STRUCTURAL RESULTS FROM METHODS OTHER THAN DIFFRACTION 21.

STUDYING REAL STRUCTURE OF CRYSTALS 21.1-1 BY DIELECTRIC SPECTROSCOPY. By A. P. Andreev, B. N. Kolodiev, S. V. Kolo-dieva, M. I. Samoilovitch, A. A. Fotchenkov and <u>A. A. Shaposhnikov</u>, Research Institute for Synthesis of Mineral Raw Materials, Alexandrov, USSR.

Data on real structure of crystals obtained

Data on real structure of crystals obtained by universally accepted techniques (EPR, op-tical spectroscopy, combinational dissipati-on etc) may be far more numerous and complete if dielectric spectroscopy is used as a further-aid. It is especially important when point defects of the studied crystals are non paramagnetic, optically inactive though electrically active in non-excited state.

Quartz is a good example of these crystals. Electrically active point defects in quartz are known to be alkali metal ions that dominate there as the type.

We have compared experimental temperature/ frequency dependencies \mathcal{E}'' at temperatures of 300-800 K with Debye theoretical curve and presented Cole-Cole and Scafe diagrams. As a result we state a presence of a broad energy spectrum of impurity alkali ions. Energy dismetry of normalized $\mathcal{E}''(\omega \mathcal{C})$ function proving it. The distribution parameter of relaxation time (α) increases with crystallization temperature and with lower rates of growth.

We have postulated presence of H20 molecules

in structure voids of cancrinite crystals zation in a temperature range of 350-500 K. We have calculated the \propto value to change from 0.05 to 0.40 at temperatures of 350-500 K. This change may result from multiple structure positions of H₂O molecules and

their reorientations at various temperatures (Kolodieva et al., Sintez miner. i exper. is-sled., Moskva (1981) 41).

Dielectric spectroscopy approach to non-rela-xation maxima $\mathcal{E}''(\mathbb{T})$ should be very careful. In interpreting of the experimental results special attention should be paid to reversi-bility of the observed change of the dielec-tric characteristics and to their correlation to other physical properties.

For example, we can state that the non-relaxation maximum \mathcal{E}'' in the region of 650-800 K in cancrinite is not reproducible at repeated heating. This fact may result from irrevers-ible changes of cancrinite structure further corroborated by substantial weakening of its piezo effect after heating to 800 K.

Similar maxima of $\mathcal{E}''(T)$ obtained during heat-ing of fluorphlogopite (up to 900 K) are re-producible though after certain intervals of time. This is a case of reversible processes with great relaxation time of the crystal to the initial state. General phenomenology may describe this process as a pre-phenomenology may describe this process as a recombination of defects that determine dielectric losses with acceptor centers thermally activated in this temperature range to form metastable pairs (or complexes) that dissociate later on (Ko-lodiev et al., Vses. nautchn. konfer. "Physi-ka dielektrikov". Sektsyia "Protsessy elekt-roperenosa". Thes. dokladov, Baku (1982) 63).

21.1-2 NMR AND EPR SPECTROSCOPY OF ALKALI SELENITES. By I.S. Vinogradova, L.V. Kirensky Institute of Physics, Academy of Sciences, Siberian Branch, Krasnoyarsk, USSR.

The results of the study of single crystals of MeHSO_3 (Me = Li, Na, K, Rb, Cs) by $^{1}\mathrm{H},~^{2}\mathrm{D},~^{7}\mathrm{Li},~^{23}\mathrm{Na},$

⁸⁷ Rb, ¹³³ Cs and ⁷⁷ Se nuclear magnetic resonance spectroscopy are reported. In addition, the electron paramagnetic resonance spectra of SeO_2^- radicals were investigated. Special attention was paid to the study of hydrogen bond networks and proton dynamics. The structural data for some crystals were first obtained by the spectroscopic methods. The present study has shown that the substitution of alkali ion leads to a change of crystal symmetry. The rubidium and potassium salts are triclinic and isostructural, the sodium salt is monoclinic, the lithium and caesium salts are orthorhombic. The $7_{\rm Li},~87_{\rm Rb}$ and $133_{\rm CS}$ NMR spectra were measured over a wide temperature range with the aim of searching for phase transitions. No phase transitions have been discovered in any of the MeHSO3 crystals investigated.

The structural parameters of hydrogen bonds were determined from deuteron magnetic resonance spectra. The lengths of hydrogen bonds are in the range 2.5-2.6Å. The deuterons are ordered in single-minimum potential wells. Only in NaDSeO3 crystals did we find disorder of deuterons in a two-minimum potential well without a transition into an ordered state up to 77K.

Unusual proton magnetic resonance spectra have been found in $\rm KHSeO_3$ single crystals. The spectra are doublets in many orientations of crystals relative to Ho. A recent ND structure investigation of KHSeO3 (B.A. Sarin, N.N. Bydanov, I.S. Vinogradova, E.E. Rider, S.P. Solovjev, <u>Sov. Phys. Crystallogr.</u> (1984) <u>29</u>) permits the observed spectra to be connected with a distinctive feature of the proton system, namely chains with p-p distances 2.6Å and 2.8Å. In the other crystals of the MeHSO3 family, as well as in MeH3(SeO3)2 crystals, the PMR spectra are single lines.

KHSeO3 and RbHSeO3 have shown unusual EPR specta of SeO₂ radicals. Besides components of spectra typical for ${\rm SeO}_2^-$ radicals we found additional lines. The nature of these is not understood. Comparisons of spectral and structural characteristics and physical properties of MeH3 (SeO3) 2 and MeHSeO3 crystals have also been made.

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