## MS27 P07

Complexes of M (carbaldehyde-oxime)( $\mathrm{M}=\mathrm{Pd}$ and Hg ): Synthesis and Structural Characterization
Ali Asgar Torabi Chemistry Department, Zanjan University, Km. 5, Tabriz Road, P. O. Box 45195-313, Zanjan, Iran.
E-mail: atorabi@mailznu.ac.ir.
Keywords: Metallic Chelates; Synthesis of New Materials; Structural Characterization of Palladium and Mercury

Syntheses and X-ray crystal structure analyses of Hg (II) and $\mathrm{Pd}(\mathrm{II})$ complexes with pyridine-2-carbaldehyde-oxime ligand are reported. The complexes are obtained from the reaction of simple $\mathrm{Hg}(\mathrm{II})$ and $\mathrm{Pd}(\mathrm{II})$ salts $\left(\mathrm{PdCl}_{2}\right.$ and $\mathrm{HgCl}_{2}$ ) with pyridine-2-aldoxime (HL). The molecular structures of complexes have been determined by singlecrystal X-ray analyses. Structural characterization reveals in Hg (II) case the presence of $\mathrm{HgL}_{2}$ complex but in Pd (II) case the formation of $\mathrm{PdL}(\mathrm{HL}) \mathrm{Cl}$ complex is observed.
[1] K. S. Murray, Adv. Inorg. Chem., 43, (1995).
[2] E. Colacio, J. M. Dominguez-Vera, A. Escuer, R. Kivekäs and A. Romerosa, Inorg. Chem., 33, (1994) and refs. therein.
[3] J. M. Dominguez-Vera, E. Colacio, A. Escuer, M. Klinga, R. Kivekäs and A. Romerosa, Polyhedron, 16, (1997) and refs. therein.
[4] E. Colacio, J. M. Dominguez-Vera, J. M. Moreno, A. Escuer, M. Klinga, R. Kivekäs and A. Romerosa, J. Chem. Soc., Dalton Trans., (1997).
[5] B. Cervera, R. Ruiz, F. Lloret, M. Julve, J. Cano, J. Faus, C. Bois and J. Mrozinski, J. Chem. Soc., Dalton Trans., (1997).
[6] F. Birkelbach, T. Weyhemüller, M. Lengen, M. Gerdan, A. X. Trautwein, K. Wieghart and P. Chaudhuri, J. Chem. Soc., Dalton Trans., (1997).
Journaux and C. Bois, Inorg. Chem., 36, (1997).
[8] R. Ruiz, F. Lloret, M. Julve, J. Faus, M. C. Muñoz and X. Solans, Inorg. Chim. Acta, 268, (1998).
[9] J. Faus, F. Lloret, M. Julve, J. M. Clemente-Juan, M. C. Muñoz, X. Solans and M. Font-Bardía, Angew. Chem., Int. Ed. Engl., 35, (1996).
[10] E. Colacio, C. López-Magaña, V. McKee and A. Romerosa, J. Chem. Soc. Dalton. Trans., (1999).
[11] P. Starynowicz and J. Lisowski, Chem. Common. 9, (1999).
[12] M. K. Sener, D. Sanchez-Garcia, M. Akkurt, S. O. yildirim, H. Fun and J. L. Sessler, Turk J. Chem. 30, (2006).


Fig. 1. ORTEP diagram of a symmetric unit of $\left[\mathrm{PdH}\left(\mathrm{C}_{6} \mathrm{~N}_{2} \mathrm{OH}_{5}\right)_{2} \mathrm{Cl}\right](\mathbf{1})$.


Fig. 2. ORTEP diagram of a unit of $\left[\mathrm{HgC}_{6} \mathrm{~N}_{2} \mathrm{OH}_{6} \mathrm{Cl}_{2}\right](\mathbf{2})$

| MS27 P08 |  |  |
| :---: | :---: | :---: |
| Synthesis and | Crystal | Structure |
| 4-Aminoquinolinium Bis(citrato)borate Monohydrate |  |  |
| Irena Zviedre ${ }^{\text {a }}$, Sergey Belyakov ${ }^{\text {b }}$, Andrey Tokmakov, ${ }^{\text {b }}$ |  |  |
| ${ }^{\text {a }}$ Institute of Inorganic Chemistry of the Riga Technical |  |  |
| University, Latvia. ${ }^{\text {b }}$ Latvian institute of Organic Synthesis, |  |  |

Keywords: single-crystal X-ray, boron compounds, hydrogen bonds

In the course of structural studies of amine salts of boron coordination compounds a new complex -4-aminoquinolinium bis(citrato)borate monohydrate (I) has been synthesized and its X-ray structural investigation has been carried out. Complex I was prepared by mixing boric and citric acids with 4 -aminoquinoline in proportions (molar ratio) 1:2:1 in water solution. Crystals of compound I were obtained by slow evaporation of water from solution.


The structure of the title compound I is formed by univalent cations $\left[\left(4-\mathrm{NH}_{2}\right) \mathrm{C}_{9} \mathrm{H}_{6} \mathrm{NH}\right]^{+}$, spiran-type complex anions $\left[\left(\mathrm{C}_{6} \mathrm{H}_{6} \mathrm{O}_{7}\right)_{2} \mathrm{~B}\right]^{-}$and crystallization water molecules (fig.). Boron atoms are tetracoordinated as confirms the $s p^{3}$ hybridization sphere around them. The bonds B-O(carb.) (average length $1.495(3) \AA$ ) are systematically longer than $\mathrm{B}-\mathrm{O}(\mathrm{hydr})$ (average length $1.448(3) \AA$ ). The values of the dihedral angles corresponding to the crossing of two main-square planes of boron-containing heterocycles - $\mathrm{C}-\mathrm{O}-\mathrm{B}-\mathrm{O}-\mathrm{C}-$ in the two independent complex anions amount to $87.88(6)^{\circ}$ and $87.87(6)^{\circ}$. The molecule of 4 -aminoquinoline molecule is protonated at the ring N 1 atom but not at the amino group nitrogen atom N 2 in acidic solutions. The $\mathrm{C} 4{ }^{\prime}-\mathrm{N} 2$ bond length (1.326(3) $\AA$ in average) is approximately equal to the $\mathrm{C}=\mathrm{N}$ double bond lenght indicating that atom N 2 of the amino group must also be $s p^{2}$ - hybridized. The $\mathrm{C}-\mathrm{C}$ bonds at C 4 ' atom are somewhat longer (1.424(3) $\AA$ in average). The N2 atoms lie almost in the quinoline ring plane (maximum deviation of $0.029(3) \AA$ ). Bond angles $\mathrm{C}-\mathrm{N}^{+}-\mathrm{C}$ are 118.6(5) ${ }^{\circ}$. The hydrogen atoms of amino groups, protonated nitrogen atoms, as well as those of hydroxyl groups from two independent complex anions and three of four hydrogen atoms from two independent water molecules are involved in formation of 17 independent $\mathrm{O}-\mathrm{H} . . . \mathrm{O}$ and $\mathrm{N}-\mathrm{H} . . . \mathrm{O}$ type hydrogen bonds. Nitrogen

