

MS16 P03

Expansion of the NASICON-type framework: from metal-sulfates to cluster-zinc cyanides Pierre Graveriau, ICMCB-CNRS, University of Bordeaux, Pessac, France. E-mail: graver@icmcb-bordeaux.cnrs.fr

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Physical properties of compounds are strongly related to crystal structures and for a 3D host-framework they can vary widely with lattice expansion. For the NASICON (Na Super Ionic CONductor*) -type skeleton a strong evolution of the cell volume from $\sim 1200 \text{ \AA}^3$ (metal sulphate*) to $\sim 12550 \text{ \AA}^3$ ((Re_6Se_8)cluster-zinc cyanides*) can be observed.

NASICON-3D framework consists of (AX_6) octahedra and (BX_4) tetrahedra sharing corners, leading to general formula $\text{A}_2\text{B}_3\text{X}_{12}$. It can be also described with structural units of 2 octahedra linked by 3 tetrahedra. In NASICON-type all these structural units have similar direction. The highest symmetry found is with space group $\text{R}\bar{3}\text{c}$ and structural units direction is \mathbf{c} axis.

If $\mathbf{X} = \mathbf{O}$, the general formula of the framework becomes $\text{A}_2(\text{BO}_4)_3$: A is metal or rare-earth and B can be S, P, (Si,P), Si, As, Ge, Mo. If necessary, additional cations could be inserted so to ensure the global electrical neutrality. The usual interstitial sites are labelled M(1) or M(2). Depending on these additional cations and on the relative A-O and X-O lengths, NASICON-structure can be also obtained in subgroups of $\text{R}\bar{3}\text{c}$: $\text{R}3\text{c}$, $\text{R}32$, $\text{R}\bar{3}$, $\text{P}\bar{3}$, $\text{C}2/\text{c}$. For these compounds, expansion of the cell volume goes from $\sim 1193 \text{ \AA}^3$ ($\text{Al}_2(\text{SO}_4)_3$; $\text{R}\bar{3}$) to $\sim 1894 \text{ \AA}^3$ ($\text{K}_{0.52}(\text{Mn}_{0.47}\text{Zr}_{0.45})(\text{MoO}_4)_3$; $\text{R}\bar{3}\text{c}$).

If $\mathbf{X} = \mathbf{C-N}$ group, the general formula of framework becomes $(\text{AC}_6)_2(\text{BN}_4)_3$ or $\text{B}_3[\text{A}(\text{CN})_6]_2$. In known examples A can be Fe or (Re_6Se_8)-cluster and B is Zn. Here also electrical neutrality can be assumed by additional cations and the expansion is so important that these compounds present zeolitic properties with inserted neutral molecules as H_2O or other gases. In these cases, expansion of the cell volume goes from $\sim 4433 \text{ \AA}^3$ ($\text{Na}_2\text{Zn}_3[\text{Fe}(\text{CN})_6]_2 \cdot 9\text{H}_2\text{O}$; $\text{R}\bar{3}\text{c}$) to $\sim 12552 \text{ \AA}^3$ ($\text{Na}_2\text{Zn}_3[(\text{Re}_6\text{Se}_8)(\text{CN})_6]_2 \cdot 24\text{H}_2\text{O}$; $\text{R}\bar{3}\text{c}$).

The poster presents a bibliographic review* and an illustration of this NASICON-type framework expansion. Beside the cristallochemistry point of view, another interest consists in the prospect for tailoring functional solid materials.

[*] list of principal references will be provided during poster session

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The dihydrogen-malonato- μ -3-calcium (II): A supramolecular framework constructed from binuclear entities ⁽¹⁾Guehria-Laidoudi A., ⁽¹⁾Aliouane K., ⁽¹⁾Rahahlia N., ⁽²⁾Dahaoui S. and ⁽²⁾Lecomte C. ^aLaboratoire de cristallographie appliquée, Institut de Chimie, U.S.T.H.B., B.P. 32, El-Alia 16111, Bab-Ezzouar, Alger, ALGERIE. ^bLCM³B-CNRS UMR 7036, Faculté des Sciences Université Henri Poincaré BP 239, 54506 Vandoeuvre-lès-Nancy, France

Keywords: polymeric structure, earth-alkaline, dicarboxylate

New generation of metal-organic coordination polymers have emerged in last years, enhancing the field of supramolecular chemistry. The anions stemming from malonic acid are the most flexible ligands for building supramolecular framework. Moreover, if the connector is the calcium cation, the resulting model complex is designed to permit the significance of interactions tacking place in biological systems [1, 2]. The title compound is one of the few hydrogen malonate complexes involving alkaline-earth metals [3]. It has been prepared under hydrothermal conditions, in single crystal form. Four μ -1,1 oxo-bridges link two crystallographically equivalent Ca cations and construct a basic binuclear building block. The distance within this cage assembly is very short (3.626(9)) \AA . The unique independent ligand is tridentate, involves all its deprotonated oxygen atoms in the coordination to the metal and displays an η^5 chelation, and both μ -1,1 and μ -1,3 bridges.

The overall network is somewhat similar to that of the barium-glutarate recently reported [4]. However, the lack of symmetry centre in the space group here, affects some features, like the basic unit which is formed of two bicapped dodecahedra sharing one face, and the orientation of the binding sites giving a complex 3D packing and replacing the two carboxylic ends in anti-parallel orientation, by end-to-end hydrogen bonding. But the substructure containing alkaline-earth atoms and forming a pseudo-eight-sided grid network is kept. This study shows that no M-O-M infinite linkage is present, confirming that this later is prevented by cage assembly around the metal.

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MS16 P05

Substituted Salicylaldiminato and Related Cobalt Complexes at 173 K Jarno Kansikas, Ahlam Shibaud, Timo Repo, Ilpo Mutikainen and Markku Leskelä, Department of Chemistry, Laboratory of Inorganic Chemistry, University of Helsinki, Finland. E-mail: jarno.kansikas@helsinki.fi

Keywords: cobalt compounds, catalyst structure, salicylaldiminato

Here will be presented two cobalt(II) and two cobalt(III) catalyst compounds formed by variously substituted salicylaldiminato ligands in three structures and an imine derivative in one structure. The studied compounds are bis[isopropyl-*N*-(3,5-di-*tert*-butylsalicylaldiminato)]cobalt(II) (**I**), bis{[(2-ethylphenylimino)ethyl]phenoxy}cobalt (II) (**II**), bis[2-phenylethyl-*N*-(3,5-di-*tert*-butylsalicylaldiminato)]bis(phenylethylamino) cobalt(III) iodide (**III**) and tris[4-fluorobenzyl-*N*-(salicylaldiminato)]cobalt(III) (**IV**). Compound (**I**) crystallizes in $P_4(3)$ (78) with $Z=8$, (**II**) in $P2_1/c$ (14) with $Z=4$, (**III**) in $C2/c$ (15) with $Z=4$ and (**IV**) in $P2_1/n$ (14) with $Z=8$.