

## MS14-03 | KAHLENBERGITE, A NEW POTASSIUM $\beta$ -ALUMINA MINERAL

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Kahlenbergite (IMA 2018-158) is a natural potassium  $\beta$ -alumina, with an empirical formula of  $(K_{0.87}Mg_{0.09}Ca_{0.03}Ba_{0.01})_{S1}(Al_{9.46}Fe^{3+}_{1.36}Mg_{0.14}Cr^{3+}_{0.02}Si_{0.02})_{S11}O_{17}$ . It occurs in small hematite segregations within wollastonite-gehlenite rocks. The mineral association suggests formation temperature between 1000 and 1200 °C (Sharygin, 2019). Kahlenbergite forms platy, light-brown crystals, epitaxially replaced and overgrown by hibonite. The unit cell dimensions ( $a=5.64860(6)$ ,  $b=22.8970(3)$  Å) and space group  $P6_3/mmc$  of kahlenbergite corresponds to that of synthetic K  $\beta$ -alumina. The crystal structure was refined using synchrotron diffraction data (beamline X06DA, SLS, PSI). Compared to synthetic K  $\beta$ -alumina, which often shows considerable amounts of positional and occupational cation disorder, the structure of kahlenbergite is fairly simple. It exhibits a fully occupied position of the K atom at  $(\frac{2}{3}, \frac{1}{3}, \frac{1}{4})$ . The structure of kahlenbergite is made of spinel blocks, divided along  $c$  into mixed ( $M$ ) layers with  $AlO_6$  octahedra and  $(Al_{0.56}Fe_{0.44})O_4$  tetrahedra, Kagome ( $K$ ) layers with  $(Al_{0.92}Fe_{0.08})O_6$ , and pillar ( $P$ ) layers with two  $AlO_4$  tetrahedra and K-atoms. The presented structure model of kahlenbergite describes an idealised ordered structure. All investigated crystals exhibit one-dimensional diffuse scattering. In one crystal additional reflections can be identified, which obviously belong to the  $Fe^{3+}$ -analog of hibonite.

The structure of kahlenbergite and the  $Fe^{3+}$ -analog of hibonite contain identical blocks, which are connected by  $P$ -layers in kahlenbergite and so-called  $R$ -layers in the  $Fe^{3+}$ -analog of hibonite. The  $R$ -layers contain Ca atoms,  $AlO_5$ -bipyramids, and further  $AlO_6$  octahedra. Therefore, the connecting layers are most likely the source of the disorder.

[1] Sharygin, (2019) Mineralogical Magazine 83, 123–135