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# 4-Methyl-1*H*-1,5-benzodiazepin-2(3*H*)-one

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In the title compound,  $C_{10}H_{10}N_2O$ , the seven-membered heterocycle displays a half-chair conformation. In the crystal,  $N-H\cdots O$  and  $C-H\cdots O$  hydrogen bonds link the molecules into chains propagating along the *a*-axis direction.



### Structure description

Benzodiazepines are a class of drugs that act on the GABA-A receptor as an allosteric modulator and increase the frequency of opening of its chloride channel (Twyman *et al.* 1989). The biological effects of these drugs include hypnotic, anticonvulsant and muscle relaxant properties (De Sarro *et al.* 1995). As part of our studies in this area, we now describe the synthesis and structure of the title compound.

The molecule of the title compound (Fig. 1) is built up from fused six- and sevenmembered rings. The seven-membered ring displays a half-chair conformation as indicated by the puckering amplitude  $Q_{\rm T} = 0.8734$  (16) Å and spherical polar angle  $\varphi_2 =$ 205.48 (11)° and  $\varphi_3 = 310.2$  (4)°. In the crystal, the molecules are linked by N–H···O and C–H···O hydrogen bonds (Table 1), generating [100] chains as shown in Fig. 2.

### Synthesis and crystallization

Ethyl acetoacetate (0.011 mmol) was added to a stirred solution of benzene-1,2-diamine (0.01 mmol) in 100 ml of xylene. The mixture was stirred at reflux for 1 h. The resulting precipitate was filtered, washed with ethanol, then dried giving a white powder. Colourless prisms of the title compound were obtained by recrystallization from diethyl ether solution.



# data reports

 $C8-H8A\cdotsO1^{ii}$ 

Table 1 Hydrogen-bond geometry (Å, °).						
$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D - H \cdots A$		
$N1 - H1N \cdots O1^{i}$	0.92 (2)	1.94 (2)	2.8481 (17)	169 (2)		

Symmetry codes: (	(i) $x - \frac{1}{2}$ ,	$-y + \frac{1}{2}$	-z + 2; (ii)	$x + \frac{1}{2}, -y$	$z + \frac{1}{2}, -z + 2.$

0.97

Yield = 90% (white solid); m.p. =  $176-178^{\circ}$ C. <sup>1</sup>H NMR (300.13 MHz; CDCl<sub>3</sub>): 2.40 (3*H*, -CH<sub>3</sub>, *s*); 3.15 (2*H*, -CH<sub>2</sub>-, *s*); 7.08–7.37 (4 $H_{arom}$ , m); 9.32 (1H, –NH, s). <sup>13</sup>C NMR (75.47 MHz; CDCl<sub>3</sub>): 28.01 (-CH<sub>3</sub>); 43.57 (-CH<sub>2</sub>-); 121.88-

2.57

3.480(2)

156



#### Figure 1

Molecular structure of the title compound with displacement ellipsoids drawn at the 30% probability level.



Figure 2

Structure of the title compound, showing molecules linked through N- $H \cdots O$  and  $C - H \cdots O$  hydrogen bonds (dashed lines).

Table	2	
Experi	mental	details.

Crystal data	
Chemical formula	$C_{10}H_{10}N_2O$
M <sub>r</sub>	174.20
Crystal system, space group	Orthorhombic, $P2_12_12_1$
Temperature (K)	296
a, b, c (Å)	7.5676 (2), 10.7802 (2), 11.4092 (3)
$V(Å^3)$	930.77 (4)
Ζ	4
Radiation type	Μο Κα
$\mu \ (\mathrm{mm}^{-1})$	0.08
Crystal size (mm)	$0.32 \times 0.28 \times 0.20$
Data collection	
Diffractometer	Bruker X8 APEXII CCD area-
	detector
No. of measured, independent and	25369, 1825, 1747
observed $[I > 2\sigma(I)]$ reflections	
R <sub>int</sub>	0.018
$(\sin \theta / \lambda)_{\max} ( A^{-1} )$	0.616
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.028, 0.086, 1.03
No. of reflections	1825
No. of parameters	123
No. of restraints	1
H-atom treatment	H atoms treated by a mixture of independent and constrained refinement
$\Delta \rho_{\rm max}, \Delta \rho_{\rm min} \ (e \ {\rm \AA}^{-3})$	0.14, -0.15
Absolute structure	Flack x determined using 705
	quotients $[(I^+)-(I^-)]/[(I^+)+(I^-)]$ (Parsons <i>et al.</i> , 2013)
Absolute structure parameter	-0.3 (2)

Computer programs: APEX2 aand SAINT (Bruker, 2009), SHELXT (Sheldrick, 2015a), SHELXL2014 (Sheldrick, 2015b), PLATON (Spek, 2009) and publCIF (Westrip, 2010).

139.62 (6C<sub>arom</sub>); 162.89 (CN); 167.36 (CO). MS-EI:  $[M + 1]^+$  = 175.

### Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

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# full crystallographic data

*IUCrData* (2018). **3**, x181718 [https://doi.org/10.1107/S2414314618017182]

# 4-Methyl-1*H*-1,5-benzodiazepin-2(3*H*)-one

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4-Methyl-1H-1,5-benzodiazepin-2(3H)-one

Crystat aata	
$C_{10}H_{10}N_2O$	
M = 174.20	

Conversal data

 $M_r = 174.20$ Orthorhombic,  $P2_12_12_1$  a = 7.5676 (2) Å b = 10.7802 (2) Å c = 11.4092 (3) Å V = 930.77 (4) Å<sup>3</sup> Z = 4F(000) = 368

## Data collection

Bruker X8 APEXII CCD area-detector<br/>diffractometerRadiation source: fine-focus sealed tubeGraphite monochromator<br/> $\omega$  and  $\varphi$  scans25369 measured reflections1825 independent reflections

## Refinement

Refinement on  $F^2$ H atoms treated by a mixture of independent Least-squares matrix: full and constrained refinement  $w = 1/[\sigma^2(F_o^2) + (0.0631P)^2 + 0.0577P]$  $R[F^2 > 2\sigma(F^2)] = 0.028$  $wR(F^2) = 0.086$ where  $P = (F_0^2 + 2F_c^2)/3$ S = 1.03 $(\Delta/\sigma)_{\rm max} < 0.001$  $\Delta \rho_{\rm max} = 0.14 \text{ e } \text{\AA}^{-3}$ 1825 reflections  $\Delta \rho_{\rm min} = -0.15 \ {\rm e} \ {\rm \AA}^{-3}$ 123 parameters 1 restraint Absolute structure: Flack x determined using Primary atom site location: dual 705 quotients  $[(I^+)-(I^-)]/[(I^+)+(I^-)]$  (Parsons *et al.*, Hydrogen site location: mixed 2013) Absolute structure parameter: -0.3 (2)

## Special details

**Geometry**. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement**. The H atoms were located in a difference map and treated as riding with C—H = 0.93–0.97 Å and N—H = 0.86 Å, and with  $U_{iso}(H) = 1.2-1.5Ueq(C, N)$ .

 $D_x = 1.243 \text{ Mg m}^{-3}$ Mo K $\alpha$  radiation,  $\lambda = 0.71073 \text{ Å}$ Cell parameters from 245 reflections  $\theta = 0.5-32^{\circ}$  $\mu = 0.08 \text{ mm}^{-1}$ T = 296 KPrism, colourless  $0.32 \times 0.28 \times 0.20 \text{ mm}$ 

1747 reflections with  $I > 2\sigma(I)$   $R_{int} = 0.018$   $\theta_{max} = 26.0^{\circ}, \ \theta_{min} = 3.6^{\circ}$   $h = -9 \rightarrow 9$   $k = -13 \rightarrow 13$  $l = -13 \rightarrow 14$ 

	x	у	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	
C1	0.0092 (2)	0.51751 (13)	0.91727 (13)	0.0373 (3)	
С9	0.2131 (2)	0.35141 (15)	0.97843 (13)	0.0379 (3)	
C6	0.1120 (2)	0.58213 (14)	0.83489 (13)	0.0410 (4)	
C8	0.3621 (2)	0.44253 (15)	0.96221 (14)	0.0402 (3)	
H8A	0.4723	0.4076	0.9906	0.048*	
H8B	0.3383	0.5185	1.0049	0.048*	
C2	-0.1472 (2)	0.56990 (16)	0.95911 (15)	0.0466 (4)	
H2	-0.2167	0.5258	1.0119	0.056*	
C5	0.0527 (3)	0.69915 (16)	0.79881 (16)	0.0550 (5)	
Н5	0.1177	0.7428	0.7433	0.066*	
C7	0.3736 (2)	0.46818 (16)	0.83226 (14)	0.0439 (4)	
C3	-0.2001 (3)	0.68650 (18)	0.92309 (17)	0.0601 (5)	
Н3	-0.3034	0.7213	0.9524	0.072*	
C4	-0.0983 (3)	0.75106 (17)	0.84313 (19)	0.0641 (6)	
H4	-0.1325	0.8300	0.8193	0.077*	
C10	0.5213 (3)	0.4081 (3)	0.76664 (19)	0.0667 (6)	
H10A	0.5044	0.4201	0.6840	0.100*	
H10B	0.5230	0.3209	0.7837	0.100*	
H10C	0.6316	0.4445	0.7900	0.100*	
N1	0.05078 (16)	0.39599 (12)	0.95461 (12)	0.0394 (3)	
N2	0.26065 (19)	0.53402 (13)	0.77731 (12)	0.0459 (3)	
01	0.23581 (16)	0.24292 (11)	1.00743 (13)	0.0519 (3)	
H1N	-0.042 (3)	0.3435 (19)	0.9707 (19)	0.057 (6)*	

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(\hat{A}^2)$ 

Atomic displacement parameters  $(Å^2)$ 

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0376 (7)	0.0352 (7)	0.0392 (7)	0.0012 (6)	-0.0087 (6)	0.0009 (6)
C9	0.0332 (7)	0.0387 (8)	0.0419 (7)	0.0000 (6)	-0.0015 (6)	0.0047 (6)
C6	0.0467 (8)	0.0381 (7)	0.0382 (7)	-0.0027 (7)	-0.0076 (6)	0.0028 (6)
C8	0.0325 (7)	0.0438 (8)	0.0443 (8)	-0.0043 (6)	-0.0057 (6)	0.0036 (6)
C2	0.0424 (8)	0.0501 (9)	0.0473 (8)	0.0056 (7)	-0.0035 (7)	0.0015 (7)
C5	0.0670 (12)	0.0448 (9)	0.0531 (10)	-0.0003 (9)	-0.0084 (9)	0.0125 (8)
C7	0.0378 (8)	0.0483 (8)	0.0454 (8)	-0.0078 (7)	0.0004 (6)	0.0011 (7)
C3	0.0616 (11)	0.0585 (11)	0.0601 (10)	0.0239 (9)	-0.0067 (9)	-0.0046 (9)
C4	0.0828 (14)	0.0448 (9)	0.0648 (12)	0.0172 (10)	-0.0152 (11)	0.0076 (9)
C10	0.0471 (10)	0.0954 (16)	0.0576 (11)	0.0069 (11)	0.0063 (8)	-0.0033 (11)
N1	0.0317 (6)	0.0351 (6)	0.0514 (7)	-0.0029 (5)	-0.0022 (5)	0.0078 (5)
N2	0.0466 (7)	0.0498 (8)	0.0413 (6)	-0.0051 (7)	-0.0003 (6)	0.0054 (6)
01	0.0383 (6)	0.0411 (6)	0.0764 (8)	0.0023 (5)	-0.0022 (6)	0.0167 (5)

## *Geometric parameters (Å, °)*

C1—C2	1.395 (2)	С2—Н2	0.9300
C1—C6	1.405 (2)	C5—C4	1.370 (3)

C1—N1	1.4131 (19)	С5—Н5	0.9300
C9—O1	1.2275 (19)	C7—N2	1.276 (2)
C9—N1	1.347 (2)	C7—C10	1.493 (3)
C9—C8	1.507 (2)	C3—C4	1.382 (3)
C6—C5	1.401 (2)	С3—Н3	0.9300
C6—N2	1.402 (2)	C4—H4	0.9300
C8—C7	1.511 (2)	C10—H10A	0.9600
C8—H8A	0.9700	C10—H10B	0.9600
C8—H8B	0.9700	C10—H10C	0.9600
C2—C3	1.382 (2)	N1—H1N	0.918 (18)
C2—C1—C6	119 86 (14)	С6—С5—Н5	119 1
$C_2 - C_1 - N_1$	117 45 (14)	$N_{2}$ $C_{7}$ $C_{10}$	119.78 (16)
C6-C1-N1	122.54 (14)	$N_2$ $C_7$ $C_8$	123.06 (15)
01-C9-N1	121.47(14)	C10-C7-C8	117.11 (16)
01	123.32 (14)	C4—C3—C2	119.55 (18)
N1-C9-C8	115.16 (13)	C4—C3—H3	120.2
C5—C6—N2	116.89 (15)	С2—С3—Н3	120.2
C5—C6—C1	117.76 (16)	C5—C4—C3	120.20 (17)
N2—C6—C1	125.06 (14)	C5—C4—H4	119.9
C9—C8—C7	106.43 (13)	C3—C4—H4	119.9
С9—С8—Н8А	110.4	C7—C10—H10A	109.5
С7—С8—Н8А	110.4	C7—C10—H10B	109.5
С9—С8—Н8В	110.4	H10A-C10-H10B	109.5
С7—С8—Н8В	110.4	C7—C10—H10C	109.5
H8A—C8—H8B	108.6	H10A-C10-H10C	109.5
C3—C2—C1	120.80 (17)	H10B-C10-H10C	109.5
С3—С2—Н2	119.6	C9—N1—C1	126.51 (13)
C1—C2—H2	119.6	C9—N1—H1N	115.8 (14)
C4—C5—C6	121.79 (18)	C1—N1—H1N	117.5 (14)
С4—С5—Н5	119.1	C7—N2—C6	120.87 (14)

# Hydrogen-bond geometry (Å, °)

D—H···A	<i>D</i> —Н	Н…А	D····A	<i>D</i> —H··· <i>A</i>
N1—H1N····O1 <sup>i</sup>	0.92 (2)	1.94 (2)	2.8481 (17)	169 (2)
C8—H8A····O1 <sup>ii</sup>	0.97	2.57	3.480 (2)	156

Symmetry codes: (i) x-1/2, -y+1/2, -z+2; (ii) x+1/2, -y+1/2, -z+2.