

2-Aminoanilinium benzene-1,2-diaminium tris(4-methylbenzene-1-sulfonate)

C. Amirthakumar,^a P. Pandi,^b R. Mohan Kumar^{a*} and G. Chakkaravarthi^{c*}

^aDepartment of Physics, Presidency College, Chennai 600 005, India, ^bDepartment of Physics, Panimalar Engineering College, Chennai 600 123, India, and ^cDepartment of Physics, CPCL Polytechnic College, Chennai 600 068, India.

*Correspondence e-mail: mohan66@hotmail.com, chakkaravarthi_2005@yahoo.com

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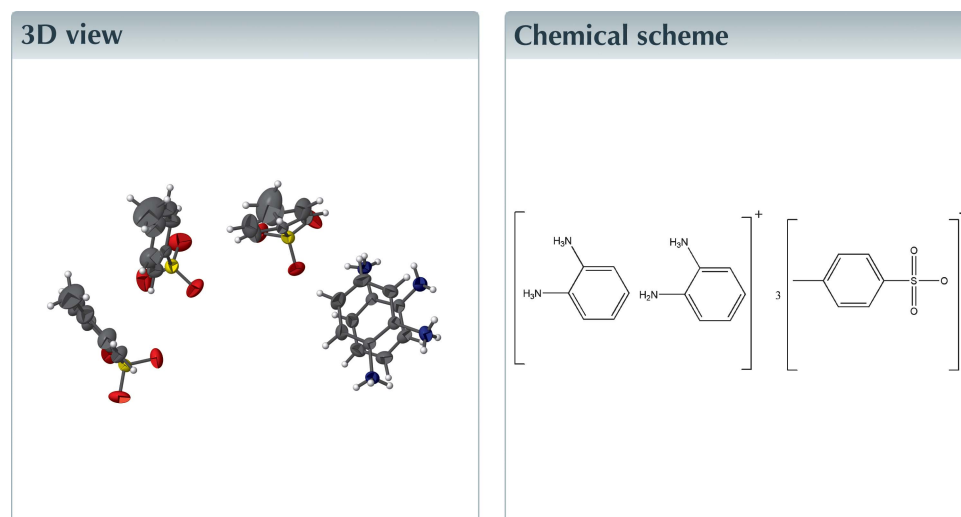
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Keywords: molecular salt; crystal structure; hydrogen bonding.

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Structural data: full structural data are available from iucrdata.iucr.org

In the title molecular salt, $C_6H_{10}N_2^{2+} \cdot C_6H_9N_2^+ \cdot 3C_7H_7O_3S^-$, one of the cations is doubly protonated and one is singly protonated with charge balance achieved by three sulfonate anions. The crystal packing features $N-H \cdots O$ and $C-H \cdots O$ hydrogen bonds. The ions are arranged into a two-dimensional network along the (010) plane and the structure is further consolidated by weak $C-H \cdots \pi$ interactions.



Structure description

A variety of pharmaceutical drugs are prepared as salts of benzenesulfonic acid. Recently, much attention has been devoted to simple molecular-ionic crystals containing organic cations and anions because of the tunability of their special structural features and their interesting physical properties (Katrusiak & Szafranski, 2006). In a continuation of our studies of molecular compounds with non-linear optical properties that are used in optoelectronic and photonic devices (Nalwa & Miyata, 1997), we herewith report the crystal structure of the title compound (Fig. 1). One of the cations is doubly protonated (at N1 and N2) and the other is singly protonated at N4. The geometric parameters agree well with reported similar structures (Jasinski *et al.*, 2011; Krishnakumar *et al.*, 2012).

The dihedral angles between the C22–C27 benzene ring of one of the cations and the C1–C6, C8–C13 and C15–C20 benzene rings of the anions are 70.63 (16), 75.67 (16) and 86.29 (16)°, respectively. The C28–C33 benzene ring of one of the cations makes dihedral angles of 63.57 (15), 69.42 (15) and 87.04 (16)°, respectively, with the C1–C6, C8–C13 and C15–C20 benzene rings of the anions.

The crystal packing features $N-H \cdots O$ and $C-H \cdots O$ hydrogen bonds (Fig. 2, Table 1). The ions are arranged into a two-dimensional network parallel to the (010) plane and the structure is further consolidated by weak $C-H \cdots \pi$ interactions (Table 1).

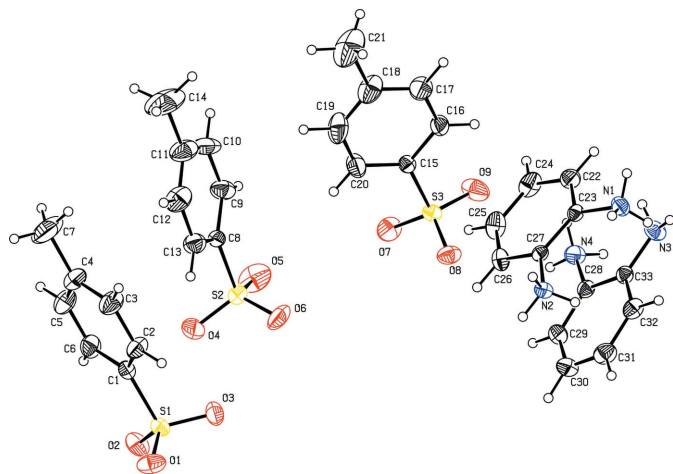


Figure 1
The molecular structure of the title molecular salt, with the atom labelling and 30% probability displacement ellipsoids.

An weak π - π stacking interaction is observed between the C22-C27 and C28-C33 benzene rings of the cations, with a centroid-to-centroid distance of 3.7565 (17) Å.

Synthesis and crystallization

o-Phenyldiamine (1.36 g) and *p*-toluenesulfonic acid (2.3 g) were mixed in a 1:2 ratio in water at ambient temperature and

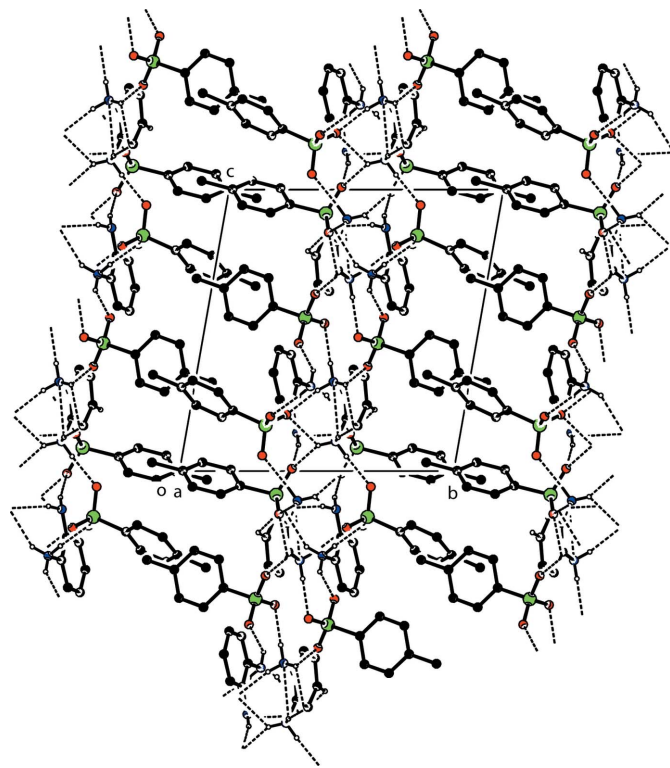


Figure 2
The crystal packing of the title molecular salt viewed along the *a* axis. The hydrogen bonds (Table 1) are shown as dashed lines. H atoms not involving in hydrogen bonding have been omitted for clarity.

Table 1
Hydrogen-bond geometry (Å, °).

*Cg*1, *Cg*2 and *Cg*3 are the centroids of the C1-C6, C8-C13 and C15-C20 rings, respectively.

| <i>D</i> -H... <i>A</i> | <i>D</i> -H | H... <i>A</i> | <i>D</i> ... <i>A</i> | <i>D</i> -H... <i>A</i> |
|---------------------------------------|-------------|---------------|-----------------------|-------------------------|
| N2-H2A...O8 ⁱ | 0.87 (1) | 1.91 (1) | 2.785 (3) | 175 (3) |
| N2-H2B...O6 ⁱ | 0.87 (1) | 1.86 (1) | 2.707 (3) | 164 (3) |
| N3-H3A...O3 ⁱ | 0.86 (1) | 2.28 (1) | 3.113 (3) | 164 (3) |
| N1-H1A...O3 ⁱ | 0.86 (1) | 2.06 (2) | 2.810 (3) | 144 (3) |
| N1-H1A...O6 ⁱ | 0.86 (1) | 2.44 (3) | 2.852 (4) | 110 (3) |
| N3-H3B...O1 ⁱⁱ | 0.86 (1) | 2.15 (1) | 2.989 (4) | 163 (3) |
| N4-H4B...O1 ⁱⁱ | 0.87 (1) | 2.12 (2) | 2.910 (4) | 150 (3) |
| C24-H24...O1 ⁱⁱⁱ | 0.93 | 2.55 | 3.197 (4) | 127 |
| N1-H1C...O4 ⁱⁱⁱ | 0.87 (1) | 1.84 (1) | 2.698 (3) | 169 (4) |
| N1-H1B...O2 ⁱⁱⁱ | 0.86 (1) | 1.83 (1) | 2.669 (3) | 164 (3) |
| N2-H2C...O2 ⁱⁱⁱ | 0.87 (1) | 2.30 (3) | 2.873 (3) | 124 (2) |
| N2-H2C...O9 ^{iv} | 0.87 (1) | 2.07 (2) | 2.833 (3) | 146 (3) |
| N4-H4A...O5 ^v | 0.87 (1) | 1.88 (1) | 2.742 (4) | 175 (4) |
| N4-H4C...O7 ^v | 0.87 (1) | 2.06 (2) | 2.826 (3) | 147 (4) |
| C5-H5... <i>Cg</i> 2 ^{vi} | 0.93 | 2.87 | 3.629 (4) | 140 |
| C10-H10... <i>Cg</i> 3 ^{vii} | 0.93 | 2.81 | 3.587 (5) | 142 |
| C13-H13... <i>Cg</i> 1 | 0.93 | 2.93 | 3.609 (3) | 131 |

Symmetry codes: (i) $-x+2, -y+1, -z-1$; (ii) $x, y, z-1$; (iii) $x+1, y, z-1$; (iv) $x+1, y, z$; (v) $-x+1, -y+1, -z-1$; (vi) $-x+1, -y+2, -z$; (vii) $-x+1, -y+2, -z-1$.

Table 2
Experimental details.

| | |
|--|--|
| Crystal data | |
| Chemical formula | $C_6H_{10}N_2^{2+} \cdot C_6H_9N_2^+ \cdot 3C_7H_7O_3S^-$ |
| M_r | 732.87 |
| Crystal system, space group | Triclinic, $P\bar{1}$ |
| Temperature (K) | 295 |
| <i>a</i> , <i>b</i> , <i>c</i> (Å) | 10.5058 (4), 12.8929 (4), 14.0425 (4) |
| α , β , γ (°) | 80.187 (2), 73.218 (1), 89.188 (2) |
| <i>V</i> (Å ³) | 1793.17 (10) |
| <i>Z</i> | 2 |
| Radiation type | Mo <i>K</i> α |
| μ (mm ⁻¹) | 0.26 |
| Crystal size (mm) | 0.24 × 0.20 × 0.18 |
| Data collection | |
| Diffractometer | Bruker APEXII CCD |
| Absorption correction | Multi-scan (<i>SADABS</i> ; Bruker, 2004) |
| No. of measured, independent and observed [$I > 2\sigma(I)$] reflections | 32987, 6325, 4091 |
| R_{int} | 0.059 |
| $(\sin \theta/\lambda)_{max}$ (Å ⁻¹) | 0.595 |
| Refinement | |
| $R[F^2 > 2\sigma(F^2)]$, $wR(F^2)$, <i>S</i> | 0.043, 0.117, 1.03 |
| No. of reflections | 6325 |
| No. of parameters | 489 |
| No. of restraints | 11 |
| H-atom treatment | H atoms treated by a mixture of independent and constrained refinement |
| $\Delta\rho_{max}$, $\Delta\rho_{min}$ (e Å ⁻³) | 0.30, -0.36 |

Computer programs: *APEX2* and *SAINT* (Bruker, 2004), *SHELXT2016* (Sheldrick, 2008), *SHELXL2016* (Sheldrick, 2015) and *PLATON* (Spek, 2009).

the solution was stirred for five hours, then filtered and placed in a beaker covered with perforated polythene. Colourless crystals of the title molecular salt were recovered after one week.

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

Acknowledgements

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full crystallographic data

IUCrData (2018). 3, x180437 [https://doi.org/10.1107/S2414314618004376]

2-Aminoanilinium benzene-1,2-diaminium tris(4-methylbenzene-1-sulfonate)

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Crystal data

$C_6H_{10}N_2^{2+} \cdot C_6H_9N_2^+ \cdot 3C_7H_7O_3S^-$

$M_r = 732.87$

Triclinic, $P\bar{1}$

$a = 10.5058(4) \text{ \AA}$

$b = 12.8929(4) \text{ \AA}$

$c = 14.0425(4) \text{ \AA}$

$\alpha = 80.187(2)^\circ$

$\beta = 73.218(1)^\circ$

$\gamma = 89.188(2)^\circ$

$V = 1793.17(10) \text{ \AA}^3$

$Z = 2$

$F(000) = 772$

$D_x = 1.357 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 7569 reflections

$\theta = 2.4\text{--}26.2^\circ$

$\mu = 0.26 \text{ mm}^{-1}$

$T = 295 \text{ K}$

Block, colourless

$0.24 \times 0.20 \times 0.18 \text{ mm}$

Data collection

Bruker APEXII CCD

diffractometer

ω and ϕ scans

Absorption correction: multi-scan

(SADABS; Bruker, 2004)

32987 measured reflections

6325 independent reflections

4091 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.059$

$\theta_{\text{max}} = 25.0^\circ$, $\theta_{\text{min}} = 2.0^\circ$

$h = -12 \rightarrow 12$

$k = -15 \rightarrow 15$

$l = -16 \rightarrow 16$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.043$

$wR(F^2) = 0.117$

$S = 1.03$

6325 reflections

489 parameters

11 restraints

Hydrogen site location: mixed

H atoms treated by a mixture of independent

and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.048P)^2 + 0.6864P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} < 0.001$

$\Delta\rho_{\text{max}} = 0.30 \text{ e \AA}^{-3}$

$\Delta\rho_{\text{min}} = -0.36 \text{ e \AA}^{-3}$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. C-bound H atoms were placed in calculated positions and allowed to ride on their carrier atoms, with C—H = 0.93 Å (aromatic CH) or 0.96 Å (methyl CH), and with $U_{\text{iso}} = 1.5U_{\text{eq}}$ (methyl C) or $U_{\text{iso}} = 1.2U_{\text{eq}}$ (aromatic C). H atoms for NH groups were located in difference-Fourier maps and refined with a distance restraint of N—H = 0.86 (1) Å.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | <i>x</i> | <i>y</i> | <i>z</i> | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|------|------------|------------|---------------|----------------------------------|
| C1 | 0.7325 (3) | 0.7693 (2) | 0.05175 (19) | 0.0359 (7) |
| C2 | 0.8554 (3) | 0.8136 (2) | -0.0052 (2) | 0.0500 (8) |
| H2 | 0.926511 | 0.770800 | -0.026143 | 0.060* |
| C3 | 0.8723 (4) | 0.9218 (3) | -0.0310 (3) | 0.0648 (10) |
| H3 | 0.955466 | 0.950946 | -0.069595 | 0.078* |
| C4 | 0.7699 (4) | 0.9875 (3) | -0.0014 (3) | 0.0671 (10) |
| C5 | 0.6483 (4) | 0.9424 (3) | 0.0555 (3) | 0.0705 (11) |
| H5 | 0.577680 | 0.985428 | 0.076773 | 0.085* |
| C6 | 0.6285 (3) | 0.8346 (3) | 0.0817 (2) | 0.0540 (8) |
| H6 | 0.544987 | 0.805795 | 0.119751 | 0.065* |
| C7 | 0.7897 (5) | 1.1056 (3) | -0.0308 (4) | 0.1138 (17) |
| H7A | 0.809013 | 1.133881 | 0.022713 | 0.171* |
| H7B | 0.710080 | 1.135453 | -0.042370 | 0.171* |
| H7C | 0.862444 | 1.122681 | -0.091320 | 0.171* |
| C8 | 0.5243 (3) | 0.8441 (2) | -0.21536 (19) | 0.0378 (7) |
| C9 | 0.4702 (3) | 0.9086 (3) | -0.2808 (2) | 0.0589 (9) |
| H9 | 0.397778 | 0.885012 | -0.297780 | 0.071* |
| C10 | 0.5252 (4) | 1.0099 (3) | -0.3215 (3) | 0.0785 (12) |
| H10 | 0.488594 | 1.053475 | -0.365877 | 0.094* |
| C11 | 0.6309 (4) | 1.0465 (3) | -0.2980 (3) | 0.0758 (11) |
| C12 | 0.6838 (4) | 0.9814 (3) | -0.2342 (3) | 0.0705 (10) |
| H12 | 0.756892 | 1.005325 | -0.218288 | 0.085* |
| C13 | 0.6323 (3) | 0.8803 (3) | -0.1919 (2) | 0.0543 (8) |
| H13 | 0.670179 | 0.837317 | -0.147991 | 0.065* |
| C14 | 0.6870 (5) | 1.1583 (3) | -0.3426 (4) | 0.141 (2) |
| H14A | 0.770246 | 1.167096 | -0.328980 | 0.211* |
| H14B | 0.625384 | 1.207543 | -0.312706 | 0.211* |
| H14C | 0.700713 | 1.170618 | -0.414314 | 0.211* |
| C15 | 0.6807 (3) | 0.7641 (2) | -0.58539 (19) | 0.0333 (6) |
| C16 | 0.7424 (3) | 0.7992 (2) | -0.6864 (2) | 0.0470 (8) |
| H16 | 0.736544 | 0.758539 | -0.733751 | 0.056* |
| C17 | 0.8127 (3) | 0.8942 (3) | -0.7177 (3) | 0.0659 (10) |
| H17 | 0.853231 | 0.917137 | -0.786234 | 0.079* |
| C18 | 0.8242 (4) | 0.9555 (3) | -0.6501 (3) | 0.0733 (11) |
| C19 | 0.7625 (4) | 0.9197 (3) | -0.5492 (3) | 0.0772 (12) |
| H19 | 0.769364 | 0.960268 | -0.502023 | 0.093* |
| C20 | 0.6906 (3) | 0.8247 (3) | -0.5163 (2) | 0.0587 (9) |
| H20 | 0.649245 | 0.802078 | -0.447786 | 0.070* |
| C21 | 0.8996 (5) | 1.0613 (3) | -0.6843 (4) | 0.131 (2) |
| H21A | 0.993507 | 1.050309 | -0.704886 | 0.196* |
| H21B | 0.878881 | 1.100075 | -0.629430 | 0.196* |
| H21C | 0.874198 | 1.100383 | -0.740054 | 0.196* |
| C22 | 1.2542 (3) | 0.5750 (2) | -0.81322 (19) | 0.0314 (6) |
| C23 | 1.1471 (3) | 0.6203 (2) | -0.8383 (2) | 0.0463 (8) |
| H23 | 1.148923 | 0.637179 | -0.905753 | 0.056* |

| | | | | |
|-----|--------------|--------------|---------------|-------------|
| C24 | 1.0372 (3) | 0.6410 (3) | -0.7643 (3) | 0.0568 (9) |
| H24 | 0.965305 | 0.673123 | -0.781748 | 0.068* |
| C25 | 1.0328 (3) | 0.6145 (3) | -0.6645 (3) | 0.0557 (9) |
| H25 | 0.957297 | 0.626978 | -0.614244 | 0.067* |
| C26 | 1.1407 (3) | 0.5694 (2) | -0.6391 (2) | 0.0435 (7) |
| H26 | 1.137998 | 0.551588 | -0.571492 | 0.052* |
| C27 | 1.2523 (2) | 0.55048 (19) | -0.71291 (18) | 0.0291 (6) |
| C28 | 0.8648 (3) | 0.3735 (2) | -0.68323 (19) | 0.0342 (6) |
| C29 | 0.8651 (3) | 0.3320 (2) | -0.5866 (2) | 0.0448 (7) |
| H29 | 0.789832 | 0.335811 | -0.532796 | 0.054* |
| C30 | 0.9761 (3) | 0.2849 (2) | -0.5691 (2) | 0.0530 (8) |
| H30 | 0.976860 | 0.256931 | -0.503717 | 0.064* |
| C31 | 1.0864 (3) | 0.2798 (2) | -0.6499 (3) | 0.0525 (8) |
| H31 | 1.162248 | 0.248455 | -0.638725 | 0.063* |
| C32 | 1.0857 (3) | 0.3202 (2) | -0.7462 (2) | 0.0441 (7) |
| H32 | 1.161153 | 0.315704 | -0.799655 | 0.053* |
| C33 | 0.9741 (3) | 0.3680 (2) | -0.76597 (19) | 0.0354 (7) |
| N1 | 1.3660 (3) | 0.5504 (2) | -0.89483 (19) | 0.0429 (6) |
| N2 | 1.3655 (2) | 0.5040 (2) | -0.68291 (18) | 0.0352 (6) |
| N3 | 0.9729 (3) | 0.4043 (2) | -0.86411 (19) | 0.0491 (7) |
| N4 | 0.7447 (3) | 0.4232 (2) | -0.6989 (2) | 0.0455 (6) |
| O1 | 0.8136 (2) | 0.59810 (16) | 0.13209 (17) | 0.0612 (6) |
| O2 | 0.57959 (19) | 0.60896 (17) | 0.15234 (14) | 0.0560 (6) |
| O3 | 0.7300 (2) | 0.59162 (16) | -0.00861 (14) | 0.0557 (6) |
| O4 | 0.4293 (3) | 0.70824 (18) | -0.05549 (16) | 0.0840 (8) |
| O5 | 0.3438 (3) | 0.7040 (2) | -0.1939 (2) | 0.0908 (9) |
| O6 | 0.5632 (2) | 0.64617 (17) | -0.20037 (18) | 0.0720 (7) |
| O7 | 0.4923 (2) | 0.64765 (17) | -0.45485 (16) | 0.0636 (6) |
| O8 | 0.6954 (2) | 0.56619 (16) | -0.52454 (15) | 0.0532 (6) |
| O9 | 0.5535 (3) | 0.61892 (18) | -0.62836 (16) | 0.0757 (8) |
| S1 | 0.71235 (7) | 0.63186 (6) | 0.08397 (5) | 0.0366 (2) |
| S2 | 0.45907 (8) | 0.71568 (6) | -0.16259 (6) | 0.0451 (2) |
| S3 | 0.59663 (7) | 0.64019 (6) | -0.54542 (5) | 0.0387 (2) |
| H1A | 1.355 (4) | 0.4892 (14) | -0.909 (3) | 0.088 (13)* |
| H1B | 1.4393 (18) | 0.557 (3) | -0.880 (2) | 0.064 (11)* |
| H1C | 1.377 (4) | 0.599 (2) | -0.9477 (17) | 0.089 (14)* |
| H2A | 1.342 (3) | 0.4814 (19) | -0.6177 (8) | 0.043 (8)* |
| H2B | 1.398 (3) | 0.4519 (18) | -0.713 (2) | 0.074 (12)* |
| H2C | 1.430 (2) | 0.5497 (19) | -0.693 (2) | 0.062 (10)* |
| H3A | 1.0508 (15) | 0.417 (2) | -0.9065 (16) | 0.048 (9)* |
| H3B | 0.921 (3) | 0.4550 (17) | -0.875 (2) | 0.058 (11)* |
| H4A | 0.714 (4) | 0.386 (3) | -0.734 (3) | 0.098 (15)* |
| H4B | 0.757 (3) | 0.4879 (12) | -0.731 (2) | 0.073 (12)* |
| H4C | 0.683 (3) | 0.423 (3) | -0.6426 (16) | 0.101 (15)* |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|-------------|-------------|-------------|--------------|--------------|--------------|
| C1 | 0.0353 (17) | 0.0430 (17) | 0.0305 (14) | 0.0038 (14) | -0.0100 (13) | -0.0083 (13) |
| C2 | 0.044 (2) | 0.0362 (19) | 0.062 (2) | 0.0026 (15) | -0.0023 (16) | -0.0098 (15) |
| C3 | 0.061 (2) | 0.043 (2) | 0.076 (2) | -0.0075 (18) | 0.0018 (19) | -0.0074 (18) |
| C4 | 0.088 (3) | 0.040 (2) | 0.071 (2) | 0.014 (2) | -0.018 (2) | -0.0114 (18) |
| C5 | 0.074 (3) | 0.058 (3) | 0.080 (3) | 0.034 (2) | -0.017 (2) | -0.024 (2) |
| C6 | 0.045 (2) | 0.061 (2) | 0.0524 (19) | 0.0132 (17) | -0.0056 (16) | -0.0168 (17) |
| C7 | 0.143 (4) | 0.040 (2) | 0.151 (4) | 0.014 (3) | -0.034 (4) | -0.014 (3) |
| C8 | 0.0439 (18) | 0.0353 (17) | 0.0313 (15) | 0.0106 (14) | -0.0072 (13) | -0.0049 (13) |
| C9 | 0.056 (2) | 0.061 (2) | 0.057 (2) | 0.0124 (18) | -0.0212 (18) | 0.0031 (18) |
| C10 | 0.087 (3) | 0.062 (3) | 0.066 (2) | 0.022 (2) | -0.011 (2) | 0.023 (2) |
| C11 | 0.069 (3) | 0.052 (2) | 0.078 (3) | 0.003 (2) | 0.010 (2) | 0.010 (2) |
| C12 | 0.058 (2) | 0.055 (2) | 0.091 (3) | -0.0095 (19) | -0.011 (2) | -0.011 (2) |
| C13 | 0.057 (2) | 0.049 (2) | 0.057 (2) | 0.0071 (17) | -0.0188 (17) | -0.0054 (16) |
| C14 | 0.135 (5) | 0.059 (3) | 0.165 (5) | -0.015 (3) | 0.025 (4) | 0.035 (3) |
| C15 | 0.0302 (16) | 0.0368 (16) | 0.0320 (15) | 0.0018 (12) | -0.0075 (12) | -0.0062 (12) |
| C16 | 0.053 (2) | 0.0442 (19) | 0.0387 (17) | -0.0027 (15) | -0.0038 (15) | -0.0094 (14) |
| C17 | 0.069 (3) | 0.052 (2) | 0.056 (2) | -0.0121 (19) | 0.0116 (18) | -0.0040 (18) |
| C18 | 0.059 (2) | 0.054 (2) | 0.095 (3) | -0.0189 (19) | 0.001 (2) | -0.021 (2) |
| C19 | 0.081 (3) | 0.076 (3) | 0.084 (3) | -0.013 (2) | -0.022 (2) | -0.044 (2) |
| C20 | 0.070 (2) | 0.064 (2) | 0.0426 (18) | -0.0095 (19) | -0.0111 (17) | -0.0163 (17) |
| C21 | 0.117 (4) | 0.078 (3) | 0.168 (5) | -0.051 (3) | 0.018 (4) | -0.040 (3) |
| C22 | 0.0325 (16) | 0.0277 (15) | 0.0343 (15) | -0.0010 (12) | -0.0093 (13) | -0.0069 (12) |
| C23 | 0.052 (2) | 0.0447 (19) | 0.0503 (18) | 0.0061 (16) | -0.0275 (17) | -0.0096 (15) |
| C24 | 0.043 (2) | 0.058 (2) | 0.085 (3) | 0.0179 (16) | -0.0356 (19) | -0.0264 (19) |
| C25 | 0.0298 (18) | 0.071 (2) | 0.069 (2) | 0.0075 (16) | -0.0082 (16) | -0.0322 (19) |
| C26 | 0.0377 (18) | 0.053 (2) | 0.0394 (16) | 0.0000 (15) | -0.0062 (14) | -0.0168 (14) |
| C27 | 0.0286 (15) | 0.0252 (14) | 0.0345 (15) | -0.0006 (12) | -0.0104 (12) | -0.0052 (11) |
| C28 | 0.0317 (16) | 0.0318 (16) | 0.0373 (16) | 0.0033 (12) | -0.0066 (13) | -0.0070 (13) |
| C29 | 0.047 (2) | 0.0477 (19) | 0.0355 (16) | 0.0009 (15) | -0.0055 (14) | -0.0062 (14) |
| C30 | 0.067 (2) | 0.051 (2) | 0.0431 (18) | 0.0044 (17) | -0.0228 (18) | -0.0009 (15) |
| C31 | 0.049 (2) | 0.049 (2) | 0.066 (2) | 0.0089 (16) | -0.0274 (18) | -0.0090 (17) |
| C32 | 0.0329 (17) | 0.0420 (18) | 0.0548 (19) | 0.0045 (14) | -0.0079 (15) | -0.0100 (15) |
| C33 | 0.0364 (17) | 0.0329 (16) | 0.0350 (16) | -0.0003 (13) | -0.0069 (13) | -0.0067 (13) |
| N1 | 0.0475 (18) | 0.0478 (18) | 0.0314 (14) | -0.0006 (14) | -0.0081 (13) | -0.0071 (13) |
| N2 | 0.0354 (15) | 0.0369 (16) | 0.0321 (14) | 0.0021 (12) | -0.0107 (12) | -0.0009 (12) |
| N3 | 0.0421 (18) | 0.062 (2) | 0.0355 (15) | 0.0089 (15) | -0.0029 (14) | -0.0019 (14) |
| N4 | 0.0370 (17) | 0.0503 (19) | 0.0405 (16) | 0.0109 (14) | -0.0014 (14) | -0.0025 (15) |
| O1 | 0.0547 (14) | 0.0508 (14) | 0.0867 (16) | -0.0012 (11) | -0.0447 (13) | 0.0072 (12) |
| O2 | 0.0408 (13) | 0.0758 (16) | 0.0386 (11) | -0.0182 (11) | 0.0041 (10) | -0.0007 (11) |
| O3 | 0.0690 (15) | 0.0557 (14) | 0.0384 (11) | -0.0137 (11) | -0.0041 (11) | -0.0157 (10) |
| O4 | 0.140 (2) | 0.0519 (15) | 0.0405 (13) | -0.0117 (15) | -0.0027 (14) | 0.0045 (11) |
| O5 | 0.0742 (18) | 0.0717 (18) | 0.139 (2) | -0.0116 (14) | -0.0529 (18) | -0.0133 (17) |
| O6 | 0.0810 (18) | 0.0399 (13) | 0.0926 (18) | 0.0209 (12) | -0.0150 (14) | -0.0237 (12) |
| O7 | 0.0456 (14) | 0.0586 (15) | 0.0630 (14) | -0.0003 (11) | 0.0181 (11) | -0.0047 (11) |
| O8 | 0.0467 (13) | 0.0462 (13) | 0.0560 (13) | 0.0120 (10) | -0.0059 (11) | 0.0041 (10) |

| | | | | | | |
|----|------------|-------------|-------------|--------------|--------------|-------------|
| O9 | 0.107 (2) | 0.0680 (16) | 0.0621 (15) | -0.0359 (14) | -0.0489 (14) | 0.0073 (12) |
| S1 | 0.0338 (4) | 0.0428 (4) | 0.0319 (4) | -0.0060 (3) | -0.0097 (3) | -0.0026 (3) |
| S2 | 0.0539 (5) | 0.0359 (4) | 0.0453 (5) | 0.0034 (4) | -0.0132 (4) | -0.0083 (3) |
| S3 | 0.0387 (4) | 0.0400 (4) | 0.0334 (4) | -0.0016 (3) | -0.0077 (3) | -0.0003 (3) |

Geometric parameters (Å, °)

| | | | |
|----------|-----------|---------|-------------|
| C1—C2 | 1.379 (4) | C22—C23 | 1.368 (4) |
| C1—C6 | 1.381 (4) | C22—C27 | 1.385 (3) |
| C1—S1 | 1.753 (3) | C22—N1 | 1.459 (4) |
| C2—C3 | 1.381 (4) | C23—C24 | 1.370 (4) |
| C2—H2 | 0.9300 | C23—H23 | 0.9300 |
| C3—C4 | 1.373 (5) | C24—C25 | 1.372 (4) |
| C3—H3 | 0.9300 | C24—H24 | 0.9300 |
| C4—C5 | 1.372 (5) | C25—C26 | 1.377 (4) |
| C4—C7 | 1.509 (5) | C25—H25 | 0.9300 |
| C5—C6 | 1.378 (5) | C26—C27 | 1.373 (4) |
| C5—H5 | 0.9300 | C26—H26 | 0.9300 |
| C6—H6 | 0.9300 | C27—N2 | 1.460 (3) |
| C7—H7A | 0.9600 | C28—C29 | 1.372 (4) |
| C7—H7B | 0.9600 | C28—C33 | 1.389 (4) |
| C7—H7C | 0.9600 | C28—N4 | 1.463 (4) |
| C8—C9 | 1.374 (4) | C29—C30 | 1.372 (4) |
| C8—C13 | 1.378 (4) | C29—H29 | 0.9300 |
| C8—S2 | 1.755 (3) | C30—C31 | 1.378 (4) |
| C9—C10 | 1.393 (5) | C30—H30 | 0.9300 |
| C9—H9 | 0.9300 | C31—C32 | 1.366 (4) |
| C10—C11 | 1.358 (5) | C31—H31 | 0.9300 |
| C10—H10 | 0.9300 | C32—C33 | 1.392 (4) |
| C11—C12 | 1.354 (5) | C32—H32 | 0.9300 |
| C11—C14 | 1.523 (5) | C33—N3 | 1.382 (4) |
| C12—C13 | 1.383 (4) | N1—H1A | 0.863 (10) |
| C12—H12 | 0.9300 | N1—H1B | 0.864 (10) |
| C13—H13 | 0.9300 | N1—H1C | 0.867 (10) |
| C14—H14A | 0.9600 | N2—H2A | 0.874 (10) |
| C14—H14B | 0.9600 | N2—H2B | 0.869 (10) |
| C14—H14C | 0.9600 | N2—H2C | 0.868 (10) |
| C15—C20 | 1.372 (4) | N3—H3A | 0.860 (10) |
| C15—C16 | 1.377 (4) | N3—H3B | 0.862 (10) |
| C15—S3 | 1.759 (3) | N4—H4A | 0.867 (10) |
| C16—C17 | 1.376 (4) | N4—H4B | 0.868 (10) |
| C16—H16 | 0.9300 | N4—H4C | 0.865 (10) |
| C17—C18 | 1.365 (5) | O1—S1 | 1.442 (2) |
| C17—H17 | 0.9300 | O2—S1 | 1.4478 (19) |
| C18—C19 | 1.378 (5) | O3—S1 | 1.4428 (19) |
| C18—C21 | 1.518 (5) | O4—S2 | 1.431 (2) |
| C19—C20 | 1.385 (5) | O5—S2 | 1.421 (2) |
| C19—H19 | 0.9300 | O6—S2 | 1.442 (2) |

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|--------------|-----------|-------------|------------|
| C20—H20 | 0.9300 | O7—S3 | 1.436 (2) |
| C21—H21A | 0.9600 | O8—S3 | 1.458 (2) |
| C21—H21B | 0.9600 | O9—S3 | 1.433 (2) |
| C21—H21C | 0.9600 | | |
| C2—C1—C6 | 119.0 (3) | C27—C22—N1 | 122.1 (2) |
| C2—C1—S1 | 119.1 (2) | C22—C23—C24 | 120.2 (3) |
| C6—C1—S1 | 121.8 (2) | C22—C23—H23 | 119.9 |
| C1—C2—C3 | 119.6 (3) | C24—C23—H23 | 119.9 |
| C1—C2—H2 | 120.2 | C23—C24—C25 | 120.2 (3) |
| C3—C2—H2 | 120.2 | C23—C24—H24 | 119.9 |
| C4—C3—C2 | 121.9 (3) | C25—C24—H24 | 119.9 |
| C4—C3—H3 | 119.0 | C24—C25—C26 | 119.7 (3) |
| C2—C3—H3 | 119.0 | C24—C25—H25 | 120.1 |
| C5—C4—C3 | 117.8 (3) | C26—C25—H25 | 120.1 |
| C5—C4—C7 | 120.9 (4) | C27—C26—C25 | 120.4 (3) |
| C3—C4—C7 | 121.3 (4) | C27—C26—H26 | 119.8 |
| C4—C5—C6 | 121.4 (3) | C25—C26—H26 | 119.8 |
| C4—C5—H5 | 119.3 | C26—C27—C22 | 119.3 (3) |
| C6—C5—H5 | 119.3 | C26—C27—N2 | 118.7 (2) |
| C5—C6—C1 | 120.2 (3) | C22—C27—N2 | 121.9 (2) |
| C5—C6—H6 | 119.9 | C29—C28—C33 | 121.8 (3) |
| C1—C6—H6 | 119.9 | C29—C28—N4 | 118.6 (3) |
| C4—C7—H7A | 109.5 | C33—C28—N4 | 119.5 (2) |
| C4—C7—H7B | 109.5 | C28—C29—C30 | 120.2 (3) |
| H7A—C7—H7B | 109.5 | C28—C29—H29 | 119.9 |
| C4—C7—H7C | 109.5 | C30—C29—H29 | 119.9 |
| H7A—C7—H7C | 109.5 | C29—C30—C31 | 119.0 (3) |
| H7B—C7—H7C | 109.5 | C29—C30—H30 | 120.5 |
| C9—C8—C13 | 119.4 (3) | C31—C30—H30 | 120.5 |
| C9—C8—S2 | 121.0 (2) | C32—C31—C30 | 120.8 (3) |
| C13—C8—S2 | 119.6 (2) | C32—C31—H31 | 119.6 |
| C8—C9—C10 | 119.3 (3) | C30—C31—H31 | 119.6 |
| C8—C9—H9 | 120.4 | C31—C32—C33 | 121.3 (3) |
| C10—C9—H9 | 120.4 | C31—C32—H32 | 119.4 |
| C11—C10—C9 | 121.6 (3) | C33—C32—H32 | 119.4 |
| C11—C10—H10 | 119.2 | N3—C33—C28 | 122.5 (3) |
| C9—C10—H10 | 119.2 | N3—C33—C32 | 120.6 (3) |
| C12—C11—C10 | 118.3 (3) | C28—C33—C32 | 116.9 (3) |
| C12—C11—C14 | 121.5 (4) | C22—N1—H1A | 111 (2) |
| C10—C11—C14 | 120.2 (4) | C22—N1—H1B | 110 (2) |
| C11—C12—C13 | 122.0 (4) | H1A—N1—H1B | 113 (3) |
| C11—C12—H12 | 119.0 | C22—N1—H1C | 109 (2) |
| C13—C12—H12 | 119.0 | H1A—N1—H1C | 111 (3) |
| C8—C13—C12 | 119.4 (3) | H1B—N1—H1C | 102 (3) |
| C8—C13—H13 | 120.3 | C27—N2—H2A | 109.6 (18) |
| C12—C13—H13 | 120.3 | C27—N2—H2B | 113 (2) |
| C11—C14—H14A | 109.5 | H2A—N2—H2B | 108 (3) |

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|-----------------|------------|-----------------|-------------|
| C11—C14—H14B | 109.5 | C27—N2—H2C | 113 (2) |
| H14A—C14—H14B | 109.5 | H2A—N2—H2C | 104 (3) |
| C11—C14—H14C | 109.5 | H2B—N2—H2C | 108 (3) |
| H14A—C14—H14C | 109.5 | C33—N3—H3A | 114.0 (19) |
| H14B—C14—H14C | 109.5 | C33—N3—H3B | 118 (2) |
| C20—C15—C16 | 119.2 (3) | H3A—N3—H3B | 110 (3) |
| C20—C15—S3 | 120.6 (2) | C28—N4—H4A | 107 (3) |
| C16—C15—S3 | 120.1 (2) | C28—N4—H4B | 115 (2) |
| C17—C16—C15 | 120.4 (3) | H4A—N4—H4B | 108 (3) |
| C17—C16—H16 | 119.8 | C28—N4—H4C | 112 (3) |
| C15—C16—H16 | 119.8 | H4A—N4—H4C | 107 (4) |
| C18—C17—C16 | 121.3 (3) | H4B—N4—H4C | 107 (3) |
| C18—C17—H17 | 119.3 | O1—S1—O3 | 112.34 (14) |
| C16—C17—H17 | 119.3 | O1—S1—O2 | 112.08 (13) |
| C17—C18—C19 | 117.9 (3) | O3—S1—O2 | 112.18 (12) |
| C17—C18—C21 | 121.6 (4) | O1—S1—C1 | 105.76 (12) |
| C19—C18—C21 | 120.5 (4) | O3—S1—C1 | 107.14 (12) |
| C18—C19—C20 | 121.6 (3) | O2—S1—C1 | 106.84 (13) |
| C18—C19—H19 | 119.2 | O5—S2—O4 | 112.67 (18) |
| C20—C19—H19 | 119.2 | O5—S2—O6 | 113.46 (16) |
| C15—C20—C19 | 119.5 (3) | O4—S2—O6 | 111.30 (16) |
| C15—C20—H20 | 120.3 | O5—S2—C8 | 106.66 (15) |
| C19—C20—H20 | 120.3 | O4—S2—C8 | 106.07 (13) |
| C18—C21—H21A | 109.5 | O6—S2—C8 | 106.05 (14) |
| C18—C21—H21B | 109.5 | O9—S3—O7 | 115.46 (15) |
| H21A—C21—H21B | 109.5 | O9—S3—O8 | 111.31 (14) |
| C18—C21—H21C | 109.5 | O7—S3—O8 | 111.09 (13) |
| H21A—C21—H21C | 109.5 | O9—S3—C15 | 106.42 (12) |
| H21B—C21—H21C | 109.5 | O7—S3—C15 | 106.38 (12) |
| C23—C22—C27 | 120.1 (3) | O8—S3—C15 | 105.44 (12) |
| C23—C22—N1 | 117.8 (2) | | |
| C6—C1—C2—C3 | 0.1 (4) | C25—C26—C27—C22 | -1.5 (4) |
| S1—C1—C2—C3 | 180.0 (2) | C25—C26—C27—N2 | 179.0 (3) |
| C1—C2—C3—C4 | 0.1 (5) | C23—C22—C27—C26 | 1.9 (4) |
| C2—C3—C4—C5 | 0.0 (5) | N1—C22—C27—C26 | -175.8 (2) |
| C2—C3—C4—C7 | -179.6 (4) | C23—C22—C27—N2 | -178.7 (2) |
| C3—C4—C5—C6 | -0.4 (5) | N1—C22—C27—N2 | 3.6 (4) |
| C7—C4—C5—C6 | 179.2 (4) | C33—C28—C29—C30 | 1.1 (4) |
| C4—C5—C6—C1 | 0.7 (5) | N4—C28—C29—C30 | -179.9 (3) |
| C2—C1—C6—C5 | -0.5 (4) | C28—C29—C30—C31 | -0.2 (4) |
| S1—C1—C6—C5 | 179.6 (2) | C29—C30—C31—C32 | -0.4 (5) |
| C13—C8—C9—C10 | -0.5 (5) | C30—C31—C32—C33 | 0.2 (5) |
| S2—C8—C9—C10 | -179.7 (3) | C29—C28—C33—N3 | 176.5 (3) |
| C8—C9—C10—C11 | -0.1 (6) | N4—C28—C33—N3 | -2.6 (4) |
| C9—C10—C11—C12 | 0.7 (6) | C29—C28—C33—C32 | -1.3 (4) |
| C9—C10—C11—C14 | -179.0 (4) | N4—C28—C33—C32 | 179.7 (2) |
| C10—C11—C12—C13 | -0.8 (6) | C31—C32—C33—N3 | -177.1 (3) |

| | | | |
|-----------------|------------|-----------------|------------|
| C14—C11—C12—C13 | 178.9 (4) | C31—C32—C33—C28 | 0.7 (4) |
| C9—C8—C13—C12 | 0.4 (4) | C2—C1—S1—O1 | 54.1 (2) |
| S2—C8—C13—C12 | 179.6 (2) | C6—C1—S1—O1 | -126.0 (2) |
| C11—C12—C13—C8 | 0.3 (5) | C2—C1—S1—O3 | -65.9 (2) |
| C20—C15—C16—C17 | -0.2 (5) | C6—C1—S1—O3 | 114.0 (2) |
| S3—C15—C16—C17 | -177.4 (3) | C2—C1—S1—O2 | 173.7 (2) |
| C15—C16—C17—C18 | 0.5 (5) | C6—C1—S1—O2 | -6.4 (3) |
| C16—C17—C18—C19 | -0.3 (6) | C9—C8—S2—O5 | -5.2 (3) |
| C16—C17—C18—C21 | -178.9 (4) | C13—C8—S2—O5 | 175.6 (2) |
| C17—C18—C19—C20 | -0.1 (6) | C9—C8—S2—O4 | -125.6 (3) |
| C21—C18—C19—C20 | 178.5 (4) | C13—C8—S2—O4 | 55.2 (3) |
| C16—C15—C20—C19 | -0.2 (5) | C9—C8—S2—O6 | 116.0 (3) |
| S3—C15—C20—C19 | 177.0 (3) | C13—C8—S2—O6 | -63.2 (3) |
| C18—C19—C20—C15 | 0.4 (6) | C20—C15—S3—O9 | 155.4 (3) |
| C27—C22—C23—C24 | -0.5 (4) | C16—C15—S3—O9 | -27.4 (3) |
| N1—C22—C23—C24 | 177.3 (3) | C20—C15—S3—O7 | 31.8 (3) |
| C22—C23—C24—C25 | -1.3 (5) | C16—C15—S3—O7 | -151.0 (2) |
| C23—C24—C25—C26 | 1.6 (5) | C20—C15—S3—O8 | -86.3 (3) |
| C24—C25—C26—C27 | -0.2 (5) | C16—C15—S3—O8 | 90.9 (2) |

Hydrogen-bond geometry (\AA , $^\circ$)

Cg1, Cg2 and Cg3 are the centroids of the C1–C6, C8–C13 and C15–C20 rings, respectively.

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|-------------------------------------|----------|-------------|-------------|---------------|
| N2—H2A \cdots O8 ⁱ | 0.87 (1) | 1.91 (1) | 2.785 (3) | 175 (3) |
| N2—H2B \cdots O6 ⁱ | 0.87 (1) | 1.86 (1) | 2.707 (3) | 164 (3) |
| N3—H3A \cdots O3 ⁱ | 0.86 (1) | 2.28 (1) | 3.113 (3) | 164 (3) |
| N1—H1A \cdots O3 ⁱ | 0.86 (1) | 2.06 (2) | 2.810 (3) | 144 (3) |
| N1—H1A \cdots O6 ⁱ | 0.86 (1) | 2.44 (3) | 2.852 (4) | 110 (3) |
| N3—H3B \cdots O1 ⁱⁱ | 0.86 (1) | 2.15 (1) | 2.989 (4) | 163 (3) |
| N4—H4B \cdots O1 ⁱⁱ | 0.87 (1) | 2.12 (2) | 2.910 (4) | 150 (3) |
| C24—H24 \cdots O1 ⁱⁱ | 0.93 | 2.55 | 3.197 (4) | 127 |
| N1—H1C \cdots O4 ⁱⁱⁱ | 0.87 (1) | 1.84 (1) | 2.698 (3) | 169 (4) |
| N1—H1B \cdots O2 ⁱⁱⁱ | 0.86 (1) | 1.83 (1) | 2.669 (3) | 164 (3) |
| N2—H2C \cdots O2 ⁱⁱⁱ | 0.87 (1) | 2.30 (3) | 2.873 (3) | 124 (2) |
| N2—H2C \cdots O9 ^{iv} | 0.87 (1) | 2.07 (2) | 2.833 (3) | 146 (3) |
| N4—H4A \cdots O5 ^v | 0.87 (1) | 1.88 (1) | 2.742 (4) | 175 (4) |
| N4—H4C \cdots O7 ^v | 0.87 (1) | 2.06 (2) | 2.826 (3) | 147 (4) |
| C5—H5 \cdots Cg2 ^{vi} | 0.93 | 2.87 | 3.629 (4) | 140 |
| C10—H10 \cdots Cg3 ^{vii} | 0.93 | 2.81 | 3.587 (5) | 142 |
| C13—H13 \cdots Cg1 | 0.93 | 2.93 | 3.609 (3) | 131 |

Symmetry codes: (i) $-x+2, -y+1, -z-1$; (ii) $x, y, z-1$; (iii) $x+1, y, z-1$; (iv) $x+1, y, z$; (v) $-x+1, -y+1, -z-1$; (vi) $-x+1, -y+2, -z$; (vii) $-x+1, -y+2, -z-1$.