

## The variety of calcium bearing efflorescence phases - An explanation by crystal chemistry

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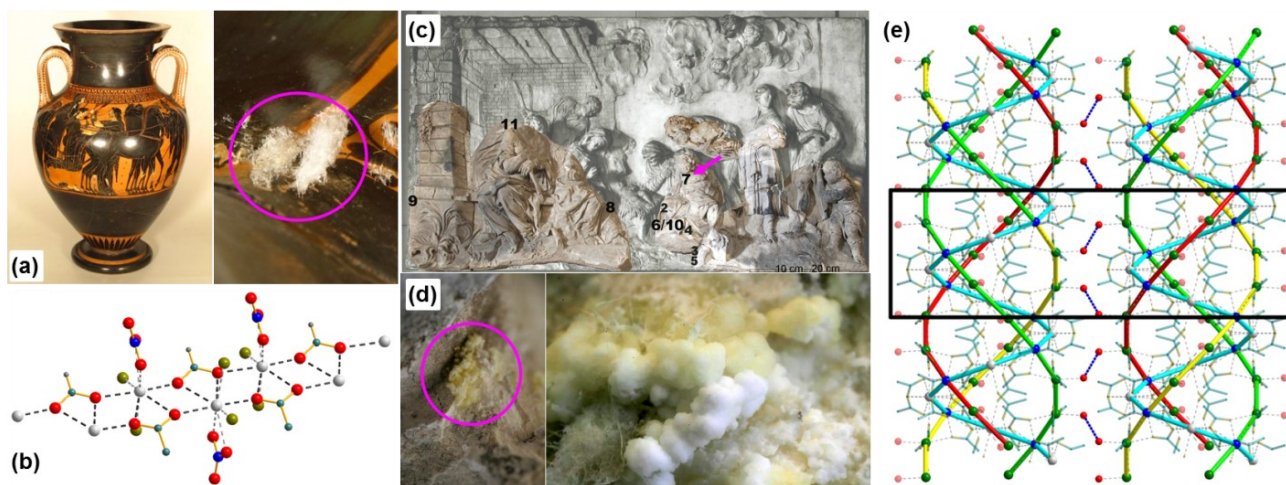
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Cultural heritage objects are affected by various corrosion processes during decades and centuries of storage in museums and collections. Atmospheric gases like CO<sub>2</sub>, moisture or - as wood emits significant amounts of formic and acetic acid<sup>1</sup> - the storage furniture itself can induce corrosion. Calcareous heritage objects like historic Mollusca shells<sup>2</sup>, eggs<sup>3</sup>, ancient pottery (Figure 1, a) or marble reliefs (Figure 1, c, d) are very sensitive to acetic and formic acid vapours. The corrosion process leads to the formation of efflorescence crystals sometimes crystallizing in pores and cracks, which can cause severe damage to the artifacts. This phenomenon has been known as “Byne’s disease” since the end of the 19<sup>th</sup> century.<sup>4</sup> Both, simple salts like Ca(CH<sub>3</sub>COO)<sub>2</sub>·H<sub>2</sub>O<sup>5</sup> or Ca(CH<sub>3</sub>COO)<sub>2</sub>·½H<sub>2</sub>O<sup>6</sup> and complex compounds like calclacite (Ca(CH<sub>3</sub>COO)Cl·5H<sub>2</sub>O)<sup>7</sup> or thecotrichite (Ca<sub>3</sub>(CH<sub>3</sub>COO)<sub>3</sub>Cl(NO<sub>3</sub>)<sub>2</sub>·6H<sub>2</sub>O)<sup>8-9</sup> were found as corrosion phases on calcareous historic objects. Many of these efflorescence phases, however, still remain poorly characterized due to their microcrystalline character and the occurrence of polyphase mixtures.

Our work focuses on the characterization of unknown or hitherto poorly characterized efflorescence phases found on heritage objects. As the amount of substance that can be removed from the artifacts is usually very small, we also describe the synthesis and structure elucidation of complex efflorescence salts like Ca<sub>2</sub>(CH<sub>3</sub>COO)(HCOO)(NO<sub>3</sub>)<sub>2</sub>·4H<sub>2</sub>O<sup>10</sup>, Ca(CH<sub>3</sub>COO)(HCOO)·2H<sub>2</sub>O and Ca<sub>3</sub>(CH<sub>3</sub>COO)<sub>4</sub>(HCOO)<sub>2</sub>·4H<sub>2</sub>O<sup>11</sup> that were found on ancient amphorae (Figure 1, a) or historic birds eggs<sup>5</sup> and seemingly simple corrosion phases like Ca(CH<sub>3</sub>COO)<sub>2</sub>·½H<sub>2</sub>O<sup>12</sup> which crystallizes on marble reliefs (Figure 1, c, d) or ceramics<sup>6</sup>. A systematic structural investigation of these efflorescence phases revealed calcium carboxylate zig-zag chains (Figure 1, b) as the common structural motif, which shows the crucial role of the carboxylic acids in the corrosion processes and explains the great chemical variety of these compounds. The seemingly simple Ca(CH<sub>3</sub>COO)<sub>2</sub>·½H<sub>2</sub>O was found to crystallize in a 11794.5(3) Å<sup>3</sup> unit cell with a triple helix motif (Figure 1, e) analogous to the collagen proteins.

In summary, the investigations on corrosion phases found on cultural heritage objects led to the discovery of many hitherto unknown or only poorly characterized solid phases with complex crystal structures. In addition, global structural motifs that were revealed in these studies indicate that a lot more compounds are to be discovered.



**Figure 1.** Attic black figured amphora from the Antikensammlung Munich with a white efflorescence phase ©Anton Buhl; (b) excerpt of the crystal structure of the efflorescence phase Ca<sub>2</sub>(CH<sub>3</sub>COO)(HCOO)(NO<sub>3</sub>)<sub>2</sub>·4H<sub>2</sub>O; (c) Fragmented relief 'Adoration of the Shepherds' by G. Torretti, (d) the relief exhibits efflorescence phases ©Rathgen Forschungslabor, Staatliche Museen zu Berlin – Preußischer Kulturbesitz; (e) excerpt of the crystal structure of the efflorescence phase Ca(CH<sub>3</sub>COO)<sub>2</sub>·½H<sub>2</sub>O exhibiting a triple helix motif.

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