03.5-19 CRYSTAL AND MOLECULAR STRUCTURE
OF 2-IMINO-1,3 DIHYDRO - 4,6
DIMETHYL PYRIMIDINUM CHLORIDE By Amit
Halder, A.K. Basak, S.K. Mazumdar of Crystallography Division, Saha Institute of
Nuclear Physics. and S. Chaudhuri, Bose
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The title compound crystallizes in the orthorhombic space group Pna2, with a=14.269(3) b= 8.022(2), c= 6.675(2)Å. Solved by heavy atom method and refined to a final R of.048. The cation exist as imino tautomer. The exocyclic C-N bond length clearly establishes its double bond character and endocyclic bond angles at N(1) and N(3) increases because of protonation. There is an indication of strong conjugation between the imino nitrogen and the pyrimidine ring. The molecules form chloride bridged hydrogen bonded chains extended infinitely along the glide direction. The chloride ions are stacked between the pyrimidine bases. Mol.Formula: $C_6H_{10}N_3Cl$. This work forms a part of the doctoral dissertation of Amit Halder.

03.5-20 THE STRUCTURE OF 168-PHENOXYBUTYLO-4-ANDROSTENE-3,17-DIONE. By W.L. Duax⁺,

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The purpose of the structure determination is

The purpose of the structure determination is to verify positions of the substituents at C(16) and C(17). The compound crystallized in the monoclinic space group P2₁ with unit cell parameters: a=19.802(4), b=7.1389(7), c=8.849(1) Å, $\beta=102.61(1)^{\circ}$, V=1220.9 Å³, Z=2, $D_{x}=1.181$ Mg m⁻³. The intensities of

2703 independent reflections were collected on a CAD4 diffractometer using CuK \propto radiation. The structure was solved by direct methods (MULTAN) using the weighted technique. The positions of all H atoms were found from difference Fouries syntheses. The structure refinement is in progress. Current R = 0.096. Dr. Solo, who synthesized the compound has assumed that the compound is a 4-cholesten derivative, but from X-ray analysis we have found, that it is 4-androsten one.

O3.5-21 ABSOLUTE CONFIGURATION AND CRYSTAL STRUCTURE OF THE NEW & -SUBSTITUTED AMINOACID DERIVATIVES. By M. Bukowska-Strzy-żewska and W. Wieczorek, Institute of General Chemistry, Technical University, 36 Żwirki, 90-924 Łódź, Poland.

Dextro-rotatory \sim -hydroxymethyl-N-benzoilo-aspartic acid lactone (I) crystallizes in the orthorhombic system: P2 $_1$ 2 $_1$ 2 $_1$, a=10.337 (1) , b=18.115 (1) , c=6.063 (1) , Z=4.Iaevorotatory S-benzylo- \sim -hydroxymethyl-N-benzoilocysteine (II) crystallizes in the monoclinic system: P2 $_1$, a=8.972 (1) , b=10.004 (1) , c=10.302 (1), β =106.37 (11) , Z=2.

03.5-22 CRYSTAL AND MOLECULAR STRUCTURES OF SOME ACYLANILIDE FUNGICIDES. By <u>J. C. J. Bart</u>, R. Scordamaglia and M. Calcaterra, Istituto G. Donegani S.p.A. (Montedison), Novara, Italy.

Some microbicidal anilide derivatives which control Phytophthora infestans on potato, Plasmopara viticola on grape, Pseudoperonospora cubensis on squash, Peronospora tabacina and Phytophthora nicotianae on tobacco, Pseudoperonospora humili on hops, and Sclerospora on corn were analyzed by x-ray diffraction in order to gain information about the molecular conformation. The following active compounds were considered:

DL-methyl-N-(2,6-dimethylphenyl)-N-(2-methoxy-acetyl)alaninate or metalaxyl (P_{2} /c, a=7.851 (3)Å, b=13.119(10)Å, c=15.107(7)Å, β =101.71(6), Z=4, R=0.079 for 1203 F(hkl));

DL-methyl-N-(2,6-dimethylphenyl)-N-(2-furoyl) alaninate or furalaxyl (Cc, a=15.997(12)Å, b-7.929(7)Å, c=14.064(11)Å, β =118.70(5)°, Z=4, R=0.068 for 852 F(hkl));

N-(2-methoxyacetyl)-N-(2,6-xylyl)-3-amino-1,3-oxazolidin-2-one (P2/n, a=9.363(1)Å, b=12.715(1)Å, c=12.360(1)Å, β =102.64(10)°, Z=4, R= 0.062 for 1175 F(hkl));

N-(2-methoxyacetyl)-N-(2,6-xylyl)- α -amino- γ -butyrolactone (P2₁/n, a=9.600(3)A, b=11.964(6) Å, c=12.802(4)Å, β = 101.25(3)°, Z=4, R=0.055 for 1686 F(hkl)).

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The molecular conformations of metalaxyl and furalaxyl are similar with the dimethylphenyl ring being almost perpendicular to the amidic plane of the molecule, as is also the case for the two isostructural compounds with the oxazolidinone and butyrolactone rings. The amidic nitrogen atom is ${\rm sp}^2$ hybridized. As in most N-substituted anilides, metalaxyl and furalaxyl exist preferentially in the exo isomeric form with the phenyl ring $\underline{\rm trans}$ with respect to the carbonyl group. The oxazolidinone ring is in a half chair form and the butyrolactone ring in a typical envelope conformation.

03.5-23 LOW TEMPERATURE STUDIES OF CHOLINE ESTERS

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The crystal structures of many salts of choline esters are known.

Y-C 0-CH₂-CH₂-H(CH₃)₃,X

Strong thermal motion or disorders are often observed in these salts, and the accuracy of the structures is generally rather low. A survey of the known structures has disclosed a strong correlation between the magnitude of some bond angles and the actual conformation. Many types of theoretical studies of molecular flexibility still make use of crystallographically observed geometry, and it was found worth while to collect low temperature data for a selection of choline ester salts, in order to get more precise knowledge about the correct values of bond lengths and angles. Results will be presented for methoxycarbonylcholine iodide, carbamoylcholine chloride and bromide, acetylcholine chloride, and succinylcholine iodide, representing the three known conformers transtrans, trans-gauche and gauche-gauche.