

Oral presentation

Structural Investigations of a Family of Perovskite-Derived Oxide Ion Conductors

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The global transition away from dependence on fossil fuels necessitates the development of devices that can utilise alternative fuel sources such as hydrogen. Solid oxide fuel cells (SOFCs) are one technology that benefits from high energy conversion efficiencies and a lack of moving parts. Traditional materials used to form the oxide ion conducting electrolyte in SOFCs (e.g., yttria-stabilised zirconia, YSZ or gadolinia-doped ceria, GDC) suffer from poor intercomponent compatibility and require high temperatures (>1000 °C) to conduct sufficiently for use in fabricated devices, which reduces cell lifetimes. This presents an opportunity for the discovery of novel materials that produce better quality devices.

Whereas the mechanisms of ionic conductivity in YSZ and GDC have been well understood through their simple defect chemistry, more recently investigated materials use more complex mechanisms of ionic conductivity. These mechanisms involve features such as flexible coordination spheres (e.g., La₂Mo₂O₉ (LAMOX) materials) and terminal oxygen sites unconnected to the wider polyhedral framework (e.g., La_{1.54}Sr_{0.46}Ga₃O_{7.27} (melilite)), which both help stabilise migrating oxide ions through local relaxation, affording greater ionic conductivity [1,2].

The A₃OhTd₂O_{7.5} materials (A = alkaline earth, Oh = octahedral, Td = tetrahedral metal ions) are a family of oxygen-deficient perovskite derivatives identified as structurally interesting potential ionic conductors because they adopt a variety of structures with chains containing Td₂O₇ moieties (figure 1a) [3,4]. These groups possess two terminal oxygens per unit and (for materials with Td = Ga) variable coordinate metal centres. There has also been evidence of order-disorder phase transitions involving the oxygen substructure, further suggesting high ionic mobility and conductivity [5].

We will present results of combined X-ray and neutron diffraction studies on members of the A₃OhTd₂O_{7.5} family including Ba_{1.5}Sr_{1.5}YGa_{2-x}Zn_xO_{7.5-δ} and Ba₃ScGa₂O_{7.5}. The former are mixed A-site materials formed by the intergrowth of Ba/Sr-rich regions and exhibit additional oxygen substructure disorder within the Ga₂O₇ units. Ba₃ScGa₂O_{7.5} has been shown to have a previously uncharacterised orthorhombic structure formed by supercell ordering of a simple distorted perovskite cell. The supercell was identified through additional electron diffraction measurements.

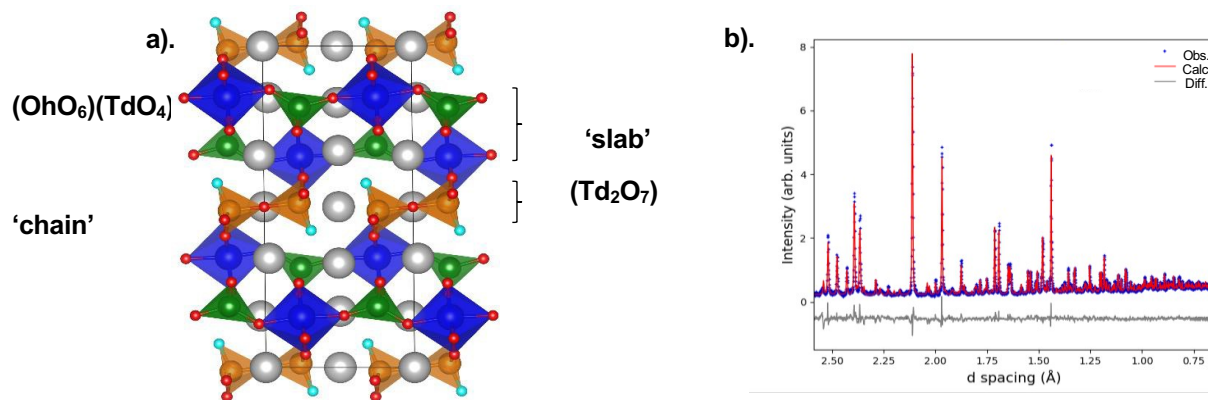


Figure 1 – a). Example of an A₃OhTd₂O_{7.5} structure (A - grey, Oh - blue, Td - green/orange, O - red. Terminal O-sites are highlighted in light blue). b). Rietveld refinement against room temperature neutron data for Ba_{1.5}Sr_{1.5}YGa₂O_{7.5}.

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