Poster Presentations

[MS24-P30] MOFs with 5Aminoisophthalate & 4,4'-Bipys or a Reduced Schiff Base Dicarboxylate.

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Hydrothermal reactions of the reduced Schiff base 4carboxy-phenylene-methyleneamino-4-benzoic acid $(L^{T}H_{2})$ with Zn(OH), and DPE [1,2-di(4pyridyl)ethylene] afforded non-porous 2D coordination polymer of the formula $\{[Zn_2(L')_2(DPE)] \cdot H_2O\}n$ (1). The analogous reaction of L^{H} , with Zn(OH), and 4,4'-azopy (4,4'-azopyridine) giving the porous 3D structure $\{[Zn_2(L^{\prime})_2(4,4'azopy)]\cdot 7.5H_2O\}$ (2), which exhibits interesting hysteretic behaviour on CO2 adsorption and desorption. The solvothermal reaction of $Zn(NO_3)_2$ · $6H_2O$ with 5aminoisophthalic acid and 4,4'-bipyridyl (4,4'-bipy) led to the self-assembly of the known 3D hybrid Hbonded/covalent structure $\{[Zn(5-AIP)(4,4'bipy)_{0.5}]$ DMF $\}n$ (3), but with dimethylformamide (DMF) in this case rather than H₂O as previously [2], with an analogous reaction using the related molecule 4,4'-azopyridine (4,4'-azopy) in place of 4,4'-bipyridyl giving the structurally related framework {[Zn(5-AIP)(4,4'-azopy)0.5] \cdot DMF}n (4). Similar solvothermal reactions of $Co(NO_3)$ · $6H_2O_3$ and $Cd(NO_2)$ · $4H_2O$ with $Mn(NO_2) \cdot 4H_2O$ 5-aminoisophthalate and the potential linkers 4,4'-bipy, 2-di(4-pyridyl)ethylene (DPE) and 4,4'-azopy afforded the porous 3D structures {[Co2(NO3)2(5-AIP)(4,4'-bipy)2] · 2EtOH} (5), $\{[Co(5AIP)(DPE)] \cdot 3DMF\}n$ n (6)

 $\{[Co(5-AIP)(4,4'azopy)] \cdot 2DMA\}n$ and (7) (DMA = dimethylacetamide){[Mn(5-AIP)(4.4'-bipy)] \cdot 2DMA}n (7),{[Mn(5-AIP)(DPE)] \cdot 1.5DMF}*n* $\{[Mn(5-AIP)]$ (8), (4,4'azopy)]· 1.5DMF}*n* (9), {[Cd(5-AIP) (4,4'bipy)]· 3DMFn (10, which has been previously reported [2]), {[Cd(5-AIP) (DPE)]· DMFn(11)and {[Cd(5-AIP) (4,4'-azopy)] DMFn (12), with structures 6-12 bearing the same network topologies with metal atoms and 5-AIP ligands in sheets, with bipy ligands acting as pillers, and solvent molecules around the bipy ligands. Carbon dioxide uptakes for these structures, whilst modest, reveal an interesting pattern, with enhanced uptakes observed for those structures which bear larger pore sizes and polar functionalities. The structures will be discussed and the specific problems associated with MOF crystallography will be explored.

[1] Kongshaug, K.O. and Fjellvåg, H., (2006). *Inorg. Chem.*, 45, 2424-2429.

[2] Wang, H. N., Meng, X., Qin, C., Wang, X. L., Yang, G.S., and Su, Z.M., (2012). *Dalton Trans.*, 41, 1047-1053.

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