

# Bis(2-formylphenyl) benzene-1,2-dicarboxylate

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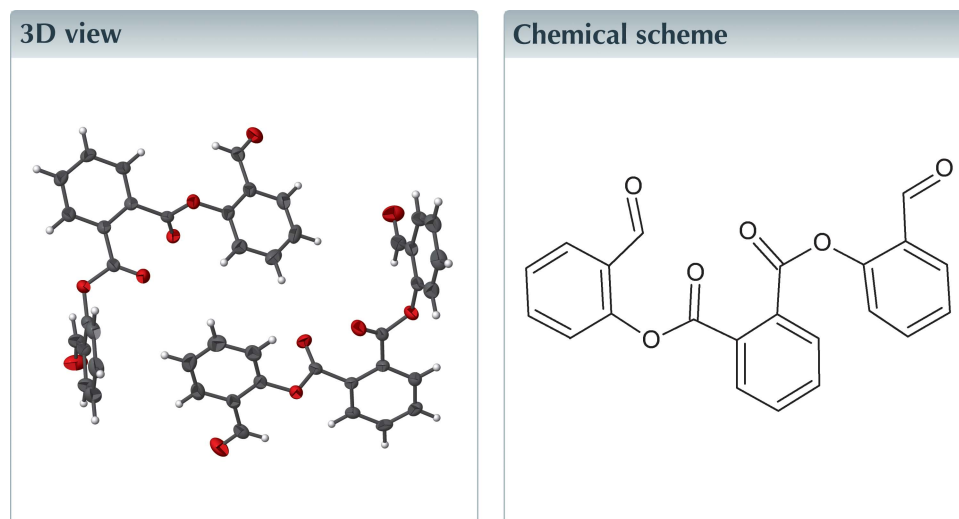
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Keywords: crystal structure; hydrogen bond;  $\pi$ - $\pi$  stacking; bis-aldehydes.

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Structural data: full structural data are available from [iucrdata.iucr.org](http://iucrdata.iucr.org)

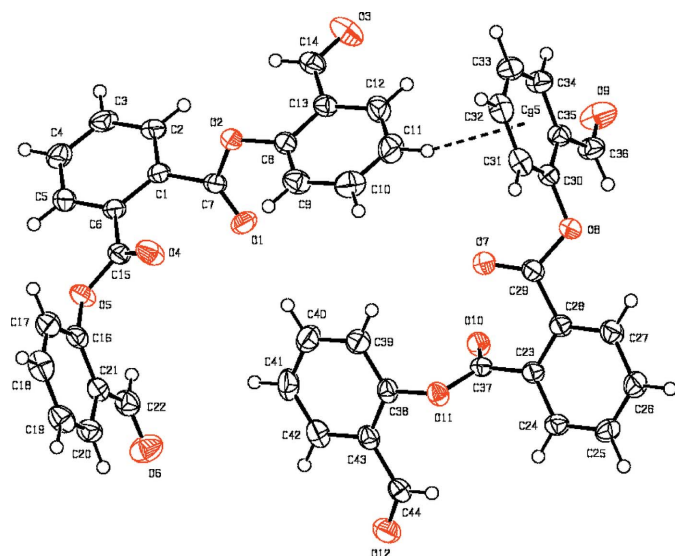
The asymmetric unit of the title compound,  $C_{22}H_{14}O_6$ , consists of two independent molecules differing in the orientations of the ester groups. In one molecule, the two terminal benzene rings are inclined to the central benzene ring by 4.99 (13) and 77.46 (13)°, while in the other the corresponding angles are 11.03 (13) and 88.09 (12)°. In the crystal, molecules are connected into a ribbon structure running along [101] via C—H...O and C—H... $\pi$  interactions. Adjacent ribbons are further linked by additional C—H...O and C—H... $\pi$  interactions. The crystal studied was a non-merohedral twin [twin law (0.986 – 0.073 – 0.008, 0.323 1.036 0.148, –0.121 – 0.102 0.942)], the ratio of components being 0.937 (4):0.063 (4).



## Structure description

The synthesis of bis-aldehydes is of great importance as this bi-functional unit is a useful synthon for the preparation of different intermediates such as bis-chalcones and bis-imines, which can be easily converted to many important pharmaceutical and biologically active heterocyclic compounds such as pyrazolines (Hawaiz & Shekh Omer, 2017) and thiazolidinones (Hussein & Azeez, 2013). Based on such findings, we herein report the synthesis and crystal structure of the title compound.

The asymmetric unit of the title compound consists of two independent molecules, which differ in the orientations of the ester groups (Fig. 1). In one molecule, the terminal C8–C13 and C16–C21 benzene rings are inclined to the central C1–C6 ring by 4.99 (13) and 77.46 (13)°, respectively, while in the other the C30–C35 and C38–C43 benzene rings are inclined to the C23–C28 ring by 88.09 (12) and 11.03 (13)°, respectively.

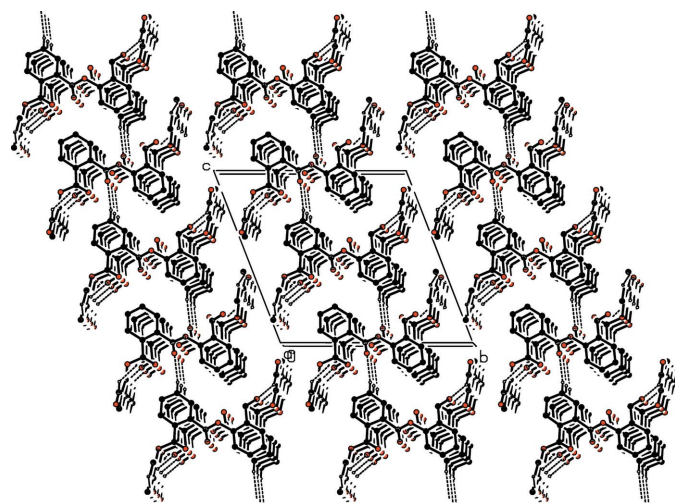


**Figure 1**  
The asymmetric unit of the title compound with labeling scheme and 50% probability ellipsoids. The C–H··· $\pi$  interaction is shown as a dashed line.

In the crystal (Fig. 2), the molecules are linked into a ribbon structure running along [101] via three predominant C–H···O hydrogen bonds and a C–H··· $\pi$  interaction (C3–H3···O10<sup>i</sup>, C22–H22···O3<sup>i</sup>, C36–H36···O12<sup>iii</sup> and C11–H11···Cg5; symmetry codes as in Table 1; Cg5 is the centroid of the C30–C35 benzene ring). The ribbons are further linked by other weak C–H···O and C–H··· $\pi$  interactions (C18–H18···O6<sup>ii</sup>, C32–H32···O9<sup>ii</sup> and C26–H26···Cg3<sup>iv</sup>; Table 1; Cg3 is the centroid of the C16–C21 benzene ring).

### Synthesis and crystallization

A mixture of phthaloyl chloride (0.5 g, 0.0025 mol), 2-hydroxybenzaldehyde (0.61 g, 0.005 mol) and triethylamine



**Figure 2**  
A packing diagram viewed along the *a* axis, showing the C–H···O hydrogen bonds (dashed lines).

**Table 1**  
Hydrogen-bond geometry (Å, °).

Cg3 and Cg5 are the centroids of the C16–C21 and C30–C35 benzene rings, respectively.

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
C3–H3···O10 <sup>i</sup>	0.95	2.35	3.221 (3)	152
C18–H18···O6 <sup>ii</sup>	0.95	2.55	3.430 (3)	155
C22–H22···O3 <sup>i</sup>	0.95	2.29	3.147 (3)	149
C32–H32···O9 <sup>ii</sup>	0.95	2.54	3.422 (3)	155
C36–H36···O12 <sup>iii</sup>	0.95	2.41	3.205 (3)	141
C11–H11···Cg5	0.95	2.75	3.664 (3)	162
C26–H26···Cg3 <sup>iv</sup>	0.95	2.68	3.546 (3)	152

Symmetry codes: (i)  $-x + 1, -y + 1, -z + 1$ ; (ii)  $x + 1, y, z$ ; (iii)  $-x, -y + 1, -z$ ; (iv)  $-x + 1, -y + 1, -z$ .

**Table 2**  
Experimental details.

Crystal data	
Chemical formula	C <sub>22</sub> H <sub>14</sub> O <sub>6</sub>
<i>M<sub>r</sub></i>	374.33
Crystal system, space group	Triclinic, <i>P</i> $\bar{1}$
Temperature (K)	150
<i>a</i> , <i>b</i> , <i>c</i> (Å)	8.1833 (2), 15.4412 (4), 15.5250 (4)
$\alpha$ , $\beta$ , $\gamma$ (°)	110.657 (1), 104.919 (1), 90.516 (1)
<i>V</i> (Å <sup>3</sup> )	1762.89 (8)
<i>Z</i>	4
Radiation type	Cu <i>K</i> $\alpha$
$\mu$ (mm <sup>-1</sup> )	0.87
Crystal size (mm)	0.17 × 0.12 × 0.08
Data collection	
Diffractometer	Bruker D8 VENTURE PHOTON 100 CMOS
Absorption correction	Multi-scan ( <i>TWINABS</i> ; Sheldrick, 2009)
<i>T<sub>min</sub></i> , <i>T<sub>max</sub></i>	0.87, 0.94
No. of measured, independent and observed [ <i>I</i> > 2 $\sigma$ ( <i>I</i> )] reflections	26732, 26732, 16950
<i>R<sub>int</sub></i>	0.032
( <i>sin</i> $\theta$ / $\lambda$ ) <sub>max</sub> (Å <sup>-1</sup> )	0.619
Refinement	
<i>R</i> [ <i>F</i> <sup>2</sup> > 2 $\sigma$ ( <i>F</i> <sup>2</sup> )], <i>wR</i> ( <i>F</i> <sup>2</sup> ), <i>S</i>	0.054, 0.153, 1.03
No. of reflections	26732
No. of parameters	507
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{max}$ , $\Delta\rho_{min}$ (e Å <sup>-3</sup> )	0.49, -0.37

Computer programs: *APEX3* and *SAINT* (Bruker, 2016), *SHELXT2014* (Sheldrick, 2015a), *SHELXL2018* (Sheldrick, 2015b), *ORTEP-3 for Windows* and *WinGX* (Farrugia, 2012) and *PLATON* (Spek, 2015).

(0.5 g, 0.005 mol) in dioxane (20 ml) was stirred for a few minutes. The solid amine salt (Et<sub>3</sub>N·HCl) was separated by filtration and the filtrate was refluxed for 30 min. On cooling, good quality crystals of the title compound were obtained, filtered and dried (0.9 g, 92%).

### Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2. The crystal studied was a non-merohedral twin [twin law (0.986 – 0.073 – 0.008, 0.323 1.036 0.148, -0.121 – 0.102 0.942)], the refined ratio of the two domains being 0.937 (4) and 0.063 (4).

### Funding information

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### References

Bruker (2016). *APEX3* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.

- Farrugia, L. J. (2012). *J. Appl. Cryst.* **45**, 849–854.  
Hawaiz, F. E. & Shekh Omer, D. A. (2017). *ARO - -The Scientific Journal of Koya University*, **5**, 30–35.  
Hussein, A. J. & Azeez, H. J. (2013). *J. Chem.*, **2013**, 1–6.  
Sheldrick, G. M. (2009). *TWINABS*. University of Göttingen, Göttingen, Germany.  
Sheldrick, G. M. (2015a). *Acta Cryst.* **A71**, 3–8.  
Sheldrick, G. M. (2015b). *Acta Cryst.* **C71**, 3–8.  
Spek, A. L. (2015). *Acta Cryst.* **C71**, 9–18.

## full crystallographic data

*IUCrData* (2018). 3, x180251 [https://doi.org/10.1107/S2414314618002511]

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## Bis(2-formylphenyl) benzene-1,2-dicarboxylate

*Crystal data*

$C_{22}H_{14}O_6$

$M_r = 374.33$

Triclinic,  $P\bar{1}$

$a = 8.1833$  (2) Å

$b = 15.4412$  (4) Å

$c = 15.5250$  (4) Å

$\alpha = 110.657$  (1)°

$\beta = 104.919$  (1)°

$\gamma = 90.516$  (1)°

$V = 1762.89$  (8) Å<sup>3</sup>

$Z = 4$

$F(000) = 776$

$D_x = 1.410$  Mg m<sup>-3</sup>

Cu  $K\alpha$  radiation,  $\lambda = 1.54178$  Å

Cell parameters from 9890 reflections

$\theta = 3.1$ – $72.5$ °

$\mu = 0.87$  mm<sup>-1</sup>

$T = 150$  K

Block, colourless

$0.17 \times 0.12 \times 0.08$  mm

*Data collection*

Bruker D8 VENTURE PHOTON 100 CMOS  
diffractometer

Radiation source: INCOATEC  $I\mu$ S micro-focus  
source

Mirror monochromator

Detector resolution: 10.4167 pixels mm<sup>-1</sup>

$\omega$  scans

Absorption correction: multi-scan  
(*TWINABS*; Sheldrick, 2009)

$T_{\min} = 0.87$ ,  $T_{\max} = 0.94$

26732 measured reflections

26732 independent reflections

16950 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.032$

$\theta_{\max} = 72.6$ °,  $\theta_{\min} = 3.2$ °

$h = -10 \rightarrow 9$

$k = -18 \rightarrow 19$

$l = -19 \rightarrow 19$

*Refinement*

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.054$

$wR(F^2) = 0.153$

$S = 1.03$

26732 reflections

507 parameters

0 restraints

Hydrogen site location: inferred from  
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0601P)^2 + 0.2628P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.49$  e Å<sup>-3</sup>

$\Delta\rho_{\min} = -0.36$  e Å<sup>-3</sup>

Extinction correction: *SHELXL2018* (Sheldrick,  
2015b),  $F_c^* = kFc[1 + 0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.0039 (5)

*Special details*

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** Refined as a 2-component twin. Analysis of 1273 reflections having  $I/\sigma(I) > 14$  chosen from the full data set with CELL\_NOW indicated the crystal to belong to the triclinic system and to contain two components rotated by  $10.9^\circ$  about the real axis [1 -1/4 -3/4]. The refined twin law was (0.986 -0.073 -0.008 0.323 1.036 0.148 -0.121 -0.102 0.942). There were 27338 single reflections, 13636 from component 1 and 13622 from component 2 and 126 full or partial overlaps. The refined twin fractions are 0.937 (4) and 0.063 (4).

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.5734 (2)	0.42829 (12)	0.41266 (11)	0.0428 (4)
O2	0.80387 (19)	0.52408 (10)	0.51789 (10)	0.0311 (3)
O3	0.6591 (3)	0.77157 (12)	0.65877 (14)	0.0543 (5)
O4	0.8409 (3)	0.30317 (11)	0.36575 (11)	0.0461 (5)
O5	0.7698 (2)	0.16274 (10)	0.36799 (10)	0.0356 (4)
O6	0.3825 (2)	0.04440 (15)	0.12472 (13)	0.0543 (5)
C1	0.7376 (3)	0.38678 (15)	0.54099 (14)	0.0300 (5)
C2	0.7252 (3)	0.42400 (16)	0.63447 (15)	0.0355 (5)
H2	0.699632	0.486288	0.659733	0.043*
C3	0.7502 (3)	0.37001 (18)	0.69075 (15)	0.0405 (6)
H3	0.741109	0.395358	0.754464	0.049*
C4	0.7882 (3)	0.27987 (17)	0.65466 (16)	0.0419 (6)
H4	0.807154	0.243657	0.693953	0.050*
C5	0.7989 (3)	0.24170 (16)	0.56124 (16)	0.0370 (5)
H5	0.823710	0.179185	0.536468	0.044*
C6	0.7734 (3)	0.29486 (15)	0.50353 (14)	0.0304 (5)
C7	0.6941 (3)	0.44492 (14)	0.48077 (14)	0.0301 (5)
C8	0.7694 (3)	0.58694 (15)	0.46973 (15)	0.0308 (5)
C9	0.7879 (3)	0.56301 (17)	0.37874 (17)	0.0398 (5)
H9	0.823531	0.504487	0.347907	0.048*
C10	0.7532 (4)	0.62625 (19)	0.33305 (18)	0.0473 (6)
H10	0.764553	0.610711	0.270116	0.057*
C11	0.7024 (4)	0.71172 (19)	0.3783 (2)	0.0485 (6)
H11	0.677560	0.754195	0.346147	0.058*
C12	0.6877 (3)	0.73525 (17)	0.47004 (19)	0.0415 (6)
H12	0.654548	0.794425	0.501197	0.050*
C13	0.7211 (3)	0.67281 (15)	0.51767 (16)	0.0322 (5)
C14	0.7065 (3)	0.69903 (16)	0.61646 (17)	0.0366 (5)
H14	0.736100	0.656566	0.648284	0.044*
C15	0.7984 (3)	0.25722 (15)	0.40591 (15)	0.0321 (5)
C16	0.8110 (3)	0.11665 (14)	0.28143 (15)	0.0312 (5)
C17	0.9792 (3)	0.10939 (17)	0.28322 (18)	0.0389 (5)
H17	1.066574	0.136746	0.341619	0.047*

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C18	1.0193 (3)	0.06153 (18)	0.19846 (19)	0.0430 (6)
H18	1.135038	0.057669	0.198343	0.052*
C19	0.8911 (3)	0.01932 (17)	0.11396 (17)	0.0411 (6)
H19	0.919116	-0.013859	0.056228	0.049*
C20	0.7233 (3)	0.02544 (16)	0.11364 (16)	0.0361 (5)
H20	0.635927	-0.004304	0.055714	0.043*
C21	0.6803 (3)	0.07516 (15)	0.19799 (15)	0.0314 (5)
C22	0.4995 (3)	0.07987 (17)	0.19666 (17)	0.0392 (5)
H22	0.473487	0.112361	0.255175	0.047*
O7	0.4612 (2)	0.68794 (11)	0.13827 (11)	0.0421 (4)
O8	0.4103 (2)	0.83121 (10)	0.13783 (10)	0.0343 (4)
O9	0.2576 (2)	0.96930 (16)	0.37361 (13)	0.0564 (5)
O10	0.1470 (2)	0.56231 (10)	0.07902 (10)	0.0344 (4)
O11	0.2885 (2)	0.47254 (10)	-0.02015 (10)	0.0307 (3)
O12	-0.0216 (2)	0.22964 (12)	-0.14418 (12)	0.0445 (4)
C23	0.1995 (3)	0.60988 (15)	-0.04291 (14)	0.0287 (4)
C24	0.1012 (3)	0.57557 (16)	-0.13781 (14)	0.0330 (5)
H24	0.050852	0.512756	-0.166770	0.040*
C25	0.0767 (3)	0.63330 (17)	-0.19027 (15)	0.0373 (5)
H25	0.010341	0.609567	-0.255340	0.045*
C26	0.1480 (3)	0.72480 (17)	-0.14857 (16)	0.0391 (5)
H26	0.130032	0.763998	-0.184761	0.047*
C27	0.2459 (3)	0.75960 (16)	-0.05380 (16)	0.0361 (5)
H27	0.294766	0.822699	-0.025161	0.043*
C28	0.2730 (3)	0.70253 (15)	-0.00049 (14)	0.0301 (5)
C29	0.3900 (3)	0.73626 (15)	0.09826 (15)	0.0306 (5)
C30	0.5356 (3)	0.87237 (14)	0.22533 (15)	0.0298 (5)
C31	0.7047 (3)	0.86742 (16)	0.22777 (17)	0.0373 (5)
H31	0.736192	0.834619	0.171062	0.045*
C32	0.8280 (3)	0.91077 (17)	0.3137 (2)	0.0421 (6)
H32	0.944924	0.907067	0.316452	0.051*
C33	0.7813 (3)	0.95957 (17)	0.39587 (18)	0.0420 (6)
H33	0.866240	0.988609	0.455021	0.050*
C34	0.6121 (3)	0.96609 (17)	0.39196 (16)	0.0377 (5)
H34	0.581216	1.000895	0.448302	0.045*
C35	0.4856 (3)	0.92221 (15)	0.30632 (15)	0.0303 (5)
C36	0.3052 (3)	0.93132 (17)	0.30308 (16)	0.0374 (5)
H36	0.222211	0.906132	0.242657	0.045*
C37	0.2092 (3)	0.54900 (14)	0.01415 (14)	0.0274 (4)
C38	0.3023 (3)	0.41139 (15)	0.03022 (15)	0.0303 (5)
C39	0.4192 (3)	0.43683 (17)	0.11928 (16)	0.0377 (5)
H39	0.488630	0.494788	0.146572	0.045*
C40	0.4332 (3)	0.37622 (19)	0.16793 (17)	0.0444 (6)
H40	0.512796	0.392745	0.229272	0.053*
C41	0.3319 (4)	0.29152 (18)	0.12787 (19)	0.0462 (6)
H41	0.341554	0.250581	0.162010	0.055*
C42	0.2169 (3)	0.26673 (17)	0.03832 (18)	0.0402 (5)
H42	0.148523	0.208449	0.010831	0.048*

C43	0.2006 (3)	0.32701 (15)	-0.01224 (15)	0.0317 (5)
C44	0.0783 (3)	0.29935 (15)	-0.10841 (16)	0.0335 (5)
H44	0.078098	0.338378	-0.143977	0.040*

*Atomic displacement parameters ( $\text{\AA}^2$ )*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0479 (10)	0.0383 (9)	0.0337 (8)	-0.0107 (8)	-0.0082 (7)	0.0175 (7)
O2	0.0334 (8)	0.0251 (8)	0.0309 (7)	-0.0009 (6)	0.0035 (6)	0.0097 (6)
O3	0.0635 (12)	0.0340 (10)	0.0613 (11)	0.0048 (9)	0.0306 (10)	0.0033 (8)
O4	0.0803 (13)	0.0286 (8)	0.0332 (8)	-0.0009 (8)	0.0221 (8)	0.0116 (7)
O5	0.0536 (10)	0.0238 (8)	0.0294 (8)	-0.0004 (7)	0.0162 (7)	0.0066 (6)
O6	0.0378 (10)	0.0713 (13)	0.0448 (10)	0.0075 (9)	0.0083 (8)	0.0129 (9)
C1	0.0334 (11)	0.0281 (11)	0.0238 (9)	-0.0020 (9)	0.0017 (8)	0.0087 (8)
C2	0.0424 (13)	0.0322 (12)	0.0245 (10)	0.0006 (10)	0.0041 (9)	0.0053 (9)
C3	0.0504 (14)	0.0449 (14)	0.0210 (10)	-0.0025 (11)	0.0054 (9)	0.0093 (9)
C4	0.0564 (15)	0.0416 (14)	0.0291 (11)	-0.0011 (12)	0.0051 (10)	0.0195 (10)
C5	0.0499 (14)	0.0293 (11)	0.0302 (11)	0.0006 (10)	0.0063 (10)	0.0124 (9)
C6	0.0370 (12)	0.0267 (11)	0.0235 (10)	-0.0026 (9)	0.0030 (8)	0.0084 (8)
C7	0.0356 (12)	0.0248 (11)	0.0252 (10)	-0.0010 (9)	0.0053 (9)	0.0060 (8)
C8	0.0302 (11)	0.0267 (11)	0.0333 (11)	-0.0022 (9)	0.0041 (9)	0.0119 (9)
C9	0.0460 (14)	0.0350 (13)	0.0379 (12)	-0.0016 (11)	0.0142 (10)	0.0110 (10)
C10	0.0552 (16)	0.0504 (16)	0.0379 (13)	-0.0102 (13)	0.0115 (11)	0.0194 (11)
C11	0.0520 (16)	0.0434 (15)	0.0540 (15)	-0.0054 (12)	0.0044 (12)	0.0300 (13)
C12	0.0390 (13)	0.0312 (12)	0.0524 (14)	0.0003 (10)	0.0060 (11)	0.0180 (11)
C13	0.0295 (11)	0.0271 (11)	0.0356 (11)	-0.0016 (9)	0.0052 (9)	0.0092 (9)
C14	0.0351 (12)	0.0287 (12)	0.0400 (12)	-0.0026 (10)	0.0091 (10)	0.0064 (9)
C15	0.0432 (13)	0.0240 (10)	0.0269 (10)	0.0006 (9)	0.0066 (9)	0.0089 (8)
C16	0.0429 (13)	0.0228 (10)	0.0303 (10)	0.0032 (9)	0.0137 (9)	0.0102 (8)
C17	0.0391 (13)	0.0350 (13)	0.0426 (13)	-0.0006 (10)	0.0084 (10)	0.0164 (10)
C18	0.0365 (13)	0.0416 (14)	0.0591 (15)	0.0083 (11)	0.0221 (12)	0.0218 (12)
C19	0.0515 (15)	0.0351 (13)	0.0428 (13)	0.0088 (11)	0.0264 (11)	0.0120 (10)
C20	0.0430 (13)	0.0326 (12)	0.0309 (11)	0.0038 (10)	0.0124 (9)	0.0080 (9)
C21	0.0381 (12)	0.0260 (11)	0.0327 (11)	0.0061 (9)	0.0136 (9)	0.0112 (9)
C22	0.0431 (14)	0.0375 (13)	0.0387 (12)	0.0112 (11)	0.0169 (11)	0.0119 (10)
O7	0.0549 (11)	0.0288 (8)	0.0346 (8)	0.0018 (8)	-0.0017 (7)	0.0123 (7)
O8	0.0441 (9)	0.0246 (8)	0.0281 (7)	0.0035 (7)	0.0041 (6)	0.0066 (6)
O9	0.0343 (9)	0.0830 (15)	0.0398 (9)	0.0076 (9)	0.0121 (7)	0.0070 (9)
O10	0.0497 (9)	0.0327 (8)	0.0259 (7)	0.0115 (7)	0.0168 (7)	0.0122 (6)
O11	0.0392 (8)	0.0275 (8)	0.0298 (7)	0.0077 (7)	0.0149 (6)	0.0118 (6)
O12	0.0445 (10)	0.0334 (9)	0.0472 (10)	0.0009 (8)	0.0099 (8)	0.0069 (7)
C23	0.0352 (11)	0.0284 (11)	0.0241 (9)	0.0058 (9)	0.0112 (8)	0.0092 (8)
C24	0.0403 (12)	0.0314 (11)	0.0244 (10)	0.0031 (10)	0.0089 (9)	0.0070 (9)
C25	0.0455 (13)	0.0430 (13)	0.0232 (10)	0.0084 (11)	0.0085 (9)	0.0126 (9)
C26	0.0532 (15)	0.0401 (13)	0.0305 (11)	0.0096 (11)	0.0135 (10)	0.0195 (10)
C27	0.0486 (14)	0.0306 (12)	0.0316 (11)	0.0044 (10)	0.0133 (10)	0.0129 (9)
C28	0.0371 (12)	0.0290 (11)	0.0254 (10)	0.0051 (9)	0.0108 (9)	0.0099 (8)
C29	0.0397 (12)	0.0252 (10)	0.0271 (10)	0.0029 (9)	0.0106 (9)	0.0089 (8)

C30	0.0340 (11)	0.0232 (10)	0.0301 (10)	0.0004 (9)	0.0053 (9)	0.0098 (8)
C31	0.0397 (13)	0.0308 (12)	0.0468 (13)	0.0070 (10)	0.0191 (10)	0.0158 (10)
C32	0.0286 (12)	0.0361 (13)	0.0645 (16)	0.0029 (10)	0.0134 (11)	0.0217 (12)
C33	0.0299 (12)	0.0375 (13)	0.0464 (13)	-0.0044 (10)	-0.0022 (10)	0.0104 (11)
C34	0.0337 (12)	0.0385 (13)	0.0333 (11)	-0.0008 (10)	0.0046 (9)	0.0075 (10)
C35	0.0281 (11)	0.0291 (11)	0.0312 (10)	-0.0011 (9)	0.0061 (8)	0.0099 (9)
C36	0.0297 (12)	0.0404 (13)	0.0341 (11)	0.0000 (10)	0.0034 (9)	0.0081 (10)
C37	0.0318 (11)	0.0249 (10)	0.0219 (9)	0.0016 (9)	0.0054 (8)	0.0057 (8)
C38	0.0350 (11)	0.0297 (11)	0.0314 (10)	0.0109 (10)	0.0143 (9)	0.0136 (9)
C39	0.0389 (13)	0.0386 (13)	0.0343 (11)	0.0081 (11)	0.0097 (9)	0.0121 (10)
C40	0.0494 (15)	0.0519 (15)	0.0328 (12)	0.0187 (13)	0.0093 (10)	0.0177 (11)
C41	0.0610 (17)	0.0452 (15)	0.0471 (14)	0.0219 (13)	0.0214 (12)	0.0296 (12)
C42	0.0494 (14)	0.0324 (12)	0.0453 (13)	0.0103 (11)	0.0196 (11)	0.0172 (10)
C43	0.0359 (12)	0.0287 (11)	0.0332 (11)	0.0098 (10)	0.0147 (9)	0.0108 (9)
C44	0.0367 (12)	0.0284 (11)	0.0353 (11)	0.0084 (10)	0.0142 (9)	0.0085 (9)

*Geometric parameters (Å, °)*

O1—C7	1.194 (3)	O7—C29	1.196 (3)
O2—C7	1.360 (3)	O8—C29	1.365 (3)
O2—C8	1.410 (3)	O8—C30	1.401 (3)
O3—C14	1.207 (3)	O9—C36	1.211 (3)
O4—C15	1.197 (3)	O10—C37	1.198 (2)
O5—C15	1.359 (3)	O11—C37	1.359 (3)
O5—C16	1.410 (2)	O11—C38	1.412 (2)
O6—C22	1.213 (3)	O12—C44	1.210 (3)
C1—C2	1.391 (3)	C23—C24	1.389 (3)
C1—C6	1.397 (3)	C23—C28	1.399 (3)
C1—C7	1.494 (3)	C23—C37	1.493 (3)
C2—C3	1.388 (3)	C24—C25	1.389 (3)
C2—H2	0.9500	C24—H24	0.9500
C3—C4	1.377 (4)	C25—C26	1.379 (3)
C3—H3	0.9500	C25—H25	0.9500
C4—C5	1.386 (3)	C26—C27	1.387 (3)
C4—H4	0.9500	C26—H26	0.9500
C5—C6	1.395 (3)	C27—C28	1.391 (3)
C5—H5	0.9500	C27—H27	0.9500
C6—C15	1.489 (3)	C28—C29	1.487 (3)
C8—C9	1.377 (3)	C30—C31	1.378 (3)
C8—C13	1.390 (3)	C30—C35	1.389 (3)
C9—C10	1.388 (4)	C31—C32	1.382 (4)
C9—H9	0.9500	C31—H31	0.9500
C10—C11	1.384 (4)	C32—C33	1.385 (4)
C10—H10	0.9500	C32—H32	0.9500
C11—C12	1.377 (4)	C33—C34	1.377 (3)
C11—H11	0.9500	C33—H33	0.9500
C12—C13	1.398 (3)	C34—C35	1.393 (3)
C12—H12	0.9500	C34—H34	0.9500



C13—C14	1.479 (3)	C35—C36	1.474 (3)
C14—H14	0.9500	C36—H36	0.9500
C16—C17	1.376 (3)	C38—C39	1.382 (3)
C16—C21	1.386 (3)	C38—C43	1.384 (3)
C17—C18	1.388 (4)	C39—C40	1.384 (3)
C17—H17	0.9500	C39—H39	0.9500
C18—C19	1.386 (4)	C40—C41	1.388 (4)
C18—H18	0.9500	C40—H40	0.9500
C19—C20	1.376 (4)	C41—C42	1.382 (4)
C19—H19	0.9500	C41—H41	0.9500
C20—C21	1.400 (3)	C42—C43	1.401 (3)
C20—H20	0.9500	C42—H42	0.9500
C21—C22	1.477 (3)	C43—C44	1.479 (3)
C22—H22	0.9500	C44—H44	0.9500
C7—O2—C8	115.88 (16)	C29—O8—C30	116.81 (16)
C15—O5—C16	117.47 (16)	C37—O11—C38	115.03 (15)
C2—C1—C6	120.02 (19)	C24—C23—C28	119.85 (19)
C2—C1—C7	117.62 (19)	C24—C23—C37	118.41 (19)
C6—C1—C7	122.07 (18)	C28—C23—C37	121.44 (18)
C3—C2—C1	119.9 (2)	C25—C24—C23	119.9 (2)
C3—C2—H2	120.0	C25—C24—H24	120.1
C1—C2—H2	120.0	C23—C24—H24	120.1
C4—C3—C2	120.3 (2)	C26—C25—C24	120.5 (2)
C4—C3—H3	119.9	C26—C25—H25	119.8
C2—C3—H3	119.9	C24—C25—H25	119.8
C3—C4—C5	120.3 (2)	C25—C26—C27	120.0 (2)
C3—C4—H4	119.8	C25—C26—H26	120.0
C5—C4—H4	119.8	C27—C26—H26	120.0
C4—C5—C6	120.2 (2)	C26—C27—C28	120.3 (2)
C4—C5—H5	119.9	C26—C27—H27	119.9
C6—C5—H5	119.9	C28—C27—H27	119.9
C5—C6—C1	119.31 (19)	C27—C28—C23	119.5 (2)
C5—C6—C15	120.6 (2)	C27—C28—C29	120.9 (2)
C1—C6—C15	119.91 (18)	C23—C28—C29	119.39 (18)
O1—C7—O2	123.6 (2)	O7—C29—O8	123.5 (2)
O1—C7—C1	125.3 (2)	O7—C29—C28	125.5 (2)
O2—C7—C1	110.90 (18)	O8—C29—C28	111.05 (18)
C9—C8—C13	122.2 (2)	C31—C30—C35	121.8 (2)
C9—C8—O2	119.8 (2)	C31—C30—O8	119.43 (19)
C13—C8—O2	118.03 (19)	C35—C30—O8	118.69 (19)
C8—C9—C10	118.6 (2)	C30—C31—C32	119.2 (2)
C8—C9—H9	120.7	C30—C31—H31	120.4
C10—C9—H9	120.7	C32—C31—H31	120.4
C11—C10—C9	120.6 (2)	C31—C32—C33	120.1 (2)
C11—C10—H10	119.7	C31—C32—H32	119.9
C9—C10—H10	119.7	C33—C32—H32	119.9
C12—C11—C10	120.1 (2)	C34—C33—C32	120.1 (2)

C12—C11—H11	120.0	C34—C33—H33	119.9
C10—C11—H11	120.0	C32—C33—H33	119.9
C11—C12—C13	120.6 (2)	C33—C34—C35	120.8 (2)
C11—C12—H12	119.7	C33—C34—H34	119.6
C13—C12—H12	119.7	C35—C34—H34	119.6
C8—C13—C12	118.0 (2)	C30—C35—C34	118.0 (2)
C8—C13—C14	121.8 (2)	C30—C35—C36	121.96 (19)
C12—C13—C14	120.2 (2)	C34—C35—C36	120.01 (19)
O3—C14—C13	123.4 (2)	O9—C36—C35	123.3 (2)
O3—C14—H14	118.3	O9—C36—H36	118.4
C13—C14—H14	118.3	C35—C36—H36	118.4
O4—C15—O5	123.68 (19)	O10—C37—O11	123.38 (19)
O4—C15—C6	125.1 (2)	O10—C37—C23	124.77 (19)
O5—C15—C6	111.23 (17)	O11—C37—C23	111.73 (16)
C17—C16—C21	121.9 (2)	C39—C38—C43	122.2 (2)
C17—C16—O5	119.1 (2)	C39—C38—O11	119.0 (2)
C21—C16—O5	118.9 (2)	C43—C38—O11	118.81 (19)
C16—C17—C18	119.0 (2)	C38—C39—C40	118.7 (2)
C16—C17—H17	120.5	C38—C39—H39	120.6
C18—C17—H17	120.5	C40—C39—H39	120.6
C19—C18—C17	120.2 (2)	C39—C40—C41	120.6 (2)
C19—C18—H18	119.9	C39—C40—H40	119.7
C17—C18—H18	119.9	C41—C40—H40	119.7
C20—C19—C18	120.1 (2)	C42—C41—C40	120.0 (2)
C20—C19—H19	120.0	C42—C41—H41	120.0
C18—C19—H19	120.0	C40—C41—H41	120.0
C19—C20—C21	120.5 (2)	C41—C42—C43	120.4 (2)
C19—C20—H20	119.7	C41—C42—H42	119.8
C21—C20—H20	119.7	C43—C42—H42	119.8
C16—C21—C20	118.2 (2)	C38—C43—C42	118.1 (2)
C16—C21—C22	121.97 (19)	C38—C43—C44	121.9 (2)
C20—C21—C22	119.8 (2)	C42—C43—C44	119.9 (2)
O6—C22—C21	123.4 (2)	O12—C44—C43	123.5 (2)
O6—C22—H22	118.3	O12—C44—H44	118.3
C21—C22—H22	118.3	C43—C44—H44	118.3
C6—C1—C2—C3	0.7 (3)	C28—C23—C24—C25	0.2 (3)
C7—C1—C2—C3	174.6 (2)	C37—C23—C24—C25	174.05 (19)
C1—C2—C3—C4	0.4 (4)	C23—C24—C25—C26	-0.6 (3)
C2—C3—C4—C5	-1.1 (4)	C24—C25—C26—C27	0.4 (4)
C3—C4—C5—C6	0.8 (4)	C25—C26—C27—C28	0.1 (4)
C4—C5—C6—C1	0.2 (4)	C26—C27—C28—C23	-0.5 (3)
C4—C5—C6—C15	175.0 (2)	C26—C27—C28—C29	174.5 (2)
C2—C1—C6—C5	-0.9 (3)	C24—C23—C28—C27	0.3 (3)
C7—C1—C6—C5	-174.6 (2)	C37—C23—C28—C27	-173.3 (2)
C2—C1—C6—C15	-175.8 (2)	C24—C23—C28—C29	-174.7 (2)
C7—C1—C6—C15	10.5 (3)	C37—C23—C28—C29	11.6 (3)
C8—O2—C7—O1	-1.1 (3)	C30—O8—C29—O7	7.1 (3)

C8—O2—C7—C1	-176.66 (17)	C30—O8—C29—C28	-171.30 (18)
C2—C1—C7—O1	-111.7 (3)	C27—C28—C29—O7	-156.0 (2)
C6—C1—C7—O1	62.1 (3)	C23—C28—C29—O7	19.0 (3)
C2—C1—C7—O2	63.7 (3)	C27—C28—C29—O8	22.4 (3)
C6—C1—C7—O2	-122.5 (2)	C23—C28—C29—O8	-162.62 (18)
C7—O2—C8—C9	-68.1 (3)	C29—O8—C30—C31	69.3 (3)
C7—O2—C8—C13	113.4 (2)	C29—O8—C30—C35	-114.7 (2)
C13—C8—C9—C10	-1.5 (4)	C35—C30—C31—C32	2.1 (3)
O2—C8—C9—C10	-179.9 (2)	O8—C30—C31—C32	177.9 (2)
C8—C9—C10—C11	0.4 (4)	C30—C31—C32—C33	-0.9 (4)
C9—C10—C11—C12	0.8 (4)	C31—C32—C33—C34	-0.8 (4)
C10—C11—C12—C13	-1.0 (4)	C32—C33—C34—C35	1.5 (4)
C9—C8—C13—C12	1.2 (3)	C31—C30—C35—C34	-1.4 (3)
O2—C8—C13—C12	179.7 (2)	O8—C30—C35—C34	-177.30 (19)
C9—C8—C13—C14	-178.1 (2)	C31—C30—C35—C36	177.1 (2)
O2—C8—C13—C14	0.4 (3)	O8—C30—C35—C36	1.2 (3)
C11—C12—C13—C8	0.1 (4)	C33—C34—C35—C30	-0.3 (3)
C11—C12—C13—C14	179.4 (2)	C33—C34—C35—C36	-178.9 (2)
C8—C13—C14—O3	-177.2 (2)	C30—C35—C36—O9	175.2 (2)
C12—C13—C14—O3	3.4 (4)	C34—C35—C36—O9	-6.3 (4)
C16—O5—C15—O4	7.3 (3)	C38—O11—C37—O10	-3.7 (3)
C16—O5—C15—C6	-171.59 (19)	C38—O11—C37—C23	-179.93 (17)
C5—C6—C15—O4	-151.2 (3)	C24—C23—C37—O10	-110.6 (2)
C1—C6—C15—O4	23.6 (4)	C28—C23—C37—O10	63.2 (3)
C5—C6—C15—O5	27.7 (3)	C24—C23—C37—O11	65.6 (3)
C1—C6—C15—O5	-157.6 (2)	C28—C23—C37—O11	-120.7 (2)
C15—O5—C16—C17	75.0 (3)	C37—O11—C38—C39	-74.2 (2)
C15—O5—C16—C21	-108.8 (2)	C37—O11—C38—C43	107.2 (2)
C21—C16—C17—C18	2.0 (3)	C43—C38—C39—C40	-0.9 (3)
O5—C16—C17—C18	178.0 (2)	O11—C38—C39—C40	-179.45 (19)
C16—C17—C18—C19	-2.0 (4)	C38—C39—C40—C41	0.1 (3)
C17—C18—C19—C20	0.6 (4)	C39—C40—C41—C42	0.6 (4)
C18—C19—C20—C21	0.8 (4)	C40—C41—C42—C43	-0.6 (4)
C17—C16—C21—C20	-0.6 (3)	C39—C38—C43—C42	0.8 (3)
O5—C16—C21—C20	-176.64 (18)	O11—C38—C43—C42	179.45 (18)
C17—C16—C21—C22	177.6 (2)	C39—C38—C43—C44	-178.7 (2)
O5—C16—C21—C22	1.6 (3)	O11—C38—C43—C44	-0.1 (3)
C19—C20—C21—C16	-0.8 (3)	C41—C42—C43—C38	-0.1 (3)
C19—C20—C21—C22	-179.1 (2)	C41—C42—C43—C44	179.4 (2)
C16—C21—C22—O6	179.9 (2)	C38—C43—C44—O12	-173.9 (2)
C20—C21—C22—O6	-1.9 (4)	C42—C43—C44—O12	6.5 (3)

### Hydrogen-bond geometry ( $\text{\AA}$ , $^\circ$ )

Cg3 and Cg5 are the centroids of the C16–C21 and C30–C35 benzene rings, respectively.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C3—H3 $\cdots$ O10 <sup>i</sup>	0.95	2.35	3.221 (3)	152
C18—H18 $\cdots$ O6 <sup>ii</sup>	0.95	2.55	3.430 (3)	155

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C22—H22…O3 <sup>i</sup>	0.95	2.29	3.147 (3)	149
C32—H32…O9 <sup>ii</sup>	0.95	2.54	3.422 (3)	155
C36—H36…O12 <sup>iii</sup>	0.95	2.41	3.205 (3)	141
C11—H11…Cg5	0.95	2.75	3.664 (3)	162
C26—H26…Cg3 <sup>iv</sup>	0.95	2.68	3.546 (3)	152

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Symmetry codes: (i)  $-x+1, -y+1, -z+1$ ; (ii)  $x+1, y, z$ ; (iii)  $-x, -y+1, -z$ ; (iv)  $-x+1, -y+1, -z$ .