MS35-P08 | Two Composite Mn(II)-squarate-dpe Supramolecular Networks Showing Interesting Water Hysteresis Phenomenon in Water Vapor Ad-/desorption Isotherms

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Assembly of two Mn(II)-squarate supramolecular networks, including composite two-dimensional (2D) $[Mn(Hdpe)_{2}(H_{2}O)_{3}][Mn(C_{4}O_{4})_{2}(H_{2}O)_{3}][Mn(C_{4}O_{4})_{2}(H_{2}O)_{3}] (1) \text{ and } [Mn(Hdpe)_{2}(H_{2}O)_{4}][Mn(C_{4}O_{4})_{2}(H_{2}O)_{3}]_{2} \times 8H_{2}O (2) (C_{4}O_{4})_{2}(H_{2}O)_{3}]$ (squarate = dianion of 3,4-dihydroxycyclobut-3-ene-1,2-dione ($H_2C_4O_4$); dpe = 1,2-bis(4-pyridyl)ethane), associated with monodentate anti-Hdpe⁺ ligands, have been synthesized and structurally characterized by single-crystal X-ray diffraction studies. Compound 1 is a [2D plus 2D] tri-layered composite polythreading network, composed of one 2D cationic $[Mn(Hdpe)(C_4O_4)_{0.5}(H_2O)_3]^{2+}$ metal-organic framework (MOF) and two 2D anionic $[Mn(C_4O_4)_2(H_2O)_3]^{2-}$ MOFs, with the penetration of the monoprotonated anti-form Hdpe⁺ ligands dangling above and below the 2D Mn(II)- $(m_{1,2,3,4}-C_4O_4)$ layer in the 2D cationic [Mn(Hdpe)(C_4O_4)_{0.5}(H₂O)₃]²⁺ MOFs into the square-grid windows of two 2D Mn(II)- $(m_{1,3}-C_4O_4)$ layers in the anionic $[Mn(C_4O_4)_2(H_2O)_2]^2$ MOFs. Compound **2** is a [0D plus 2D] bi-layered composite polythreading network composed of a cationic $[Mn(Hdpe)_2(H_2O)_4]^{2+}$ monomer and a 2D layered anionic $[Mn(C_4O_4)_2(H_2O)_2]^{2^-}$ MOF, with the penetration of the monoprotonated *anti*-form Hdpe⁺ ligands in the cationic $[Mn(Hdpe)_2(H_2O)_4]^{2+}$ monomers into the square-grid windows of two 2D Mn(II)- $(m_{1,3}-C_4O_4)$ layers in the anionic $[Mn(C_4O_4)_2(H_2O)_2]^{2^*}$ MOF. Both of the 2D composite polythreading networks in **1** and **2** are further extended to their 3D supramolecular architectures via the combination of intermolecular hydrogen bonds and p-p stacking interactions. Compounds 1 and 2 both exhibit interesting hysteresis phenomenon in water vapor ad-/de-sorption isotherms with chemisorption on the Mn(II) sites found in 1 and 2 and physisorption of water chains found in 2. The de/re-hydration procedures by cyclic TG analysis and powder X-ray diffraction measurements of 1 and 2 evidence reversible sponge-like water de-/ad-sorption property associated with a dynamic structural transformation.