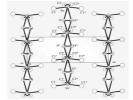
CRYSTALLOGRAPHY OF ORGANOMETALLIC, CO-ORDINATION AND MAIN GROUP

We took a X-ray structure analysis of title compound to get a information of incommensurateness using a Rigaku RAXIS-RAPID Imaging Plate diffractometer with graphite Mo-K α radiation.

As this compound is very unstable in air, crystal was sealed in a capillary under nitrogen atmosphere in a glove box.

Crystal data of (CH₃)₃SnOH: Orthorhombic, $Pmn2_1$, a = 11,207(1) Å, b = 4.171(1) Å, c = 6.652(1) Å, V = 310.9(1) Å 3 , Z = 2, R = 0.140.

The obtained crystal structure is shown in figure. This complex has an infinite chain structure throughthe bridging trimethyltin groups by hydroxide groups along the b axis. The co-ordination of each tin atom is approximate trigonal bipyramidal with two hydroxide groups and three methyl groups essentially. Due



to the mirror symmetry perpendicular to the a axis passing through the tin atom, this complex takes the disordered structure consequently.

[1] Kasai N., et al., J. Organometal. Chem., 1965, 3, 172.

Keywords: tin, hydroxide, disorder

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Crystal Structure of $[RuCl(dpphen)(terpy)]PF_6$ and $[Ru(CH_3CN)(phen) (terpy)](PF_6)_2$

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The desired complex was prepared by a sequential procedure with

ligand replacement. [RuCl₃]'H₂O and 2,2':6',2''-terpyridine (terpy) were mixed in ethyleneglycol (15 ml). The suspended mixture was refluxed for 5 min. in a microwave oven under a purging nitrogen atmosphere. 4,7-Diphenyl-1,10-phenanthroline (dpphen) was added to the refluxing red solution for 10 min. A saturated aqueous solution of KPF₆ (20 ml) was added, and a black–red product began to



precipitate. We examined complexes 1 and 2 using X-ray analysis, CV, and UV.

Crystal data of [RuCl(dpphen)(terpy)]PF₆ (1) [1] Monoclinic, C2/c, a = 42.071(1) Å, b = 8.6042(1) Å, c = 19.9088(1) Å, β = 96.096(2), V = 7165.9(2) Å³, Z = 8, R(F2)= 0.083, wR(F2) = 0.201 for 28548 measured reflections.

Crystal data of [Ru(CH₃CN)(phen)(terpy)](PF₆)₂ (**2**) [2] Triclinic, P $\bar{1}$, a = 8.7861(3) Å, b = 10.3590(9) Å, c = 17.9636(7) Å, α = 99.192(7), β =90.389(2), γ = 105.774(3), V = 1551.0 (2) Å³, Z = 2, R(F2)= 0.089, wR(F2) = 0.172 for 14369 measured reflections.

[1] Yoshikawa N., et al., *Acta Cryst.*, 2005, **E61**, m545-m547. [2] Yoshikawa N., et al., *Acta Cryst.*, 2005, **E61**, m55-m56.

Keywords: microwave, terpyridine, ruthenium compounds

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Structural Studies on Copper(II) Carboxylate Complexes containing Pyrazole

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Polynuclear copper derivatives are the subject of an increasing number of studies [1, 2]. Trinuclear copper derivatives characterized by the presence of the triangular core $[Cu_3(\mu_3-OH)(\mu-pz)_3(RCOO)_2(R=H, C_2H_5, C_3H_7)]$ have been prepared and characterized by X-ray studies. Copper(II) formate gives $[Cu_3(\mu_3-OH)(\mu-pz)_3(HCOO)_2-(Hpz)_2]$, 1, whereas copper propionate and butyrate afford $[Cu_3(\mu_3-OH)(\mu-pz)_3(HCOO)_2-(Hpz)_2]$) to the subject of an increasing number of studies of the subject of an increasing number of studies of the subject of an increasing number of studies of the subject of an increasing number of studies of the subject of an increasing number of studies of the subject of an increasing number of studies of the subject of an increasing number of studies of the subject of an increasing number of studies of the subject of an increasing number of studies of the subject of the subject

OH)(μ -pz)₃(C₂H₅COO)₂-(EtOH)], **2**, and [Cu₃(μ ₃-OH)(μ -pz)₃-(C₃H₇COO)₂(MeOH)(H₂O)], **3**, respectively, both containing solvent molecules coordinated to the copper atoms.

Here we report a comparison of their geometries and their supramolecular architectures which show the formation of interesting 1D coordination polymers.

[1] Solomon E.I., Sundaram U.M., Machonin T.E., Chem. Rev., 1996, 96, 2563. [2] La Monica G., Ardizzoia G.A., Progr. Inorg. Chem., 1997, 46, 151. Keywords: copper complexes, X-ray crystal structure, carboxylates

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Crystal Packing of Three-Aqua Sodium Maleate

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Three-Aqua Sodium Maleate was crystallized as a side product in the process of the synthesis of the manganese complexes with mixed ligands [1]. The crystal of NaC₄H₉O₇ belongs to the space group P-1 with unit cell parameters a = 5.9609(14) Å, b = 6.3907(16) Å, c = 11.2308(27) Å, $\alpha = 104.178(4)^{\circ}$, $\beta = 91.574(4)^{\circ}$, $\gamma = 100.241(4)^{\circ}$ and $D_x = 1.567 \text{ Mgm}^{-3}$ for Z = 2. The maleate monoanion is the ligand bonded through an oxygen atom to the Na⁺ ion. The intramolecular O-H..O hydrogen bond in the maleate monoanion was analysed and compared with literature data [2,3]. The central Na atom is surrounded by six oxygen atoms and forms an octahedral polyhedron with the shape of a deformed quadratic bipyramid. Molecules built in this manner form a polymerized infinite chain along the a crystallographic axis. Maleate monoanions alternate with Na ion chains and form a layer in the ac crystallographic plane. Viewed along the a axis, the packing shows that two chains are related by the inversion center, so that they form a channel along the crystallographic a axis.

[1] Stamenković J., Cakić S.,Nikolić G., *Chemical Industry*, 2003, **57**, 559. [2] James M.N.G., Williams G.J.B., *Acta Crystallogr.*, 1974, **B30**, 1249. [3] James M.N.G., Williams G.J.B., *Acta Cryst.*, 1974, **B30**, 1257.

Keywords: coordination geometry, hydrogen bonds, crystal packing

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Structural Parameters of Several Lanthanide Clusters

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This work presents key structural parameters for a series of amino acid-chelated μ_3 -hydroxide-containing polygadolinium (and lanthanide analog) complexes with potential as high relaxivity MRI contrast agents. The structures of perchlorate salts of D_{4h} -symmetry $\rm Ln_{14}(\mu_4\text{-}OH)_2(\mu_3\text{-}OH)_{16}(OH_2)_8(serine)_{20}^{3+}, \quad D_{5h}$ -symmetry $\rm Ln_{15}(\mu_3\text{-}Br)(\mu_3\text{-}OH)_{20}(\mu_3\text{-}histidinate)_{15}(\mu\text{-}OH_2)_8(OH_2)_20^{9+}, \quad T_d$ -symmetry $\rm Ln_4(\mu_3\text{-}OH)_4(valine)_4(OH_2)_8^{8+}$ and $\rm Ln_2(alanine)_4$ (OH_2)_8^{6+} complexes will be presented.

Structural parameters (Gd-O and Gd. H distances, Gd-water tilt angles, and H-bonding) that may play a significant role in MRI relaxation mechanisms for these novel complexes will be presented. We will collect neutron diffraction data on the dysprosium analog of one of the complexes at the Intense Pulsed Neutron Source (IPNS) at Argonne National Laboratory in April and hope to include these results.

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Keywords: clusters, lanthanides, materials magnetic resonance imaging