9. STRUCTURES OF ORGANIC, ORGANOMETALLIC AND COORDINATION COMPOUNDS

On the basis of presented structure data (see the projection below), it is known that chlorination of $\alpha$-phosphorylsulfoxides occurs with prevailing retention of the configuration at the carbon atom.

$I$


II
09. 2-47 CONFORMATIONAL ANALYSES OF THO 8B-t-BUTYL-CIS-DECAHYDROQUTNOLINE SALTS:
TWO DIFFERENT RING CONFORMATIONS
By Kay D. Onan and Martha Estes, Department of Chemistry, Northeastern University, Boston, MA 02115 and Friedrich W. Vierhapper, Institut für Organische Chemie, Universität Wien, A-1090 Wien, Austria

As a part of our investigation of the conformations of saturated six-membered rings bearing axial t-butyl groups, we have prepared the picrate (I) and chloride (II) salts of $8 \beta$-t-butyl-cis-decahydroquinoline and have determined their crystal structures. Both I and II crystallize in monoclinic space groups, $I_{0}$ in $P 2 / / \mathrm{c}$ with $\mathrm{a}=$ $13.335(2), b=8.584(1), c=22.419(3) \AA, \quad B=125.29(1)^{\circ}, Z=4$. Salt II crystallizes in space group C2 with $a=17.035(I)$, $j=7.2880(5), c=25.048(2) A, \beta=109.90(1) ; Z=8$. The structures were solved by direct methods and refined by fullmatrix, anisotropic ( $\mathrm{Cl}, \mathrm{O}, \mathrm{N}, \mathrm{C}$ ) least-squares to a present $R$ of $9.3 \%$ over 2284 reflections for $I$ and to $R=3.9 \%$ over 2296 reflections for II. The rather high R value for I is due to disorder in the picrate anion. Surprisingly, the two molecules which compose the asymmetric unit of II are a racemic pair which displays essentially no difference in magnitudes of torsion angles between the molecules.

Chair conformations have been adopted by the heterocyclic rings in both salts. In II the saturated cyclohexane ring, which bears an axial t-butyl group, adopts a chair conformation flattened near this group. In I, however, the conformation adopted is a distorted twist-boat form, with the t-butyl group in an isoclinal orientation. This result corroborates earlier evidence given for a small energy barrier between a flattened chair form and a distorted twist-boat form in such a strained, flexible system (Onan and Vierhapper, Tet. Lett. (1984) 00).
09. 2-48 CRYSTAL AND MOLECULAR STRUCTURE OF 3-(0-HYDROXYPHENYL)-1-PHENYLTRIAZENE 1-OXIDE, $\mathrm{C}_{12} \mathrm{H}_{11} \mathrm{~N}_{3} \mathrm{O}_{2}$.
By S.K. Talapatra, S.B. Sarkar, S.C. Saha, P.K. De and Chitra Samanta , Department of Physics, Jadavpur University, Calcutta-700032, India.
In order to confirm the $N$-oxide form as revealed from IR studies and to establish the nature of hydrogen bonds, if any, it was considered necessary to investigate the structure of triazenes having orthosubstituents. From Weissenberg photographs and single crystal diffractometry at room temperature $\left(300^{\circ} \mathrm{K}\right)$, the deep brown crystals of the title compound were shown to be
monoclinic, $\mathrm{P}_{1} / \mathrm{n}, \mathrm{M}_{\mathrm{r}}=229.0$, $\mathrm{a}=6.569(2)$,
$b=11.870(4), \ldots c=13.889(4) \AA_{r} \quad \beta=94.95(2)^{\circ}$,
$\mathrm{V}=1079(1) \mathrm{A}^{3}, \mathrm{z}=4, \mathrm{D}_{\mathrm{x}}=1.38 \mathrm{~g} \mathrm{~cm}{ }^{-3}$ (aq. KI ),
$D_{C}=1.41 \mathrm{~g} \mathrm{~cm}^{-3}, \lambda(\mathrm{MOKa})=0.71073 \AA, \mu=1.06 \mathrm{~mm}^{-1}$.
1997 independent reflections with $2 \theta<50^{\circ}$ were recorded. The structure was solved by direct methods using the MULAN programs. The H-atoms were located by Fourier methods and also from the known geometry around c-atoms. The non-hydrogen atoms were refined anisotropically and H-atoms isotropically. The final discrepancy indices for 1449 observed reflections were $R=0.040$ and $R_{i j}=0.062$.

The molecules are nearly planar and have intramolecular $\mathrm{N}-\mathrm{H} . . . \mathrm{O}$ hydrogen bonds. The crystal structure is stabilized by a network of intermolecular 0-H... 0 hydrogen bonds. The structure establishes the N -oxide form of triazene as in the case of 3-(o-carboxyphenyl)-1-phenyltriazene 1-oxide previously reported (Acta cryst., 1983, c39, 1075).
09. 2-49 STRUCTURB OF DI-SYNRPHRI NB RTHER DIHYDROBROMIDB ( $\alpha, \alpha^{\prime}-$ N-METHYLAMI HO METHYL-4, $4^{\prime}$ DIHYDROXY DIBENZYI BTHER DIHYDROBROMIDE) By B.P.Mukhopedhyay and J.K.Dattagupta, C. e M.B.Division, Saha Institute of Nuclear Physics, Sector-I, Block-'AF', Bidhen Negar, Calcutta-700 064, India.
Synephrine is a well known sympathomimetic amine. Structure of the phosphate complex salt of it has already been studied by us (Acta Cryst. (1982)B38, 2830-2834) where two molecules in the asymmetric unit exhibited different conformations. Structure analysis of the hydrobromide salt of this compound was undertaken by us. Synephrine has been treated: with $30 \%$ aqueous HBr in $1.5: 2$ mole ratio in presence of ethyl alcohol at $27^{\circ} \mathrm{C}$. The single crystals of the compound thus formed belong to monoclinic space group C2/c with unit cell
dimensions $a=7.645, b=19.777, c=13.704 \AA$, $\beta=98 \cdot 32^{\circ}$. Intensity data has been collected on a CAD-4 diffractometer and the structure has been solved by heavy atom method. The structural parameters have been refined by full-matrix least squares method upto an $R$ value of $8.5 \%$ with isotropic temperature factors and without hydrogen atoms. The chemical formula of this compound is found to be ( $\left.\mathrm{C}_{9} \mathrm{ONH}_{12} \cdot \mathrm{HBr}\right)_{2} \mathrm{O}$. Mass spectra of the
compound has also been taken which supporta this chemical structure. Details will be presented.

