

S,S-Diphenyl-*N*-tosyl sulfone diimine

Md. Chanmiya Sheikh,^a Toshiaki Yoshimura^{a*} and Ryuta Miyatake^b

^aDepartment of Applied Chemistry, Faculty of Engineering, University of Toyama, 3190 Gofuku, Toyama, 930-8555, Japan, and ^bCenter for Environmental Conservation and Research Safety, University of Toyama, 3190 Gofuku, Toyama, 930-8555, Japan. *Correspondence e-mail: by4ut6@bma.biglobe.ne.jp

Received 4 April 2019

Accepted 16 April 2019

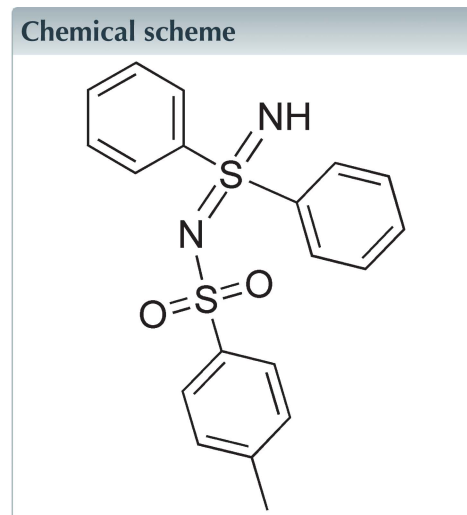
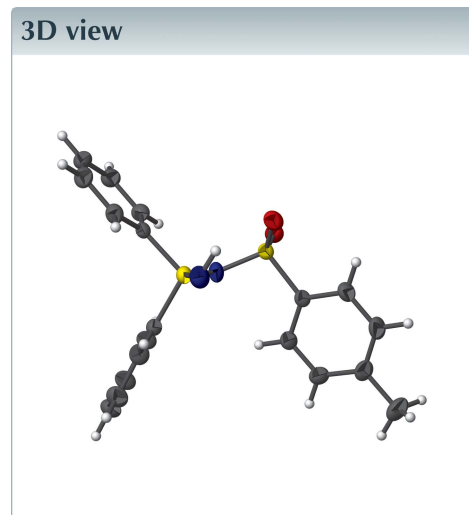
Edited by H. Ishida, Okayama University, Japan

Keywords: crystal structure; *S,S*-diphenyl-sulfilimide; *N*-tosyl-sulfonediimines; hydrogen bonding.

CCDC reference: 1452477

Structural data: full structural data are available from iucrdata.iucr.org

In the title compound [systematic name: *N*-(iminodiphenyl- λ^6 -sulfanylidene)-4-methylbenzene-1-sulfonamide], C₁₉H₁₈N₂O₂S₂, the configuration around the sulfur atom of the sulfonediimine group is a slightly distorted tetrahedron structure with two S=N bonds and two S—C bonds. The bond lengths of S=N (*p*-toluenesulfonyl) and S=N (NH) are 1.5785 (15) and 1.5158 (18) Å, respectively. The S—N=S plane makes a dihedral angle of 74.24 (14)° with the *p*-toluene ring, while it makes dihedral angles of 73.43 (13) and 41.98 (14)° with the phenyl rings. The two torsion angles of S—N=S—C (phenyl) and the S—N=S=N angle are 105.86 (12), −144.54 (11) and −25.67 (17)°, respectively. In the crystal, molecules are connected by pairs of N—H···O hydrogen bonds, forming inversion dimers with an *R*₂²(12) ring motif. The dimers are linked by C—H···O interactions, forming a tape structure along the *a*-axis direction.



Structure description

The aza analogues of sulfones such as sulfoximines and sulfonediimines have interesting pharmaceutical properties (Kennewell & Tavlour, 1975; Haake, 1976; Sellinger *et al.*, 1969). However, their chemical reactivities have not been fully explored as yet except for a few scattered utilizations in organic syntheses for example as an alkylidene-transfer reagent (Johnson *et al.*, 1973). To date, some related crystal structures have been reported (Yoshimura *et al.*, 2008). As part of our studies in this area, we report herein on the synthesis and crystal structure of the title compound.

The molecular structure of the title compound is illustrated in Fig. 1. The S2—N1 and S2—N2 bond lengths are 1.5785 (15) and 1.5158 (18) Å, respectively. These bond lengths are significantly longer than the S≡N triple bond of triphenylsulfanenitrile (1.462 Å; Yoshimura *et al.*, 1997), close to the S=N double bond lengths of *S,S*-dimethyl-sulfonediimine (1.533 Å, electron diffraction study; Oberhammer & Zeil, 1970), *S,S*-

Table 1
Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N2-H8\cdots O2^i$	0.89 (3)	2.20 (3)	3.018 (3)	152 (2)
$C12-H12\cdots O1^{ii}$	0.95	2.34	3.283 (2)	169

Symmetry codes: (i) $-x, -y, -z + 2$; (ii) $x - 1, y, z$.

diphenyl-*S*-pyrrolidinoiminosulfonium perchlorate [1.503 (2) Å for S–N (NH), X-ray; Sheikh *et al.*, 2017] and *S,S*-diphenylsulfodiimide-*N*-phenyl [1.526 Å for S–N (NH) and 1.546 (1) Å for S–N (Ph), X-ray; Yoshimura *et al.*, 2008], and shorter than that of *S,S*-diphenyl-*N*-tosylsulfilimine (1.628 Å, X-ray; Kálmán *et al.*, 1971). The N1–S2–N2 bond angle [126.05 (9)°] is larger than 109.5°, while the N–S–C and C–S–C bond angles [102.92 (9)–112.96 (19)°] are close to 109.5°. In the crystal, the molecules are linked through N–H⋯O and C–H⋯O hydrogen bonds (Table 1, Fig. 2), forming a tape along the *a*-axis direction (Fig. 3).

Synthesis and crystallization

The title compound was prepared by the method previously reported (Furukawa *et al.*, 1984) using *S,S*-diphenylsulfilimide mono hydrate with chloramine-T in the presence of excess sodium salt of tosylamide in anhydrous acetonitrile and was crystallized from an ethanol solution (yield 90%; m.p. 153–154°C).

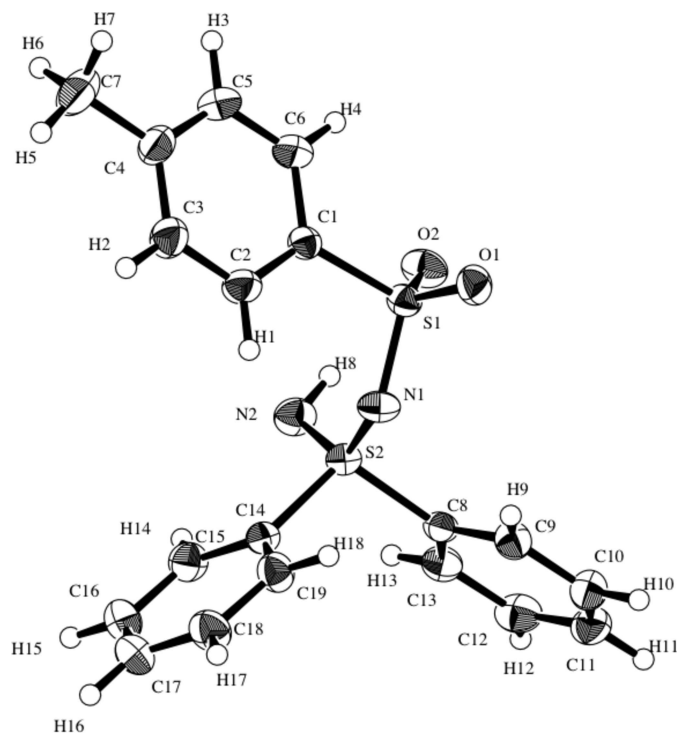


Figure 1
The molecular structure of the title compound, showing the atom labelling. Displacement ellipsoids are drawn at the 50% probability level.

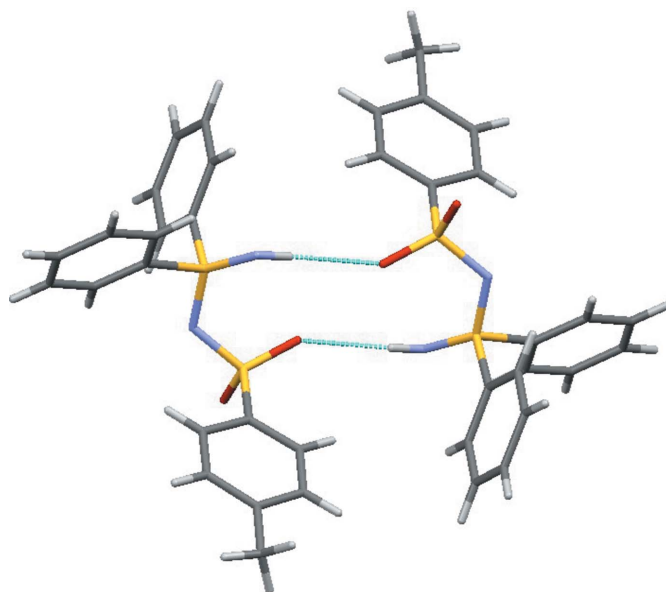


Figure 2
A view of the title compound showing a pair of N–H⋯O hydrogen bonds (blue dashed lines) forming an $R_2^2(12)$ ring motif.

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

Acknowledgements

The authors are grateful to the Department of Applied Chemistry, Faculty of Engineering, University of Toyama, for the provision of laboratory facilities and the Center for Environmental Conservation and Research Safety, University of Toyama, for providing facilities for single-crystal X-ray analysis.

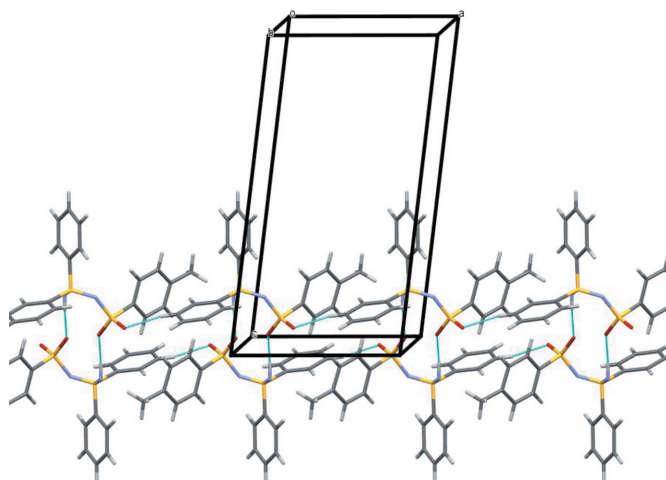


Figure 3
A partial packing diagram of the title compound, showing the tape along the *a*-axis direction formed via N–H⋯O and C–H⋯O hydrogen bonds (blue dashed lines).

Table 2
Experimental details.

Crystal data	
Chemical formula	C ₁₉ H ₁₈ N ₂ O ₂ S ₂
<i>M</i> _r	370.48
Crystal system, space group	Monoclinic, <i>P</i> 2 ₁ / <i>c</i>
Temperature (K)	173
<i>a</i> , <i>b</i> , <i>c</i> (Å)	9.47865 (17), 10.38462 (19), 18.0852 (4)
β (°)	97.4526 (7)
<i>V</i> (Å ³)	1765.13 (6)
<i>Z</i>	4
Radiation type	Cu <i>K</i> α
μ (mm ⁻¹)	2.86
Crystal size (mm)	0.56 × 0.52 × 0.14
Data collection	
Diffractometer	Rigaku R-Axis RAPID
Absorption correction	Multi-scan (<i>ABSCOR</i> ; Higashi, 1995)
<i>T</i> _{min} , <i>T</i> _{max}	0.500, 0.670
No. of measured, independent and observed [<i>F</i> ² > 2.0 σ (<i>F</i> ²)] reflections	19250, 3226, 2893
<i>R</i> _{int}	0.086
(sin θ/λ) _{max} (Å ⁻¹)	0.602
Refinement	
<i>R</i> [<i>F</i> ² > 2 σ (<i>F</i> ²)], <i>wR</i> (<i>F</i> ²), <i>S</i>	0.043, 0.127, 1.09
No. of reflections	3226
No. of parameters	231
H-atom treatment	H atoms treated by a mixture of independent and constrained refinement
$\Delta\rho_{\max}$, $\Delta\rho_{\min}$ (e Å ⁻³)	0.83, -0.52

Computer programs: *RAPID-AUTO* (Rigaku, 2001), *SIR92* (Altomare *et al.*, 1993), *SHELXL97* (Sheldrick, 2008) and *CrystalStructure* (Rigaku, 2010).

Funding information

This work was supported in part by the Japan Society for the Promotion of Science, JSPS (No. P11336).

References

- Altomare, A., Cascarano, G., Giacovazzo, C. & Guagliardi, A. (1993). *J. Appl. Cryst.* **26**, 343–350.
- Furukawa, N., Akutagawa, K. & Oae, S. (1984). *Phosphorus Sulfur Relat. Elem.* **20**, 1–14.
- Haake, M. (1976). *Topics in Sulfur Chemistry*, Vol. 1, edited by A. Senning. Stuttgart: George Thieme Verlag.
- Higashi, T. (1995). *ABSCOR*. Rigaku Corporation, Tokyo, Japan.
- Johnson, C. R., Kirchhoff, R. A., Reischer, R. J. & Katekar, G. F. J. (1973). *J. Am. Chem. Soc.* **95**, 4287–4291.
- Kálmán, A., Duffin, B. & Kucsman, Á. (1971). *Acta Cryst.* **B27**, 586–594.
- Kennewell, P. D. & Taylor, J. B. (1975). *Chem. Soc. Rev.* **4**, 189–209.
- Oberhammer, H. & Zeil, W. (1970). *Z. Naturforsch. Teil A*, **25**, 845–849.
- Rigaku (2001). *RAPID-AUTO*. Rigaku Corporation, Tokyo, Japan.
- Rigaku (2010). *CrystalStructure*. Rigaku Corporation, Tokyo, Japan.
- Sellinger, O. Z. & Ohlsson, W. G. (1969). *J. Neurochem.* **16**, 1193–1195.
- Sheikh, M. C., Yoshimura, T., Takata, E., Fujii, T. & Miyatake, R. (2017). *IUCrData*, **2**, x171251.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Yoshimura, T., Hamada, K., Imado, M., Hamata, K., Tomoda, T., Fujii, T., Morita, H., Shimasaki, S., Ono, S., Tsukurimichi, E., Furukawa, N. & Kimura, T. (1997). *J. Org. Chem.* **62**, 3802–3803.
- Yoshimura, T., Ishikawa, H., Fujie, T., Takata, E., Miyatake, R., Kita, H. & Tsukurimichi, E. (2008). *Synthesis*, pp. 1835–1840.

full crystallographic data

IUCrData (2019). 4, x190523 [https://doi.org/10.1107/S2414314619005236]

***S,S*-Diphenyl-*N*-tosyl sulfone diimine**

Md. Chanmiya Sheikh, Toshiaki Yoshimura and Ryuta Miyatake

N-(Iminodiphenyl- λ^6 -sulfanylidene)-4-methylbenzene-1-sulfonamide

Crystal data

$C_{19}H_{18}N_2O_2S_2$

$M_r = 370.48$

Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

$a = 9.47865$ (17) Å

$b = 10.38462$ (19) Å

$c = 18.0852$ (4) Å

$\beta = 97.4526$ (7)°

$V = 1765.13$ (6) Å³

$Z = 4$

$F(000) = 776.00$

$D_x = 1.394$ Mg m⁻³

Cu $K\alpha$ radiation, $\lambda = 1.54187$ Å

Cell parameters from 18540 reflections

$\theta = 4.3$ – 68.2 °

$\mu = 2.86$ mm⁻¹

$T = 173$ K

Platelet, colorless

$0.56 \times 0.52 \times 0.14$ mm

Data collection

Rigaku R-AXIS RAPID
diffractometer

Detector resolution: 10.000 pixels mm⁻¹

ω scans

Absorption correction: multi-scan
(ABSCOR; Higashi, 1995)

$T_{\min} = 0.500$, $T_{\max} = 0.670$

19250 measured reflections

3226 independent reflections

2893 reflections with $F^2 > 2.0\sigma(F^2)$

$R_{\text{int}} = 0.086$

$\theta_{\max} = 68.2$ °

$h = -11 \rightarrow 11$

$k = -12 \rightarrow 12$

$l = -21 \rightarrow 21$

Refinement

Refinement on F^2

$R[F^2 > 2\sigma(F^2)] = 0.043$

$wR(F^2) = 0.127$

$S = 1.09$

3226 reflections

231 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H atoms treated by a mixture of independent
and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0797P)^2 + 0.2545P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.83$ e Å⁻³

$\Delta\rho_{\min} = -0.52$ e Å⁻³

Special details

Geometry. ENTER SPECIAL DETAILS OF THE MOLECULAR GEOMETRY

Refinement. Refinement was performed using all reflections. The weighted R-factor (wR) and goodness of fit (S) are based on F^2 . R-factor (gt) are based on F. The threshold expression of $F^2 > 2.0 \sigma(F^2)$ is used only for calculating R-factor (gt).

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.17740 (4)	0.17061 (4)	0.93189 (2)	0.02550 (17)
S2	-0.08500 (4)	0.16500 (4)	0.84447 (2)	0.02510 (17)
O1	0.27017 (14)	0.27580 (12)	0.95689 (7)	0.0345 (4)
O2	0.10643 (14)	0.10862 (14)	0.98795 (7)	0.0375 (4)
N1	0.07009 (15)	0.22168 (15)	0.86291 (8)	0.0299 (4)
N2	-0.13277 (18)	0.03177 (16)	0.86480 (10)	0.0349 (4)
C1	0.28321 (18)	0.05373 (16)	0.89394 (10)	0.0251 (4)
C2	0.3026 (2)	0.05765 (18)	0.81932 (10)	0.0332 (5)
C3	0.3949 (3)	-0.02824 (19)	0.79274 (11)	0.0375 (5)
C4	0.4692 (2)	-0.11932 (18)	0.83931 (11)	0.0357 (5)
C5	0.4450 (3)	-0.1227 (2)	0.91330 (12)	0.0416 (5)
C6	0.3530 (3)	-0.03744 (19)	0.94066 (11)	0.0362 (5)
C7	0.5717 (3)	-0.2107 (3)	0.81015 (14)	0.0517 (6)
C8	-0.19435 (18)	0.29055 (17)	0.87394 (9)	0.0253 (4)
C9	-0.1404 (2)	0.41329 (18)	0.88810 (10)	0.0311 (4)
C10	-0.2267 (3)	0.50666 (19)	0.91370 (11)	0.0371 (5)
C11	-0.3646 (3)	0.4763 (2)	0.92469 (11)	0.0391 (5)
C12	-0.4183 (2)	0.3542 (2)	0.90947 (11)	0.0387 (5)
C13	-0.33348 (19)	0.25966 (19)	0.88366 (10)	0.0315 (5)
C14	-0.1138 (2)	0.16802 (16)	0.74561 (10)	0.0261 (4)
C15	-0.1989 (2)	0.07367 (18)	0.70885 (11)	0.0344 (5)
C16	-0.2169 (3)	0.0731 (2)	0.63126 (12)	0.0417 (5)
C17	-0.1524 (3)	0.1666 (2)	0.59297 (12)	0.0408 (5)
C18	-0.0697 (3)	0.2607 (2)	0.63049 (11)	0.0387 (5)
C19	-0.0498 (2)	0.26312 (18)	0.70765 (10)	0.0335 (5)
H1	0.2527	0.1189	0.7868	0.0398*
H2	0.4083	-0.0253	0.7416	0.0450*
H3	0.4929	-0.1851	0.9458	0.0500*
H4	0.3377	-0.0415	0.9915	0.0435*
H5	0.5742	-0.1943	0.7570	0.0620*
H6	0.5408	-0.2995	0.8170	0.0620*
H7	0.6669	-0.1981	0.8375	0.0620*
H8	-0.124 (3)	0.019 (3)	0.9139 (14)	0.046 (7)*
H9	-0.0456	0.4332	0.8804	0.0373*
H10	-0.1913	0.5913	0.9237	0.0445*
H11	-0.4230	0.5402	0.9429	0.0469*
H12	-0.5135	0.3349	0.9167	0.0464*
H13	-0.3698	0.1755	0.8729	0.0378*
H14	-0.2440	0.0108	0.7359	0.0413*
H15	-0.2734	0.0084	0.6047	0.0501*
H16	-0.1654	0.1660	0.5400	0.0489*
H17	-0.0260	0.3243	0.6033	0.0464*
H18	0.0063	0.3283	0.7340	0.0402*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0227 (3)	0.0297 (3)	0.0239 (3)	0.00039 (16)	0.0021 (2)	0.00052 (15)
S2	0.0231 (3)	0.0256 (3)	0.0264 (3)	0.00086 (15)	0.0024 (2)	0.00271 (15)
O1	0.0313 (7)	0.0336 (8)	0.0374 (8)	-0.0019 (6)	-0.0003 (6)	-0.0083 (6)
O2	0.0347 (8)	0.0512 (9)	0.0277 (7)	0.0014 (7)	0.0079 (6)	0.0089 (6)
N1	0.0219 (8)	0.0340 (9)	0.0328 (9)	0.0008 (7)	0.0002 (7)	0.0077 (7)
N2	0.0360 (9)	0.0301 (9)	0.0380 (10)	-0.0026 (7)	0.0028 (8)	0.0061 (7)
C1	0.0233 (9)	0.0246 (9)	0.0267 (9)	-0.0021 (7)	0.0004 (7)	-0.0006 (7)
C2	0.0386 (11)	0.0321 (10)	0.0287 (10)	0.0064 (9)	0.0041 (9)	0.0061 (8)
C3	0.0440 (12)	0.0363 (11)	0.0334 (11)	0.0061 (9)	0.0094 (10)	-0.0005 (8)
C4	0.0336 (11)	0.0270 (10)	0.0458 (12)	0.0029 (8)	0.0023 (9)	-0.0043 (9)
C5	0.0459 (12)	0.0348 (11)	0.0415 (11)	0.0123 (10)	-0.0049 (10)	0.0055 (9)
C6	0.0416 (12)	0.0371 (11)	0.0285 (10)	0.0066 (9)	-0.0010 (9)	0.0046 (8)
C7	0.0517 (14)	0.0382 (13)	0.0662 (16)	0.0148 (11)	0.0114 (13)	-0.0060 (11)
C8	0.0233 (9)	0.0305 (10)	0.0215 (9)	0.0032 (8)	0.0008 (7)	0.0013 (7)
C9	0.0281 (9)	0.0350 (11)	0.0304 (10)	-0.0016 (8)	0.0043 (8)	-0.0022 (8)
C10	0.0439 (12)	0.0323 (11)	0.0348 (10)	0.0032 (9)	0.0040 (9)	-0.0021 (9)
C11	0.0411 (12)	0.0445 (13)	0.0323 (10)	0.0139 (10)	0.0078 (9)	0.0009 (9)
C12	0.0268 (10)	0.0532 (13)	0.0372 (11)	0.0054 (9)	0.0089 (9)	0.0065 (10)
C13	0.0265 (9)	0.0355 (11)	0.0320 (10)	-0.0013 (8)	0.0017 (8)	0.0032 (8)
C14	0.0265 (9)	0.0262 (10)	0.0254 (9)	0.0036 (7)	0.0024 (8)	0.0007 (7)
C15	0.0354 (11)	0.0298 (10)	0.0375 (11)	-0.0027 (8)	0.0029 (9)	0.0010 (8)
C16	0.0449 (12)	0.0389 (12)	0.0386 (11)	-0.0024 (10)	-0.0052 (10)	-0.0086 (9)
C17	0.0487 (13)	0.0457 (13)	0.0266 (10)	0.0035 (10)	-0.0004 (10)	-0.0022 (9)
C18	0.0468 (12)	0.0403 (12)	0.0292 (10)	-0.0018 (9)	0.0052 (9)	0.0049 (8)
C19	0.0391 (11)	0.0305 (10)	0.0306 (10)	-0.0045 (8)	0.0031 (9)	-0.0007 (8)

Geometric parameters (Å, °)

S1—O1	1.4380 (13)	C15—C16	1.391 (3)
S1—O2	1.4394 (15)	C16—C17	1.381 (4)
S1—N1	1.5947 (15)	C17—C18	1.375 (3)
S1—C1	1.7688 (18)	C18—C19	1.384 (3)
S2—N1	1.5785 (15)	N2—H8	0.89 (3)
S2—N2	1.5158 (18)	C2—H1	0.950
S2—C8	1.7893 (19)	C3—H2	0.950
S2—C14	1.7735 (19)	C5—H3	0.950
C1—C2	1.386 (3)	C6—H4	0.950
C1—C6	1.379 (3)	C7—H5	0.980
C2—C3	1.379 (3)	C7—H6	0.980
C3—C4	1.395 (3)	C7—H7	0.980
C4—C5	1.387 (3)	C9—H9	0.950
C4—C7	1.502 (4)	C10—H10	0.950
C5—C6	1.379 (3)	C11—H11	0.950
C8—C9	1.385 (3)	C12—H12	0.950
C8—C13	1.390 (3)	C13—H13	0.950

C9—C10	1.386 (3)	C15—H14	0.950
C10—C11	1.384 (3)	C16—H15	0.950
C11—C12	1.380 (3)	C17—H16	0.950
C12—C13	1.388 (3)	C18—H17	0.950
C14—C15	1.384 (3)	C19—H18	0.950
C14—C19	1.386 (3)		
S1…N2	3.3570 (17)	C12…H2 ⁱⁱⁱ	3.0176
S2…O2	3.0244 (13)	C12…H3 ^{iv}	3.3035
O1…C2	3.408 (3)	C12…H5 ⁱⁱⁱ	3.2164
O1…C6	3.368 (3)	C12…H11 ^{xiv}	3.4108
O2…N2	3.070 (2)	C12…H15 ^{xv}	3.3133
O2…C6	3.002 (3)	C12…H16 ^v	3.1456
N1…C2	2.972 (3)	C13…H2 ⁱⁱⁱ	3.1945
N1…C9	2.896 (3)	C13…H4 ^{iv}	3.2026
N1…C19	2.921 (3)	C13…H5 ⁱⁱⁱ	3.2245
N2…C13	3.083 (3)	C13…H16 ^v	3.1559
N2…C15	2.842 (3)	C15…H6 ⁱⁱⁱ	3.4729
C1…C4	2.786 (3)	C15…H9 ^{vi}	3.3276
C2…C5	2.761 (3)	C15…H18 ^{vi}	3.2253
C3…C6	2.757 (3)	C16…H6 ⁱⁱⁱ	3.5778
C8…C11	2.748 (3)	C16…H9 ^{vi}	2.9117
C8…C19	3.475 (3)	C16…H11 ^{xvi}	3.5185
C9…C12	2.780 (3)	C16…H12 ^{xvi}	3.5816
C10…C13	2.785 (3)	C17…H3 ⁱⁱⁱ	3.5623
C13…C14	3.581 (3)	C17…H9 ^{vi}	3.0652
C14…C17	2.737 (3)	C17…H10 ^{vi}	3.4010
C15…C18	2.781 (3)	C18…H8 ⁱⁱⁱ	3.40 (3)
C16…C19	2.783 (3)	C18…H9 ^{vi}	3.5850
O1…C10 ⁱ	3.317 (3)	C18…H10 ^{vi}	3.2861
O1…C11 ⁱ	3.394 (3)	H1…C7 ^x	3.1191
O1…C12 ⁱⁱ	3.283 (3)	H1…H5 ^x	2.7234
O1…C16 ⁱⁱⁱ	3.481 (3)	H1…H6 ^x	3.0033
O2…O2 ^{iv}	3.095 (2)	H1…H7 ^x	3.1121
O2…N2 ^{iv}	3.018 (3)	H2…C7 ^x	3.4100
O2…C18 ^v	3.520 (3)	H2…C8 ^{vi}	3.3221
N2…O2 ^{iv}	3.018 (3)	H2…C9 ^{vi}	3.2879
N2…C18 ^{vi}	3.402 (3)	H2…C10 ^{vi}	3.1156
C5…C6 ^{vii}	3.478 (3)	H2…C11 ^{vi}	2.9821
C5…C17 ^{vi}	3.523 (4)	H2…C12 ^{vi}	3.0176
C6…C5 ^{vii}	3.478 (3)	H2…C13 ^{vi}	3.1945
C6…C6 ^{vii}	3.380 (3)	H2…H5 ^x	3.4413
C6…C18 ^{vi}	3.517 (3)	H2…H6 ^x	2.6426
C10…O1 ⁱ	3.317 (3)	H2…H11 ^{vi}	3.4262
C11…O1 ⁱ	3.394 (3)	H2…H12 ^{vi}	3.4696
C12…O1 ^{viii}	3.283 (3)	H2…H14 ⁱⁱ	3.3315
C16…O1 ^{vi}	3.481 (3)	H3…S1 ^{vii}	3.5901
C17…C5 ⁱⁱⁱ	3.523 (4)	H3…O1 ^{vii}	2.8289

C18...O2 ^{ix}	3.520 (3)	H3...C6 ^{vii}	3.3034
C18...N2 ⁱⁱⁱ	3.402 (3)	H3...C12 ^{iv}	3.3035
C18...C6 ⁱⁱⁱ	3.517 (3)	H3...C17 ^{vi}	3.5623
S1...H1	2.8584	H3...H4 ^{vii}	2.9880
S1...H4	2.8101	H3...H11 ^{xii}	2.9642
S1...H8	3.24 (3)	H3...H12 ^{iv}	2.9179
S1...H9	3.5019	H3...H16 ^{vi}	3.5068
S2...H1	3.5259	H4...N2 ^{iv}	3.4420
S2...H9	2.8735	H4...C5 ^{vii}	3.0320
S2...H13	2.8142	H4...C6 ^{vii}	3.1367
S2...H14	2.8145	H4...C13 ^{iv}	3.2026
S2...H18	2.8406	H4...H3 ^{vii}	2.9880
O1...H1	3.4659	H4...H4 ^{vii}	3.1712
O1...H4	3.3990	H4...H8 ^{iv}	2.8271
O1...H9	3.5316	H4...H13 ^{iv}	2.8027
O2...H4	2.6838	H4...H16 ^{vi}	3.4609
O2...H8	2.58 (3)	H4...H17 ^{vi}	3.5072
N1...H1	2.5786	H5...C2 ^{xi}	3.2128
N1...H8	3.02 (3)	H5...C12 ^{vi}	3.2164
N1...H9	2.4937	H5...C13 ^{vi}	3.2245
N1...H18	2.5815	H5...H1 ^{xi}	2.7234
N2...H13	2.7166	H5...H2 ^{xi}	3.4413
N2...H14	2.4402	H5...H12 ^{vi}	3.1335
C1...H2	3.2430	H5...H13 ^{vi}	3.1487
C1...H3	3.2400	H5...H14 ⁱⁱ	2.7970
C2...H4	3.2554	H6...C2 ^{xi}	3.3804
C3...H3	3.2424	H6...C3 ^{xi}	3.2034
C3...H5	2.5618	H6...C10 ^{xii}	3.3114
C3...H6	3.1440	H6...C11 ^{xii}	3.0923
C3...H7	3.1412	H6...C15 ^{vi}	3.4729
C4...H1	3.2751	H6...C16 ^{vi}	3.5778
C4...H4	3.2666	H6...H1 ^{xi}	3.0033
C5...H2	3.2414	H6...H2 ^{xi}	2.6426
C5...H5	3.3078	H6...H10 ^{xii}	3.1916
C5...H6	2.7645	H6...H11 ^{xii}	2.8057
C5...H7	2.7686	H6...H14 ^{vi}	3.4652
C6...H1	3.2555	H7...N2 ⁱⁱ	3.0502
C7...H2	2.6731	H7...C10 ^{xii}	3.4584
C7...H3	2.6691	H7...H1 ^{xi}	3.1121
C8...H8	2.97 (3)	H7...H8 ⁱⁱ	3.1947
C8...H10	3.2493	H7...H10 ^{xii}	2.9120
C8...H12	3.2507	H7...H11 ^{xii}	3.4890
C8...H18	3.3808	H7...H14 ⁱⁱ	3.0321
C9...H11	3.2533	H7...H17 ^{xi}	3.4409
C9...H13	3.2776	H7...H18 ^{xi}	3.5180
C9...H18	3.3896	H8...S1 ^{iv}	3.50 (3)
C10...H12	3.2577	H8...O2 ^{iv}	2.20 (3)
C11...H9	3.2581	H8...C18 ^{vi}	3.40 (3)

C11···H13	3.2598	H8···H4 ^{iv}	2.8271
C12···H10	3.2581	H8···H7 ^{viii}	3.1947
C13···H8	3.20 (3)	H8···H17 ^{vi}	2.5120
C13···H9	3.2772	H9···C15 ⁱⁱⁱ	3.3276
C13···H11	3.2546	H9···C16 ⁱⁱⁱ	2.9117
C14···H1	3.4944	H9···C17 ⁱⁱⁱ	3.0652
C14···H8	3.43 (3)	H9···C18 ⁱⁱⁱ	3.5850
C14···H13	3.5550	H9···H15 ⁱⁱⁱ	3.1003
C14···H15	3.2456	H9···H16 ⁱⁱⁱ	3.3423
C14···H17	3.2413	H9···H16 ^v	3.3965
C15···H16	3.2554	H10···S1 ⁱ	3.5867
C15···H18	3.2797	H10···O1 ⁱ	2.7476
C16···H17	3.2519	H10···O2 ⁱ	3.5465
C17···H14	3.2617	H10···C7 ^{xiii}	3.5062
C17···H18	3.2544	H10···C17 ⁱⁱⁱ	3.4010
C18···H15	3.2523	H10···C18 ⁱⁱⁱ	3.2861
C19···H1	3.3835	H10···H6 ^{xiii}	3.1916
C19···H9	3.5843	H10···H7 ^{xiii}	2.9120
C19···H14	3.2802	H10···H16 ⁱⁱⁱ	3.4474
C19···H16	3.2478	H10···H16 ^v	3.3903
H1···H2	2.3234	H10···H17 ⁱⁱⁱ	3.2546
H1···H18	3.2430	H11···O1 ⁱ	2.8900
H2···H5	2.3486	H11···C7 ^{xiii}	3.5250
H2···H6	3.3322	H11···C11 ^{xiv}	3.3262
H2···H7	3.3361	H11···C12 ^{xiv}	3.4108
H3···H4	2.3207	H11···C16 ^{xv}	3.5185
H3···H5	3.5985	H11···H2 ⁱⁱⁱ	3.4262
H3···H6	2.7061	H11···H3 ^{xiii}	2.9642
H3···H7	2.7235	H11···H6 ^{xiii}	2.8057
H8···H13	2.8597	H11···H7 ^{xiii}	3.4890
H8···H14	3.2723	H11···H11 ^{xiv}	2.8055
H9···H10	2.3440	H11···H12 ^{xiv}	2.9812
H9···H18	2.9627	H11···H15 ^{xv}	2.9136
H10···H11	2.3297	H11···H15 ^v	3.1221
H11···H12	2.3240	H11···H16 ^v	3.5384
H12···H13	2.3460	H12···S1 ^{viii}	3.4321
H13···H14	3.3530	H12···O1 ^{viii}	2.3445
H14···H15	2.3540	H12···C1 ^{viii}	3.4930
H15···H16	2.3251	H12···C16 ^{xv}	3.5816
H16···H17	2.3167	H12···H2 ⁱⁱⁱ	3.4696
H17···H18	2.3434	H12···H3 ^{iv}	2.9179
S1···H3 ^{vii}	3.5901	H12···H5 ⁱⁱⁱ	3.1335
S1···H8 ^{iv}	3.50 (3)	H12···H11 ^{xiv}	2.9812
S1···H10 ⁱ	3.5867	H12···H15 ^{xv}	2.6972
S1···H12 ⁱⁱ	3.4321	H13···C1 ^{viii}	3.5904
O1···H3 ^{vii}	2.8289	H13···C2 ^{viii}	3.3617
O1···H10 ⁱ	2.7476	H13···C3 ^{viii}	3.2731
O1···H11 ⁱ	2.8900	H13···C4 ^{viii}	3.4401

O1...H12 ⁱⁱ	2.3445	H13...H4 ^{iv}	2.8027
O1...H15 ⁱⁱⁱ	2.6614	H13...H5 ⁱⁱⁱ	3.1487
O2...H8 ^{iv}	2.20 (3)	H14...C7 ^{viii}	3.2799
O2...H10 ⁱ	3.5465	H14...H2 ^{viii}	3.3315
O2...H17 ^{vi}	3.4199	H14...H5 ^{viii}	2.7970
O2...H17 ^v	2.6629	H14...H6 ⁱⁱⁱ	3.4652
N1...H15 ⁱⁱⁱ	3.5536	H14...H7 ^{viii}	3.0321
N2...H4 ^{iv}	3.4420	H14...H18 ^{vi}	2.9405
N2...H7 ^{viii}	3.0502	H15...O1 ^{vi}	2.6614
N2...H17 ^{vi}	2.6480	H15...N1 ^{vi}	3.5536
N2...H18 ^{vi}	3.1056	H15...C10 ^{ix}	3.5418
C1...H12 ⁱⁱ	3.4930	H15...C11 ^{xvi}	3.4196
C1...H13 ⁱⁱ	3.5904	H15...C11 ^{ix}	3.2616
C1...H17 ^{vi}	3.4140	H15...C12 ^{xvi}	3.3133
C2...H5 ^x	3.2128	H15...H9 ^{vi}	3.1003
C2...H6 ^x	3.3804	H15...H11 ^{xvi}	2.9136
C2...H13 ⁱⁱ	3.3617	H15...H11 ^{ix}	3.1221
C3...H6 ^x	3.2034	H15...H12 ^{xvi}	2.6972
C3...H13 ⁱⁱ	3.2731	H16...C6 ⁱⁱⁱ	3.5956
C4...H13 ⁱⁱ	3.4401	H16...C8 ^{ix}	3.0140
C5...H4 ^{vii}	3.0320	H16...C9 ^{ix}	2.9069
C6...H3 ^{vii}	3.3034	H16...C10 ^{ix}	2.9029
C6...H4 ^{vii}	3.1367	H16...C11 ^{ix}	3.0131
C6...H16 ^{vi}	3.5956	H16...C12 ^{ix}	3.1456
C6...H17 ^{vi}	3.4151	H16...C13 ^{ix}	3.1560
C7...H1 ^{xi}	3.1191	H16...H3 ⁱⁱⁱ	3.5068
C7...H2 ^{xi}	3.4100	H16...H4 ⁱⁱⁱ	3.4609
C7...H10 ^{xii}	3.5062	H16...H9 ^{vi}	3.3423
C7...H11 ^{xii}	3.5250	H16...H9 ^{ix}	3.3965
C7...H14 ⁱⁱ	3.2799	H16...H10 ^{vi}	3.4474
C8...H2 ⁱⁱⁱ	3.3221	H16...H10 ^{ix}	3.3903
C8...H16 ^v	3.0140	H16...H11 ^{ix}	3.5384
C9...H2 ⁱⁱⁱ	3.2879	H17...O2 ⁱⁱⁱ	3.4199
C9...H16 ^v	2.9069	H17...O2 ^{ix}	2.6629
C10...H2 ⁱⁱⁱ	3.1156	H17...N2 ⁱⁱⁱ	2.6480
C10...H6 ^{xiii}	3.3114	H17...C1 ⁱⁱⁱ	3.4140
C10...H7 ^{xiii}	3.4584	H17...C6 ⁱⁱⁱ	3.4151
C10...H15 ^v	3.5418	H17...H4 ⁱⁱⁱ	3.5072
C10...H16 ^v	2.9029	H17...H7 ^x	3.4409
C11...H2 ⁱⁱⁱ	2.9821	H17...H8 ⁱⁱⁱ	2.5120
C11...H6 ^{xiii}	3.0923	H17...H10 ^{vi}	3.2546
C11...H11 ^{xiv}	3.3262	H18...N2 ⁱⁱⁱ	3.1056
C11...H15 ^{xv}	3.4196	H18...C15 ⁱⁱⁱ	3.2253
C11...H15 ^v	3.2616	H18...H7 ^x	3.5180
C11...H16 ^v	3.0131	H18...H14 ⁱⁱⁱ	2.9405
O1—S1—O2	116.16 (8)	C17—C18—C19	120.3 (2)
O1—S1—N1	107.18 (8)	C14—C19—C18	118.38 (18)

O1—S1—C1	106.79 (8)	S2—N2—H8	112.7 (15)
O2—S1—N1	113.01 (8)	C1—C2—H1	120.300
O2—S1—C1	107.86 (9)	C3—C2—H1	120.313
N1—S1—C1	105.12 (8)	C2—C3—H2	119.289
N1—S2—N2	126.05 (9)	C4—C3—H2	119.290
N1—S2—C8	103.00 (8)	C4—C5—H3	119.292
N1—S2—C14	102.92 (9)	C6—C5—H3	119.301
N2—S2—C8	112.96 (10)	C1—C6—H4	120.122
N2—S2—C14	104.42 (9)	C5—C6—H4	120.124
C8—S2—C14	105.56 (8)	C4—C7—H5	109.473
S1—N1—S2	121.56 (10)	C4—C7—H6	109.475
S1—C1—C2	120.74 (14)	C4—C7—H7	109.479
S1—C1—C6	118.96 (15)	H5—C7—H6	109.468
C2—C1—C6	120.20 (18)	H5—C7—H7	109.464
C1—C2—C3	119.39 (17)	H6—C7—H7	109.469
C2—C3—C4	121.42 (19)	C8—C9—H9	120.460
C3—C4—C5	117.80 (19)	C10—C9—H9	120.466
C3—C4—C7	121.01 (19)	C9—C10—H10	120.070
C5—C4—C7	121.19 (19)	C11—C10—H10	120.067
C4—C5—C6	121.41 (19)	C10—C11—H11	119.635
C1—C6—C5	119.75 (19)	C12—C11—H11	119.641
S2—C8—C9	120.77 (14)	C11—C12—H12	119.913
S2—C8—C13	117.72 (14)	C13—C12—H12	119.912
C9—C8—C13	121.50 (18)	C8—C13—H13	120.669
C8—C9—C10	119.07 (18)	C12—C13—H13	120.677
C9—C10—C11	119.86 (19)	C14—C15—H14	120.818
C10—C11—C12	120.7 (2)	C16—C15—H14	120.817
C11—C12—C13	120.18 (19)	C15—C16—H15	120.052
C8—C13—C12	118.65 (18)	C17—C16—H15	120.052
S2—C14—C15	118.40 (15)	C16—C17—H16	119.551
S2—C14—C19	119.46 (14)	C18—C17—H16	119.556
C15—C14—C19	122.13 (17)	C17—C18—H17	119.844
C14—C15—C16	118.37 (19)	C19—C18—H17	119.842
C15—C16—C17	119.90 (19)	C14—C19—H18	120.805
C16—C17—C18	120.9 (2)	C18—C19—H18	120.818
O1—S1—N1—S2	-150.69 (11)	S1—C1—C6—C5	-174.74 (12)
O1—S1—C1—C2	-88.83 (13)	C2—C1—C6—C5	1.6 (3)
O1—S1—C1—C6	87.52 (13)	C6—C1—C2—C3	-1.6 (3)
O2—S1—N1—S2	-21.45 (14)	C1—C2—C3—C4	0.1 (3)
O2—S1—C1—C2	145.64 (12)	C2—C3—C4—C5	1.3 (3)
O2—S1—C1—C6	-38.00 (14)	C2—C3—C4—C7	-178.79 (16)
N1—S1—C1—C2	24.81 (14)	C3—C4—C5—C6	-1.3 (3)
N1—S1—C1—C6	-158.83 (12)	C7—C4—C5—C6	178.79 (17)
C1—S1—N1—S2	95.93 (12)	C4—C5—C6—C1	-0.2 (3)
N2—S2—N1—S1	-25.67 (17)	S2—C8—C9—C10	-177.44 (11)
N1—S2—C8—C9	14.99 (14)	S2—C8—C13—C12	177.35 (11)
N1—S2—C8—C13	-163.57 (11)	C9—C8—C13—C12	-1.2 (3)

C8—S2—N1—S1	105.86 (12)	C13—C8—C9—C10	1.1 (3)
N1—S2—C14—C15	148.19 (13)	C8—C9—C10—C11	0.0 (3)
N1—S2—C14—C19	-31.18 (15)	C9—C10—C11—C12	-0.9 (3)
C14—S2—N1—S1	-144.54 (11)	C10—C11—C12—C13	0.7 (3)
N2—S2—C8—C9	153.90 (12)	C11—C12—C13—C8	0.3 (3)
N2—S2—C8—C13	-24.66 (14)	S2—C14—C15—C16	-177.49 (12)
N2—S2—C14—C15	15.17 (16)	S2—C14—C19—C18	177.70 (12)
N2—S2—C14—C19	-164.20 (13)	C15—C14—C19—C18	-1.6 (3)
C8—S2—C14—C15	-104.15 (13)	C19—C14—C15—C16	1.9 (3)
C8—S2—C14—C19	76.49 (15)	C14—C15—C16—C17	-1.2 (3)
C14—S2—C8—C9	-92.61 (13)	C15—C16—C17—C18	0.3 (4)
C14—S2—C8—C13	88.83 (13)	C16—C17—C18—C19	-0.1 (4)
S1—C1—C2—C3	174.68 (11)	C17—C18—C19—C14	0.7 (3)

Symmetry codes: (i) $-x, -y+1, -z+2$; (ii) $x+1, y, z$; (iii) $-x, y+1/2, -z+3/2$; (iv) $-x, -y, -z+2$; (v) $x, -y+1/2, z+1/2$; (vi) $-x, y-1/2, -z+3/2$; (vii) $-x+1, -y, -z+2$; (viii) $x-1, y, z$; (ix) $x, -y+1/2, z-1/2$; (x) $-x+1, y+1/2, -z+3/2$; (xi) $-x+1, y-1/2, -z+3/2$; (xii) $x+1, y-1, z$; (xiii) $x-1, y+1, z$; (xiv) $-x-1, -y+1, -z+2$; (xv) $-x-1, y+1/2, -z+3/2$; (xvi) $-x-1, y-1/2, -z+3/2$.

Hydrogen-bond geometry ($\text{\AA}, ^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N2—H8 \cdots O2 ^{iv}	0.89 (3)	2.20 (3)	3.018 (3)	152 (2)
C12—H12 \cdots O1 ^{viii}	0.95	2.34	3.283 (2)	169

Symmetry codes: (iv) $-x, -y, -z+2$; (viii) $x-1, y, z$.