## Poster Presentations

[MS10-P19] trans-Bis(perchlorato-κΟ) tetrakis(1-vinyl-1H-imidazole-κN3) copper(II). Yunus Celika, Kamil Kucukb Ramazan Şahinc, Bunyamin Karabulutd, Omer Andac c, Necmi Degea

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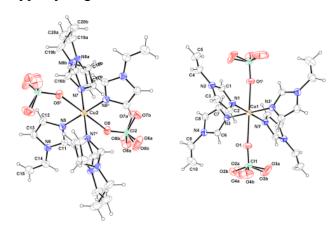
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A new copper(II) complex of 1-vinylimidazole (1-Vim) and perchlorate has been synthesized and structurally characterized. The molecular structure of the complex exhibits six coordination with a distorted octahedron geometry. The 1-Vim ring is coordinated through the imidazole nitrogen, as expected from earlier spectroscopic studies of related adducts [1-3]. The complex crystallizes in triclinic system and the space group is P-1 with the following parameters; a=9.6242(4), b=10.4140(4), c=15.2662(6) Å,  $\alpha$ =70.278(3) $_{\rm i}$ ,  $\beta$ =80.056(3) $_{\rm i}$ ,  $\gamma$ =74.662(3) $_{\rm i}$ , V=1383.15(10) Å3, and Z=2.

In the title compound, [Cu(ClO<sub>4</sub>)<sub>2</sub>(C<sub>5</sub>H<sub>6</sub>N<sub>2</sub>)<sub>4</sub>], the Cu(II) ion is located on an inversion centre. It features a Jahn–Teller distorted octahedral coordination geometry, defined by four N atoms of four 1-vinylimidazole ligands in the equatorial plane and two nitrate O atoms in the axial positions. The structure contains chemically identical but crystallographically independent two neutral molecules as shown in Figure 1. All perchlorate ligands and two of the 1-Vim ligands were shown to be disordered over two sites with a ratio of 0.45(1):0.55(1) but nitrogen and oxygen atoms coordinated to Cu(II) metal ions were not disordered. In the crystal, the complex molecules

are linked by weak intermolecular C—HááO type hydrogen bonds.



**Figure 1.** ORTEP III view of [Cu(ClO4)2(C5H6N2)4] with the atomnumbering scheme (10% probability ellipsoids). The disordered part of the atoms has been shown. [symmetry codes (i):1-x, 1-y, 1-z; (ii):-x, -y, -z].

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**Keywords:** Cu(II) complex; 1-Vinylimidazole; Single crystal; XRD