (4)	0.046 (4)	-0.002 (4)	0.005 (3)	0.004 (3)
C11	0.043 (5)	0.055 (5)	0.043 (4)	0.012 (4)	0.001 (3)	0.002 (3)
C12	0.040 (4)	0.055 (5)	0.053 (5)	0.012 (4)	0.015 (3)	0.020 (4)
C13	0.091 (7)	0.049 (5)	0.072 (6)	0.014 (5)	0.015 (5)	0.008 (4)
C14	0.078 (7)	0.057 (6)	0.086 (7)	0.025 (5)	0.016 (5)	0.022 (5)
C15	0.079 (7)	0.093 (8)	0.076 (7)	0.021 (6)	0.008 (5)	0.039 (6)
C16	0.073 (6)	0.101 (8)	0.054 (5)	0.027 (6)	0.000 (4)	0.017 (5)
C17	0.049 (5)	0.050 (5)	0.049 (5)	-0.005 (4)	0.010 (4)	-0.012 (4)
C18	0.069 (6)	0.053 (5)	0.057 (5)	0.014 (4)	-0.001 (4)	-0.008 (4)
C19	0.092 (7)	0.056 (6)	0.072 (6)	0.011 (5)	0.008 (5)	-0.010 (5)
C20	0.079 (7)	0.083 (7)	0.064 (6)	0.003 (6)	0.004 (5)	-0.035 (5)
C21	0.098 (7)	0.072 (7)	0.052 (5)	0.025 (6)	-0.003 (5)	-0.014 (5)
C69	0.103 (8)	0.082 (7)	0.040 (5)	0.025 (7)	0.000 (5)	-0.010 (4)
C70	0.054 (6)	0.065 (6)	0.048 (5)	0.007 (5)	0.008 (4)	-0.008 (4)
Zn2	0.0525 (6)	0.0569 (6)	0.0868 (7)	0.0081 (5)	0.0192 (5)	0.0252 (5)
S3	0.0770 (17)	0.0734 (16)	0.0919 (17)	-0.0114 (13)	0.0371 (13)	0.0076 (12)
S4	0.0651 (18)	0.0662 (18)	0.204 (3)	0.0199 (15)	0.0143 (19)	0.0062 (18)
N4	0.059 (4)	0.038 (4)	0.086 (5)	-0.005 (3)	0.021 (4)	0.009 (3)
N5	0.044 (4)	0.043 (4)	0.061 (4)	0.009 (3)	0.010 (3)	0.005 (3)
N6	0.060 (5)	0.081 (5)	0.070 (5)	0.007 (4)	0.016 (4)	0.025 (4)
N12	0.052 (5)	0.075 (5)	0.100 (6)	0.000 (4)	0.020 (4)	0.027 (4)
N13	0.065 (5)	0.064 (5)	0.097 (5)	0.022 (4)	0.023 (4)	0.021 (4)
C22	0.108 (8)	0.040 (6)	0.107 (8)	0.009 (6)	0.035 (7)	0.008 (5)
C23	0.110 (8)	0.059 (6)	0.074 (6)	0.013 (6)	0.006 (5)	0.011 (5)
C24	0.100 (7)	0.044 (6)	0.070 (6)	-0.007 (5)	0.002 (5)	0.006 (4)

C25	0.045 (5)	0.039 (4)	0.066 (5)	-0.002(4)	0.006 (4)	-0.001 (4)
C26	0.094 (7)	0.058 (6)	0.072 (6)	-0.019 (5)	0.014 (5)	0.002 (5)
C27	0.130 (10)	0.059 (6)	0.095 (8)	-0.029 (6)	0.027 (7)	-0.021 (6)
C28	0.035 (4)	0.046 (5)	0.057 (5)	0.010 (4)	0.014 (3)	0.010 (4)
C29	0.058 (5)	0.052 (5)	0.055 (5)	-0.008 (4)	0.005 (4)	0.004 (4)
C30	0.048 (5)	0.040 (5)	0.058 (5)	0.006 (4)	0.013 (4)	0.002 (4)
C31	0.051 (5)	0.047 (5)	0.065 (5)	0.002 (4)	0.013 (4)	-0.011 (4)
C32	0.047 (5)	0.050 (5)	0.050 (4)	0.009 (4)	0.011 (4)	0.002 (4)
C33	0.048 (5)	0.062 (6)	0.054 (5)	0.002 (4)	0.014 (4)	0.000 (4)
C34	0.060 (6)	0.086 (7)	0.070 (6)	0.013 (5)	0.006 (5)	0.009 (5)
C35	0.079 (7)	0.143 (11)	0.063 (6)	-0.004 (8)	0.004 (5)	0.000 (6)
C36	0.085 (8)	0.154 (12)	0.063 (7)	0.005 (8)	-0.001 (5)	0.037 (8)
C37	0.085 (7)	0.086 (7)	0.081 (7)	-0.005 (6)	0.014 (6)	0.031 (6)
C38	0.059 (5)	0.046 (5)	0.072 (6)	0.000 (4)	0.020 (4)	0.003 (4)
C39	0.072 (6)	0.051 (5)	0.066 (6)	-0.008(5)	0.012 (4)	0.001 (4)
C40	0.082 (7)	0.070 (7)	0.084 (6)	-0.008 (6)	0.009 (5)	-0.006(5)
C41	0.077 (7)	0.058 (6)	0.095 (7)	-0.008(5)	-0.002(6)	-0.009(5)
C43	0.072 (6)	0.037 (5)	0.112 (8)	-0.007(5)	0.019 (6)	0.015 (5)
C65	0.071 (7)	0.027 (4)	0.096 (6)	0.005 (5)	0.006 (5)	0.010 (4)
C66	0.054 (5)	0.053 (5)	0.057 (5)	0.002 (4)	0.002 (4)	0.012 (4)
Zn3	0.0495 (6)	0.0417 (5)	0.0650 (6)	-0.0042(4)	0.0047 (4)	-0.0081 (4)
S5	0.0707 (17)	0.109 (2)	0.1016 (19)	-0.0259 (16)	-0.0078 (14)	-0.0296 (16)
S6	0.0731 (17)	0.0666 (16)	0.121 (2)	0.0156 (13)	0.0259 (15)	-0.0121 (14)
N7	0.053 (4)	0.040 (4)	0.073 (4)	-0.013(3)	0.017 (3)	-0.001 (3)
N8	0.035 (3)	0.039 (3)	0.046 (3)	0.001 (3)	0.007 (3)	-0.003(3)
N9	0.078 (5)	0.053 (4)	0.045 (4)	0.001 (4)	-0.002(3)	-0.006(3)
N14	0.054 (5)	0.059 (5)	0.099 (6)	-0.013 (4)	0.007 (4)	-0.016 (4)
N15	0.068 (5)	0.045 (4)	0.080 (5)	-0.003 (4)	0.000 (4)	-0.014 (3)
C44	0.108 (8)	0.043 (6)	0.103 (8)	-0.024(5)	0.033 (6)	-0.026(5)
C45	0.159 (11)	0.046 (6)	0.082 (7)	-0.018 (6)	0.009 (7)	0.001 (5)
C46	0.103 (7)	0.058 (6)	0.058 (5)	-0.022 (5)	0.002 (5)	-0.002(4)
C47	0.058 (5)	0.040 (5)	0.048 (4)	-0.002 (4)	0.006 (4)	-0.002(3)
C48	0.066 (5)	0.037 (5)	0.061 (5)	0.003 (4)	0.010 (4)	-0.008(4)
C49	0.082 (7)	0.051 (6)	0.071 (5)	-0.007 (5)	0.004 (5)	-0.015 (4)
C50	0.037 (4)	0.037 (4)	0.048 (4)	0.006 (3)	-0.002(3)	-0.005 (3)
C51	0.047 (5)	0.038 (4)	0.053 (4)	0.005 (4)	0.005 (3)	0.005 (3)
C52	0.053 (5)	0.036 (4)	0.050 (4)	0.000 (4)	0.005 (3)	0.004 (3)
C53	0.056 (5)	0.034 (4)	0.041 (4)	0.003 (4)	0.005 (3)	0.001 (3)
C54	0.049 (5)	0.041 (4)	0.042 (4)	-0.001 (4)	0.003 (3)	0.003 (3)
C55	0.050 (5)	0.024 (4)	0.066 (5)	-0.012 (3)	0.014 (4)	-0.003 (3)
C56	0.090 (7)	0.050 (5)	0.057 (5)	-0.020(5)	0.006 (4)	0.003 (4)
C57	0.131 (9)	0.072 (7)	0.056 (5)	-0.021 (6)	0.012 (5)	0.003 (5)
C58	0.114 (8)	0.059 (6)	0.068 (6)	-0.008 (6)	0.020 (5)	0.018 (5)
C59	0.059 (6)	0.038 (5)	0.088 (6)	-0.004 (4)	0.012 (4)	-0.012 (4)
C60	0.046 (5)	0.038 (5)	0.050 (4)	0.011 (4)	0.006 (3)	-0.001 (3)
C61	0.076 (6)	0.051 (5)	0.049 (5)	0.005 (4)	0.006 (4)	-0.001 (4)
C62	0.086 (7)	0.081 (7)	0.054 (5)	-0.004 (6)	0.011 (5)	0.008 (5)
C63	0.130 (9)	0.079 (7)	0.046 (5)	-0.009 (7)	0.001 (5)	-0.008 (5)
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C64	0.113 (8)	0.067 (6)	0.062 (6)	-0.016 (6)	0.003 (5)	-0.010 (5)
C67	0.066 (6)	0.030 (4)	0.059 (5)	0.006 (4)	0.001 (4)	-0.009 (3)
C68	0.079 (7)	0.040 (5)	0.061 (5)	0.002 (5)	0.025 (5)	-0.009 (4)

Geometric parameters (Å, °)

Zn1—N10	1.949 (7)	С27—Н27	0.9300
Zn1—N11	1.975 (8)	C28—C29	1.384 (10)
Zn1—N2	2.071 (5)	C28—C31	1.404 (9)
Zn1—N3	2.181 (6)	C29—C30	1.363 (9)
Zn1—N1	2.219 (6)	С29—Н29	0.9300
S1—C69	1.654 (12)	C30—C38	1.492 (10)
S2C70	1.611 (9)	C31—C32	1.390 (10)
N1—C16	1.339 (9)	C31—H31	0.9300
N1—C12	1.347 (9)	C32—C33	1.481 (10)
N2—C10	1.339 (8)	C33—C34	1.361 (11)
N2—C9	1.353 (8)	C34—C35	1.393 (12)
N3—C21	1.317 (9)	С34—Н34	0.9300
N3—C17	1.328 (8)	C35—C36	1.345 (13)
N10—C69	1.108 (10)	С35—Н35	0.9300
N11—C70	1.125 (9)	C36—C37	1.377 (13)
C1—C2	1.356 (11)	С36—Н36	0.9300
C1—C6	1.375 (12)	С37—Н37	0.9300
C1—H1	0.9300	C38—C39	1.370 (10)
C2—C3	1.383 (10)	C39—C40	1.361 (11)
С2—Н2	0.9300	С39—Н39	0.9300
C3—C4	1.382 (9)	C40—C41	1.376 (12)
С3—Н3	0.9300	C40—H40	0.9300
C4—C5	1.394 (10)	C41—C43	1.362 (11)
C4—C7	1.478 (9)	C41—H41	0.9300
C5—C6	1.373 (10)	C43—H43	0.9300
С5—Н5	0.9300	Zn3—N14	1.958 (7)
С6—Н6	0.9300	Zn3—N15	1.982 (8)
C7—C8	1.391 (9)	Zn3—N8	2.085 (5)
C7—C11	1.391 (9)	Zn3—N7	2.145 (6)
C8—C9	1.368 (9)	Zn3—N9	2.188 (6)
С8—Н8	0.9300	S5—C68	1.616 (10)
C9—C12	1.492 (9)	S6—C67	1.624 (9)
C10—C11	1.390 (8)	N7—C55	1.344 (8)
C10—C17	1.488 (9)	N7—C59	1.358 (9)
C11—H11	0.9300	N8—C54	1.348 (8)
C12—C13	1.371 (10)	N8—C52	1.350 (8)
C13—C14	1.373 (10)	N9—C60	1.335 (8)
С13—Н13	0.9300	N9—C64	1.336 (9)
C14—C15	1.361 (11)	N14—C68	1.128 (9)
C14—H14	0.9300	N15—C67	1.125 (9)
C15—C16	1.386 (12)	C44—C49	1.358 (11)
C15—H15	0.9300	C44—C45	1.371 (12)

C16—H16	0.9300	C44—H44	0.9300
C17—C18	1.370 (10)	C45—C46	1.383 (11)
C18—C19	1.367 (10)	C45—H45	0.9300
C18—H18	0.9300	C46—C47	1.375 (10)
C19—C20	1.376 (10)	C46—H46	0.9300
С19—Н19	0.9300	C47—C48	1.370 (9)
C20—C21	1.351 (11)	C47—C50	1.480 (9)
С20—Н20	0.9300	C48—C49	1.384 (10)
C21—H21	0.9300	C48—H48	0.9300
Zn2—N13	1.951 (8)	C49—H49	0.9300
Zn2—N12	1.963 (7)	C50—C53	1.387 (9)
Zn2—N5	2.086 (6)	C50—C51	1.408 (9)
Zn2—N6	2.195 (7)	C51—C52	1.373 (9)
Zn2—N4	2.209 (7)	C51—H51	0.9300
S3—C66	1.603 (9)	C52—C60	1.483 (9)
S4—C65	1.628 (10)	C53—C54	1.381 (9)
N4—C43	1.336 (10)	С53—Н53	0.9300
N4—C38	1.348 (9)	C54—C55	1.492 (9)
N5—C32	1.324 (9)	C55—C56	1.379 (9)
N5—C30	1.332 (8)	C56—C57	1.360 (11)
N6—C37	1.325 (10)	C56—H56	0.9300
N6—C33	1.344 (9)	C57—C58	1.386 (11)
N12—C66	1.140 (9)	С57—Н57	0.9300
N13—C65	1.122 (9)	C58—C59	1.368 (10)
C22—C23	1.345 (12)	C58—H58	0.9300
C22—C27	1.352 (12)	С59—Н59	0.9300
С22—Н22	0.9300	C60—C61	1.379 (9)
C23—C24	1.399 (11)	C61—C62	1.369 (10)
С23—Н23	0.9300	C61—H61	0.9300
C24—C25	1.372 (10)	C62—C63	1.366 (11)
C24—H24	0.9300	С62—Н62	0.9300
C25—C26	1.364 (10)	C63—C64	1.378 (11)
C25—C28	1.482 (10)	С63—Н63	0.9300
C26—C27	1.390 (12)	C64—H64	0.9300
С26—Н26	0.9300		
N10—Zn1—N11	116.2 (3)	С30—С29—Н29	119.4
N10—Zn1—N2	129.6 (3)	С28—С29—Н29	119.4
N11—Zn1—N2	114.1 (2)	N5-C30-C29	121.6 (7)
N10—Zn1—N3	100.5 (3)	N5-C30-C38	114.3 (6)
N11—Zn1—N3	97.7 (3)	C29—C30—C38	124.0 (7)
N2—Zn1—N3	75.3 (2)	C32—C31—C28	119.7 (7)
N10—Zn1—N1	94.4 (3)	C32—C31—H31	120.1
N11—Zn1—N1	99.3 (3)	C28—C31—H31	120.1
N2—Zn1—N1	74.7 (2)	N5—C32—C31	121.3 (6)
N3—Zn1—N1	149.5 (2)	N5—C32—C33	114.5 (7)
C16—N1—C12	118.5 (7)	C31—C32—C33	124.1 (7)
C16—N1—Zn1	125.2 (6)	N6-C33-C34	120.7 (7)
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116.0 (5)	N6—C33—C32	115.0 (7)
120.2 (6)	C34—C33—C32	124.3 (8)
119.4 (5)	C33—C34—C35	120.8 (9)
120.4 (4)	C33—C34—H34	119.6
118.1 (7)	C35—C34—H34	119.6
125.8 (5)	C36—C35—C34	117.0 (10)
115.7 (5)	C36—C35—H35	121.5
146.8 (8)	C34—C35—H35	121.5
170.4(7)	$C_{35} - C_{36} - C_{37}$	121.5 120.6(9)
1194(8)	$C_{35} = C_{36} = H_{36}$	119.7
119.4 (0)	$C_{33} = C_{30} = H_{30}$	110.7
120.3	$N_{6} = C_{27} = C_{26}$	119.7
120.5	$N_0 = C_3 7 = C_{30}$	121.8 (9)
120.5 (8)	$N_0 - C_3 / - H_3 / C_2 (C_2 - H_3 - H_3)$	119.1
119.7	$C_{30} - C_{37} - H_{37}$	119.1
119.7	N4—C38—C39	120.8 (7)
121.0 (7)	N4—C38—C30	115.1 (7)
119.5	C39—C38—C30	124.0 (7)
119.5	C40—C39—C38	120.9 (8)
117.6 (6)	C40—C39—H39	119.5
121.7 (6)	С38—С39—Н39	119.5
120.8 (6)	C39—C40—C41	117.9 (9)
120.7 (7)	C39—C40—H40	121.0
119.6	C41—C40—H40	121.0
119.6	C43—C41—C40	119.4 (8)
120.6 (8)	C43—C41—H41	120.3
119.7	C40—C41—H41	120.3
119.7	N4—C43—C41	122.6 (8)
116.8 (6)	N4—C43—H43	118 7
121 4 (6)	C41 - C43 - H43	118.7
121.7(6)	N13_C65_S4	175.7(8)
121.7(0) 120.9(6)	N12 C66 S3	176.2(7)
120.9 (0)	N12 - C00 - 55 N14 - 7n3 - N15	170.2(7)
119.5	$\frac{114}{2}$	107.3(3)
119.5	$\frac{1}{14} - \frac{1}{215} - \frac{1}{100}$	140.2(2)
120.9 (6)	N15—Zn5—N8	112.0(2)
114.9 (6)	N14— $Zn3$ — $N/$	101.1(3)
124.2 (7)	N15— $Zn3$ — $N/$	103.8 (3)
120.3 (6)	N8—Zn3—N7	75.5 (2)
114.4 (6)	N14—Zn3—N9	96.3 (3)
125.2 (6)	N15—Zn3—N9	96.7 (3)
120.8 (6)	N8—Zn3—N9	73.7 (2)
119.6	N7—Zn3—N9	147.6 (2)
119.6	C55—N7—C59	116.7 (6)
121.5 (6)	C55—N7—Zn3	116.8 (5)
113.8 (6)	C59—N7—Zn3	126.2 (5)
124.7 (7)	C54—N8—C52	120.2 (6)
120.3 (8)	C54—N8—Zn3	118.8 (4)
119.9	C52—N8—Zn3	120.8 (4)
119.9	C60—N9—C64	116.9 (7)
	116.0(5) $120.2(6)$ $119.4(5)$ $120.4(4)$ $118.1(7)$ $125.8(5)$ $115.7(5)$ $146.8(8)$ $170.4(7)$ $119.4(8)$ 120.3 120.3 120.3 120.3 120.3 $120.5(8)$ 119.7 119.7 119.5 $117.6(6)$ $121.7(6)$ $120.8(6)$ $120.7(7)$ 119.6 $120.6(8)$ 119.7 119.7 119.6 $120.9(6)$ $114.4(6)$ $122.2(7)$ $120.9(6)$ $114.9(6)$ $124.2(7)$ $120.3(6)$ $114.4(6)$ $125.2(6)$ $120.8(6)$ 119.6 $121.5(6)$ $113.8(6)$ $124.7(7)$ $120.3(8)$ 119.9 119.9	116.0 (5) N6-C33-C32 120.2 (6) C34-C33-C32 119.4 (5) C33-C34-C35 120.4 (4) C33-C34-H34 118.1 (7) C35-C34-H34 125.8 (5) C36-C35-C34 115.7 (5) C36-C35-H35 146.8 (8) C34-C35-H35 170.4 (7) C35-C36-C37 119.4 (8) C35-C36-H36 120.3 C37-C36-H36 120.3 C37-C36-H36 120.3 C37-C36-H36 120.3 N6-C37-H37 119.7 C36-C37-H37 119.7 C36-C37-H37 119.7 C36-C37-H37 119.7 N4-C38-C30 119.5 C39-C38-C30 119.5 C39-C40-C41 120.6 (6) C39-C40-C41 120.7 (7) C39-C40-C41 120.8 (6) C39-C40-C41 120.6 (8) C41-C40-H40 119.6 C41-C40-H40 120.6 (8) C43-C41-H41 119.7 C40-C43-H43 121.7 (6) N13-C65-S4 120.9 (6) N12-C66-S3 119.5

C15—C14—C13	118.2 (8)	C60—N9—Zn3	117.0 (4)
C15—C14—H14	120.9	C64—N9—Zn3	125.7 (6)
C13—C14—H14	120.9	C68—N14—Zn3	168.2 (8)
C14—C15—C16	119.9 (8)	C67—N15—Zn3	169.3 (7)
C14—C15—H15	120.0	C49—C44—C45	119.8 (8)
C16—C15—H15	120.0	C49—C44—H44	120.1
N1—C16—C15	121.5 (8)	C45—C44—H44	120.1
N1—C16—H16	119.2	C44—C45—C46	119.6 (8)
C15—C16—H16	119.2	C44—C45—H45	120.2
N3—C17—C18	121.8 (6)	C46—C45—H45	120.2
N3—C17—C10	114.7 (7)	C47—C46—C45	121.5 (8)
C18—C17—C10	123.4 (6)	C47—C46—H46	119.2
C19—C18—C17	119.4 (7)	C45—C46—H46	119.2
C19—C18—H18	120.3	C48—C47—C46	117.5 (7)
C17—C18—H18	120.3	C48—C47—C50	120.6 (6)
C18—C19—C20	118.5 (8)	C46—C47—C50	121.9 (7)
C18—C19—H19	120.7	C47—C48—C49	121.6(7)
C20-C19-H19	120.7	C47—C48—H48	119.2
C_{21} C_{20} C_{19} C_{19}	118 4 (8)	C49—C48—H48	119.2
$C_{21} = C_{20} = H_{20}$	120.8	C44-C49-C48	120.0 (8)
C19 - C20 - H20	120.8	C44—C49—H49	120.0
N3-C21-C20	123.8 (8)	C48—C49—H49	120.0
N3-C21-H21	118.1	C_{53} C_{50} C_{51}	116.8 (6)
C_{20} C_{21} H_{21}	118.1	C_{53} C_{50} C_{47}	121.9 (6)
N10-C69-S1	176 9 (10)	$C_{51} - C_{50} - C_{47}$	121.3 (6)
N11-C70-S2	177.8 (8)	C52-C51-C50	120.0 (6)
N13 - 7n2 - N12	115.6 (3)	C52—C51—H51	120.0
N13— $Zn2$ — $N5$	112.9 (3)	C50—C51—H51	120.0
N12 - Zn2 - N5	131.5 (3)	N8—C52—C51	121.4 (6)
N13 - Zn2 - N6	96.6 (3)	N8—C52—C60	113.2 (6)
N12—Zn2—N6	98.1 (3)	C51—C52—C60	125.4 (6)
N5—Zn2—N6	74.9 (3)	C54—C53—C50	121.5 (6)
N13—Zn2—N4	100.0 (3)	С54—С53—Н53	119.3
N12— $Zn2$ — $N4$	97.6 (3)	С50—С53—Н53	119.3
N5—Zn2—N4	75.1 (2)	N8—C54—C53	120.1 (6)
N6—Zn2—N4	149.5 (2)	N8—C54—C55	114.1 (6)
C43—N4—C38	118.2 (7)	C53—C54—C55	125.8 (6)
C43—N4—Zn2	126.9 (6)	N7—C55—C56	123.4 (6)
C38—N4—Zn2	114.8 (5)	N7—C55—C54	114.1 (6)
C32—N5—C30	119.8 (6)	C56—C55—C54	122.5 (6)
C32—N5—Zn2	119.5 (5)	C57—C56—C55	118.7 (7)
C30—N5—Zn2	119.7 (5)	С57—С56—Н56	120.7
C37—N6—C33	118.9 (8)	С55—С56—Н56	120.7
C37—N6—Zn2	125.9 (7)	C56—C57—C58	119.6 (8)
C33—N6—Zn2	115.1 (5)	C56—C57—H57	120.2
C66—N12—Zn2	170.0 (8)	C58—C57—H57	120.2
C65—N13—Zn2	170.5 (8)	C59—C58—C57	118.6 (8)
C23—C22—C27	119.6 (9)	C59—C58—H58	120.7

C23—C22—H22	120.2	С57—С58—Н58	120.7
С27—С22—Н22	120.2	N7—C59—C58	123.0 (7)
C22—C23—C24	119.6 (9)	N7—C59—H59	118.5
С22—С23—Н23	120.2	С58—С59—Н59	118.5
C24—C23—H23	120.2	N9—C60—C61	122.4 (7)
C25—C24—C23	122.0 (8)	N9—C60—C52	114.2 (6)
C25—C24—H24	119.0	C61—C60—C52	123.2 (7)
C23—C24—H24	119.0	C62—C61—C60	118.6 (8)
C26—C25—C24	116.6 (7)	С62—С61—Н61	120.7
C26—C25—C28	122.4 (7)	С60—С61—Н61	120.7
C_{24} C_{25} C_{28}	120.9 (7)	C63 - C62 - C61	120.8 (8)
C25—C26—C27	121.5 (8)	C63—C62—H62	119.6
C25—C26—H26	119.3	C61—C62—H62	119.6
C27—C26—H26	119.3	C62 - C63 - C64	116.3 (8)
$C_{22} - C_{27} - C_{26}$	120 7 (9)	C62 - C63 - H63	121.9
C22—C27—H27	1197	C64 - C63 - H63	121.9
$C_{26} = C_{27} = H_{27}$	119.7	N9-C64-C63	124.9 (8)
$C_{20} = C_{28} = C_{31}$	116.1 (6)	N9_C64_H64	117 5
$C_{29} = C_{28} = C_{25}$	$122 \ 9 \ (7)$	C_{63} C_{64} H_{64}	117.5
$C_{23} = C_{23} = C_{23}$	122.9(7) 120.9(7)	N15 C67 S6	176.1 (8)
$C_{31} = C_{20} = C_{23}$	120.9(7) 121.2(7)	N13 - C68 - S5	170.1(8) 178.5(9)
030 025 028	121.2(7)	114 000 55	170.5 ())
N10—Zn1—N1—C16	-52.1 (7)	C32—N5—C30—C29	0.0 (10)
N11—Zn1—N1—C16	65.4 (7)	Zn2—N5—C30—C29	-168.9 (6)
N2—Zn1—N1—C16	178.0 (7)	C32—N5—C30—C38	-179.7 (6)
N3—Zn1—N1—C16	-171.5 (6)	Zn2—N5—C30—C38	11.4 (8)
N10—Zn1—N1—C12	133.5 (5)	C28—C29—C30—N5	0.3 (11)
N11—Zn1—N1—C12	-109.1 (5)	C28—C29—C30—C38	179.9 (7)
N2—Zn1—N1—C12	3.5 (5)	C29—C28—C31—C32	5.3 (10)
N3—Zn1—N1—C12	14.1 (8)	C25—C28—C31—C32	-178.6(6)
N10—Zn1—N2—C10	95.6 (6)	C30—N5—C32—C31	2.5 (10)
N11—Zn1—N2—C10	-87.9 (5)	Zn2—N5—C32—C31	171.4 (5)
N3—Zn1—N2—C10	4.1 (5)	C30—N5—C32—C33	179.5 (6)
N1— $Zn1$ — $N2$ — $C10$	178.6 (5)	Zn2—N5—C32—C33	-11.5(8)
N10—Zn1—N2—C9	-84.9 (6)	C28—C31—C32—N5	-5.3 (11)
N11—Zn1—N2—C9	91.6 (5)	C28—C31—C32—C33	178.0 (6)
N3— $Zn1$ — $N2$ — $C9$	-176.4(5)	C37—N6—C33—C34	-0.8(12)
N1— $Zn1$ — $N2$ — $C9$	-1.9(5)	Zn2—N6—C33—C34	-177.6(6)
N10-Zn1-N3-C21	52.8 (8)	C_{37} N6 C_{33} C_{32}	177.8 (7)
$N_{11} - Z_{n1} - N_{3} - C_{21}$	-657(7)	$Zn^2 - N6 - C33 - C32$	10(8)
$N_2 - Z_n - N_3 - C_2 $	-1787(7)	N5-C32-C33-N6	65(9)
N1 - Zn1 - N3 - C21	170.8 (6)	$C_{31} - C_{32} - C_{33} - N_6$	-176.6(7)
N10-7n1-N3-C17	-1346(5)	N_{5} C_{32} C_{33} C_{34}	-175.0(7)
$N11_7n1_N3_C17$	106.9 (5)	C_{31} C_{32} C_{33} C_{34}	19(12)
N_{2}^{-1} N_{3}^{-1} $N_{$	-61(5)	N6-C33-C34-C35	0.1(12)
$N_1 = Zn_1 = N_3 = C_{17}$	-166(8)	C_{32} C_{33} C_{34} C_{35}	-1783(8)
$N11_7n1_N10_C60$	-130.0(15)	$C_{32} = C_{33} = C_{35} = C$	-1.3(14)
$N_2 = 7n1 = N_{10} = C_{00}$	150.0(15)	C_{34} C_{35} C_{36} C_{37}	1.5(14)
IN2-LIII-INIU-C09	40.3 (17)	034 - 033 - 030 - 037	5.1 (10)

N3—Zn1—N10—C69	126.0 (16)	C33—N6—C37—C36	2.7 (13)
N1—Zn1—N10—C69	-27.4 (16)	Zn2—N6—C37—C36	179.1 (7)
N10—Zn1—N11—C70	176 (5)	C35—C36—C37—N6	-4.0 (16)
N2—Zn1—N11—C70	-1(5)	C43—N4—C38—C39	-3.6(11)
N3—Zn1—N11—C70	-78 (5)	Zn2—N4—C38—C39	179.2 (6)
N1—Zn1—N11—C70	77 (5)	C43—N4—C38—C30	-179.6 (7)
C6—C1—C2—C3	2.9 (14)	Zn2—N4—C38—C30	3.1 (8)
C1—C2—C3—C4	0.1 (14)	N5-C30-C38-N4	-9.2 (9)
C2—C3—C4—C5	-1.9 (12)	C29—C30—C38—N4	171.1 (7)
C2—C3—C4—C7	177.2 (7)	N5-C30-C38-C39	174.8 (7)
C3—C4—C5—C6	0.9 (12)	C29—C30—C38—C39	-4.8 (12)
C7—C4—C5—C6	-178.2 (8)	N4—C38—C39—C40	2.9 (12)
C4—C5—C6—C1	2.0 (14)	C30-C38-C39-C40	178.7 (8)
$C_{2}-C_{1}-C_{6}-C_{5}$	-3.9(15)	C38—C39—C40—C41	-2.4(13)
$C_{3}-C_{4}-C_{7}-C_{8}$	-19.4(11)	C39—C40—C41—C43	2.6 (14)
C5—C4—C7—C8	159.7 (7)	C38—N4—C43—C41	3.9 (13)
C3-C4-C7-C11	161.3 (7)	Zn2—N4—C43—C41	-179.2 (6)
C5-C4-C7-C11	-19.6(11)	C40-C41-C43-N4	-3.4(14)
C11—C7—C8—C9	0.0 (10)	Zn2—N13—C65—S4	-62(15)
C4—C7—C8—C9	-179.3(6)	Z_{n2} N12—C66—S3	97 (13)
C10-N2-C9-C8	-1.9(10)	N14—Zn3—N7—C55	146.3 (5)
Zn1—N2—C9—C8	178.5 (5)	N15—Zn3—N7—C55	-102.5(5)
C10 - N2 - C9 - C12	179.7 (6)	N8—Zn3—N7—C55	7.1 (5)
Zn1-N2-C9-C12	0.1 (8)	N9—Zn3—N7—C55	25.1 (8)
C7—C8—C9—N2	1.0 (10)	N14—Zn3—N7—C59	-40.3(6)
C7—C8—C9—C12	179.3 (6)	N15—Zn3—N7—C59	70.9 (6)
C9—N2—C10—C11	1.7 (10)	N8—Zn3—N7—C59	-179.6 (6)
Zn1—N2—C10—C11	-178.7(5)	N9—Zn3—N7—C59	-161.5 (6)
C9—N2—C10—C17	178.7 (6)	N14—Zn3—N8—C54	-97.6 (6)
Zn1—N2—C10—C17	-1.7 (8)	N15—Zn3—N8—C54	91.7 (5)
N2—C10—C11—C7	-0.7 (10)	N7—Zn3—N8—C54	-7.5(5)
C17—C10—C11—C7	-177.3 (6)	N9—Zn3—N8—C54	-177.5(5)
C8—C7—C11—C10	-0.2 (10)	N14—Zn3—N8—C52	88.5 (6)
C4—C7—C11—C10	179.2 (6)	N15—Zn3—N8—C52	-82.2(5)
C16—N1—C12—C13	1.5 (11)	N7—Zn3—N8—C52	178.5 (5)
Zn1—N1—C12—C13	176.3 (6)	N9—Zn3—N8—C52	8.5 (5)
C16—N1—C12—C9	-179.4 (7)	N14—Zn3—N9—C60	-150.2 (5)
Zn1—N1—C12—C9	-4.5 (8)	N15—Zn3—N9—C60	101.5 (5)
N2—C9—C12—N1	3.0 (9)	N8—Zn3—N9—C60	-9.5 (5)
C8—C9—C12—N1	-175.4 (6)	N7—Zn3—N9—C60	-27.8(8)
N2-C9-C12-C13	-177.9 (7)	N14—Zn3—N9—C64	37.2 (7)
C8—C9—C12—C13	3.8 (11)	N15—Zn3—N9—C64	-71.1 (7)
N1—C12—C13—C14	-0.9(12)	N8—Zn3—N9—C64	177.9 (7)
C9—C12—C13—C14	180.0 (7)	N7—Zn3—N9—C64	159.6 (6)
C12—C13—C14—C15	0.3 (13)	N15—Zn3—N14—C68	111 (3)
C13—C14—C15—C16	-0.2 (14)	N8—Zn3—N14—C68	-60 (4)
C12—N1—C16—C15	-1.4 (13)	N7—Zn3—N14—C68	-141 (3)
Zn1—N1—C16—C15	-175.7 (7)	N9—Zn3—N14—C68	12 (3)
			- (-)

C14—C15—C16—N1	0.8 (14)	N14—Zn3—N15—C67	-124 (4)
C21—N3—C17—C18	1.8 (11)	N8—Zn3—N15—C67	50 (4)
Zn1—N3—C17—C18	-171.4 (6)	N7—Zn3—N15—C67	129 (4)
C21—N3—C17—C10	-179.7 (7)	N9—Zn3—N15—C67	-25 (4)
Zn1—N3—C17—C10	7.1 (8)	C49—C44—C45—C46	1.7 (15)
N2-C10-C17-N3	-3.7(9)	C44—C45—C46—C47	-1.8(16)
C11—C10—C17—N3	173.1 (6)	C45—C46—C47—C48	0.3 (13)
N2-C10-C17-C18	174.8 (7)	C45—C46—C47—C50	-177.8(8)
C11—C10—C17—C18	-8.4 (11)	C46—C47—C48—C49	1.3 (12)
N3-C17-C18-C19	-1.5(12)	C50-C47-C48-C49	179.3 (7)
C10-C17-C18-C19	-1799(7)	$C_{45} - C_{44} - C_{49} - C_{48}$	-0.3(14)
C17 - C18 - C19 - C20	0.4(13)	C47-C48-C49-C44	-1.3(13)
C18 - C19 - C20 - C21	0.3(14)	C_{48} C_{47} C_{50} C_{53}	-180(13)
$C_{17} = 0.000 = 0.00000 = 0.00000 = 0.00000 = 0.00000 = 0.00000 = 0.00000 = 0.00000 = 0.00000 = 0.00000 = 0.00000 = 0.00000 = 0.00000 = 0.00000 = 0.00000 = 0.00000000$	-10(14)	$C_{46} - C_{47} - C_{50} - C_{53}$	160.0(11)
7n1 - N3 - C21 - C20	1.0(14) 1714(7)	$C_{48} - C_{47} - C_{50} - C_{51}$	160.0(0)
C19-C20-C21-C20	0.0(15)	C46-C47-C50-C51	-181(11)
7n1 N10 C69 S1	-127(15)	$C_{40} = C_{47} = C_{50} = C_{51}$	-0.4(10)
2n1 - N10 - C09 - S1	-51(27)	$C_{33} = C_{30} = C_{31} = C_{32}$	177.8(6)
2111 - 111 - C/0 - S2	-31(27)	$C_{4} = C_{50} = C_{51} = C_{52}$	-15(10)
N13 - Z12 - N4 - C43	(7)	C_{34} N8 C_{52} C_{51}	-1.3(10)
N12 - Zn2 - N4 - C43	-44.1(7)	$2\pi 3 - \pi 8 - C 52 - C 51$	172.4 (5)
$N_{3} = 2n_{2} = N_{4} = C_{43}$	-1/5.1(7)	$C_{34} = N_{8} = C_{52} = C_{60}$	1/9./(6)
N6-Zn2-N4-C43	-164.3 (6)	2n3 - N8 - C52 - C60	-6.4 (8)
N13 - Zn2 - N4 - C38	-109.3 (6)	C50—C51—C52—N8	1./(11)
N12—Zn2—N4—C38	132.9 (6)	C50—C51—C52—C60	-179.7 (6)
N5—Zn2—N4—C38	1.9 (5)	C51—C50—C53—C54	-1.1 (10)
N6—Zn2—N4—C38	12.7 (8)	C47—C50—C53—C54	-179.3 (6)
N13—Zn2—N5—C32	-81.7 (6)	C52—N8—C54—C53	0.0 (10)
N12—Zn2—N5—C32	96.6 (6)	Zn3—N8—C54—C53	-174.0(5)
N6—Zn2—N5—C32	9.2 (5)	C52—N8—C54—C55	-179.1 (6)
N4—Zn2—N5—C32	-176.5 (6)	Zn3—N8—C54—C55	6.9 (8)
N13—Zn2—N5—C30	87.3 (6)	C50—C53—C54—N8	1.4 (11)
N12—Zn2—N5—C30	-94.5 (6)	C50—C53—C54—C55	-179.7 (7)
N6—Zn2—N5—C30	178.1 (6)	C59—N7—C55—C56	1.3 (11)
N4—Zn2—N5—C30	-7.5 (5)	Zn3—N7—C55—C56	175.3 (6)
N13—Zn2—N6—C37	-69.5 (7)	C59—N7—C55—C54	-179.8 (6)
N12—Zn2—N6—C37	47.5 (7)	Zn3—N7—C55—C54	-5.7 (8)
N5—Zn2—N6—C37	178.4 (7)	N8—C54—C55—N7	-0.5 (9)
N4—Zn2—N6—C37	167.6 (6)	C53—C54—C55—N7	-179.6 (7)
N13—Zn2—N6—C33	106.9 (6)	N8—C54—C55—C56	178.4 (7)
N12—Zn2—N6—C33	-136.0 (6)	C53—C54—C55—C56	-0.6(12)
N5—Zn2—N6—C33	-5.1 (5)	N7—C55—C56—C57	-1.0(13)
N4—Zn2—N6—C33	-15.9 (8)	C54—C55—C56—C57	-179.9 (8)
N13—Zn2—N12—C66	133 (4)	C55—C56—C57—C58	0.6 (14)
N5—Zn2—N12—C66	-45 (4)	C56—C57—C58—C59	-0.4 (15)
N6—Zn2—N12—C66	32 (4)	C55—N7—C59—C58	-1.1 (11)
N4—Zn2—N12—C66	-122 (4)	Zn3—N7—C59—C58	-174.5 (6)
N12—Zn2—N13—C65	-133 (4)	C57—C58—C59—N7	0.7 (14)
N5-Zn2-N13-C65	45 (5)	C64—N9—C60—C61	-2.1(11)
	(-)		()

N6—Zn2—N13—C65	-31 (5)	Zn3—N9—C60—C61	-175.4 (6)
N4—Zn2—N13—C65	123 (4)	C64—N9—C60—C52	-177.4 (7)
C27—C22—C23—C24	-0.4 (15)	Zn3—N9—C60—C52	9.3 (8)
C22—C23—C24—C25	1.5 (14)	N8—C52—C60—N9	-2.3 (9)
C23—C24—C25—C26	-1.8 (13)	C51—C52—C60—N9	179.0 (7)
C23—C24—C25—C28	-178.8 (8)	N8-C52-C60-C61	-177.5 (7)
C24—C25—C26—C27	1.1 (13)	C51—C52—C60—C61	3.8 (11)
C28—C25—C26—C27	178.0 (8)	N9-C60-C61-C62	1.9 (12)
C23—C22—C27—C26	-0.3 (16)	C52—C60—C61—C62	176.8 (7)
C25—C26—C27—C22	-0.1 (15)	C60—C61—C62—C63	-1.4 (13)
C26—C25—C28—C29	-170.9 (8)	C61—C62—C63—C64	1.1 (14)
C24—C25—C28—C29	5.9 (11)	C60—N9—C64—C63	2.0 (13)
C26—C25—C28—C31	13.3 (11)	Zn3—N9—C64—C63	174.6 (8)
C24—C25—C28—C31	-169.9 (7)	C62—C63—C64—N9	-1.5 (15)
C31—C28—C29—C30	-2.9 (10)	Zn3—N15—C67—S6	-106 (11)
C25—C28—C29—C30	-178.9 (7)	Zn3—N14—C68—S5	0 (36)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H··· <i>A</i>	D—H	H···A	D····A	<i>D</i> —H··· <i>A</i>
C5—H5…S6 ⁱ	0.93	2.87	3.748 (8)	159
C41—H41···S6 ⁱⁱ	0.93	2.82	3.612 (9)	144

Symmetry codes: (i) x+1/2, -y+1/2, z-1/2; (ii) -x+1/2, y-1/2, -z+3/2.