Zinc complexes with nitroderivatives of quinolin-8-ol

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Five new zinc(II) complexes, $\{Na[Zn(ClNQ)(SO_4)(H_2O)]\}_n$ (1), $[Zn(dNQ)_2(H_2O)_2] \cdot 1,4$ -dioxane (2), $[Zn(dNQ)_2(H_2O)_2]$ (3), $NH_2(CH_3)_2[Zn(ClNQ)_3] \cdot DMF$ (4) and $K[Zn(ClNQ)_3] \cdot 2DMF$ (5), (HClNQ = 5-chloro-7-nitroquinolin-8-ol, HdNQ = 5,7-dinitroquinolin-8-ol (Fig.1)) have been prepared. All complexes were characterized by IR spectroscopy, elemental analysis and X-ray structure analysis.

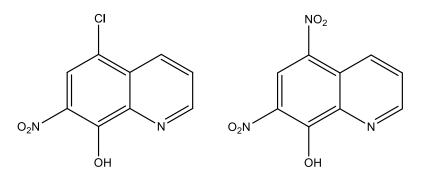


Figure 1. Chemical structures of HClNQ (left) and HdNQ (right)

Complex 1 has a polymeric structure. Zn(II) atom is penta-coordinated by one bidentate molecule of ClNQ ligand, one molecule of water and a pair of crystallographically equivalent sulfate anions, which interconnect adjacent zinc atoms to form a *zig-zag* chain. In addition, the central atoms are also connected through ionic interactions between oxygen atoms with a partial negative charge and sodium cation.

Complexes 2 and 3 have similar molecular structures, Zn(II) atom sits at the center of the symmetry, therefore only a half of the molecule is independent. In their crystal structures, there are two *trans*-coordinated dNQ molecules in the equatorial plane while two water molecules occupy axial positions, forming a deformed octahedral geometry. Complex 2 also contains one uncoordinated molecule of 1,4-dioxane.

Complexes 4 and 5 are ionic compounds with very similar structures, in which Zn(II) atom is tris-coordinated by molecules of deprotonated 5-chloro-7-nitro-quinolin-8-ol with nitrogen and oxygen donor atoms coordinated in *mer*-fashion. The negative charge of the complex anions is counterbalanced by uncoordinated dimethylammonium and potassium cations, respectively, and interesting orientation of the oxygen atoms to $NH_2(CH_3)_2^+$ (4) and K^+ (5) ions is observed. In addition, the complex 5 contains one more solvated molecule of DMF molecule.

Keywords: 5-chloro-7-nitroquinolin-8-ol; 5,7-dinitroquinolin-8-ol; zinc

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