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MS24-P3 Electron density analysis in Quantum Magnets

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We have investigated the correlation between the accurate electron density distribution and the magnetic properties of two metal-organic polymeric quantum magnets, the $\{[Cu(pyz)_2CI]BF_4\}_n$ and the $\{[Cu(pyz)_2CI]BF_4\}_n$ and the $\{[Cu(pyz)_2Br]BF_4\}_n$ (pyz = pyrazine), using high resolution single crystal X-ray diffraction and density functional theory (DFT) calculations in the crystalline state as well as in the gas phase on selected fragments of the framework. Topological Analysis based on Quantum Theory of Atoms in Molecules (OTAIM) has been applied to characterize the possible magnetic exchange coupling constants. Electron density analysis confirmed the orientation of the magnetic orbital. The magnetic properties have been examined and correlated with the topological and integrated properties of the electronic distribution. This has enabled the detailed rationalization of the experimental antiferromagnetic exchange coupling constants in terms of the interchain Cu-Cu interactions. Molecular orbital and spin density analysis have been used to identify the atomic and group sources of magnetism. In particular, halogens and pyrazine act as typical non-innocent ligands, with large part of the spin density developed on their atoms. In both cases, the experimentally observed antiferromagnetic coupling can be explained by the copper-copper super-exchange coupling mediated by the pyrazine ligands. Moreover, our results suggest a non-negligible coupling through the halogens although the electron density shows a much weaker interaction than the one observed along pyrazine. From this study, it is clear that systematic electron density analysis on transition metal compounds could lead to a better understanding of the super exchange mechanism with a topological description of the involved interactions.

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