

Poster

Quaternary compounds in $R-T-Al-M$ systemsS. Pukas¹, R. Gladyshevskii¹

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We are at present carrying out a systematic search for quaternary aluminosilicides and alumogermanides containing a rare-earth metal (R) and a 3d-metal (T). Aluminosilicides of two compositions, $R_2TAAl_4Si_2$, $R_3TAAl_3Si_2$, and alumogermanides of four compositions, $RTAl_4Ge_2$, $R_2TAAl_4Ge_2$, $R_3TAAl_3Ge_2$, $R_5T_3Al_3Ge_4$, were synthesized by arc melting under argon and annealed at 873 K for up to three months. The crystal structures were studied by X-ray powder diffraction. 43 compounds with Si and 92 with Ge were found to adopt the rhombohedral $SmNiAl_4Ge_2$ (Pearson symbol $hR24$, space group $R-3m$), tetragonal $Tb_2NiAl_4Ge_2$ ($tI18$, $I4/mmm$), hexagonal $Y_3NiAl_3Ge_2$ ($hP9$, $P-62m$), or orthorhombic $Er_5Ni_3Al_3Ge_4$ ($oP30$, $Pmmn$) structure types (see the table). The distribution of atoms was in all cases fully ordered.

The formation of rows of isotypic compounds $R_2CoAl_4Si_2$ and $R_3FeAl_3Si_2$ was observed for the first time (shaded in the table). Complete series appear for the late rare-earth elements, sometimes interrupted by the systems with Yb (Sm or Lu). The maximum number of isotypic compounds (85) corresponds to the structure type $Y_3NiAl_3Ge_2$, which is a quaternary substitution variant of the binary type Fe_2P . This structure type is observed not only with transition metals of the iron triad, but also with Mn and Cu.

Rows of isotypic quaternary compounds in $R-T-Al-M$ systems

(+ existence established (* in Lviv); – no isotypic compound observed at 873 K)

[Pearson's Crystal Data, Crystal Structure Database for Inorganic Compounds, Release 2023/24, Eds. P. Villars, K. Cenzual, ASM International, Materials Park, Ohio, USA]

Structure type	T	R									
		Y	Sm	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
Aluminosilicides											
$Tb_2NiAl_4Ge_2$	Co	+	–	+	+	+	+	+	+	–	+
$Y_3NiAl_3Ge_2$	Fe	+	+	+	+	+	+	+	+	–	+
	Co	+	+	+	+	+	+	+	+	–	+
	Ni	+	–	+	+	+	+	+	+	–	+
	Cu	+	+	+	+	+	+	+	+	–	+
Alumogermanides											
$SmNiAl_4Ge_2$	Ni	+	+	+	+	+	+	+	+	–	+
$Tb_2NiAl_4Ge_2$	Fe	+	+	+	+	+	+	+	+	–	+
	Co	+	+	+	+	+	+	+	+	+	+
	Ni	+	+	+	+	+	+	+	+	+	–
$Y_3NiAl_3Ge_2$	Mn	+	+	+	+	+	+	+	+	+	+
	Fe	+	+	+	+	+	+	+	+	+	+
	Co	+	+	+	+	+	+	+	+	+	+
	Ni	+	+	+	+	+	+	+	+	+	+
	Cu	+	+	+	+	+	+	+	+	+	+
$Er_5Ni_3Al_3Ge_4$	Ni					+	+	+	+	+	

This work was carried out under the project "Search for new structure types" of the company Material Phases Data System, Vitznau, Switzerland.