

Crystal structure and hydrogenation properties of the MgNi₂B₆ ternary compound

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The intermetallic compounds based on magnesium are widely investigated now due to their excellent hydrogen storage and electrochemical properties [1]. During the systematic study of Mg–Ni–B alloys, the tetragonal phase MgNi₂B₆ was detected. The MgNi₂B₆ ternary compound was prepared in a tantalum crucible in a resistance furnace with a thermocouple controller. The single crystal of MgNi₂B₆ was investigated by means of an Oxford Diffraction Xcalibur3 diffractometer with a CCD detector. The crystal structure of the MgNi₂B₆ compound was successfully solved by direct methods and refined in the space group *Immm* (N°71). The refined lattice parameters are $a=3.0985(8)$ Å, $b=6.5572(13)$ Å, $c=8.3176(17)$ Å, $V=168.99(6)$ Å³. The starting atomic parameters were taken from an automatic interpretation of direct methods followed by difference Fourier syntheses using the SHELX-97 package programs [2]. Finally, all parameters are refined to $R1 = 0.047$ and $wR2 = 0.122$ using 129 independent reflections with $I > 2\sigma(I)$ and presented in Table 1. The ternary boride MgNi₂B₆ is isostructural to CeCr₂B₆ [3]. Boron nets are presented in Figure 1.

The prepared MgNi₂B₆ sample was hydrogenated by hydrogen gas at a pressure up to 20 bar and a temperature of 623 K. Under these conditions, MgNi₂B₆ absorbs up to 2.4 wt% H₂.

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å²) for MgNi₂B₆

Atoms	Sites	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{eq}
B1	4 <i>h</i>	0	0.139(3)	1/2	0.041(3)
B2	8 <i>l</i>	0	0.290(1)	0.332(1)	0.034(2)
Ni	4 <i>j</i>	1/2	0	0.3484(3)	0.0492(7)
Mg	2 <i>a</i>	0	0	0	0.035(1)

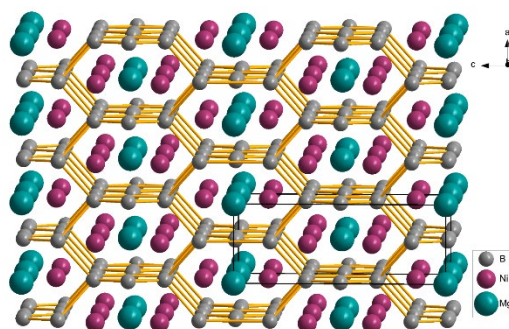


Figure 1. Boron nets in the MgNi₂B₆ structure

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[3] Kuz'ma Y.B., Svarichevskaya S.I., Fomenko V.N. *Inorg. Mater.* (1973) **9**, 1372-1374.