PS02.05.14 CRYSTAL STRUCTURE DETERMINATION OF TWO POLYMORPHIC PHASES OF LANTHANUM NITRATE TETRAHYDRATE FROM X-RAY POWDER DIFFRACTION. M. Louër, A.-E. Gobichon, J. P. Auffrédic, D. Louër, Laboratoire de Cristallochimie, CSIM (URA CNRS 1495), Université de Rennes, 35042 Rennes cedex, France

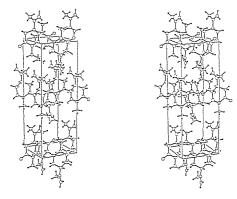
Two polymorphic varieties of La(NO₃)₃.4H₂O have been identified from temperature dependent X-ray diffraction experiments of the hexahydrate carried out in different water pressures. The two phases α and β are obtained for P(H₂O) greater and lower than 4 Torr, respectively. Powder diffraction data for the two phases were collected with a high resolution powder diffractometer using monochromatic radiation (λ = 1.5406 Å). The indexing of the powder patterns was performed by the successive dichotomy method (DICVOL91), the structure solutions were obtained from Patterson maps and subsequent Fourier analyses and, finally, the models were refined by the Rietveld method (FULLPROF).

cr-La(NO₃)₃.4H₂O: $P2_1/m$. a=6.778(1), b=11.367(2), c=6.585(1) Å, $\beta=90.639(5)^\circ$, V=507.3 ų, Z=4, $M_{20}=28$, $F_{30}=59(0.011.44)$. The Rietveld refinement (49 varying parameters) converged to $R_F=0.05$ and $R_{wp}=0.08$. The structure consists of infinite chains, formed by 11-coordinated La atoms, running along [001]. The La polyhedra, formed by three bidentate, one monodentate nitrate groups and four water molecules, are connected through a bridging nitrate group. The cohesion of the structure is ensured by a network of hydrogen bonds. The 11-coordination of lanthanum has also been reported for the structure of the hexahydrate precursor.

β-La(NO₃)₃.4H₂O: Pbca, a=13.531(1), b=11.834(1), c=12.973(1) Å, V=2027.3 Å³, Z=8, M₂₀=60, F₃₀=85(0.0048,74). The Rietveld refinement (71 varying parameters) converged to R_F=0.06 and R_{wp}=0.12. The structure consists of infinite chains, formed by 10-coordinated La atoms, running along [001]. The polyhedra are formed by two bidentate, two bridging monodentate nitrate groups and four water molecules.

PS02.05.15 AB INITIO CRYSTAL STRUCTURE DETERMINATION FROM LOW TEMPERATURE X-RAY POWDER DATA: THE CHIRAL COMPOUND L-CARVONE. C. Miravitilles*, J. Rius*, J. Sañé*, M. A. Cuevast † & T. Calvet† (*) Institut de Ciencia de Materials de Barcelona, camp. UAB, O8193Bellaterra, Spain (†) Fac. de Geologia, Univ. de Barcelona, c/ Marti Franques s/n, Barcelona, Spain

The *ab initio* structure determination of the chiral compound l-carvone ($C_{10}H_{14}O$) at 218 K is presented. The pattern was measured with $CuK\alpha_1$ radiation, a position sensitive detector and the sample, liquid at room temperature, included in a glass capillary. The crystal data are: orthorhombic $P2_12_12_1$ with a=6.8576(3), b=6.8831(5), c=19.988(2) Å, Z=4. The structure was solved by Patterson-search methods using the 44 available integrated intensities. A rigid-body Rietveld refinement with preferred orientation correction and allowing the variation of one torsion angle converged to Rp= 0.03 and Rwp=0.04.



Perspective view of the l-carvone crystal structure at 218 K.

PS02.05.16 CRYSTAL STRUCTURE OF A NEW COOR-DINATION POLYMER, {[Ni₂Cl₂(L)(CH₃OH)₂(H₂O)₂]Cl₂}_n, FROM X-RAY POWDER DATA. A. Neels, B. Mathez Neels and H. Stoeckli-Evans, Département de Chimie-Physique II, Université de Neuchâtel, Switzerland. A. Clearfield and D. M. Poojary, Department of Chemistry, Texas A & M University, USA.

Coordination metal complexes with polymeric structures are of special interest in the development of new materials. Molecular ferromagnets and non-linear optical materials are the most important applications of such coordination polymers.

2,5-Bis(2-pyridyl)pyrazine (L) has been used as molecular bridge between metal centers. Some binuclear complexes and a two-dimensional polymer with 3d transition metals are characterized by their single-crystal X-ray structure [1].

[[Ni₂Cl₂(L)(CH₃OH)₂(H₂O)₂]Cl₂]_n (1), a new one-dimensional coordination polymer, was obtained in micro-crystalline form. The structure was determined *ab initio* from X-ray powder diffraction data. The compound belong to the triclinic space group P1 with a = 8.7014(4)Å, b = 10.1465(5)Å, c = 8.0303(3)Å, α = 116.095(2)°, β = 112.713(3)°, γ = 64.056(3)°. The final agreement factors are R_{wp} = 0.124, R_p = 0.095, and R_F = 0.045

The octahedral coordination of the nickel atom is achieved by two nitrogens of the ligand 2,5-bis(2-pyridyl)pyrazine, two chlorines and two oxygens of the coordinated solvent molecules. The bridging nature of chloride ions and bis-bidentate ligands (L) leads to a one-dimensional polymer.

[1] Neels, A., Stoeckli-Evans, H.; Escuer, A.; Vicente, R.; Inorg. Chem. 34(1995) 1964.

PS02.05.17 AlMethylPO₄.H₂O:THE *AB INITIO* STRUCTURE SOLUTION OF A NEW LAYERED COMPOUND. Sawers, L-J. M., Carter, V., Armstrong, A. R., Bruce, P. G., Wright, P. A., Dept. Of Chemistry, University of St. Andrews, St. Andrews, Fife, Scotland

Micro-porous and layered alumino-methyl phosphonates offer wide-ranging possibilities in separation and catalysis. The title compound is a novel structure made up of alumino phosphonate layers containing bound water, the layers being separated by methyl groups. The synthesis and structure solution of AlmethylPO₄.H₂O is described. The structure was solved ab initio from laboratory X-ray data using a combination of Direct Methods, molecular modeling and Rietveld refinement. Indexing gave a monoclinic unit cell of: a = 9.45 Å, b = 7.06 Å, c = 7.85 Å and $\beta = 106.6^{\circ}$ in the space group P2₁/a. A default run of the Direct Methods package SIRPOW gave a starting structure consisting of 1P, 1Al and 3O's. The remaining atoms were found through the use of difference fourier maps and the Biosym modeling package, Insight. Which, combined with a prior knowledge of bond angles and distances for both octahedral and tetrahedral aluminium sites was used in the latter stages of the Rietveld refinement to determine starting positions for 2O's and 5H's. The result was a final structure solution with an R_{Wp} of 7.85%, an R_p of 5.87% and a chi2 of 5.39%.