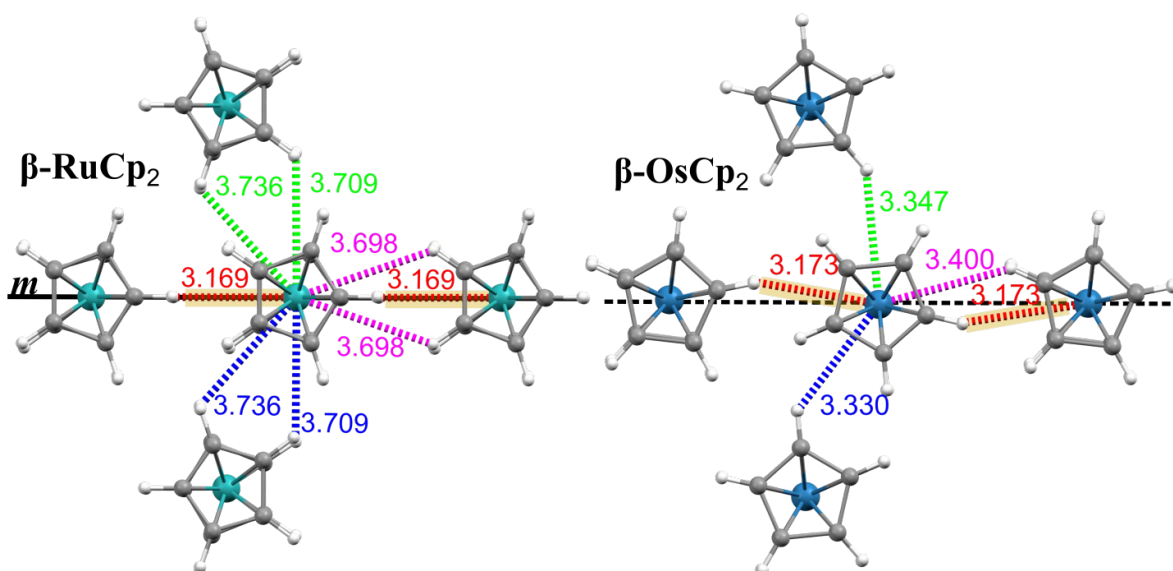


## Poster

Competition between  $\text{CH} \cdots \pi$  and  $\text{CH} \cdots \text{M}$  contacts in metallocenes structuresI. Moszczyńska<sup>1</sup>, A. Katrusiak<sup>1</sup><sup>1</sup>Adam Mickiewicz University, Uniwersytetu Poznańskiego 8 61-614

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Ferrocene ( $\text{FeCp}_2$ ), [1] ruthenocene ( $\text{RuCp}_2$ ) [2] and osmocene ( $\text{OsCp}_2$ ) [3] are the only known simple metallocenes, which crystallize in orthorhombic system with eclipsed conformation of cyclopentadienyl rings. The structures at 0.1 MPa/ 296 K of  $\text{RuCp}_2$  ( $\alpha\text{-RuCp}_2$ ) and  $\text{OsCp}_2$  ( $\alpha\text{-OsCp}_2$ ) and low-temperature phase of  $\text{FeCp}_2$  are isostructural (space group  $Pnma$ ) and mostly stabilize by  $\text{CH} \cdots \pi$  contacts. We also discovered a new high-pressure phase of  $\text{RuCp}_2$  ( $\beta\text{-RuCp}_2$ ) [4] and  $\text{OsCp}_2$  ( $\beta\text{-OsCp}_2$ ) [5] which differ from each other. In spite of differences in crystal symmetry of  $\beta\text{-RuCp}_2$  and  $\beta\text{-OsCp}_2$ , the reason of occurring phase transitions was the same - higher preference to  $\text{CH} \cdots \text{M}$  bond creations at high-pressure conditions. In  $\beta\text{-RuCp}_2$  (space group  $Pcmb$ ) one ruthenocene molecule creates only one close  $\text{CH} \cdots \text{M}$  contact, while in  $\beta\text{-OsCp}_2$  (space group  $Pcab$ ) there are four symmetry independent  $\text{CH} \cdots \text{M}$  contacts per one molecule (Fig. 1).



**Figure 1.**  $\text{CH} \cdots \text{M}$  bonds in  $\beta\text{-RuCp}_2$  at 1.00 GPa and in  $\beta\text{-OsCp}_2$  at 1.05 GPa. The interacting molecules are viewed along their axes, lying in the mirror planes for ruthenocene and close to glide plane  $a$  for osmocene. The indicated distances are given in angstroms.

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