MS14-P21 | INVERSE CRYSTAL STRUCTURE BEHAVIOUR OF CA₃AL₄ZNO₁₀ AT HIGH PRESSURE

AND HIGH TEMPERATURE

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Ca₃Al₄ZnO₁₀ was initially found in an investigation on the incorporation of ZnO into Portland cement clinker phases [1]. In contrast to a previous description of the structure in space group *Pbc*2₁ [2] Ca₃Al₄ZnO₁₀ was found to be isotypic with Ca₃Al₄MgO₁₀ [3]. Ca₃Al₄ZnO₁₀, a = 5.1364(3), b = 16.7403(9), c = 10.7019(6) Å, V = 920.20(8) Å³, Z = 4, crystallizes in *Pbcm*. Crystals were obtained in a crystal growth experiment and studied in-situ between ambient pressure and 6.8(1) GPa as well as 25(2)-797(2) °C. Lattice parameters change continuously in the examined P-T-range, thus allowing computation of a second-order Birch-Murnagham equation-of-state. The structure represents a three-dimensional network of corner-sharing [AlO₄]- and [ZnO₄]-tetrahedra with Ca ions in six- and eightfold coordination. Analysis of the pressure- and temperature-induced response of the structure shows an inverse relationship of all observed mechanism contributing to overall structural changes. To give an example: The only Ca-O bond distance increasing with increasing pressure, i.e. Ca2-O3, is found to decrease with increasing temperature.

[1] H. Bolio-Arceo, F.P. Glasser; Adv. Cem. Res. 1998, 10, 25-32.

[2] V.D. Barbanyagre, T.I. Timoshenko, A.M. Ilyinets, V.M. Shamshurov; Powder Diffr., 1997, 12, 22–26.

[3] V. Kahlenberg, R. Albrecht, D. Schmidmair, H. Krüger, B. Krüger, M. Tribus, A. Pauluhn; J. Am. Ceram Soc. 2019, 102, 2084-2093.