

## 09. STRUCTURES OF ORGANIC, ORGANOMETALLIC AND COORDINATION COMPOUNDS

The compound crystallizes in the space group  $P2_1/c$  with  $a = 9.922(2)$ ,  $b = 8.462(2)$ ,  $c = 16.068(2)$  Å,  $\beta = 100.29(1)^\circ$ ,  $D_m = 1.208$  gm/cm<sup>3</sup>,  $D_c = 1.197$  gm/cm<sup>3</sup>,  $\mu(\text{CuK}\alpha) = 6.175$  cm<sup>-1</sup> and  $Z = 4$ . The structure was solved by MULTAN with 1705 unique reflections collected on a diffractometer. 170 reflections with  $E_{\text{max}} = 4.549$  and  $E_{\text{min}} = 1.682$  were used to solve the phase problem and all the non-hydrogen atoms were shown by the search map of MULTAN. The positional and thermal parameters of non-hydrogen atoms were initially refined by full-matrix least square method. A difference map at  $R = 0.12$  revealed some of the hydrogen atoms and the rest were generated. They were assigned isotropic temperature factors of non-hydrogen atoms to which they were attached. The positional parameters of hydrogen and non-hydrogen atoms and their isotropic and anisotropic thermal parameters respectively were refined by block diagonal least square method and the final  $R$ -value was 0.0432. The bond lengths and angles were well within the range of chemically accepted values and the non-bonded contacts were always greater than the sum of the Vander Waals radii.

The derived three-dimensional molecular architecture confirms the stereospecificity of the reduction process and the formation of a CIS Ketone. The ring fusion methyl group is axial to the B-ring and equatorial to the Ketone containing A-ring of the molecule, whereas the ring fusion hydrogen atom is equatorial to the B-ring and axial to the A-ring of the molecule. The estimated distance between the hydrogen atom attached to C<sub>4</sub> and the hydrogen atom attached to C<sub>5</sub> of the aromatic benzene ring reveals that the alternative conformer with the methyl group equatorial to B-ring and the hydrogen axial to it is not favoured in the crystalline state.

09.2-7 CRYSTAL AND MOLECULAR STRUCTURE OF 2-METHYL-1,2,3,9,10,10a-HEXAHYDRO-2 $\alpha$ ,10 $\alpha$ ,11-OXOETHANOPHENANTHRENE. By A.K. Pal\*, S.C. Kundadas and B.S. Basak, X-ray Laboratory, Presidency College, Calcutta-73, India.

The title compound (C<sub>17</sub>H<sub>18</sub>O) is one of a number of important organic compounds studied in our laboratory. The compound crystallized in space group  $P2_1/c$ ,  $a = 10.109(4)$ ,  $b = 11.725(5)$ ,  $c = 11.159(1)$  Å,  $\beta = 94.24(5)^\circ$ ,  $V = 1319$  Å<sup>3</sup>,  $D_c = 1.20$  g cm<sup>-3</sup> for  $Z = 4$ ,  $D_m = 1.18$  g cm<sup>-3</sup>,  $F(000) = 512$ ,  $\mu(\text{MoK}\alpha) = 0.78$  cm<sup>-1</sup>,  $\lambda(\text{MoK}\alpha) = 0.7107$  Å.

Three-dimensional diffraction data were recorded on a four-circle diffractometer by  $\omega$ -2 $\theta$  scans. A total of 2314 reflections were collected, including 1058 with  $|F_o| \leq 4\sigma(F_o)$ . The structure was solved by direct methods using the 1978 version of MULTAN. The structure was refined isotropically and also anisotropically using the program MAMIE for full-matrix least-squares and the program BLOK for block-diagonal least-squares calculations, respectively. Prior to anisotropic refinement, the positions of the hydrogen atoms were determined by difference synthesis. The refinement ended with  $R = 0.048$ .

\*Present address: Department of Physics, Jogesh Chandra Chaudhuri College, 30 Prince Anwar Shah Road, Calcutta-700033, India.

09.2-8 CRYSTAL AND MOLECULAR STRUCTURE OF cis-METHYL-3,4,4a,9,10,10a-HEXAHYDROPHENANTHRENE-2(1H)-ONE-4a-ACETATE.

By A.K. Pal\*, S.C. Kundadas and B.S. Basak, X-ray Laboratory, Presidency College, Calcutta-73, India.

The crystal structure analysis of the title compound is part of a structural study of biologically important organic compounds. The compound (C<sub>17</sub>H<sub>20</sub>O<sub>3</sub>) crystallised in space group  $P\bar{1}$ ,  $a = 6.413(1)$ ,  $b = 10.501(3)$ ,  $c = 10.862(4)$  Å,  $\alpha = 82.71(2)$ ,  $\beta = 87.73(2)$ ,  $\gamma = 85.29(2)^\circ$ ,  $V = 724$  Å<sup>3</sup>,  $D_c = 1.24$  g cm<sup>-3</sup> for  $Z = 2$ ,  $D_m = 1.23$  g cm<sup>-3</sup>,  $F(000) = 584$ ,  $\mu(\text{MoK}\alpha) = 0.91$  cm<sup>-1</sup>,  $\lambda(\text{MoK}\alpha) = 0.7107$  Å. Three-dimensional counter data were recorded using  $\omega$ -2 $\theta$  scans. A total of 1880 reflections were measured, including 944 with  $|F_o| \leq 4\sigma(F_o)$ . The structure was solved by direct method using the 1978 version of MULTAN. It was refined first isotropically and then anisotropically using the programs MAMIE and BLOK for full-matrix least-squares and block-diagonal least-squares calculations, respectively. The hydrogen atoms were located prior to the anisotropic refinement. The refinement converged with  $R = 0.055$ .

\* Present address: Jogesh Chandra Chaudhuri College, 30 Prince Anwar Shah Road, Calcutta-700033, India.

09.2-9 CRYSTAL AND MOLECULAR STRUCTURE OF 6-(3-HYDROXY-3-METHYL-1-BUTINYL)-7-METHOXY-2H-1-BENZOPYRAN-2-ONE (SUBERINOL).

By R.R. Bandyopadhyay\* and B.S. Basak, X-ray Laboratory, Presidency College, Calcutta-700073, India.

The crystal structure of suberinol (C<sub>15</sub>H<sub>16</sub>O<sub>4</sub>) has been determined as part of a study of the structures of biologically important organic compounds. Suberinol crystallized in space group  $P2_1/c$ ,  $a = 7.94$ ,  $b = 15.64$ ,  $c = 10.99$  Å,  $\beta = 103.97^\circ$ ,  $V = 1324$  Å<sup>3</sup>,  $D_c = 1.30$  g cm<sup>-3</sup> for  $Z = 4$ ,  $D_m = 1.29$  g cm<sup>-3</sup>,  $F(000) = 552$ ,  $\mu(\text{CuK}\alpha) = 7.87$  cm<sup>-1</sup>,  $\lambda(\text{CuK}\alpha) = 1.5418$  Å. Three-dimensional X-ray diffraction data (OKl to 6kl, h0l, hl $\bar{l}$ ) were recorded photographically using a Unicam Weissenberg goniometer. The structure was solved by direct methods. Isotropic followed by anisotropic refinement calculations were made by means of the full-matrix least-squares program MAMIE and the block-diagonal least-squares program BLOK, respectively. The final  $R = 0.09$ . Bond lengths and bond angles agree with the values in similar compounds.

\*Present address: Department of Physics, M.U.C. Women's College, Burdwan, West Bengal, India.