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# (*E*)-2-Methyl-6-{[(5-methylpyridin-2-yl)imino]methyl}phenol

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In the title compound,  $C_{14}H_{14}N_2O$ , the dihedral angle between the aromatic rings is 5.54 (9)°. The conformation is reinforced by an intramolecular  $O - H \cdots N$  hydrogen bond, which closes an S(6) ring. The pyridine N atom and methyl group lie to opposite sides of the molecule. In the crystal, the molecules are linked into a zigzag chain propagating in  $[0\overline{1}1]$  by weak  $C - H \cdots O$  hydrogen bonds.



## **Structure description**

As part of our ongoing studies of phenolic Schiff-base compounds (Adam *et al.* (2015), we now describe the synthesis and structure of the title compound (Fig. 1), which features an intramolecular  $O-H \cdots N$  hydrogen bond (Table 1), which helps to establish near-coplanarity of the aromatic rings [dihedral angle = 5.54 (9)°]. In the crystal, the molecules are linked by a C2–H2A–O1 hydrogen bond into a zigzag *C*(9) chain propagating in [011] (Table 1, Fig. 2). Adjacent molecules in the chain are related by *c*-glide symmetry.

Synthesis and crystallization

The synthesis scheme is shown in Fig. 3. 2-Hydroxy-3-methylbenzaldehyde (0.681 g, 5 mmol) was dissolved in 20 ml toluene and after adding 0.2 ml acetic acid, the mixture was refluxed for 30 min. Then, 5-methylpyridin-2-amine 0.541 g (5 mmol) solution in 20 ml toluene was added dropwise with stirring to the aldehyde solution. The resulting light-yellow solution was refluxed for 4 h with stirring. The solvent was allowed to evaporate. The crude product was washed with benzene and *n*-hexane. Orange blocks were obtained by slow evaporation of a solution in toluene; m.p.: 363–364 K; yield: 95%.





Figure 1

The molecular structure of the title compound showing 50% displacement ellipsoids.



Figure 2 The packing of the title compound viewed along [010].



Figure 3 Synthesis of the title compound

## Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

## Acknowledgements

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Table 1		
Hydrogen-bond geometry	′ (Å, °).	

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
$O1-H101\cdots N2$	0.87 (3)	1.82 (3)	2.590 (2)	146 (3)
$C2-H2A\cdots O1^{i}$	0.95	2.52	3.322 (3)	142

Symmetry code: (i)  $-x + \frac{3}{2}, y + 1, z - \frac{1}{2}$ .

# Table 2Experimental details.

-	
Crystal data	
Chemical formula	$C_{14}H_{14}N_2O$
Mr	226.27
Crystal system, space group	Orthorhombic, $Pca2_1$
Temperature (K)	100
a, b, c (Å)	23.440 (3), 4.6307 (6), 10.6408 (13)
$V(A^3)$	1155.0 (3)
Z	4
Radiation type	Μο Κα
$\mu (\mathrm{mm}^{-1})^{31}$	0.08
Crystal size (mm)	$0.59\times0.18\times0.14$
Data collection	
Diffractometer	Bruker APEXII CCD
Absorption correction	-
No. of measured, independent and observed $[I > 2\sigma(I)]$ reflections	9247, 3106, 2792
R <sub>int</sub>	0.031
$(\sin \theta / \lambda)_{\max} ( \text{\AA}^{-1} )$	0.685
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.041, 0.112, 1.09
No. of reflections	3106
No. of parameters	160
No. of restraints	1
H-atom treatment	H atoms treated by a mixture of independent and constrained
$\Lambda_0 \qquad \Lambda_0 \qquad (e \stackrel{\circ}{A}^{-3})$	0.25 = 0.18
$\Delta \rho \max$ , $\Delta \rho \min (0.11)$	0.20, 0.10

Computer programs: APEX2 and SAINT (Bruker, 2004), SHELXS97 and SHELXTL (Sheldrick, 2008), SHELXL2014/7 (Sheldrick, 2015).

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# full crystallographic data

## *IUCrData* (2017). **2**, x162009 [https://doi.org/10.1107/S2414314616020095]

# (E)-2-Methyl-6-{[(5-methylpyridin-2-yl)imino]methyl}phenol

## Md. Azharul Arafath, Farook Adam and Mohd. R. Razali

(E)-2-Methyl-6-{[(5-methylpyridin-2-yl)imino]methyl}phenol

Crystal data	
$C_{14}H_{14}N_{2}O$ $M_{r} = 226.27$ Orthorhombic, $Pca2_{1}$ $a = 23.440 (3) Å$ $b = 4.6307 (6) Å$ $c = 10.6408 (13) Å$ $V = 1155.0 (3) Å^{3}$ $Z = 4$ $F(000) = 480$	$D_x = 1.301 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ Å}$ Cell parameters from 2856 reflections $\theta = 2.6-29.1^{\circ}$ $\mu = 0.08 \text{ mm}^{-1}$ T = 100  K Block, orange $0.59 \times 0.18 \times 0.14 \text{ mm}$
Data collection	
Bruker APEXII CCD diffractometer $\varphi$ and $\omega$ scans 9247 measured reflections 3106 independent reflections 2792 reflections with $I > 2\sigma(I)$	$R_{int} = 0.031$ $\theta_{max} = 29.1^{\circ}, \ \theta_{min} = 1.7^{\circ}$ $h = -31 \rightarrow 31$ $k = -6 \rightarrow 6$ $l = -14 \rightarrow 14$
Refinement	
Refinement on $F^2$ Least-squares matrix: full $R[F^2 > 2\sigma(F^2)] = 0.041$ $wR(F^2) = 0.112$ S = 1.09 3106 reflections 160 parameters 1 restraint	Primary atom site location: structure-invariant direct methods Hydrogen site location: mixed H atoms treated by a mixture of independent and constrained refinement $w = 1/[\sigma^2(F_o^2) + (0.0653P)^2 + 0.0393P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 0.25$ e Å <sup>-3</sup> $\Delta\rho_{min} = -0.18$ e Å <sup>-3</sup>

## Special details

**Geometry**. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(Å^2)$ 

	x	у	Ζ	$U_{\rm iso}$ */ $U_{\rm eq}$
01	0.65226 (6)	0.0538 (3)	0.69407 (15)	0.0236 (4)

N1	0.61387 (7)	0.6480 (3)	0.31428 (17)	0.0194 (4)
N2	0.63432 (7)	0.3591 (4)	0.49405 (16)	0.0177 (4)
C1	0.70576 (8)	0.6813 (4)	0.4124 (2)	0.0204 (4)
H1A	0.7307	0.6230	0.4781	0.025*
C2	0.72340 (9)	0.8817 (4)	0.3237 (2)	0.0212 (4)
H2A	0.7608	0.9600	0.3273	0.025*
C3	0.68583 (8)	0.9677 (4)	0.22889 (19)	0.0185 (4)
C4	0.63168 (9)	0.8438 (4)	0.2302 (2)	0.0197 (4)
H4A	0.6055	0.9022	0.1669	0.024*
C5	0.65087 (8)	0.5669 (4)	0.40364 (19)	0.0170 (4)
C6	0.70294 (9)	1.1822 (4)	0.1299 (2)	0.0228 (4)
H6A	0.6696	1.2327	0.0788	0.034*
H6B	0.7324	1.0974	0.0760	0.034*
H6C	0.7180	1.3566	0.1702	0.034*
C7	0.58560 (9)	0.2307 (4)	0.48190 (19)	0.0178 (4)
H7A	0.5619	0.2782	0.4125	0.021*
C8	0.56627 (8)	0.0167 (4)	0.57120 (18)	0.0168 (4)
C9	0.51292 (9)	-0.1140 (4)	0.5548 (2)	0.0205 (4)
H9A	0.4899	-0.0613	0.4849	0.025*
C10	0.49335 (9)	-0.3183 (4)	0.6386 (2)	0.0232 (4)
H10A	0.4570	-0.4051	0.6272	0.028*
C11	0.52755 (9)	-0.3965 (4)	0.7406 (2)	0.0222 (4)
H11A	0.5139	-0.5377	0.7981	0.027*
C12	0.58079 (9)	-0.2740 (4)	0.76055 (19)	0.0205 (4)
C13	0.60016 (8)	-0.0652 (4)	0.67508 (19)	0.0181 (4)
C14	0.61778 (10)	-0.3582 (5)	0.8700 (2)	0.0290 (5)
H14A	0.6269	-0.1862	0.9198	0.043*
H14B	0.6532	-0.4453	0.8388	0.043*
H14C	0.5974	-0.4980	0.9226	0.043*
H101	0.6583 (13)	0.190 (7)	0.639 (3)	0.046 (9)*

Atomic displacement parameters  $(Å^2)$ 

_	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
01	0.0218 (7)	0.0269 (8)	0.0221 (8)	-0.0048 (6)	-0.0036 (6)	0.0055 (7)
N1	0.0210 (8)	0.0201 (8)	0.0170 (8)	0.0000 (7)	-0.0007 (7)	0.0006 (7)
N2	0.0182 (8)	0.0172 (8)	0.0178 (8)	0.0001 (6)	0.0024 (6)	0.0005 (7)
C1	0.0190 (10)	0.0206 (9)	0.0217 (11)	-0.0005 (7)	-0.0031 (8)	0.0039 (8)
C2	0.0204 (9)	0.0210 (9)	0.0222 (10)	-0.0015 (8)	0.0014 (8)	0.0014 (9)
C3	0.0233 (10)	0.0163 (9)	0.0159 (9)	0.0013 (7)	0.0029 (7)	-0.0011 (8)
C4	0.0234 (10)	0.0186 (9)	0.0170 (9)	0.0021 (7)	-0.0009 (8)	0.0006 (8)
C5	0.0211 (9)	0.0148 (9)	0.0150 (9)	0.0005 (7)	0.0013 (7)	-0.0003 (7)
C6	0.0288 (11)	0.0200 (10)	0.0196 (10)	0.0007 (8)	0.0037 (8)	0.0039 (8)
C7	0.0197 (9)	0.0168 (8)	0.0169 (10)	0.0018 (7)	0.0016 (7)	-0.0009 (8)
C8	0.0184 (9)	0.0165 (8)	0.0156 (10)	0.0020 (7)	0.0023 (7)	-0.0006 (7)
C9	0.0199 (10)	0.0195 (9)	0.0222 (10)	0.0016 (7)	0.0009 (8)	-0.0015 (8)
C10	0.0190 (9)	0.0224 (10)	0.0281 (11)	-0.0001 (8)	0.0034 (8)	-0.0007(9)
C11	0.0268 (10)	0.0183 (9)	0.0217 (11)	0.0008 (8)	0.0081 (8)	0.0008 (8)

# data reports

C12	0.0248 (10)	0.0209 (10)	0.0158 (10)	0.0022 (8)	0.0030 (8)	0.0014 (8)
C13	0.0195 (9)	0.0183 (8)	0.0164 (10)	0.0012 (7)	0.0014 (7)	-0.0025 (8)
C14	0.0349 (12)	0.0318 (12)	0.0202 (11)	-0.0005 (9)	-0.0005 (9)	0.0069 (10)

Geometric parameters (Å, °)

01—C13	1.355 (2)	С6—Н6С	0.9800	
O1—H101	0.87 (4)	C7—C8	1.446 (3)	
N1-C4	1.341 (3)	C7—H7A	0.9500	
N1C5	1.341 (3)	C8—C9	1.401 (3)	
N2—C7	1.294 (3)	C8—C13	1.413 (3)	
N2—C5	1.415 (2)	C9—C10	1.379 (3)	
C1—C2	1.387 (3)	С9—Н9А	0.9500	
C1—C5	1.395 (3)	C10-C11	1.397 (3)	
C1—H1A	0.9500	C10—H10A	0.9500	
С2—С3	1.397 (3)	C11—C12	1.387 (3)	
C2—H2A	0.9500	C11—H11A	0.9500	
C3—C4	1.393 (3)	C12—C13	1.403 (3)	
С3—С6	1.503 (3)	C12—C14	1.503 (3)	
C4—H4A	0.9500	C14—H14A	0.9800	
С6—Н6А	0.9800	C14—H14B	0.9800	
С6—Н6В	0.9800	C14—H14C	0.9800	
C13-O1-H101	110 (2)	С8—С7—Н7А	119.1	
C4—N1—C5	117.50 (17)	C9—C8—C13	118.92 (17)	
C7—N2—C5	119.12 (17)	C9—C8—C7	119.61 (18)	
C2—C1—C5	118.93 (19)	C13—C8—C7	121.47 (17)	
C2—C1—H1A	120.5	C10—C9—C8	120.8 (2)	
C5—C1—H1A	120.5	С10—С9—Н9А	119.6	
C1—C2—C3	119.62 (18)	С8—С9—Н9А	119.6	
C1—C2—H2A	120.2	C9—C10—C11	119.32 (19)	
C3—C2—H2A	120.2	C9—C10—H10A	120.3	
C4—C3—C2	116.74 (18)	C11—C10—H10A	120.3	
C4—C3—C6	121.49 (18)	C12—C11—C10	121.93 (19)	
С2—С3—С6	121.77 (18)	C12—C11—H11A	119.0	
N1—C4—C3	124.66 (19)	C10-C11-H11A	119.0	
N1—C4—H4A	117.7	C11—C12—C13	118.30 (19)	
C3—C4—H4A	117.7	C11—C12—C14	122.09 (19)	
N1-C5-C1	122.52 (18)	C13—C12—C14	119.61 (19)	
N1-C5-N2	119.71 (17)	O1—C13—C12	118.38 (18)	
C1—C5—N2	117.77 (18)	O1—C13—C8	120.94 (17)	
С3—С6—Н6А	109.5	C12—C13—C8	120.67 (17)	
С3—С6—Н6В	109.5	C12—C14—H14A	109.5	
H6A—C6—H6B	109.5	C12—C14—H14B	109.5	
С3—С6—Н6С	109.5	H14A—C14—H14B	109.5	
H6A—C6—H6C	109.5	C12—C14—H14C	109.5	
H6B—C6—H6C	109.5	H14A—C14—H14C	109.5	
N2	121.73 (18)	H14B—C14—H14C	109.5	

N2—C7—H7A	119.1		
C5-C1-C2-C3 $C1-C2-C3-C4$ $C1-C2-C3-C4$ $C5-N1-C4-C3$ $C2-C3-C4-N1$ $C6-C3-C4-N1$ $C4-N1-C5-C1$ $C4-N1-C5-N2$ $C2-C1-C5-N1$ $C2-C1-C5-N2$ $C7-N2-C5-N1$ $C7-N2-C5-C1$ $C5-N2-C7-C8$ $N2-C7-C8-C9$	$\begin{array}{c} 0.9 (3) \\ 0.2 (3) \\ -179.50 (18) \\ 0.2 (3) \\ -0.8 (3) \\ 178.89 (19) \\ 1.0 (3) \\ -179.48 (17) \\ -1.6 (3) \\ 178.90 (18) \\ 6.5 (3) \\ -174.01 (18) \\ -179.94 (17) \\ 179.30 (18) \\ 0.0 (18) \end{array}$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.4 (3) -179.86 (18) -0.5 (3) 0.2 (3) 0.2 (3) -179.8 (2) -179.71 (18) 0.3 (3) -0.3 (3) 179.73 (18) 179.42 (17) -0.3 (3) 0.0 (3) -179.77 (18)
N2-C/-C0-C15	0.9 (3)		

# Hydrogen-bond geometry (Å, °)

<i>D</i> —H··· <i>A</i>	D—H	Н…А	D····A	<i>D</i> —H··· <i>A</i>
O1—H101…N2	0.87 (3)	1.82 (3)	2.590 (2)	146 (3)
C2—H2A····O1 <sup>i</sup>	0.95	2.52	3.322 (3)	142

Symmetry code: (i) -*x*+3/2, *y*+1, *z*-1/2.