

PS07.02.14 CRYSTAL STRUCTURE AND EPR DATA OF A DINUCLEAR Cu(II) COMPLEX, [Cu(N₃)(NCO)diEten]₂. J. Zukerman-Schpector,¹ O.E. Piro,² E.E. Castellano,³ C.A. De Simone,⁴ D.M. Martino⁵ and C.A. Steren⁵ ¹Dept. Química, Univ. Federal de Sao Carlos, C.P. 676, 13565-905 Sao Carlos(SP), Brazil, ²Dept. Física, Ftad. C. Exactas, Univ. Nacional de La Plata, C.C. 67, 1900 La Plata, Argentina ³Inst. Física de Sao Carlos, Univ. de Sao Paulo, C.P. 369, 13560 Sao Carlos, Brazil ⁴Dept. Química, Univ. Federal de Alagoas, 57000 Maceio (AL), Brazil, ⁵INTEC (CONICET-UNL), Guemes 3450, 3000 Santa Fe Argentina

The [Cu(N₃)(NCO)(C₁₆N₂)₂]₂ molecules are arranged as centrosymmetric dimers in which two azide ligands bridge neighboring copper ions in an asymmetric head-on fashion. The Cu(II) ion is coordinated to five nitrogen atoms which form a distorted tetragonal pyramid. At the pyramid base are the two N atoms of a diEten molecule [d(Cu-N)=2.00(1), 2.12(1) Å], an azide end atom [d(Cu-N)=1.99(1) Å] and a NCO group [d(Cu-N)=1.95(1) Å]. At the pyramid apex is the other, inversion related to the first one, azide N atom in the dimer [d(Cu-N)=2.38(1) Å]. This Cu-N contact links the monomers within a dimer providing a electronic path to transmit the superexchange coupling between the Cu(II) unpaired electrons. Neighboring dimers are coupled by weak N-H...O contacts. Single crystal EPR data at X-band show that the pair of resonances expected for neighboring, magnetically nonequivalents dimers, collapse into a single line, a signature of interdimer superexchange coupling. The observed crystal gyromagnetic tensor is used to disclose the electronic and magnetic structure around Cu(II) ions. The compound crystallizes in the space P21/n, with *a*=8.336(1), *b*=17.405(3), *c*=8.376(1) Å, β=109.73(2)° and *Z*=4.

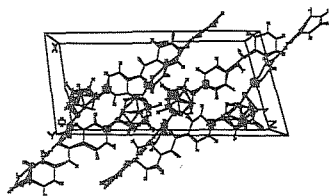
Metal Clusters

PS07.03.01 CRYSTAL STRUCTURE OF 6,9-BIS(4-STILBAZOLE)-NIDO-DECABORANE(12), 6,9-[4-(PhC₂H₂-Py)]₂B₁₀H₁₂. V.I. Alekseev, T.M. Polyanskaya, S.T. Dunaev, E.A. Il'inchik, Institute of Inorganic Chemistry Sib. Branch RAS, Novosibirsk, Russia

A single-crystal X-ray study has been performed for the title decaborane(14) derivative. The sample used was prepared by the well known type of displacement reaction of (SMe₂)₂B₁₀H₁₂ with 4-stilbazole and was recrystallized from hot dimethylformamide.

The most interesting feature of the compound is that four crystallographically independent asymmetrical molecules exist in the unit cell due to different arrangements of both the two cycle aromatic pyridine and phenol fragments and those in respect to the three-dimensional aromatic B₁₀H₁₂ nido-cluster.

The nitrogen atoms of pyridine are linked to the 6,9-boron atoms on the open face of the B₁₀ basket at the mean distance of 1.60 Å. The following mean bond lengths are B-B 1.80, C-N 1.33, C-C 1.40 and 1.38 Å for Py and Ph, respectively.

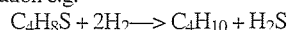


Crystal data: dark red color, triclinic P1, *a*=11.285(1), *b*=13.061(2), *c*=21.176(2) Å, α=85.37(1), β=102.26(1), γ=115.58(1)°, *V*=2750.7(7) Å³, *Z*=4, *D*_c=1.165 g·cm⁻³, CAD-4, λ MoKα.

As a solid, this compound exhibits a photoluminescence (λ_f=653 nm) with the quantum yield approximating that of rhodamine 6G.

PS07.03.02 INCOMPLETE CUBANES M₃X₄⁺⁺. Alan Hazell^a, Michael Brorson^b, Claus Jacobsen^b, Birgitte Jespersen^b and Iver Schmidt^b ^aChemistry Dept. Aarhus University, DK-8000 Århus C, Denmark. ^bHaldor Topsøe A/S, Nymøllevej 55, DK2800 Lyngby, Denmark.

Cluster compounds of the type M₃X₄⁺⁺ and M₃X₄M'⁺⁺ (M=Mo,W; X=O,S; M'=Co,Ni,Fe) are being studied as catalysts for homogeneous hydrosulphurization e.g.

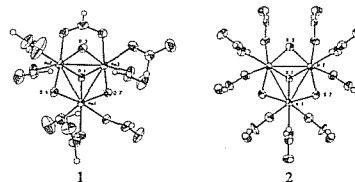


The structures of two such compounds have been studied:

1 (NH₄)_{4-x}K_x[Mo₃O₄(CHOO)₈·(H₃O)⁺·(CHOO)·H₂O (*x*=0.80) and **2** K₅[W₃S₄(CHOO)₉]⁺·(CHOO)·3H₂O.

1 is triclinic, space group P-1 with *a* = 11.011(2), *b* = 13.310(2), *c* = 9.993(1) Å, α = 106.817(7), β = 91.651(9), γ = 88.340(9)°, *Z* = 2, *R*(*F*) = 0.026 for *N*_o = 3972 and *N*_v = 451. **2** is monoclinic, space group C2/m with *a* = 19.606(6), *b* = 14.348(7), *c* = 13.627(5) Å and β = 118.94(2)°, *Z* = 4, *R*(*F*) = 0.049 for *N*_o = 2656 and *N*_v = 226.

1 contains isolated [Mo(IV)₃O₄(CHOO)₈]⁺ anions with one of the formate ions bridging two molybdenum atoms whereas in **2** all nine monodentate formate ions are bound to a W(IV)₃S₄ cluster. In both compounds there is a free formate ion. In **1** two of the cation sites are disordered and are occupied by either potassium or ammonium ions. **2** is either disordered or has a lower symmetry than C2/m, but refinement in lower symmetry space groups did not give a lower *R*-value.



PS07.03.03 SYNTHESIS AND STRUCTURE OF DI-μ-ACETATOBIS [DICARBONYLPYRAZOLERUTHENIUM(I)] (Ru-Ru) By C. H. Huang¹, T. H. Lu¹, W. M. Lee² and K. B. Shiu² ¹Department of Physics, National Tsing Hua University, Hsinchu, Taiwan, 300; ²Department of Chemistry, National Cheng Kung University, Tainan, Taiwan 701, Republic of China.

Nucleophilic substitution of [Ru₂(CO)₄(μ-O₂CMe)₂(NCMe)₂] with excess pyrazole (HPz) gave almost quantitatively the title compound [Ru₂(CO)₄(μ-O₂CMe)₂(HPz)₂].

X-ray structure of the compound was determined after crystal dimensions of 0.30 x 0.35 x 0.45 mm had been selected. The cell dimensions are *a*=8.332(1), *b*=9.5236(7), *c*=13.539(2) Å, α=90.020(8), β=100.33(2), γ=111.34(1)° with centrosymmetric space group of P 1 bar. The cell volume is 982.0(2) Å³ containing two molecules. Calculated density is 1.922 Mg m⁻³. The linear absorption coefficient of the crystal is 1.55 mm⁻¹. Absorption corrections were made. The final *R*/*R*_w factors were 0.026/0.042 for 3886 reflections with *I*>2.5σ(*I*) collected on a Nonius diffractometer, using the theta/ two theta scan mode. Weights based on counting-statistics were used. The maximum shift/sigma ratio was 0.001. In the last difference Fourier map, the deepest hole was -0.54 e/Å³ and the highest peak 0.88 e/Å³. The secondary extinction coefficient is 0.13(2). The Ru-Ru distance is 2.6746(6) Å Both rutheniums have six bonds. The distances of Ru-O range from 2.110(2) to 2.136(2) Å, those of Ru-C span from 1.822(3) to 1.829(3) Å and those of Ru-N from 2.203(2) to 2.205(3) Å. The intramolecular hydrogen bonds of N2-H...O4 and N4H...O6 are 2.790(3) and 2.774(3) Å respectively, and help stabilize the crystal structure. The conformations and intersections of five member rings will be presented.

